

Electronic Supplementary Information

Pd-Impregnated NiCo_2O_4 nanosheets/porous carbon composites as a free-standing and binder-free catalyst for a high energy lithium–oxygen battery

*Daniel A. Agyeman[†], Mihui Park[‡] and Yong-Mook Kang**

Department of Energy and Materials Engineering, Dongguk University-Seoul, Seoul, 04620, Republic of Korea.

*E-mail: dake1234@dongguk.edu

[†]These authors contributed equally to this work

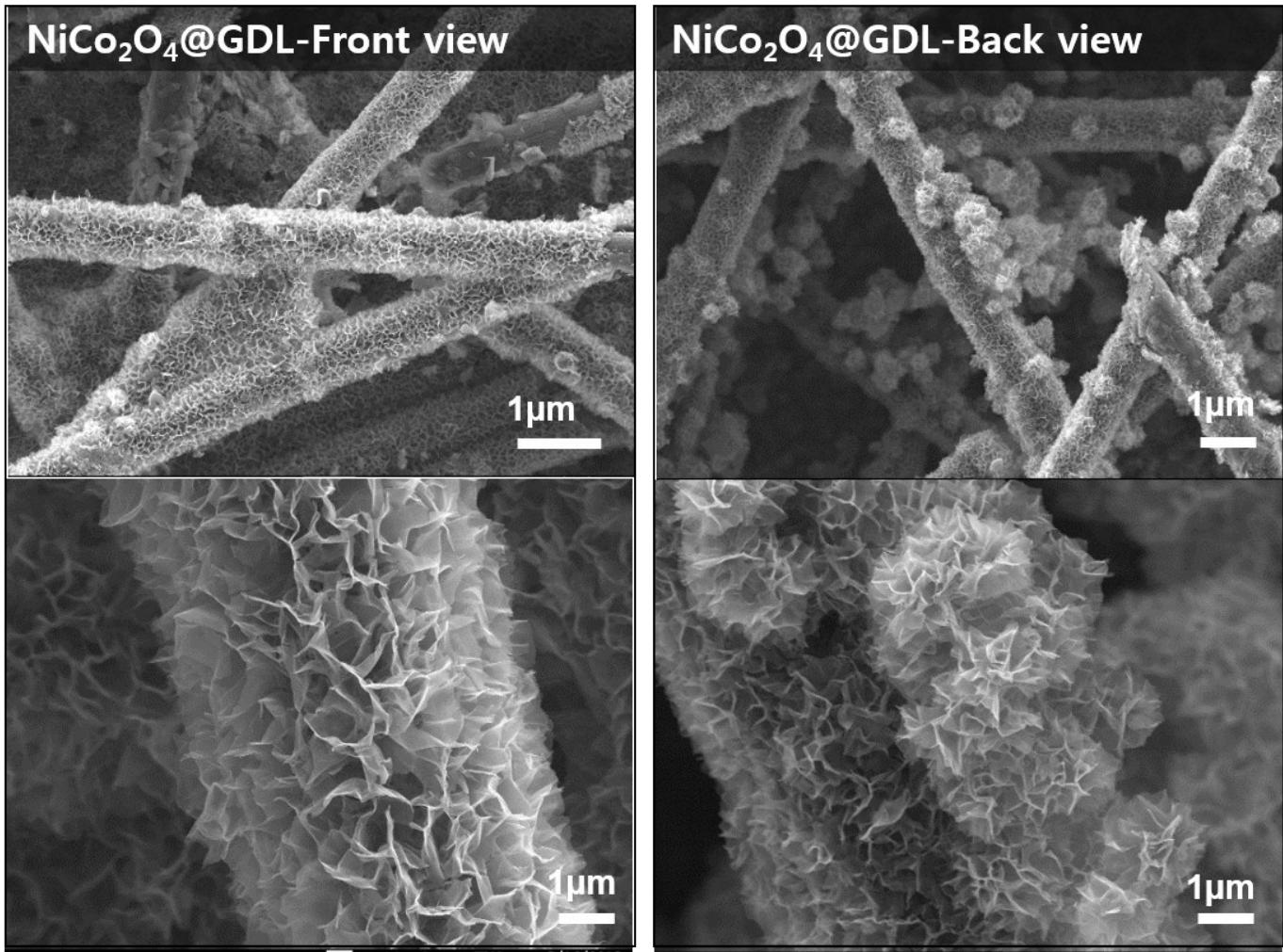


Fig. S1 FESEM images of as-prepared NiCo_2O_4 nanosheets directly grown on both sides of GDL as a result of vertically immersion of GDL in reaction solution

