

Supporting Information

**Biomass willow catkins-derived Co₃O₄/N-doped hollow
hierarchical porous carbon microtubes as the effective tri-
functional electrocatalyst**

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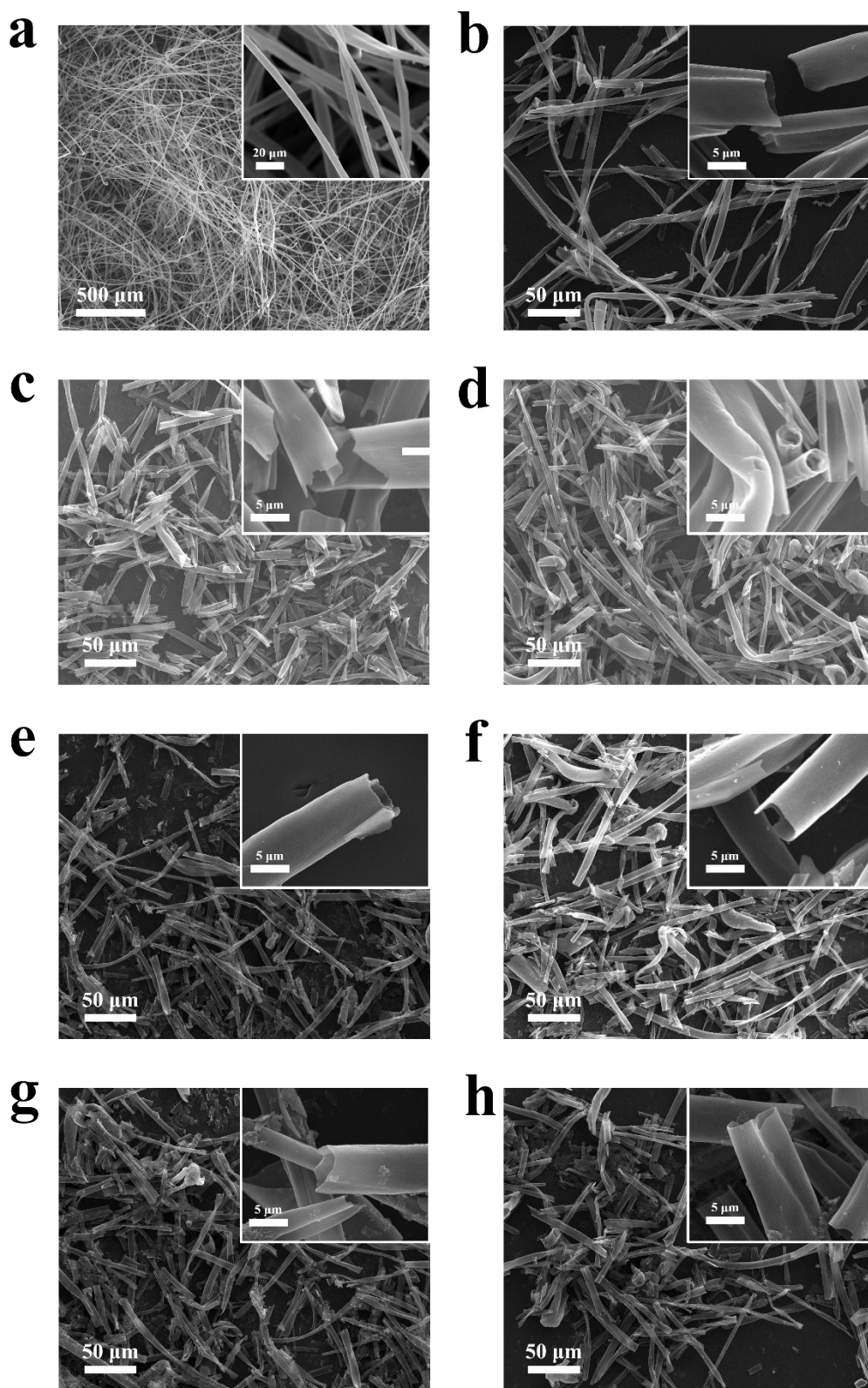


Fig. S1 SEM images of (a) pristine catkin, (b) CMTs-600, (c) CMTs-800, (d) NCMTs-800, (e) $\text{Co}_3\text{O}_4/\text{CMTs-800}$, (f) $\text{Co}_3\text{O}_4/\text{NCMTs-700}$, (g) $\text{Co}_3\text{O}_4/\text{NCMTs-800}$, (h) $\text{Co}_3\text{O}_4/\text{NCMTs-900}$.

