Electronic Supplementary Information

In situ growth of TiO₂ layer on flexible Ti substrate targeting interface recombination issue of BiVO₄ photoanodes for efficient

solar water splitting

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Fig. S1 SEM images of Ti foil-BiOI electrode at low (a) and high (b) magnifications.



Fig. S2 J-V curves of Mo-BiVO₄ photoanodes based on Ti substrate varied from concentration of molybdenum source measured in 0.1 M phosphate buffer solution (pH 7.4).



Fig. S3 SEM images of Ti foil before (a) and after (b) annealing process. Digital photographs (c) and XRD patterns (d) of non-annealed Ti foil and annealed Ti foil samples. When comparing the SEM images of Ti foil before and after the annealing treatment, one can notice that the surface turns to be compact and smooth for the annealed Ti foil. The XRD pattern of annealed Ti foil also shows one weak peak of TiO₂ (JCPDS 21-1276), confirming the in-situ growth of TiO₂ on the Ti substrate after the annealing process.



Fig. S4 XPS spectra of (a) Mo_{3d} and (b) N_{1s} peaks from the Mo, N-BiVO₄ sample.



Fig. S5 UV-vis absorption spectra of pristine and modified BiVO₄ photoanodes based on Ti substrate.



Fig. S6 Charge separation efficiency (η_{sep}) of BiVO₄ and co-doped BiVO₄ photoanodes based on FTO substrate and Ti substrate, respectively.



Fig. S7 PL spectra of pristine and modified $BiVO_4$ photoanodes based on Ti substrate (with excitation wavelength of 330 nm).



Fig. S8 *J-V* curves of pristine and modified $BiVO_4$ photoanodes based on Ti substrate for solar sulfite oxidation collected under AM 1.5G illumination in 0.1 M phosphate buffer solution (pH 7.4) containing 1 M Na₂SO₃ as hole scavenger.

Table S1. Fitting equivalent circuit and EIS fitting results of pristine and modifiedBiVO4 photoanodes based on Ti substrate.

	CPE1			
Sample	BiVO ₄	Mo-BiVO ₄	N-BiVO ₄	Mo, N-BiVO ₄
$R_{s}\left(\Omega\right)$	44	33	48	35
CPE-T (µF)	180	205	206	273
CPE-T (µF)	0.75	0.83	0.80	0.82
$R_{ct}(k\Omega)$	40.90	15.22	18.78	0.38



Fig. S9 SEM images at high (a) and low magnifications (b) and XRD pattern (c) of the CoO_x -Mo, N-BiVO₄ sample.



Fig. S10 *J-V* curves of pristine and modified BiVO₄ photoanodes based on the Ti substrate for electrical water oxidation collected in dark. The cathodic shift of the dark current onset potential for CoO_x -Mo, N-BiVO₄ indicates the excellent OER kinetics of CoO_x catalyst for electrochemical water splitting. A 0.1 M phosphate buffer solution (pH 7.4) was used as the electrolyte.



Fig. S11 *J-V* curves of the pristine $BiVO_4$, Mo, N- $BiVO_4$ and CoO_x -Mo, N- $BiVO_4$ photoanodes based on Ti substrate for solar sulfite oxidation collected under AM 1.5G illumination in 0.1 M phosphate buffer solution (pH 7.4) containing 1 M Na₂SO₃ as hole scavenger.