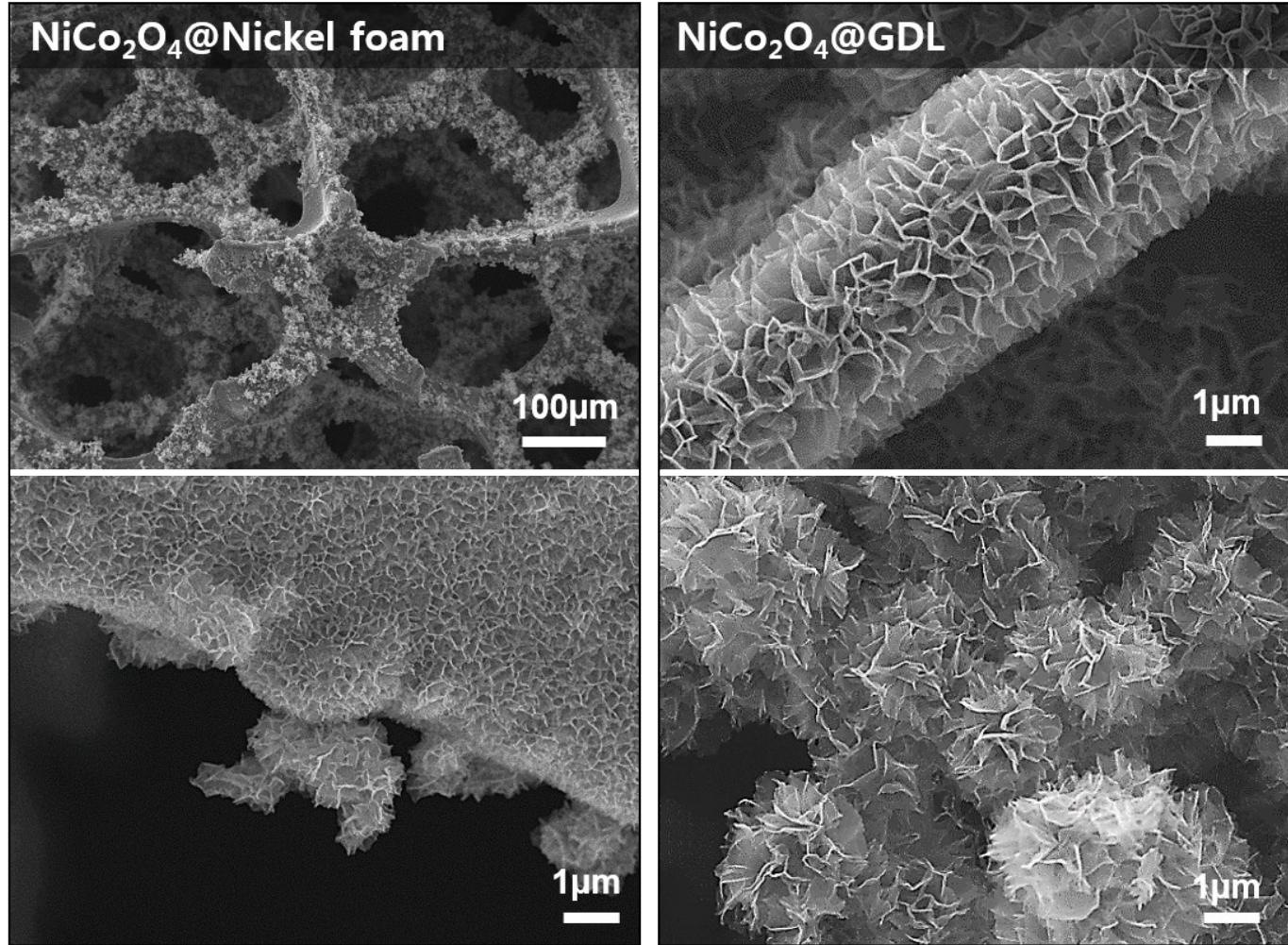


Fig. S2 FESEM images of NiCo_2O_4 nanosheets grown on two substrates (i.e. Nickel foam and GDL) illustrating the compatibility of nanosheets on various substrates