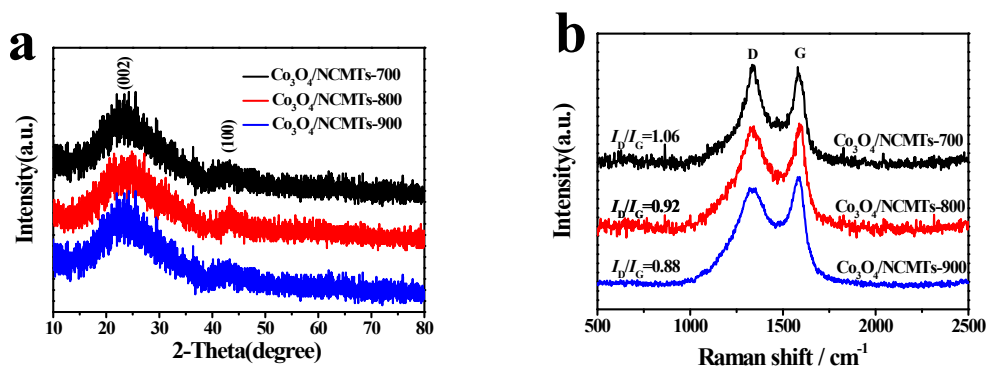


Fig. S2 (a) XRD survey and (b) Raman of the $\text{Co}_3\text{O}_4/\text{NCMTs-X}$ ($X = 700, 800, 900$)

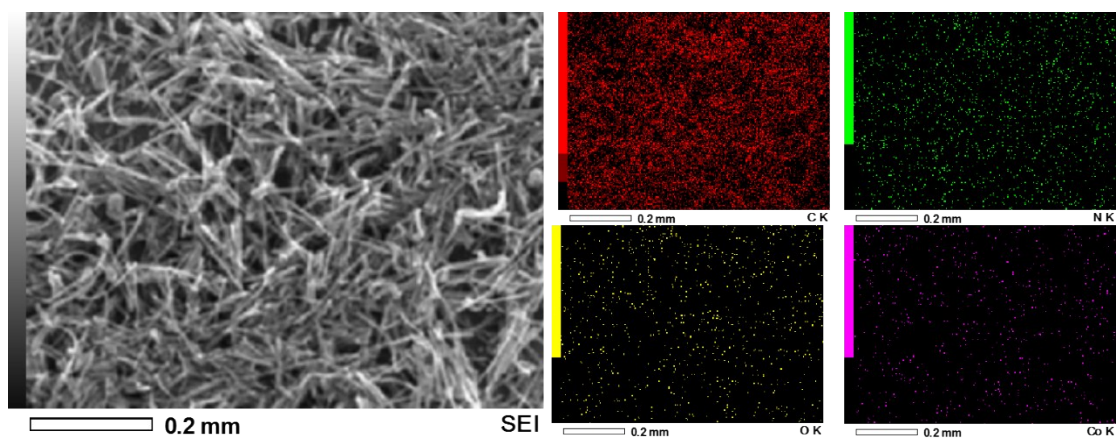


Fig. S3 Typical SEM images of $\text{Co}_3\text{O}_4/\text{NCMTs-800}$ materials and corresponding elemental mapping images of C, N, O and Co.

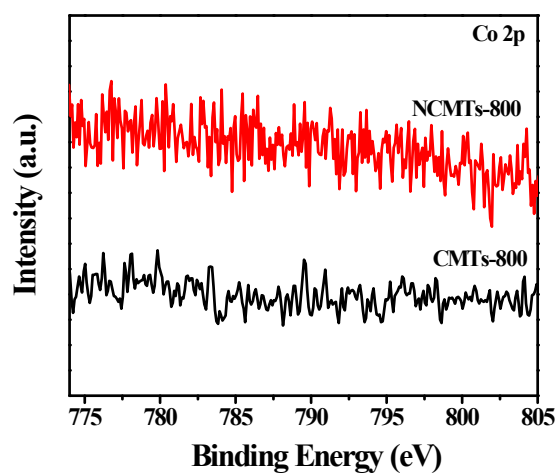


Fig. S4 the high-resolution Co 2p XPS spectrum of NCMTs-800 and CMTs-800.

