

Electronic Supplementary Information

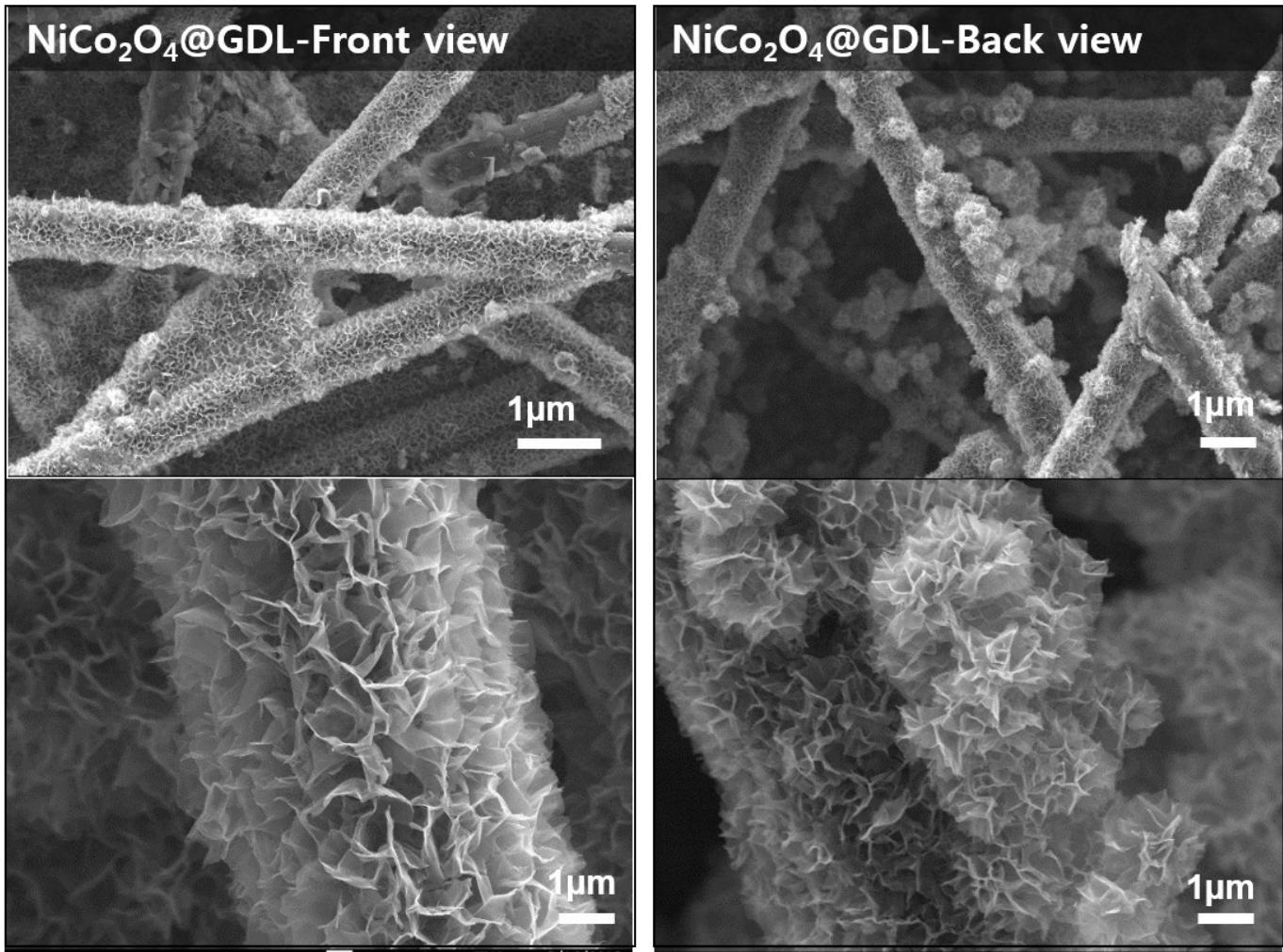
**Pd-Impregnated  $\text{NiCo}_2\text{O}_4$  nanosheets/porous carbon composites as a free-standing and binder-free catalyst for a high energy lithium–oxygen battery**

