

Fig. S1 The schematic diagram of the device for the liquid flow plating method and the way to deposit a dense nickel layer on the surface and sidewall of Si-MCP. Under the action of gravity and water pump, the deposition fluid flows continuously through the surface and interior of the Si-MCP that obtain the OMEP with a dense and uniform nickel layer.

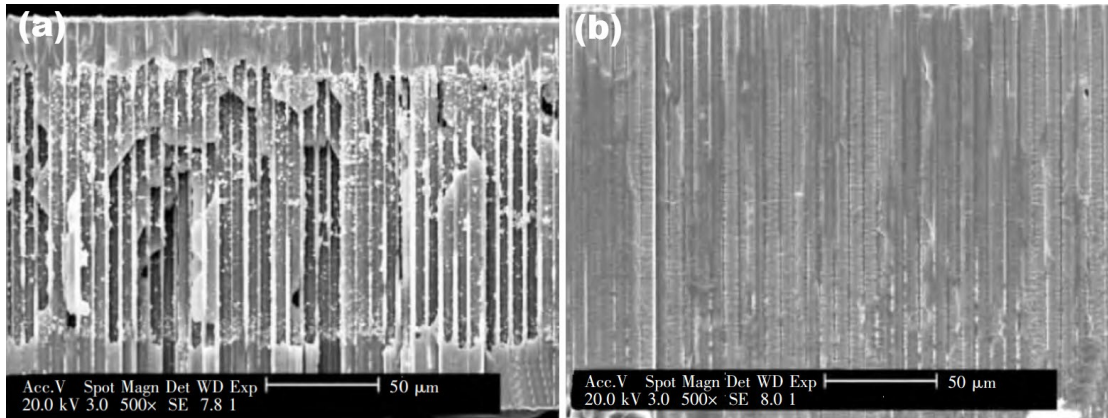


Fig. S2 (a) The nickel layer fabricated through the traditional electroless plating method; (b) The nickel layer fabricated through the liquid flow plating method.

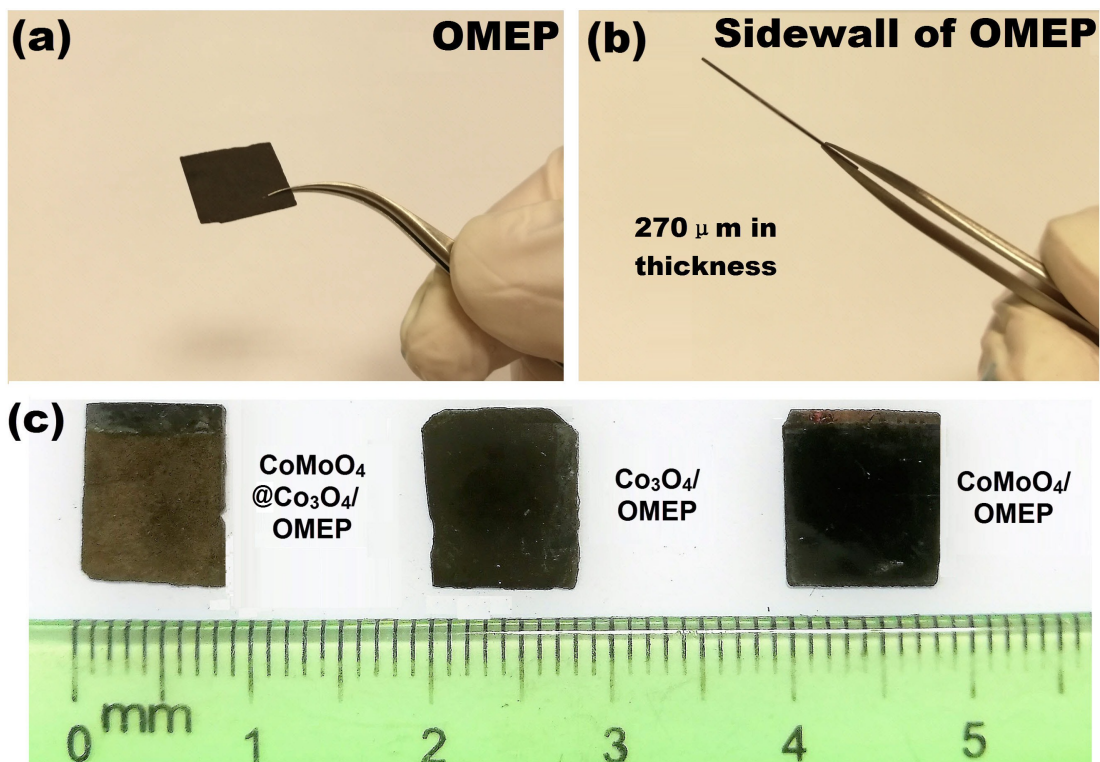
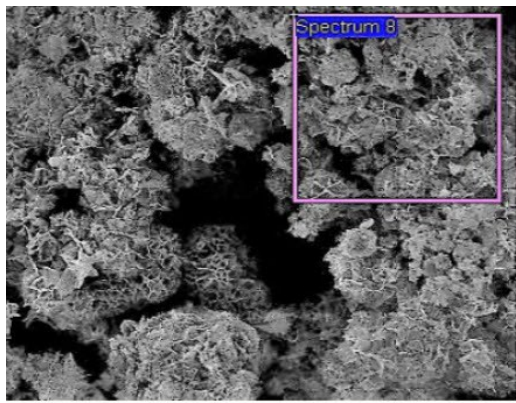