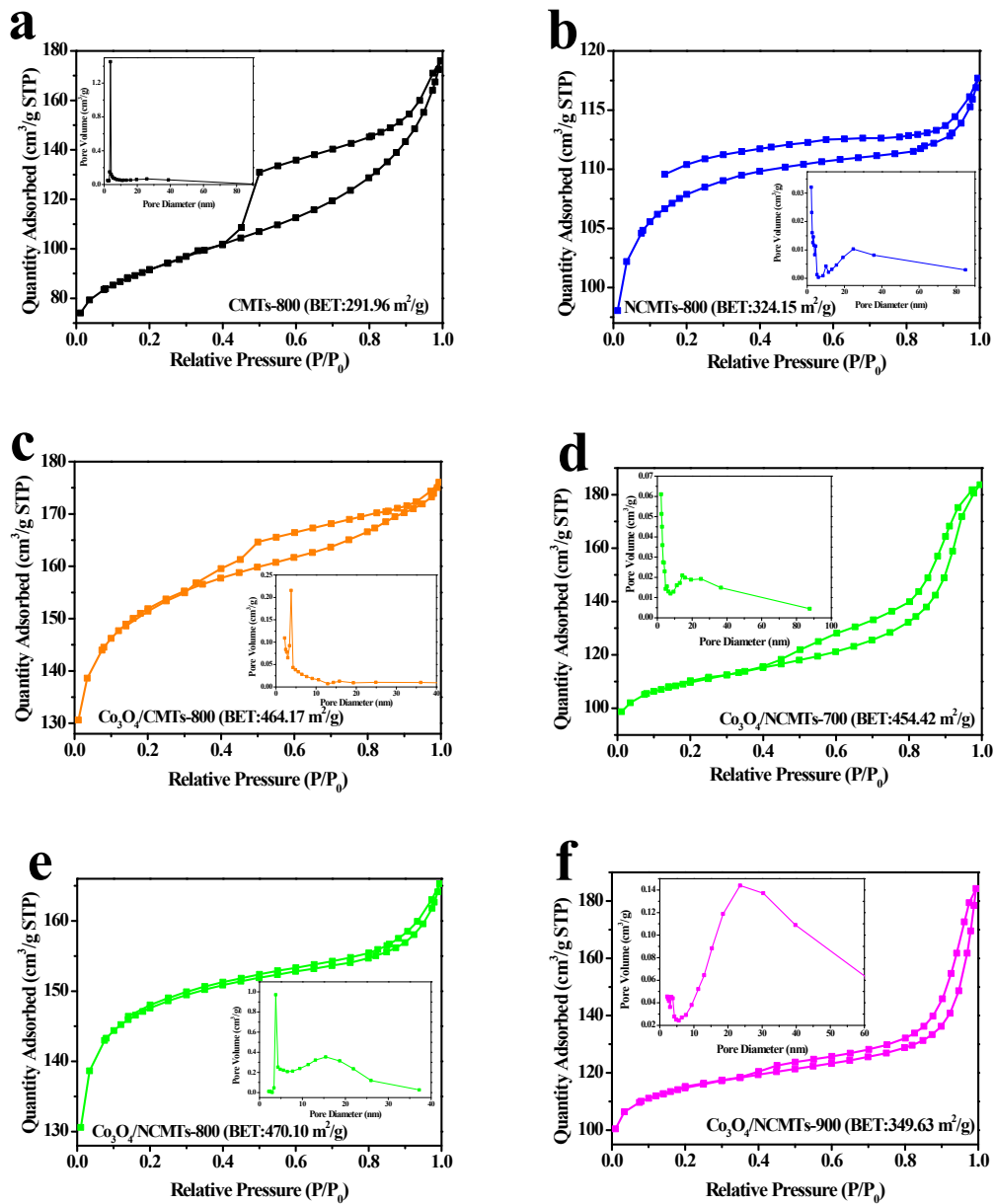


Fig. S5 The nitrogen adsorption-desorption isotherms and the corresponding pore-size distribution of the CMTs-800 (a), NCMTs-800 (b), Co₃O₄/CMTs-800 (c), Co₃O₄/NCMTs-700 (d), Co₃O₄/NCMTs-800 (e) and Co₃O₄/NCMTs-900 (f).

