Supporting information:

Thermally stable nanosized LEV zeolite synthesized by hydrothermal conversion of FAU zeolite in the presence of N,N-dimethylpiperidinium cation

Natsumi Funasea, Takuya Tanigawa, Yoshitaka Yamasakia, Nao Tsunoji*a, Masahiro

Sadakane^a, and Tsuneji Sano*^a

^aDepartment of Applied Chemistry, Graduate School of Engineering, Hiroshima University, Higashi-

Hiroshima 739-8527, Japan

Corresponding Author

*Corresponding authors: Nao Tsunoji, tel.: +81-82-424-7606, e-mail: tnao7373@hiroshima-u.ac.jp; Tsuneji Sano, tel.: +81-82-424-7607, e-mail: tsano@hiroshima-u.ac.jp

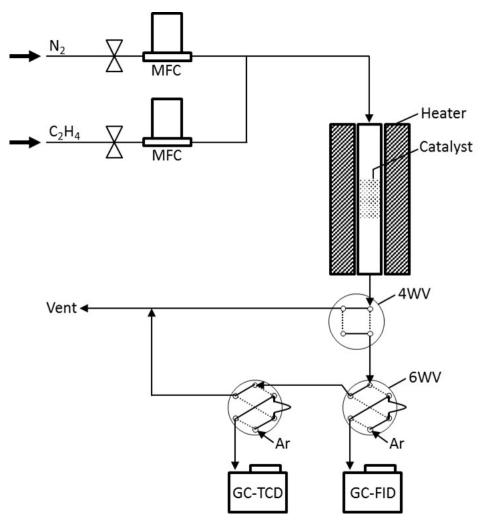


Figure S1. Illustration of setup of fixed-bed flow reactor for ethylene conversion. MFC, 4WV, 6WV, GC-FID, and GC-TCD indicate mass flow controller, four-way valve, six-way valve, gas chromatography with thermal conductivity detector, and gas chromatography with flame ionization detector, respectively.

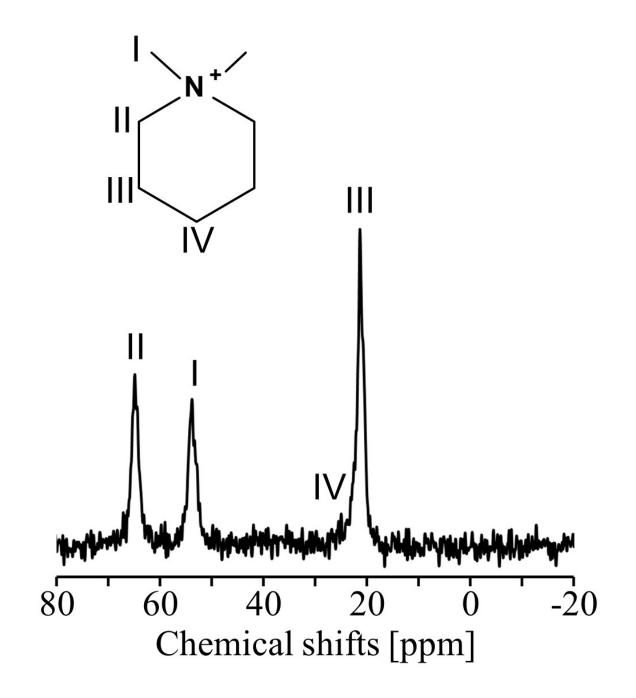


Figure S2. ¹³C CP MAS NMR spectrum of as-synthesized LEV zeolite obtained from FAU zeolite in the presence of DMPOH (sample no. 5).

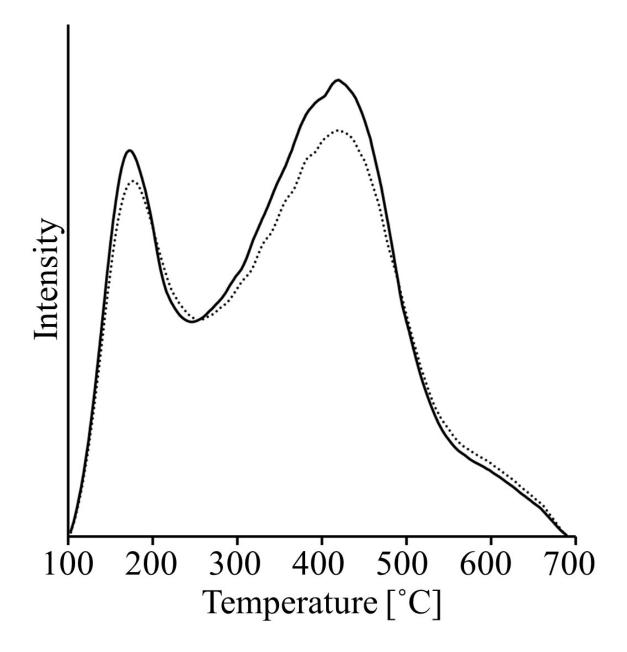


Figure S3. NH₃-TPD curves of LEV zeolites (H-form) obtained from FAU zeolite in the presence of (solid line) DMPOH (sample no. 5) and (dotted line) cholineOH (sample no. 11).

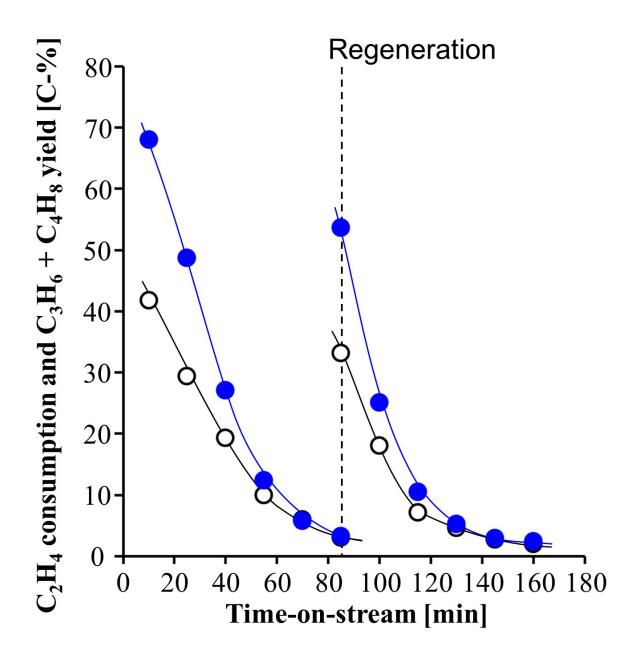


Figure S4. Time-on-stream of (•) C_2H_4 consumption and (•) $C_3H_6 + C_4H_8$ yield for ethylene conversion over nanosized LEV zeolite (H-form) obtained from FAU zeolite in the presence of DMPOH through catalyst regeneration (calcination at 500 °C for 10h).