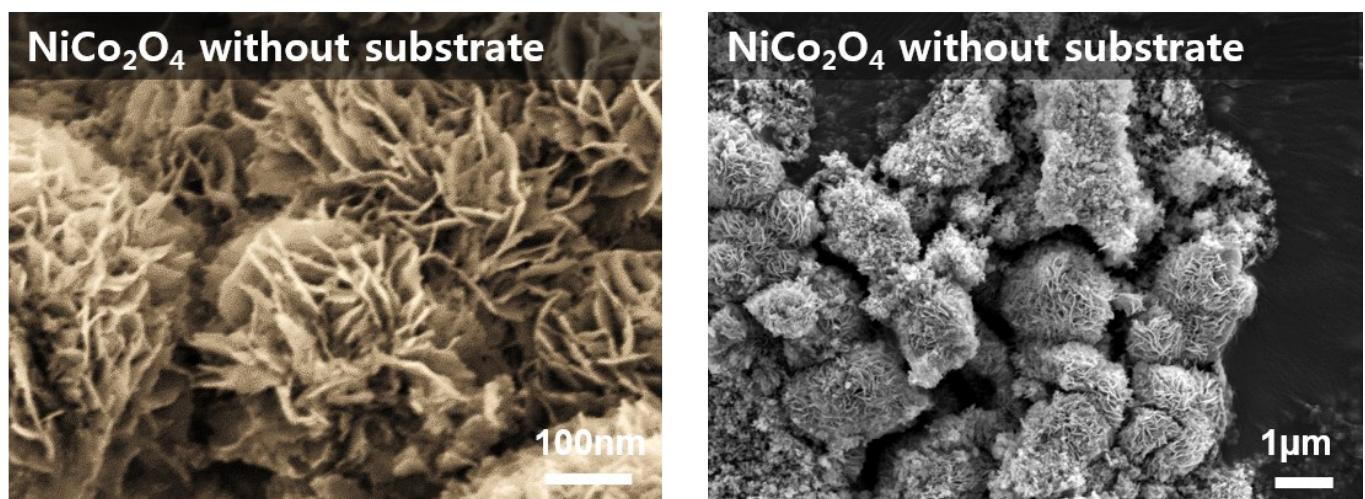
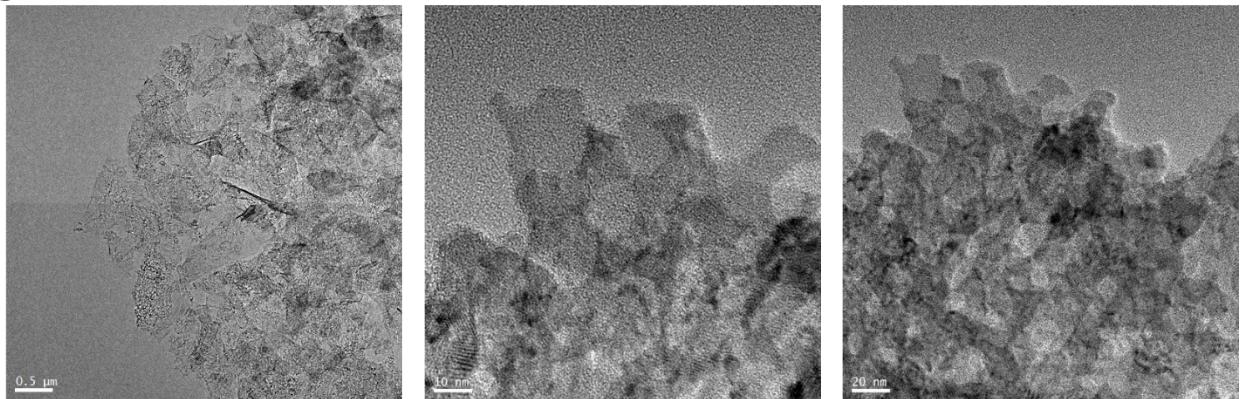


Fig. S3 FESEM images of NiCo₂O₄ nanosheets assembled into flower-like morphology, which is grown without substrate