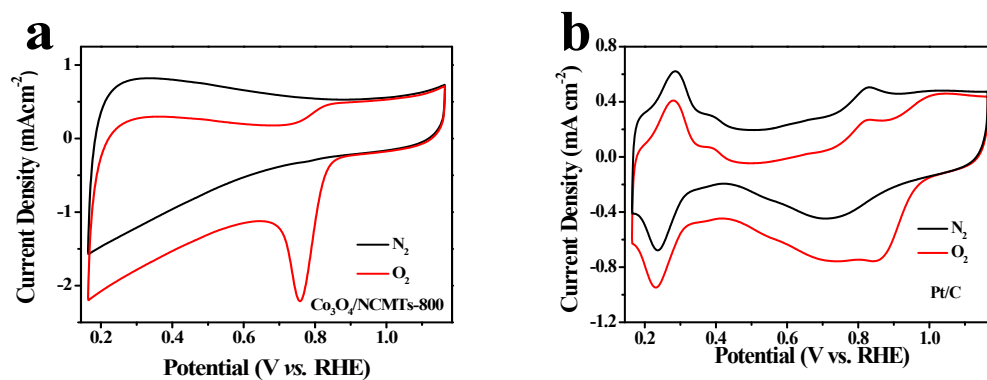


Fig. S6 CV curves of (a) $\text{Co}_3\text{O}_4/\text{NCMTs-800}$ and (b) Pt/C in N_2/O_2 -saturated 0.1 M KOH solution at a scan rate of 50 mV/s.

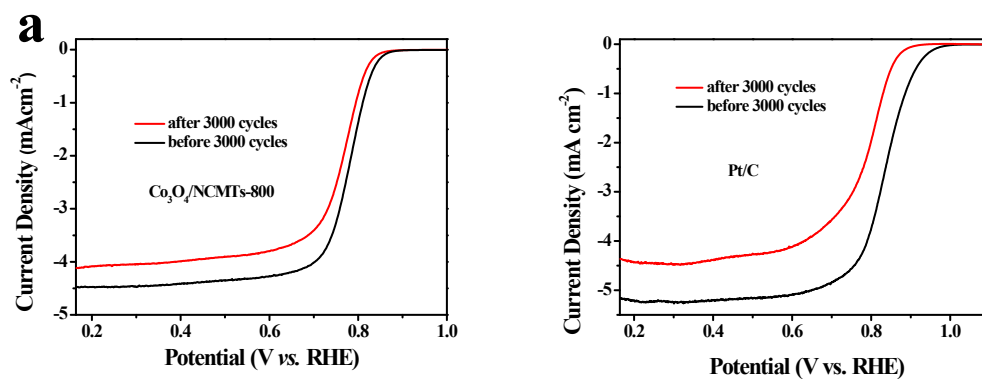


Fig. S7 Linear sweep voltammetric (LSV) curves of (a) $\text{Co}_3\text{O}_4/\text{NCMTs-800}$ and (b) Pt/C for ORR in O_2 -saturated 0.10 M KOH before and after 3000 cycles.