Photoanodes	Substrate	Photocurrent density at 1.23 V_{RHE}	ABPE value	Ref.
BiOI/BiVO ₄	FTO glass	3.27 mA cm ⁻²	0.97%	(5)
FeOOH-NiOOH/N-BiVO ₄	FTO glass	$\sim 5 \text{ mA cm}^{-2}$	2.2%	(7)
Co-Pi/W: BiVO ₄	FTO glass	~1.4 mA cm ⁻²	~	(9)
NiFeO _x -Bi-BiVO ₄ (two electrode stack)	ITO glass	\sim 5 mA cm ⁻²	2.25%	(10)
BiVO ₄ /FeOOH/NiOOH	FTO glass	2.73 mA cm^{-2} at 0.6 V _{RHE}	1.75%	(11)
CoPi/BiVO ₄ /TiO ₂	FTO glass	1.61 mA cm ⁻²	~	(14)
Mo-BiVO ₄	FTO glass	1.2 mA cm ⁻²	~	(15)
NiO/CoO _x /BiVO ₄	ITO glass	$\begin{array}{c} 2.5 \text{ mA cm}^{-2} \text{ at} \\ 0.6 \text{ V}_{\text{RHE}} \end{array}$	1.5%	(16)
BiVO ₄ /WO ₃ /SnO ₂	FTO glass	2 mA cm ⁻²	~	(17)
NiO/CoO _x /BiVO ₄	Ti	3.5 mA cm ⁻²	1.5%	(23)
NiOOH/FeOOH/CQDs/BiVO ₄	FTO glass	5.99 mA cm ⁻²	2.29%	(29)
CoO _x -Mo, N-BiVO ₄	Ti	5.04 mA cm ⁻²	1.41%	This work

Table S2. Summary of recent key advances in BiVO₄ based photoanodes for PEC solar water splitting.



Fig. S12 The Chronoamperometry (*J-t*) curve of Ti foil-BiVO₄ collected at 0.6 V_{RHE} in 0.1 M phosphate buffer solution containing 1 M Na₂SO₃ (pH 7.4) under AM 1.5G illumination.



Fig. S13 The Chronoamperometry (*J-t*) curve of CoO_x -Mo, N-BiVO₄ collected at 0.6 V_{RHE} in 0.1 M phosphate buffer solution (pH 7.4) without hole scavenger under AM 1.5G illumination.



Fig. S14 O_2 produced on the CoO_x-Mo, N-BiVO₄ photoanode at 0.6 V_{RHE} in 0.1 M phosphate buffer solution (pH 7.4) under AM 1.5 G illumination.



Fig. S15 Faradaic efficiency of CoO_x -Mo, N-BiVO₄ photoanode for water oxidation. Solution, 0.1 M phosphate buffer solution (pH 7.4); scan rate, 10 mV s⁻¹; and front illumination (intensity: 100 mW cm⁻²).



Fig. S16 Electron microscopic characterization of the Ti foil-BiVO₄ photoanode.(a) Top-view SEM images of the Ti foil-BiVO₄ photoanode and (b-e) SEM-EDS elemental mapping images for Bi, V, O and Ti, respectively; (f) TEM image of Ti foil-BiVO₄ and (g-j) TEM-EDS elemental mapping images for Bi, V, O and Ti, respectively.



Fig. S17 CV curves measured in a non-Faradaic region of 0.95-1.15 V at various scan rates for (a) BiVO₄, (b) Mo, N-BiVO₄, (c) CoO_x-Mo, N-BiVO₄ with a geometric area of 1 cm², respectively. Charging current differences ($\Delta I = I_a - I_c$) measured at 1.05 V plotted against scan rate for BiVO₄, Mo,N-BiVO₄, and CoO_x-Mo,N-BiVO₄, respectively. I_a and I_c are the anodic and cathodic current, respectively, and the linear slope is twice of the double-layer capacitance (C_{dl}).



Fig. S18 Double-layer capacitance (C_{dl}) measurements for determining the specific capacitance (C_s) of Ti substrate from cyclic voltammetry (CV) in 0.1 M phosphate buffer solution (a) CV curves measured in a non-Faradaic region of 0.95–1.15 V at various scan rates, (b) Charging current density differences ($\Delta I = I_a - I_c$) measured at 1.05 V plotted against scan rate. I_a and I_c are the anodic and cathodic current density, respectively, and the linear slope is twice of the C_s .



Fig. S19 (a) ECSA normalized photocurrent density vs. voltage curves, and (b) a comparison of the photocurrent density normalized by the geometrical area and the ECSA at $1.23 V_{RHE}$ for BiVO₄, Mo,N-BiVO₄, and CoO_x-Mo,N-BiVO₄, respectively.