Supporting Information

Mg-Ti co-doping behavior of porous LiFePO₄ microspheres for highrate lithium-ion batteries

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Fig. S1. XPS spectra of Fe 2p (a) and P 2p (b) of LFP microspheres over different etching times.



Fig. S2. XPS spectra of Fe 2p (a), P 2p (b), Mg 1s (c) and Ti 2p (d) of MT-LFP microspheres over different etching

times.



Fig. S3. The linear fitting curves of $I_p-v^{1/2}$ of the as-prepared LFP (a) and MT-LFP (b), and the histogram of Li-ion diffusion coefficients of LFP and MT-LFP cathode during the charge and discharge process with different scan rates (c).

Table S1. Lattice parameters and unit cell volume of the LFP and MT-LFP samples.

Samples	a (Å)	b (Å)	<i>c</i> (Å)	Volume (ų)
LFP	10.3274	6.0065	4.6915	291.02
MT-LFP	10.3259	6.0057	4.6916	290.95

Etching time (s)	O 1s	O 1s Fe 2p3/2	
0	73.16	7.95	18.99
60	69.68	10.08	20.24
180	68.01	11.31	20.68
240	66.65	12.74	20.61
780	65.55	13.67	20.78

Table S2. The atomic ratio of primary orbital peak of undoped LiFePO₄ microspheres under different etching time.

Etching time (s) 0 1s Fe 2p3/2 P 2p Mg 1s Ti 2p3/2 0 68.37 9.28 21.66 0.37 0.30 0.47 60 63.12 14.94 21.20 0.27 660 60.82 17.29 20.84 0.72 0.33

Table S3. The atomic ratio of primary orbital peak of Mg-Ti co-doped LiFePO₄ microspheres under different etching time.

Samples		0.1 mV s ⁻¹	0.2 mV s⁻¹	0.5 mV s ⁻¹	1 mV s ⁻¹
LFP	Charge	5.08E-9	5.89E-9	4.95E-9	4.72E-9
	Discharge	2.98E-9	4.17E-9	3.96E-9	3.62E-9
MT-LFP	Charge	6.21E-9	6.68E-9	6.97E-9	6.71E-9
	Discharge	4.42E-9	4.65E-9	4.99E-9	4.88E-9

Table S4. The Li-ion diffusion coefficients of the as-prepared LFP and MT-LFP cathodes at different scan rates.