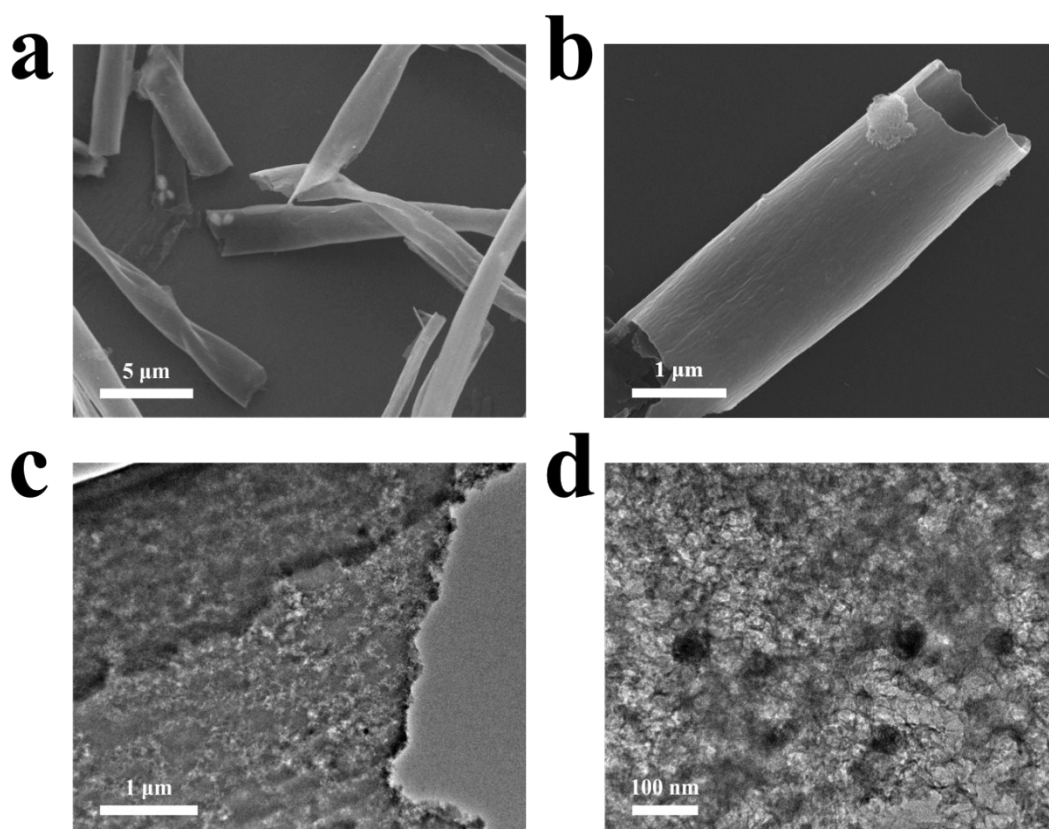


Fig. S8 SEM (a, b) and TEM (c, d) images of $\text{Co}_3\text{O}_4/\text{NCMTs-800}$ after 3000 cycles ORR stability test.

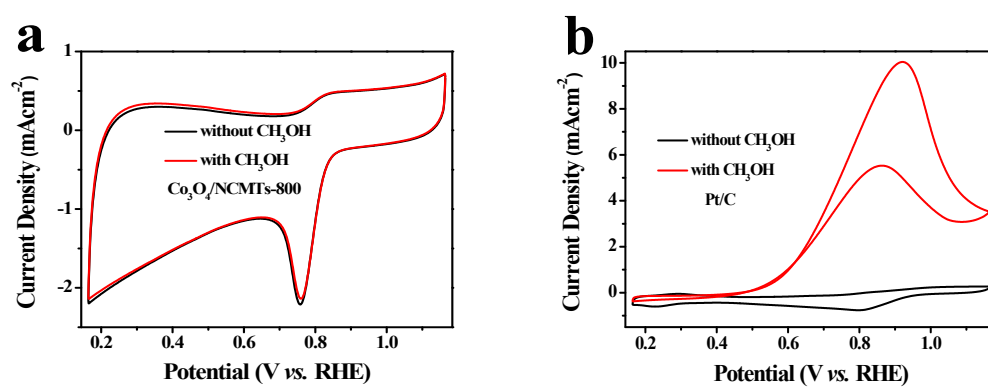


Fig. S9 The CVs of the (a) $\text{Co}_3\text{O}_4/\text{NCMTs-800}$ and (b) Pt/C in O_2 -saturated 0.10 M KOH without and with 1.0 M CH_3OH .

Table S1 XPS results for CMTs-800, NCMTs-800, Co₃O₄/CMTs-800 and

Sample	N content [at.%] ^a	Pyridinic N [%] ^b	Pyrrolic N [%] ^b	graphitic N [%] ^b	oxidized N [%] ^b
CMTs-800	0.43	32.53	8.43	48.19	10.85
NCMTs-800	3.87	52.06	10.80	27.95	9.19
Co ₃ O ₄ /CMTs-800	0.73	31.28	9.92	46.36	12.44
Co ₃ O ₄ /NCMTs-800	1.81	53.64	7.60	31.16	7.60

Co₃O₄/NCMTs-800

a: The total N content (at.%) in the CMTs-800, NCMTs-800, Co₃O₄/CMTs-800 and Co₃O₄/NCMTs-800 determined by XPS elemental analysis, respectively.

b: The percentage of pyridinic-, pyrrolic-, graphitic- and oxidied-type nitrogen for the doped N

Table S2 Textual parameters of the materials

Sample	BET surface area [m ² g ⁻¹]	Average pore diameter [nm]	Total pore volume [cm ³ g ⁻¹]
CMTs-800	291.96	4.04	0.18
NCMTs-800	324.15	5.27	0.26
Co ₃ O ₄ /CMTs-800	464.17	5.47	0.26
Co ₃ O ₄ /NCMTs-700	454.42	5.48	0.27
Co ₃ O ₄ /NCMTs-800	470.10	7.39	0.29
Co ₃ O ₄ /NCMTs-900	349.63	11.32	0.35

Table S3. Comparison of electrochemical activity of Co₃O₄/NCMTs-800 to the other reported ORR catalysts in alkaline electrolyte