(a) NCO@GDL



(b) Pd-NCO@GDL

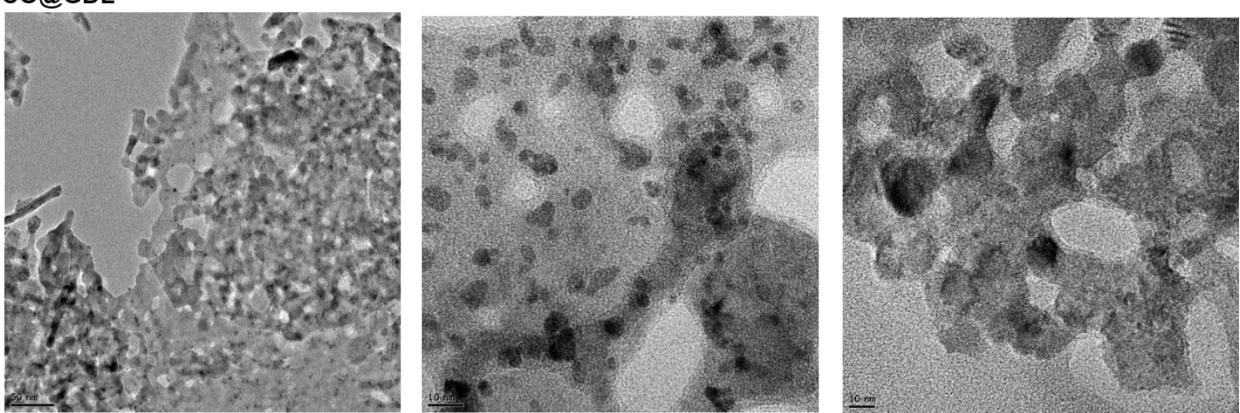


Fig. S4 TEM images of NiCo₂O₄ nanosheets before and after Pd deposition

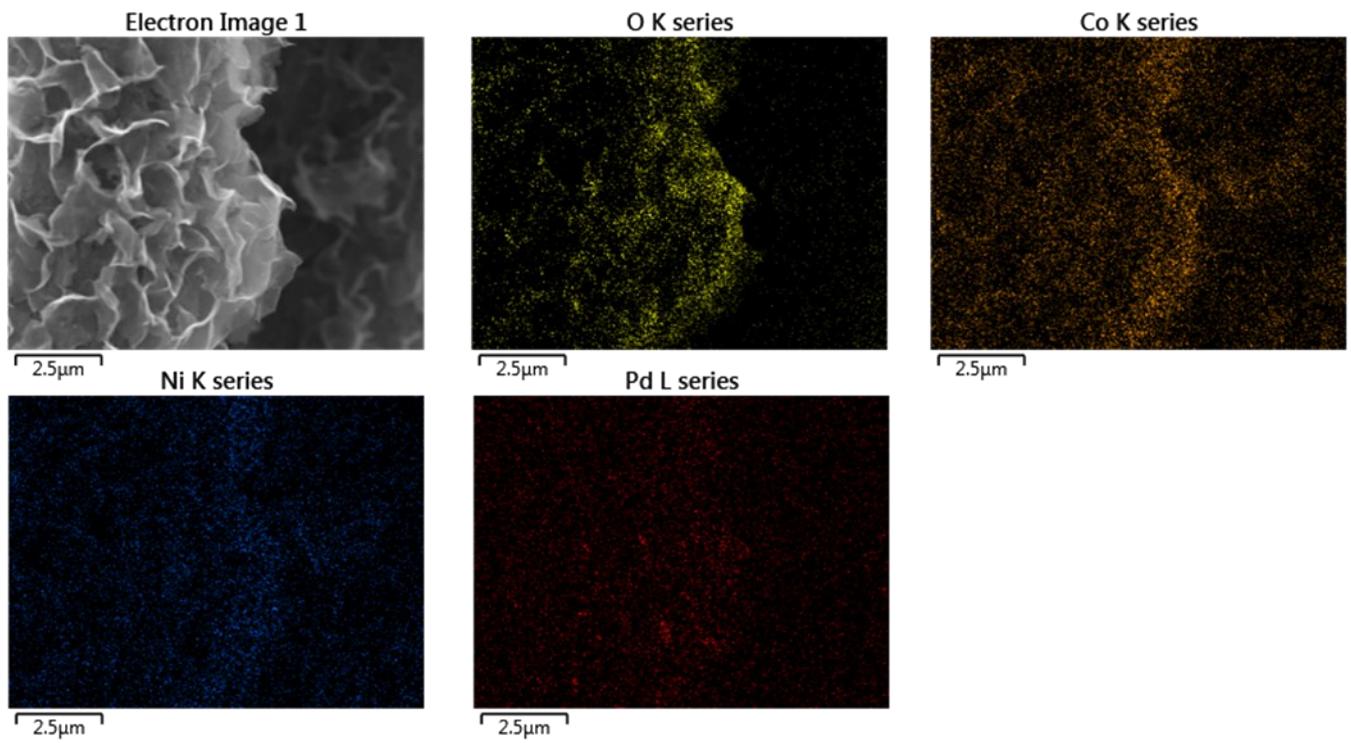


Fig. S5 FESEM-EDS elemental mapping of $\text{Pd}@\text{NiCo}_2\text{O}_4/\text{GDL}$ showing the various elements present in the composition