*Daniel A. Agyeman<sup>†</sup>, Mihui Park<sup>‡</sup> and Yong-Mook Kang\**

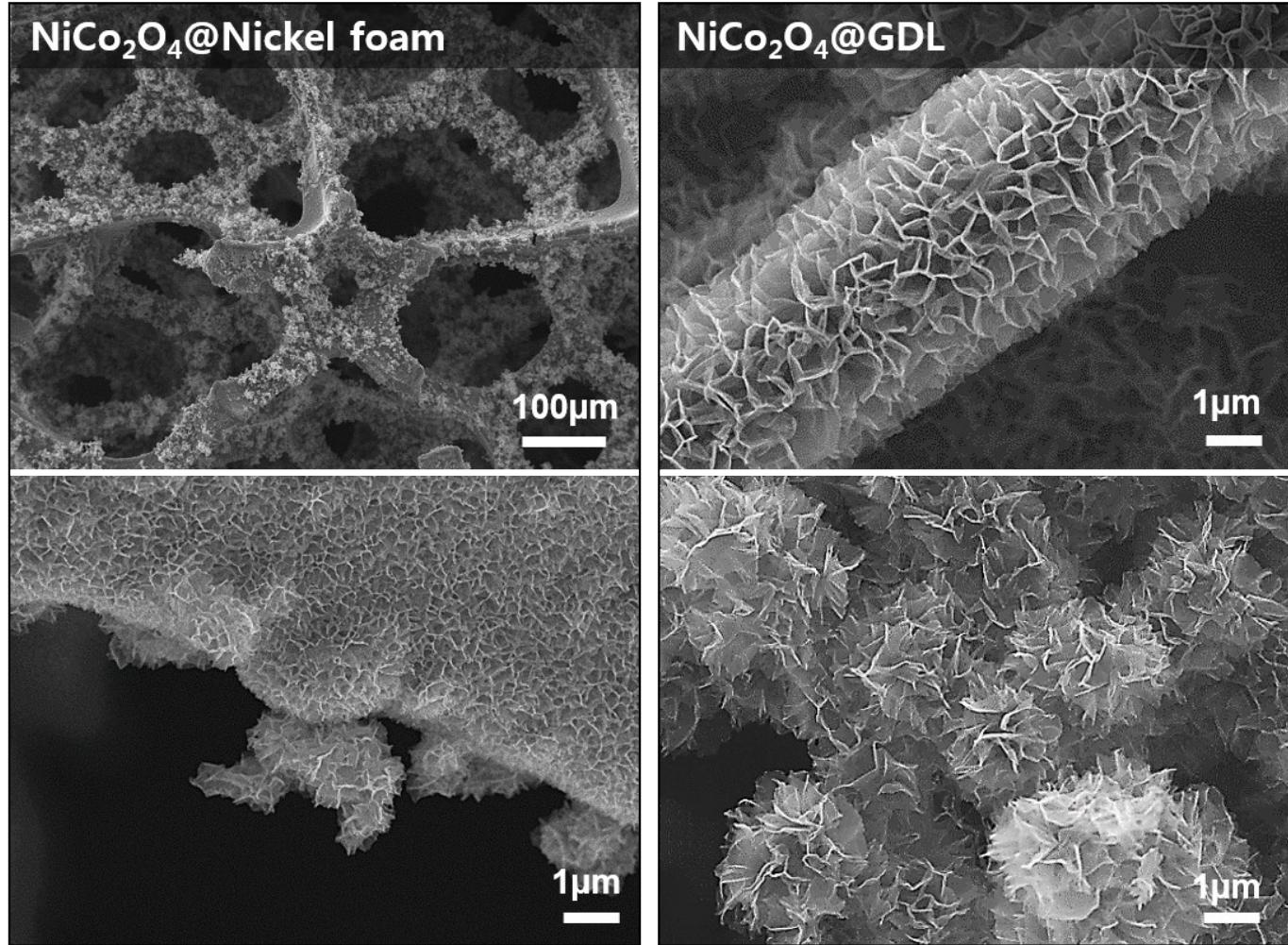
Department of Energy and Materials Engineering, Dongguk University-Seoul, Seoul, 04620, Republic of Korea.

\*E-mail: [dake1234@dongguk.edu](mailto:dake1234@dongguk.edu)

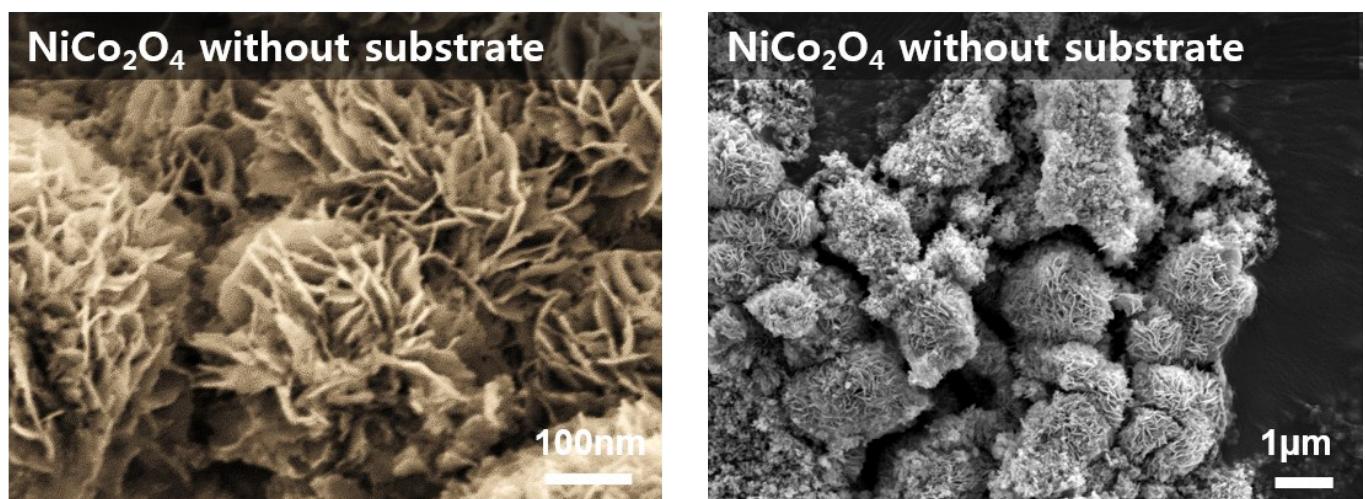
<sup>†</sup>These authors contributed equally to this work