Catalysts [Ⓢ]	Mass loading [Ⓢ] /mg cm ⁻² [Ⓢ]	onset potential (V vs.RHE) [Ⓢ]	E _{1/2} [Ⓢ] (V vs.RHE) [Ⓢ]	Biomass [Ⓢ]	Reference [Ⓢ]
Co ₃ O ₄ /NCMTs-800 [Ⓢ]	0.28 [Ⓢ]	0.906 [Ⓢ]	0.778 [Ⓢ]	Catkins [Ⓢ]	This work [Ⓢ]
BP350@C-1000 [Ⓢ]	2.55 [Ⓢ]	0.90 [Ⓢ]	0.78 [Ⓢ]	Pig blood [Ⓢ]	[S1] [Ⓢ]
N,P-carbon [Ⓢ]	0.31 [Ⓢ]	~0.944 [Ⓢ]	~0.754 [Ⓢ]	Coconut shells [Ⓢ]	[S2] [Ⓢ]
N,S- carbon-800 (PHC) [Ⓢ]	0.66 [Ⓢ]	~0.912 [Ⓢ]	~0.77 [Ⓢ]	Honeysuckles [Ⓢ]	[S3] [Ⓢ]
egg-CMS [Ⓢ]	0.136 [Ⓢ]	~0.86 [Ⓢ]	~0.69 [Ⓢ]	Egg [Ⓢ]	[S4] [Ⓢ]
GLBS-1000 [Ⓢ]	0.4 [Ⓢ]	0.971 [Ⓢ]	0.842 [Ⓢ]	Keratin [Ⓢ]	[S5] [Ⓢ]
Ni/NiO/NiCo ₂ O ₄ /N-CNT-As [Ⓢ]	0.24 [Ⓢ]	0.89 [Ⓢ]	0.74 [Ⓢ]	Seaweed [Ⓢ]	[S6] [Ⓢ]
Sewage sludge-derived porous carbon (SN-AW) [Ⓢ]	0.1 [Ⓢ]	0.914 [Ⓢ]	0.819 [Ⓢ]	Sewage sludge [Ⓢ]	[S7] [Ⓢ]
NFe-NPCS-900-4 [Ⓢ]	0.200 [Ⓢ]	0.955 [Ⓢ]	0.836 [Ⓢ]	Starch [Ⓢ]	[S8] [Ⓢ]
N and S comodified 3D porous graphene (NSG) [Ⓢ]	- [Ⓢ]	0.788 [Ⓢ]	0.778 [Ⓢ]	Horn [Ⓢ]	[S9] [Ⓢ]
NCS-800 [Ⓢ]	0.05 [Ⓢ]	~0.90 [Ⓢ]	~0.76 [Ⓢ]	<u>Typha orientalis</u> [Ⓢ]	[10] [Ⓢ]
WHC-700 [Ⓢ]	0.16 [Ⓢ]	0.98 [Ⓢ]	~0.86 [Ⓢ]	<u>Eichhornia crassipes</u> [Ⓢ]	[11] [Ⓢ]
N-CNAs [Ⓢ]	0.076 [Ⓢ]	~0.884 [Ⓢ]	0.774 [Ⓢ]	Dried Grass [Ⓢ]	[12] [Ⓢ]
Fe/N/CNT@PCF [Ⓢ]	~0.23 [Ⓢ]	0.866 [Ⓢ]	0.770 [Ⓢ]	catkin [Ⓢ]	[S13] [Ⓢ]
CMTs-750 [Ⓢ]	0.20 [Ⓢ]	~0.793 [Ⓢ]	~0.523 [Ⓢ]	catkin [Ⓢ]	[S14] [Ⓢ]
CMTs-950 [Ⓢ]	0.20 [Ⓢ]	~0.793 [Ⓢ]	~0.523 [Ⓢ]	catkin [Ⓢ]	

Table S4. Comparison of electrochemical activity of Co₃O₄/NCMTs-800 to the other reported OER catalysts in alkaline electrolyte