Fig. S3 (a) Front view of OMEP; (b) Sectional view of OMEP; (c) Samples based on OMEP after the electrochemistry tests (from left to right: CoMoO₄@Co₃O₄/OMEP, Co₃O₄/OMEP, and CoMoO₄/OMEP).



| Element | Weight% | Atomic% |
|---------|---------|---------|
| O K | 20.65 | 50.77 |
| Co K | 52.52 | 3.06 |
| Ni K | 12.19 | 8.17 |
| Mo L | 14.65 | 6.01 |
| Totals | 100.00 | |

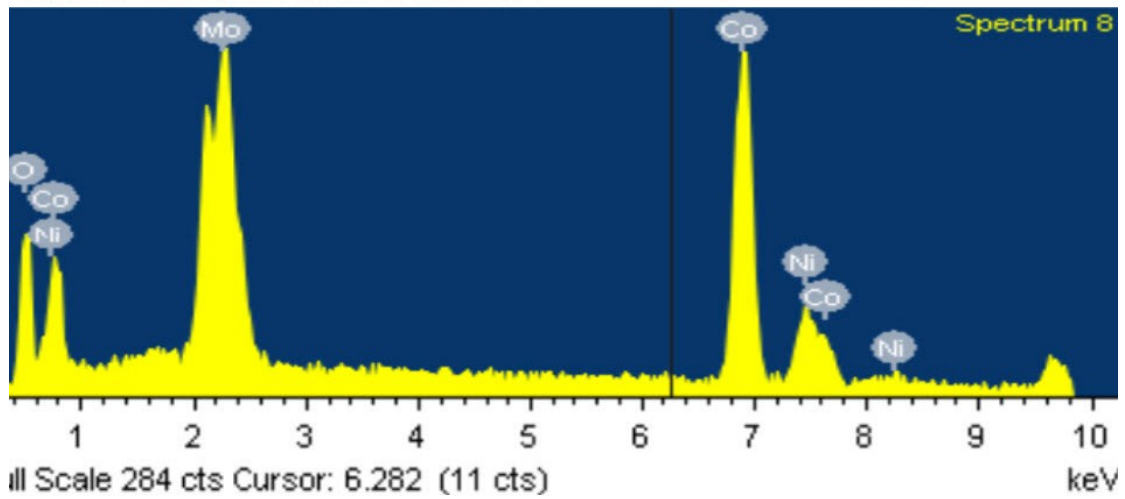


Fig. S4 EDS spectrum of the $\text{CoMoO}_4@\text{Co}_3\text{O}_4/\text{OMEP}$ electrode.