**Fig. S1** FESEM images of as-prepared  $\text{NiCo}_2\text{O}_4$  nanosheets directly grown on both sides of GDL as a result of vertically immersion of GDL in reaction solution

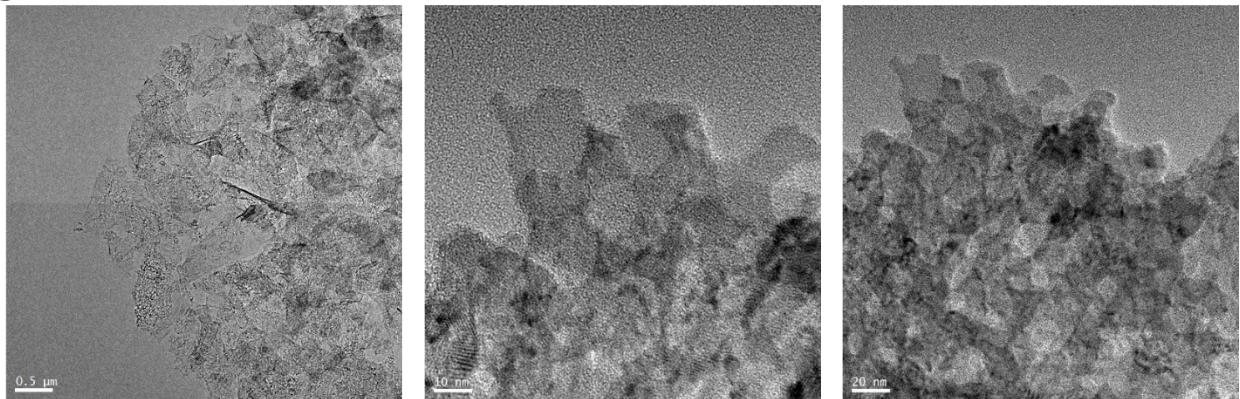


**Fig. S2** FESEM images of  $\text{NiCo}_2\text{O}_4$  nanosheets grown on two substrates (i.e. Nickel foam and GDL) illustrating the compatibility of nanosheets on various substrates

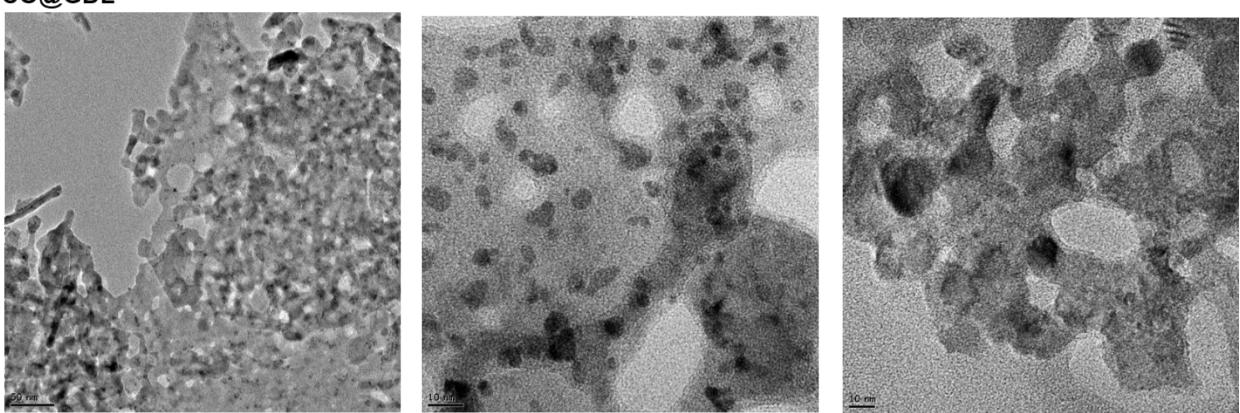


**Fig. S3** FESEM images of NiCo<sub>2</sub>O<sub>4</sub> nanosheets assembled into flower-like morphology, which is grown without substrate