Catalysts [Ⓢ]	Mass loading [Ⓢ] /mg cm ⁻² [Ⓢ]	η_{10}/V (j=10 mA cm ⁻²) [Ⓢ]	Biomass [Ⓢ]	Reference [Ⓢ]
Co ₃ O ₄ /NCMTs-800 [Ⓢ]	1 [Ⓢ]	0.35 [Ⓢ]	biomass [Ⓢ]	This work [Ⓢ]
nitrogen-doped carbon hollow cubes (NCHCs) [Ⓢ]	0.401 [Ⓢ]	~0.48 [Ⓢ]	Lysine [Ⓢ]	[S15] [Ⓢ]
N and S comodified 3D porous graphene (NSG) [Ⓢ]	10 [Ⓢ]	0.468 [Ⓢ]	Horn [Ⓢ]	[S9] [Ⓢ]
Co ₉ S ₈ @NC [Ⓢ]	~0.41 [Ⓢ]	0.400 [Ⓢ]	Shrimp shell [Ⓢ]	[S16] [Ⓢ]
Sewage sludge-derived porous carbon (SN-AW) [Ⓢ]	0.1 [Ⓢ]	0.384 [Ⓢ]	Sewage sludge [Ⓢ]	[S7] [Ⓢ]
egg-CMS [Ⓢ]	2 [Ⓢ]	0.3 [Ⓢ]	Egg [Ⓢ]	[S4] [Ⓢ]
Ni/NiO/NiCo ₂ O ₄ /N-CNT-As [Ⓢ]	0.24 [Ⓢ]	0.24 [Ⓢ]	Seaweed [Ⓢ]	[S6] [Ⓢ]
LiCoO ₂ NSs [Ⓢ]	0.1 [Ⓢ]	0.530 [Ⓢ]	– [Ⓢ]	[S17] [Ⓢ]
Delithiation LiCoO ₂ NSs [Ⓢ]	0.1 [Ⓢ]	0.500 [Ⓢ]	– [Ⓢ]	
LiCoO ₂ NPs [Ⓢ]	0.1 [Ⓢ]	0.510 [Ⓢ]	– [Ⓢ]	
Delithiation LiCoO ₂ NPs [Ⓢ]	0.1 [Ⓢ]	0.390 [Ⓢ]	– [Ⓢ]	
NiCo LDH [Ⓢ]	0.17 [Ⓢ]	0.474 [Ⓢ]	– [Ⓢ]	[S18] [Ⓢ]
MnCo ₂ Ox [Ⓢ]	0.142 [Ⓢ]	>0.410 [Ⓢ]	– [Ⓢ]	[S19] [Ⓢ]
NiMoO ₄ NW/CC [Ⓢ]	5 [Ⓢ]	0.400 [Ⓢ]	– [Ⓢ]	[S20] [Ⓢ]
NiMoP ₂ NW/CC [Ⓢ]	5 [Ⓢ]	0.330 [Ⓢ]	– [Ⓢ]	
Co ₂ P [Ⓢ]	0.284 [Ⓢ]	0.370 [Ⓢ]	– [Ⓢ]	[S21] [Ⓢ]
Co-P films [Ⓢ]	2.52 [Ⓢ]	0.345 [Ⓢ]	– [Ⓢ]	[S22] [Ⓢ]
Ni(OH) ₂ NSs [Ⓢ]	– [Ⓢ]	0.350 [Ⓢ]	– [Ⓢ]	[S23] [Ⓢ]
NiMn HNSs [Ⓢ]	– [Ⓢ]	0.312 [Ⓢ]	– [Ⓢ]	
Co ₃ O ₄ /rmGO [Ⓢ]	1 [Ⓢ]	0.31 [Ⓢ]	– [Ⓢ]	[S24] [Ⓢ]
CoOx@CN [Ⓢ]	1 [Ⓢ]	0.260 [Ⓢ]	– [Ⓢ]	[S25] [Ⓢ]

Table S5. Comparison of electrochemical activity of Co₃O₄/NCMTs-800 to the other reported HER catalysts in alkaline electrolyte