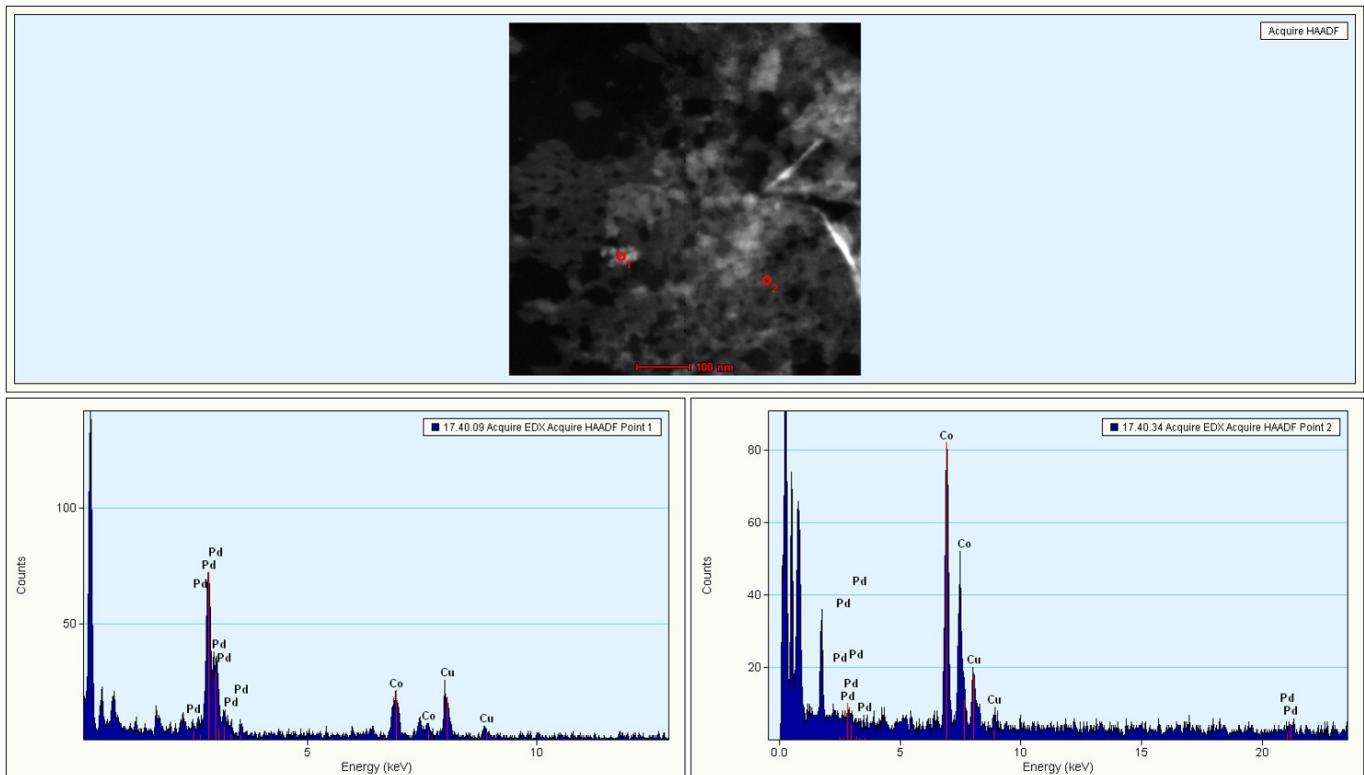


Fig. S6 HAADF-STEM image and point EDX spectra showing the existence of Pd nanoparticles adsorbed onto the NiCo_2O_4 nanosheets

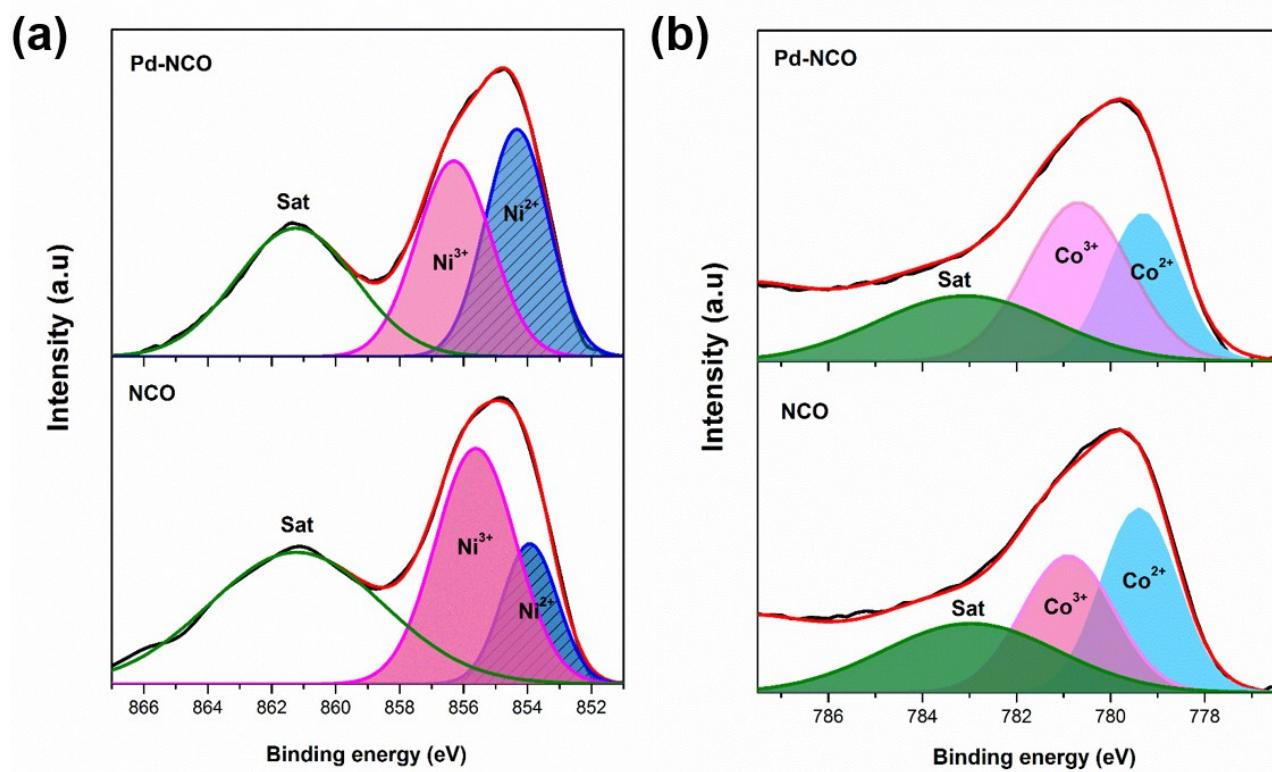


Fig. S7 XPS analysis of (a) Ni 1s and (b) Co 2p spectra demonstrating the change of oxidation states before and after Pd introduction

