## **Electronic Supplementary Information**

## Photothermal Nanocomposite Membranes for Direct Solar Membrane Distillation

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## 1. TEM measurements of the nanoparticles

A transmission electron microscope operated at 100 kV (JEM-2010F FasTEM, JEOL USA, Peabody, MA) was used to characterize the morphology of SiO<sub>2</sub>/Au NSs and CB NPs. The CB NPs were dispersed in chloroform at 5 mg/L. The SiO<sub>2</sub>/Au NS aqueous suspension was prepared by diluting the stock suspension  $(4.45 \times 10^9 \text{ particles/mL})$  100 times to reach a final concentration of  $4.45 \times 10^7$  particles/mL. Both nanoparticle suspensions were then sonicated for 10 min at 100 W using a sonication probe (Vibra Cell, VCX 500, Sonics and Material, Newtown, CT). For each sample, one droplet of the suspension was applied on a 400-mesh, carbon lacey film supported 400-mesh copper grid (Ted Pella, Redding, CA), and dried at room temperature before imaging. TEM images of the SiO<sub>2</sub>/Au NSs and CB NPs are presented in Figure S1. Consistent with previously reported results,<sup>1, 2</sup> the SiO<sub>2</sub>/Au NSs were spherical, and had a fairly uniform particle size of 160 nm, i.e., a 20 nm thick Au coating on the 120 nm diameter SiO<sub>2</sub> core. The CB NP particles exist mostly in small aggregates, with individual particle sizes of approximately 10 nm.



Figure S1. TEM images of the SiO<sub>2</sub>/Au NS (A & B) and CB NP (C & D) nanoparticles.

2. Permeate mass change vs. time of the modified membranes

Representative curves of permeate mass vs. time are presented in Figure S2. The figures demonstrate good linearity of the curves ( $R^2 > 0.998$ , except for *I*NS-m), indicating good stability of membrane performance over the testing period.



Figure S2. Representative curves of the permeate mass change over time of the modified membranes.

## References

- 1. O. Neumann, A. S. Urban, J. Day, S. Lal, P. Nordlander and N. J. Halas, ACS Nano, 2013, 7, 42-49.
- 2. S. J. Oldenburg, R. D. Averitt, S. L. Westcott and N. J. Halas, *Chemical Physics Letters*, 1998, **288**, 243-247.