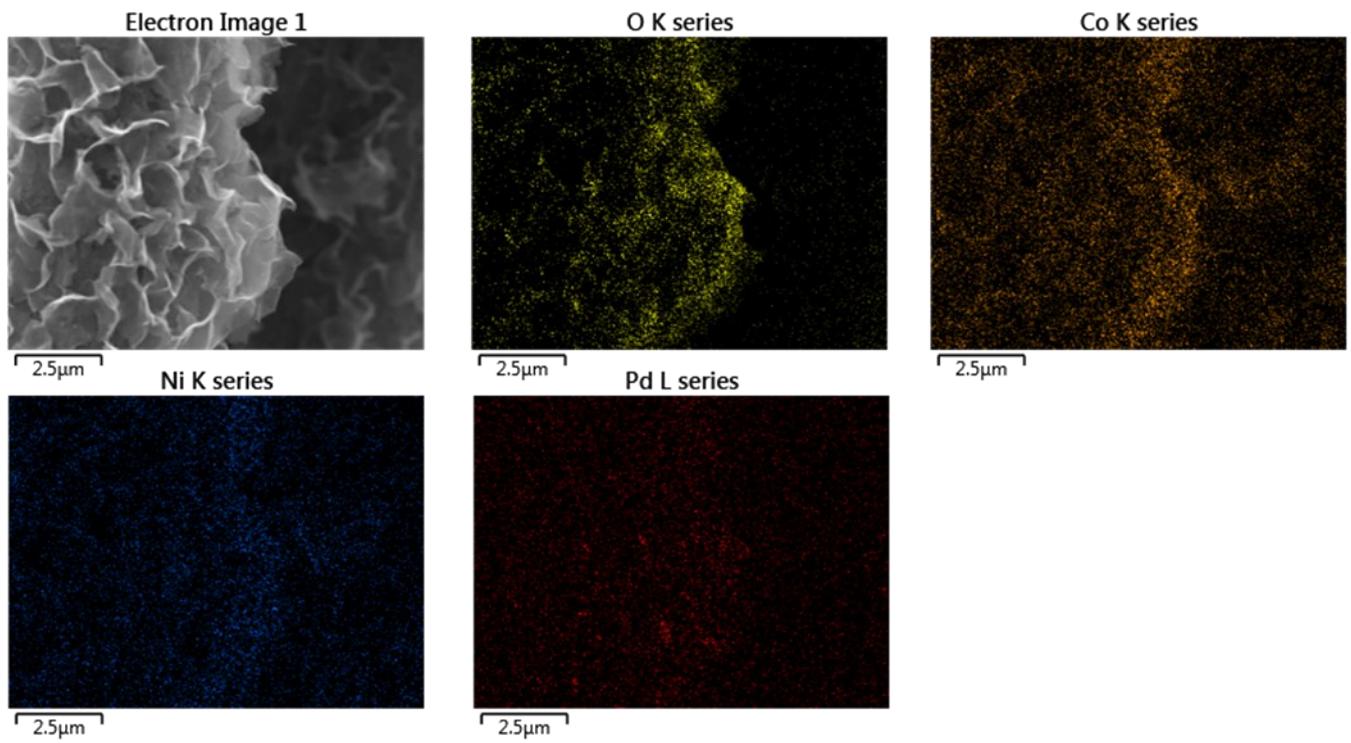
**(a) NCO@GDL**



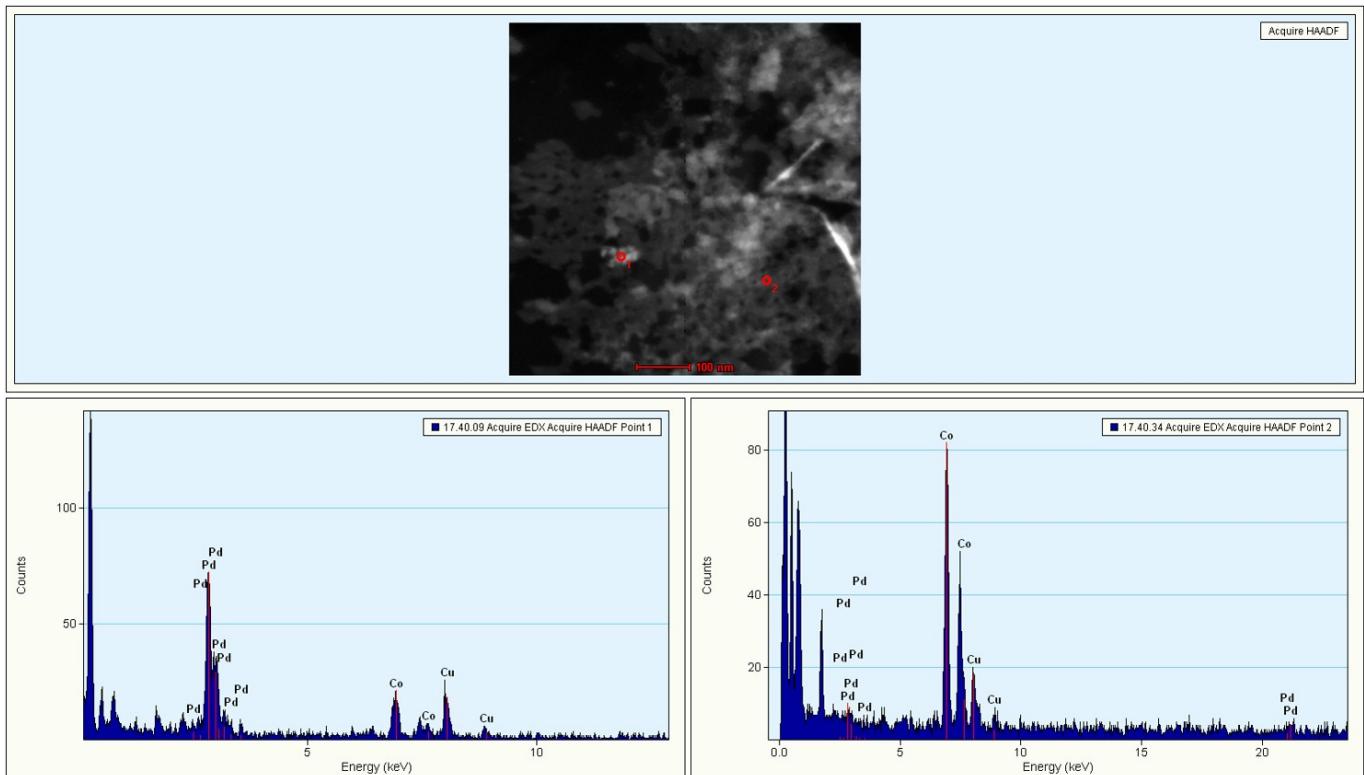
**(b) Pd-NCO@GDL**



