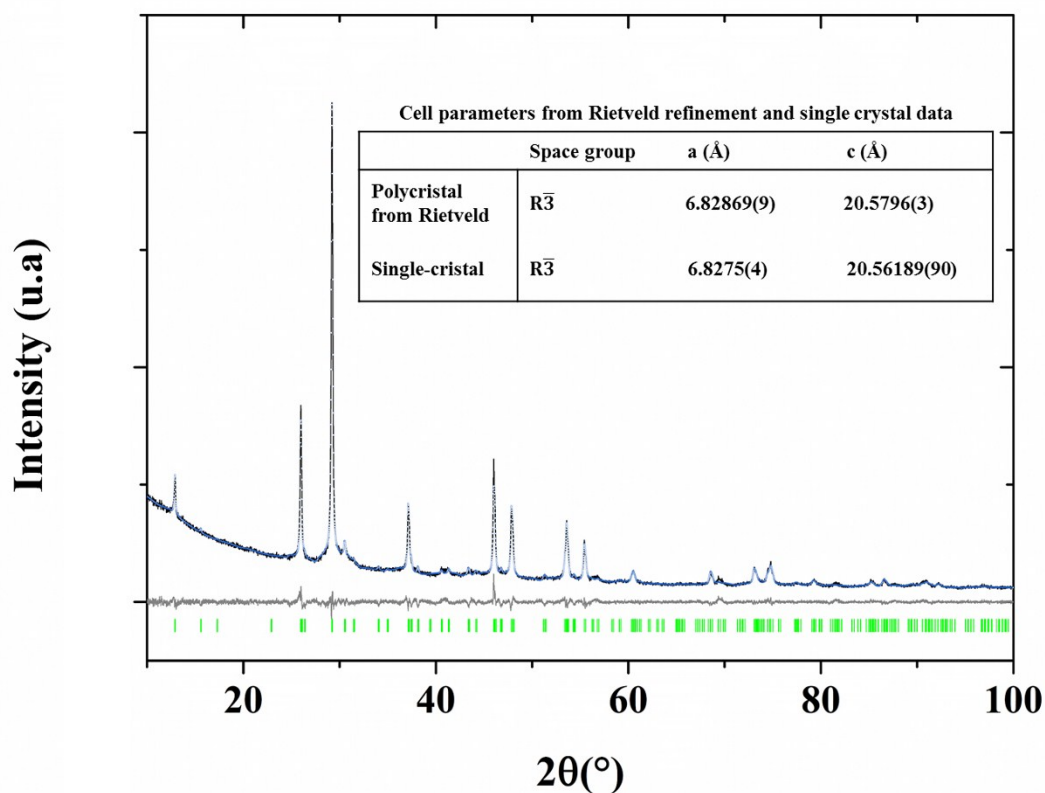


SFig.1



SFigure 1. Rietveld refinement plot for polycrystalline $\text{Cr}_2\text{Ge}_2\text{Te}_6$. Plot is represented as follows: observed (black line), calculated (blue dot), and difference (grey line) intensities. Bragg positions are indicated by green vertical bars. The included table shows the correspondence between the parameters of single-crystal characterization¹⁶ and the present Rietveld refinement.

STable 1. SPS Conditions Used to Densify the Four Polycrystalline Samples.

Compositions	Heating time	Heating temperature	Remaining time	Cooling time	Pressure at all steps
InSiTe_3	2h30 (Step by Step)	773 K	30 min	60 min	28 MPa
$\text{Cr}_2\text{Si}_2\text{Te}_6$	2h (Step by Step)				
$\text{In}_2\text{Ge}_2\text{Te}_6$	45 min	710 K		45 min	64 MPa
$\text{Cr}_2\text{Ge}_2\text{Te}_6$		823 K			

STable 2. Details on the Structure Refinement of Polycrystalline Cr₂Ge₂Te₆ and Atomic Coordinates (x,y,z), Site Occupation Factors, Isotropic Displacement Parameters (Å²) and their Estimated Standard Deviations Obtained from the Rietveld Refinement of Cr₂Ge₂Te₆.

Physical, Crystallographic and Analytic data						
compound		Cr ₂ Ge ₂ Te ₆				
diffractometer radiation (Å)		CuK α , β = 1.540600 and 1.544400				
2 θ range (deg), no. of points		5 – 120, 8758				
step size (deg)		0.013130				
no. of background points		93				
profile function		Thompson-Cox-Hastings pseudo-Voigt				
no. of refined parameters		16				
space group		R $\bar{3}$				
unit-cell parameters (Å)		a = b	6.82869(9)			
		c =	20.5796(3)			
cell volume (Å ³)		831.08(2)				
reliability factors		R _F (%)	7.83			
		R _B (%)	6.06			
		R _{WP} (%)	4.08			
		χ^2	2.43			
Refined coordinates, Atomic Displacement Parameters and their Estimated Standard Deviations from Rietveld Refinement						
atom	Wyckoff position	occ	x	y	z	U _{iso} (Å ²)
Cr	6c	1	0	0	0.3283(5)	0.054(3)
Ge	6c	1	0	0	0.0612(3)	0.076(3)
Te	18f	1	0.0025(7)	0.3534(3)	0.08442(1)	0.049(0)

CAPTIONS:

Svid:

Svid1. X-ray diffraction evolution in temperature of the reflection intensities corresponding to InSiTe₃, In₂Te₃ and Si₂Te₃, resulting from the incongruent melting of the InSiTe₃ compound