Catalysts [⊖]	Mass loading [⊖] /mg cm ⁻² [⊖]	η_{10}/V (j=10 mA cm ⁻²) [⊖]	Biomass [⊖]	Reference [⊖]
Co ₃ O ₄ /NCMTs-800 [⊖]	1 [⊖]	0.21 [⊖]	biomass [⊖]	This work [⊖]
Co ₉ S ₈ @NC [⊖]	~0.41 [⊖]	0.343 [⊖]	Shrimp shell [⊖]	[S16] [⊖]
Co ₉ S ₈ @NPC-10 [⊖]	~0.41 [⊖]	0.261 [⊖]	Shrimp shell [⊖]	
human hair-derived porous carbons (HCP-800) [⊖]	0.285 [⊖]	~0.26 [⊖]	human hair [⊖]	[S26] [⊖]
FeP NPs@NPC [⊖]	1.4 [⊖]	0.22 [⊖]	Phytic acid [⊖]	[S27] [⊖]
Co-NRCNTs [⊖]	0.28 [⊖]	0.370 [⊖]	- [⊖]	[S28] [⊖]
MoS _{2+x} /FTO [⊖]	0.02 [⊖]	-0.310 [⊖]	- [⊖]	[S29] [⊖]
Co-P nanocube [⊖]	0.286 [⊖]	0.280 [⊖]	- [⊖]	[S30] [⊖]
EG/Co _{0.85} Se/NiFe-LDH [⊖]	4 [⊖]	0.260 [⊖]	- [⊖]	[S31] [⊖]
CoOx@CN [⊖]	0.42 [⊖]	0.232 [⊖]	- [⊖]	[S24] [⊖]
CoP/CC [⊖]	0.92 [⊖]	0.209 [⊖]	- [⊖]	[S32] [⊖]
Ni ₂ P [⊖]	1.0 [⊖]	0.205 [⊖]	- [⊖]	[S33] [⊖]
Co-MoS ₂ [⊖]	0.89 [⊖]	0.203 [⊖]	- [⊖]	[S34] [⊖]
NiFe LDH/Ni foam [⊖]	- [⊖]	0.200 [⊖]	- [⊖]	[S35] [⊖]
CoSe ₂ /carbon cloth [⊖]	- [⊖]	0.190 [⊖]	- [⊖]	[S36] [⊖]
NiCo HNSs [⊖]	- [⊖]	0.230 [⊖]	- [⊖]	[S23] [⊖]
NiMn HNSs [⊖]	- [⊖]	0.302 [⊖]	- [⊖]	
Ni(OH) ₂ NSs [⊖]	- [⊖]	0.374 [⊖]	- [⊖]	
MoC _x /C [⊖]	0.80 [⊖]	0.151 [⊖]	- [⊖]	[S37] [⊖]

Table S6. Comparison of the electrocatalytic performance of the Co₃O₄/NCMTs-800|| Co₃O₄/NCMTs-800 electrode pair for two-electrode overall water splitting with that of reported non-precious metal electrocatalysts and precious metal electrocatalysts tested under similar conditions.

Catalysts [⊖]	Mass loading /mg cm ⁻² ⊖	η_{10}/V (j=10 mA cm ⁻²) [⊖]	Reference [⊖]
Co ₃ O ₄ /NCMTs-800 [⊖]	1 [⊖]	1.74 [⊖]	This work [⊖]
NFPGNS [⊖]	2.55 [⊖]	1.91 [⊖]	[S38] [⊖]
Pristine NiFeOx/CFP [⊖]	1.6 [⊖]	~1.88 [⊖]	[S39] [⊖]
Ni(OH) ₂ /NiSe [⊖]	- [⊖]	1.78 [⊖]	[S40] [⊖]
Ni ₁₂ P ₅ /Ni ₃ (PO ₄) ₂ -HS [⊖]	0.71 [⊖]	1.76 [⊖]	[S41] [⊖]
Co-P film on Au foil [⊖]	0.143 [⊖]	1.73 [⊖]	[S42] [⊖]
NiCo ₂ O ₄ Ni _{0.33} Co _{0.67} S ₂ [⊖]	0.3 [⊖]	1.72 [⊖]	[S43] [⊖]
3.5 nm Pt /NF [⊖]	- [⊖]	1.71 [⊖]	[S35] [⊖]
NiFe LDH/NF [⊖]	- [⊖]	1.70 [⊖]	
Ni ₅ P ₄ on Ni foil [⊖]	13.9 [⊖]	1.7 [⊖]	[S44] [⊖]
Co-P/NC [⊖]	0.283 [⊖]	1.7 [⊖]	[S45] [⊖]
NiFe HNSS [⊖]	- [⊖]	1.67 [⊖]	[S23] [⊖]
EG/Co _{0.85} Se/NiFe-LDH [⊖]	4 [⊖]	1.67 [⊖]	[S31] [⊖]
NiMoP ₂ /PE-NiMoP ₂ [⊖]	5 [⊖]	1.67 [⊖]	[S46] [⊖]
α -NiOOH/NF [⊖]	0.8172 [⊖]	1.66 (η_{10}/V j=1 mA cm ⁻²) [⊖]	[S47] [⊖]
2-cycle NiFeOx/CFP [⊖]	1.6 [⊖]	1.61 [⊖]	[S40] [⊖]

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