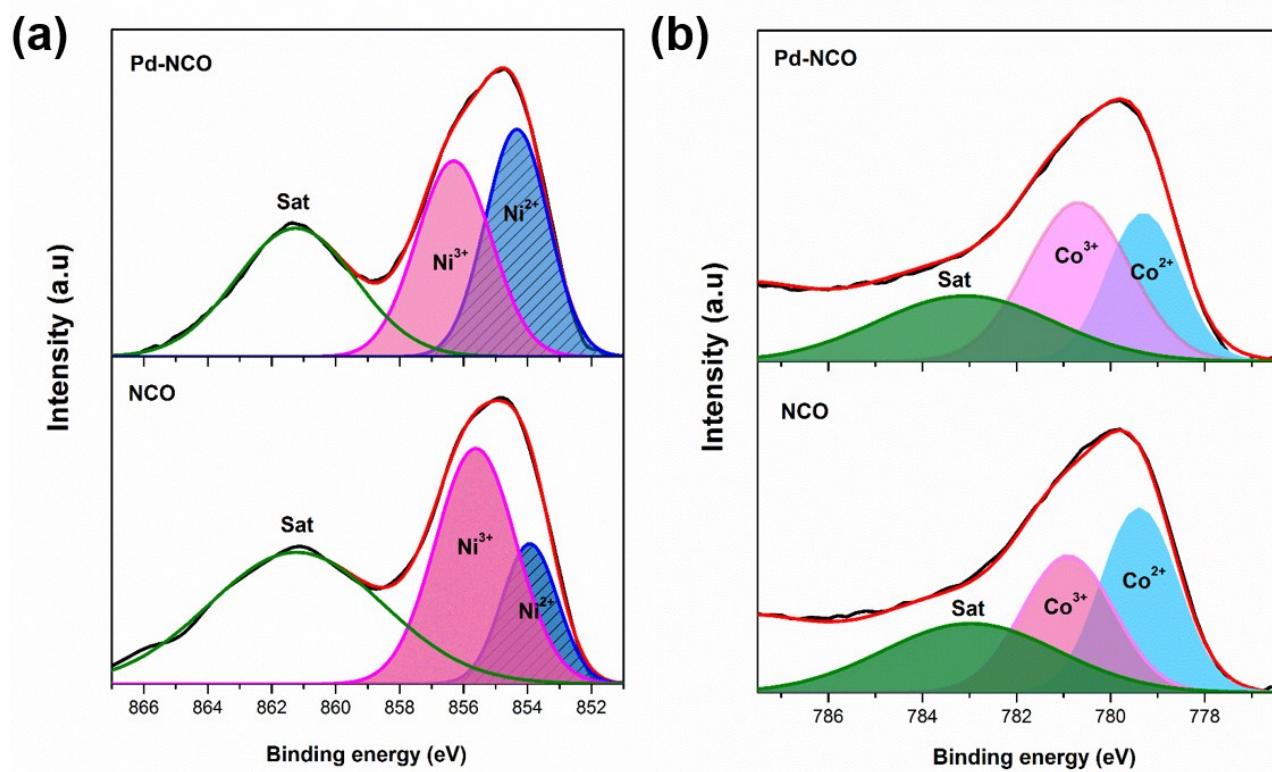
**Fig. S4** TEM images of NiCo<sub>2</sub>O<sub>4</sub> nanosheets before and after Pd deposition