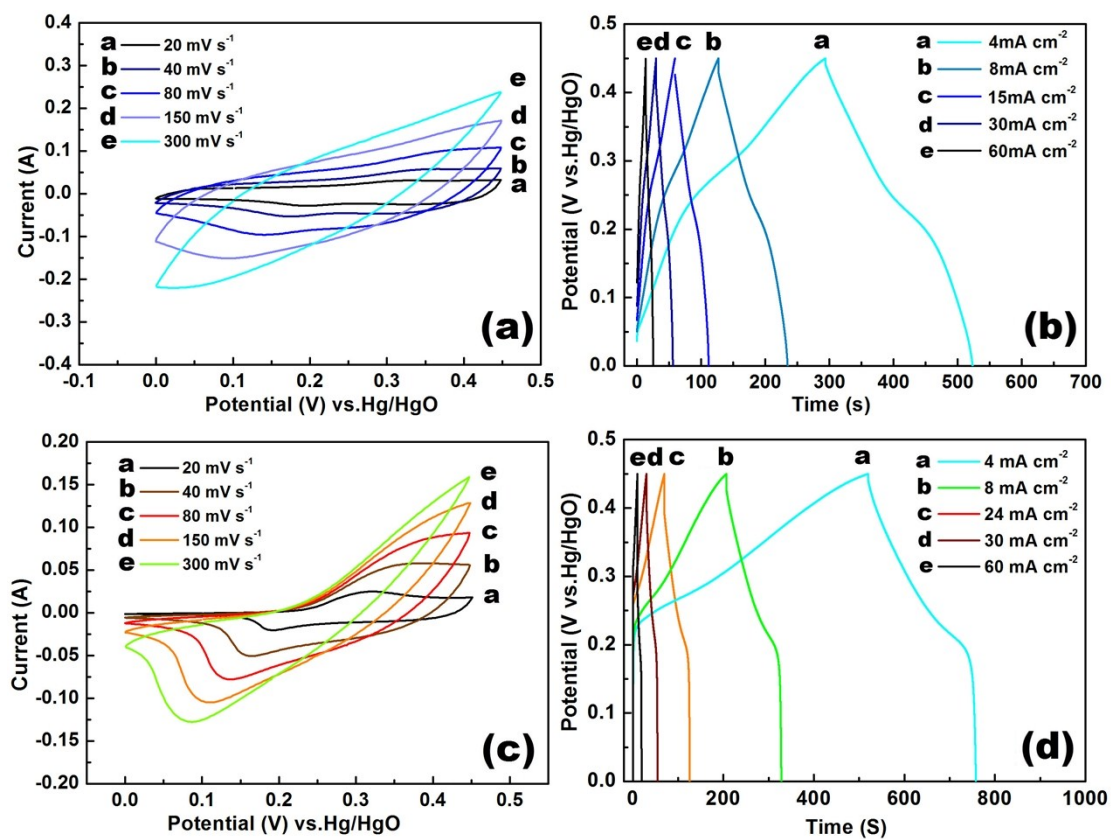


Fig. S5 (a) CV curves of CoMoO₄/OMEP at different scanning rate; (b) First discharge curves of CoMoO₄/OMEP at different current density; (c) CV curves of Co₃O₄/OMEP at different scanning rates; (d) First discharge curves of Co₃O₄/OMEP at different current densities.

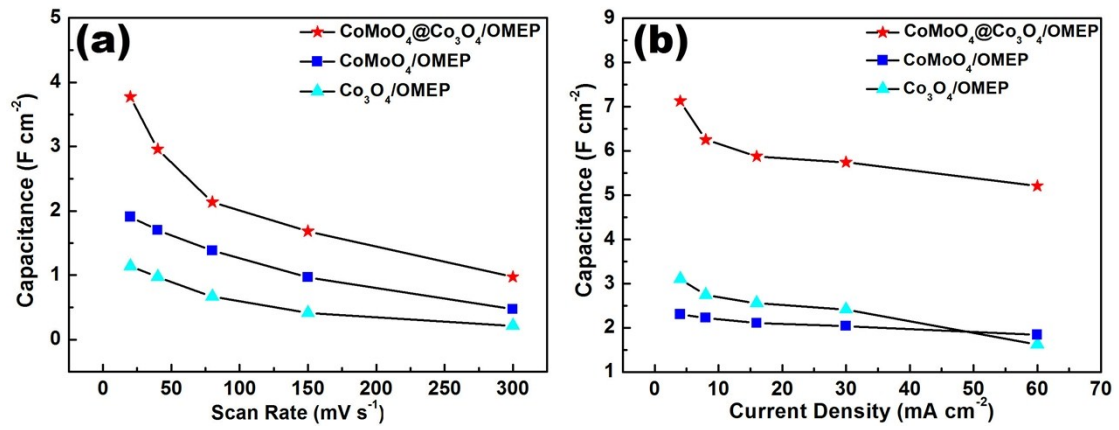


Fig. S6 (a) Variation of the interfacial capacitances of the CoMoO₄@Co₃O₄/OMEP, CoMoO₄/OMEP, Co₃O₄/OMEP electrodes at different scanning rates; (b) Variation of the interfacial capacitances of the CoMoO₄@Co₃O₄/OMEP, CoMoO₄/OMEP, Co₃O₄/OMEP electrodes at different current densities.

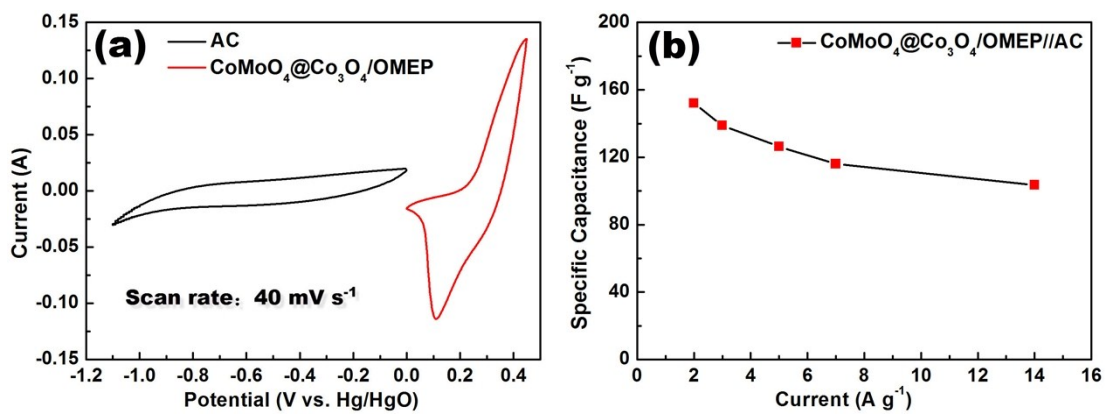


Fig. S7 (a) CV curves of the composite CoMoO₄@Co₃O₄/OMEF (0 to 0.45 V) and AC (-1.1 to 0 V) at a scanning rate of 40 mV s⁻¹; (b) Specific capacitance values of the CoMoO₄@Co₃O₄/OMEF//AC asymmetrical supercapacitor as a function of current density.