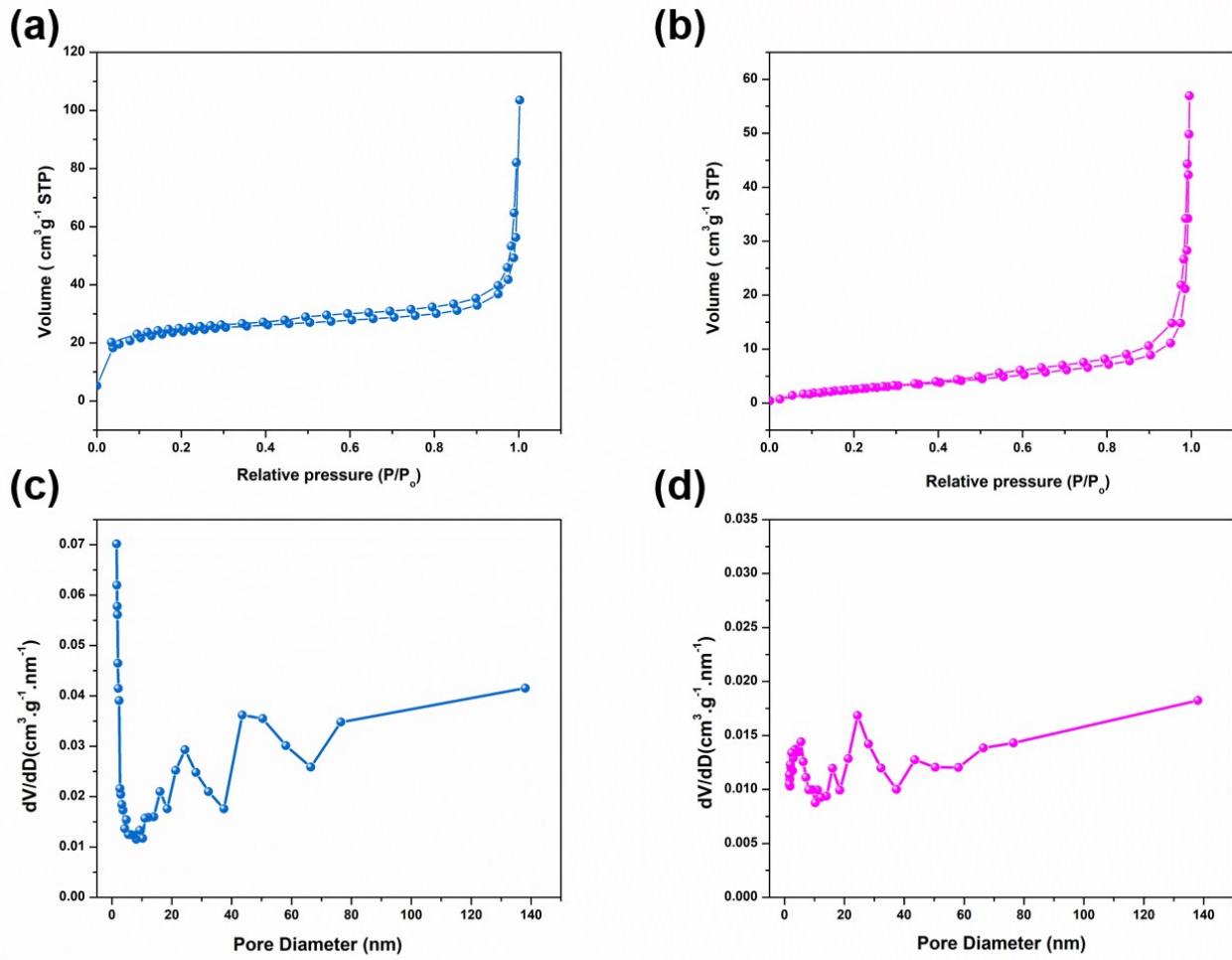


Fig. S8. Nitrogen adsorption-desorption isotherms and Pore size distributions of (a & c) Pd@NiCo₂O₄/GDL and (b & d) NiCo₂O₄/GDL