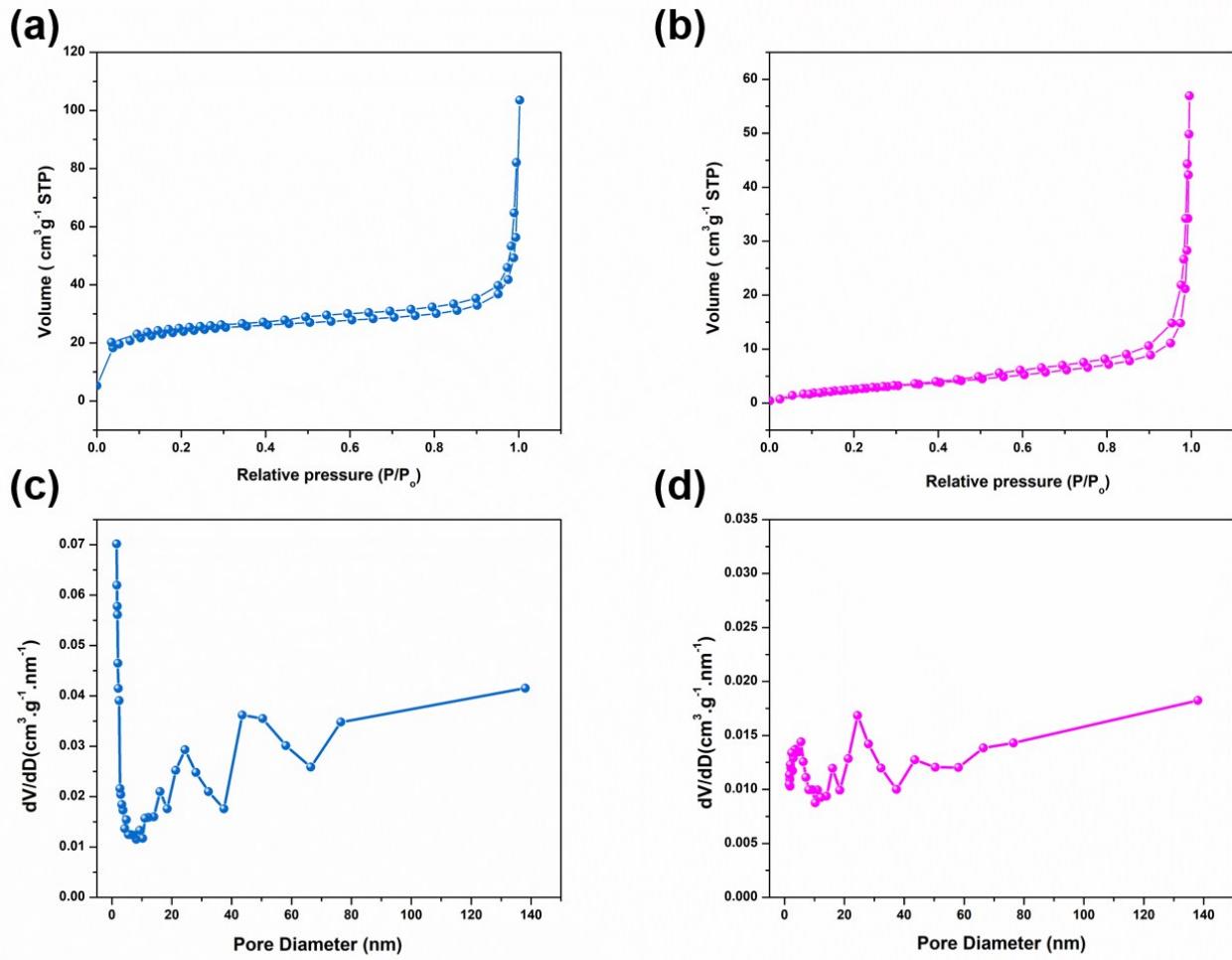
**Fig. S5** FESEM-EDS elemental mapping of  $\text{Pd}@\text{NiCo}_2\text{O}_4/\text{GDL}$  showing the various elements present in the composition



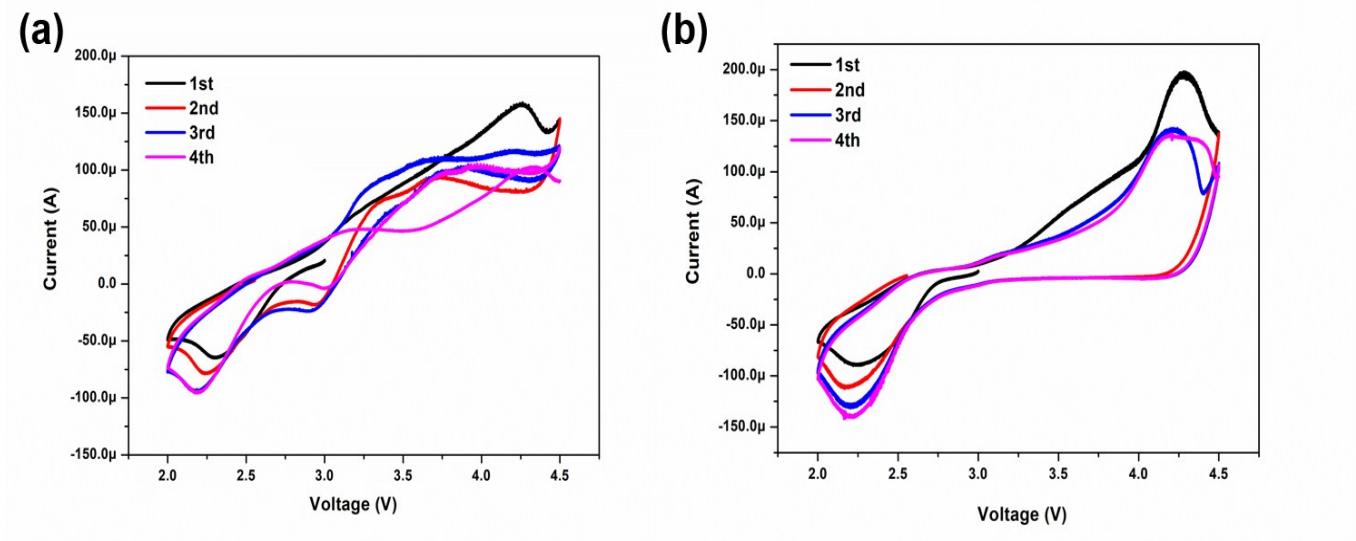
**Fig. S6** HAADF-STEM image and point EDX spectra showing the existence of Pd nanoparticles adsorbed onto the  $\text{NiCo}_2\text{O}_4$  nanosheets



