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Supporting information

Boosting the Bifunctional Electrocatalytic Oxygen Activities of CoO_x

Nanoarrays with Porous N-doped Carbon Coating and the

Application in Zn-Air Batteries

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Fig. S1 XRD pattern of $Co(CO_3)_{0.5}(OH)$ on carbon fiber paper.



Fig. S2 (a) Thermogravimetric analysis of PVDF. (b-d) TEM image (b), N₂ adsorption-desorption isotherms (c), and pore size distribution curve (d) of PVDF derived carbon.



Fig. S3 SEM images of CoO_x (a) and CoO_x@C (b) nanoarray.



Fig. S4 TEM images of CoO_x (a) and $CoO_x@C$ (b) nanoarray. HRTEM image of $CoO_x@C$ nanoarray (c).



Fig. S5 The N 1s scan XPS patterns of $CoO_x@C$ and carbon cloth.



Fig. S6 The OER (a, c) and ORR (b, d) performance $CoO_x@NC$ with different calcination temperature (a, b) and nitrogen source (c, d) in 0.1 M KOH solution.



Fig. S7 OER (a) and ORR (b) performance of $CoO_x@NC$ and its acid treated electrode in 0.1 M KOH solution.



Fig. S8 CV plots of CoO_x, CoO_x@C, and CoO_x@NC under different scan rate in 0.1 M KOH solution.



Fig. S9 Co 2p XPS spectrum of CoOx@NC nanoarray after 12 h stability test in 0.1 M KOH solution.



Fig. S10 Comparison of OER and ORR bifunctional activities of samples in this work with some representative electrocatalysts in recent references (from 2015 to now). The dash lines show the ΔE at constant values.



Fig. S11 OER (a) and ORR (b) performance of CoO_x@NC nanoarray, Pt/C and IrO₂/C in 0.1 M KOH solution.



Fig. S12 Galvanostatic cycling discharge and charge curves of carbon cloth.



Fig. S13 Galvanostatic cycling discharge and charge curves obtained using air under ambient conditions of Pt/C without Nafion.