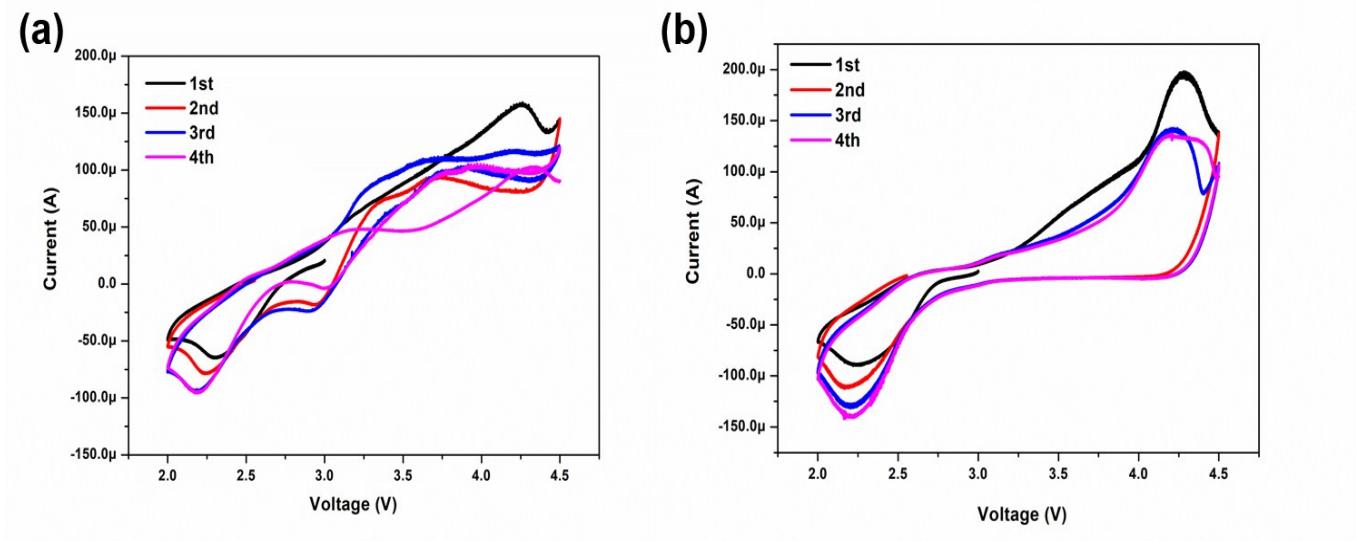


Fig. S9 Cyclic Voltammetry (CV) curves of Li-O₂ cells with (a) Pd@NiCo₂O₄/GDL and (b) NiCo₂O₄/GDL at the scan rate of 0.2mVs⁻¹