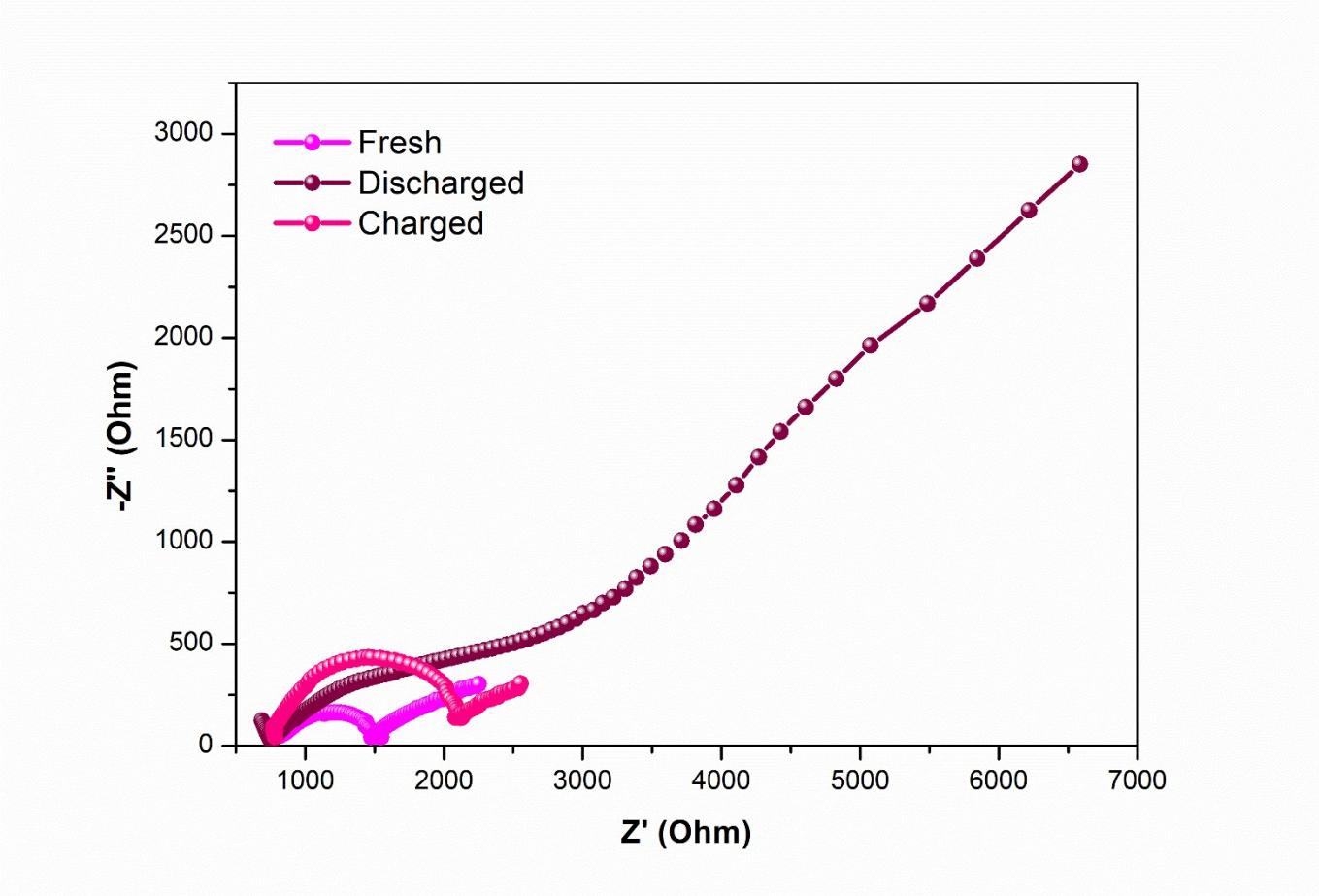
**Fig. S7** XPS analysis of (a) Ni 1s and (b) Co 2p spectra demonstrating the change of oxidation states before and after Pd introduction



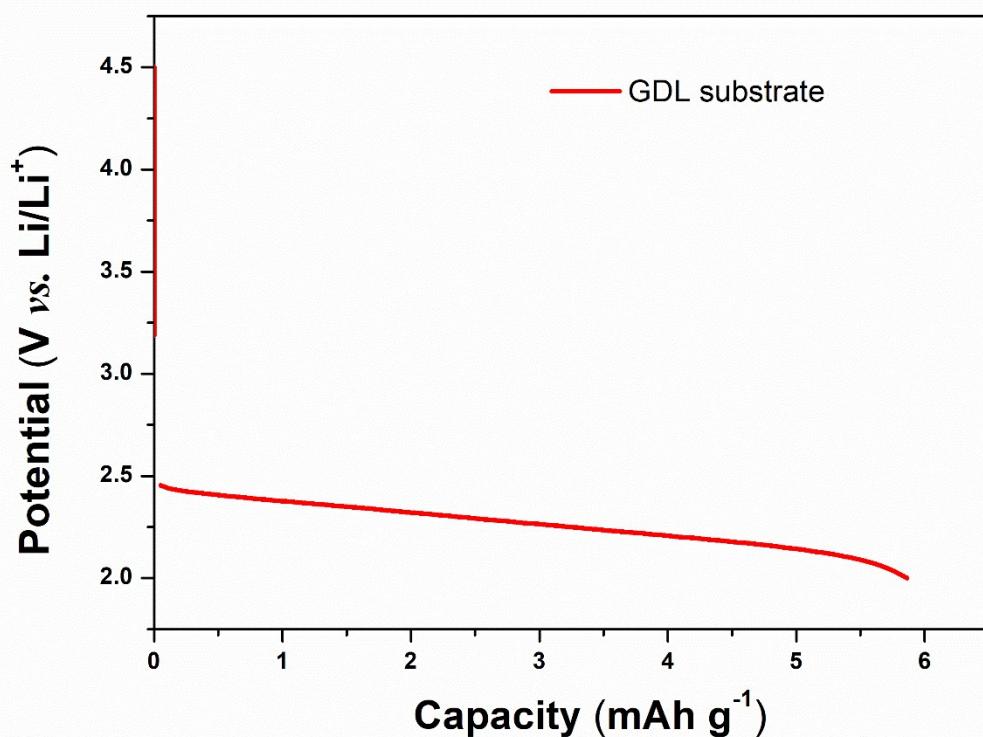
**Fig. S8.** Nitrogen adsorption-desorption isotherms and Pore size distributions of (a & c) Pd@NiCo<sub>2</sub>O<sub>4</sub>/GDL and (b & d) NiCo<sub>2</sub>O<sub>4</sub>/GDL



**Fig. S9** Cyclic Voltammetry (CV) curves of Li-O<sub>2</sub> cells with (a) Pd@NiCo<sub>2</sub>O<sub>4</sub>/GDL and (b) NiCo<sub>2</sub>O<sub>4</sub>/GDL at the scan rate of 0.2mVs<sup>-1</sup>



**Fig. S10** Nyquist plots of the Li-O<sub>2</sub> cells containing NiCo<sub>2</sub>O<sub>4</sub>/GDL cathode



**Fig. S11** Discharge curve of bare GDL substrate

Sample Name	Discharge capacity(mAhg <sup>-1</sup> )	Cut-Off capacity (mAhg <sup>-1</sup> )/ Cycling	Discharge product	Reference
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	<sup>1)</sup> /current density	stability	morphology	
Au@NiCo <sub>2</sub> O <sub>4</sub> /3D-G	1275 / 42.5mA $\text{g}^{-1}$	500 / 40	Flakes	23
Flower-like NiCo <sub>2</sub> O <sub>4</sub> microsphere	3163 / 0.08mAcm <sup>-1</sup>	500 / 60	Thin film	22
Mesoporous spinel NiCo <sub>2</sub> O <sub>4</sub>	7309 / 0.2mAcm <sup>-1</sup>	1000 / 35	precipitate	14
3D Foam-like NiCo <sub>2</sub> O <sub>4</sub>	10137 / 200mA $\text{g}^{-1}$	1000 / 80	precipitate	13
Porous NiCo <sub>2</sub> O <sub>4</sub> nanotube	6000 / 100mA $\text{g}^{-1}$	1000 / 110	toroidal	44
Macroporous/mesoporous NiCo <sub>2</sub> O <sub>4</sub> nanosheets	11860 / 200mA $\text{g}^{-1}$	500 / 50	plates	45
Pd@NiCo <sub>2</sub> O <sub>4</sub> /GDL	4000 / 200mA $\text{g}^{-1}$	1000 / 100	Flower-like	This work

**Table S1** Comparison between the electrochemical properties of various NiCo<sub>2</sub>O<sub>4</sub> based electrodes for Li-O<sub>2</sub> batteries

## References

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