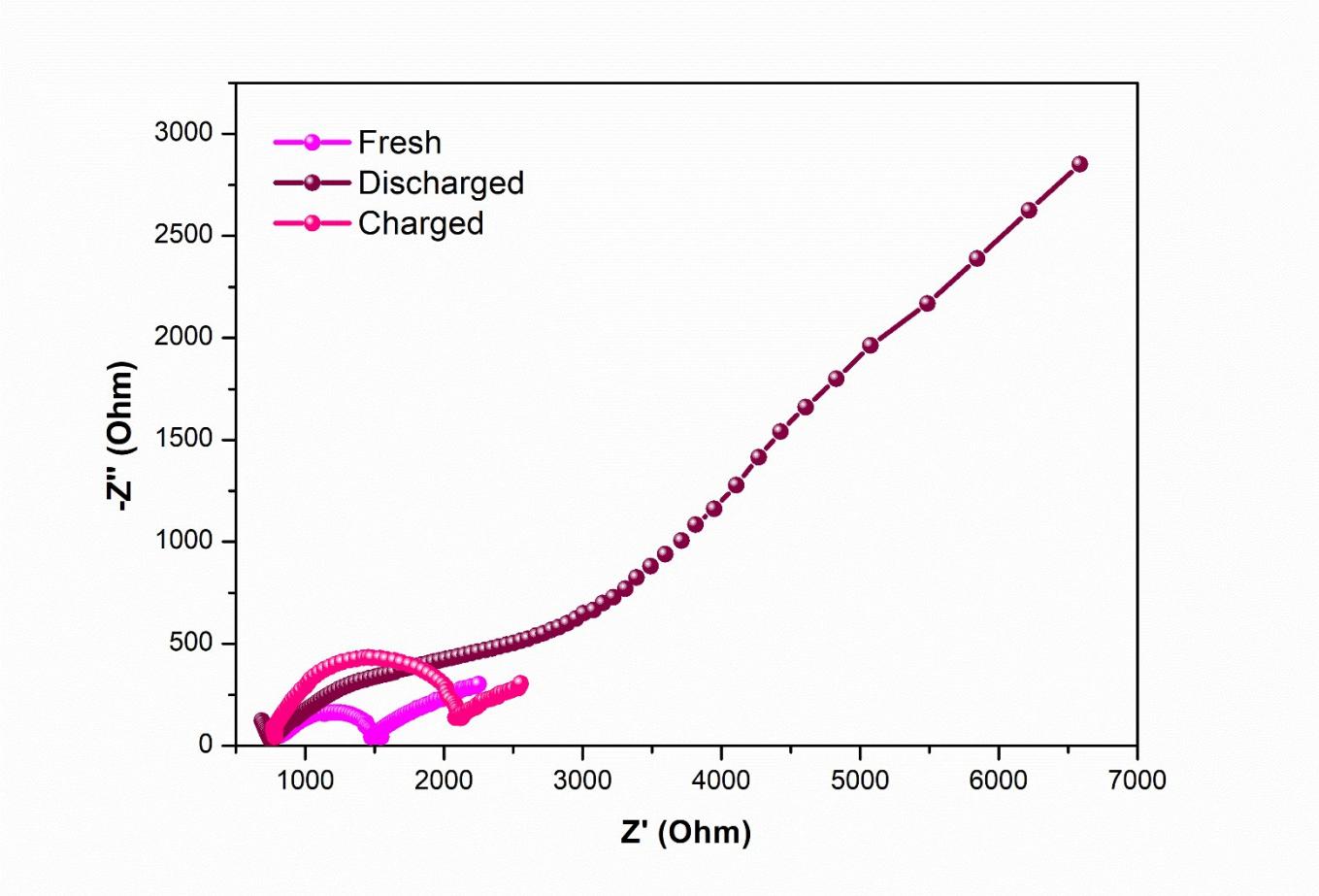


Fig. S10 Nyquist plots of the Li-O₂ cells containing NiCo₂O₄/GDL cathode

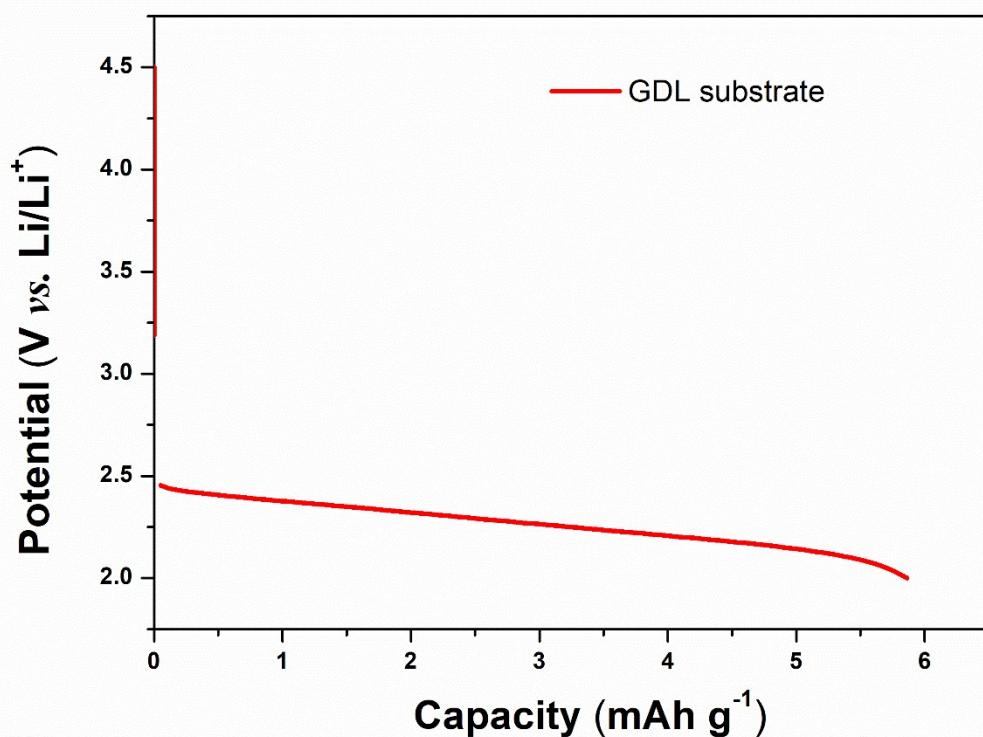


Fig. S11 Discharge curve of bare GDL substrate

Sample Name	Discharge capacity(mAhg ⁻¹)	Cut-Off capacity (mAhg ⁻¹)/ Cycling	Discharge product	Reference
-------------	---	---	-------------------	-----------

	¹⁾ /current density	stability	morphology	
Au@NiCo ₂ O ₄ /3D-G	1275 / 42.5mA g^{-1}	500 / 40	Flakes	23
Flower-like NiCo ₂ O ₄ microsphere	3163 / 0.08mAcm ⁻¹	500 / 60	Thin film	22
Mesoporous spinel NiCo ₂ O ₄	7309 / 0.2mAcm ⁻¹	1000 / 35	precipitate	14
3D Foam-like NiCo ₂ O ₄	10137 / 200mA g^{-1}	1000 / 80	precipitate	13
Porous NiCo ₂ O ₄ nanotube	6000 / 100mA g^{-1}	1000 / 110	toroidal	44
Macroporous/mesoporous NiCo ₂ O ₄ nanosheets	11860 / 200mA g^{-1}	500 / 50	plates	45
Pd@NiCo ₂ O ₄ /GDL	4000 / 200mA g^{-1}	1000 / 100	Flower-like	This work

Table S1 Comparison between the electrochemical properties of various NiCo₂O₄ based electrodes for Li-O₂ batteries

References

13. F. Deng, L. Yu, G. Cheng, T. Lin, M. Sun, F. Ye and Y. Li, J. Power Sources, 2014, **251**, 202–207.
14. H. Guo, L. Liu, T. Li, W. Chen, J. Liu, Y. Guo and Y. Guo, Nanoscale, 2014, **6**, 5491–5497.
22. Z. Li, X. Li, L. Xiang, X. Xie, X. Li, D.-R. Xiao, J. Shen, W. Lu, L. Lu and S. Liu, J. Mater. Chem. A, 2016, **4**, 18335–18341.
23. H.-Q. Wang, J. Chen, S.-J. Hu, X.-H. Zhang, X.-P. Fan, J. Du, Y.-G. Huang and Q.-Y. Li, RSC Adv., 2015, **5**, 72495–72499.
44. L. Li, L. Shen, P. Nie, G. Pang, J. Wang, H. Li, S. Dong and X. Zhang, J. Mater. Chem. A, 2015, **3**, 224309–224314.
45. B. Sun, X. Huang, S. Chen, Y. Zhao, J. Zhang, P. Munroe and G. Wang, J. Mater. Chem. A, 2014, **2**, 12053–12059