

Electronic Supplementary Information for

Effects of co-coordinated auxiliary ligands on photoelectrochemical behaviours of titanium-alkoxide-dyes

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Figures

Fig. S1. Experimental XRD patterns of **1** (a), **2** (b) and **3** (c) along with the calculated patterns from the crystal data.

Fig. S2. Ball-stick plots of compound **2** (a), **3** (b), showing the molecular structure.

Fig. S3. The surface morphologies of the nano TiO₂ film (a) and films of **1** (b), **2** (c), **3** (d), showing the smooth and uniform surfaces.

Fig. S4. The absorption spectra of **1**, dye Az and Az treated TiO₂ substrate.

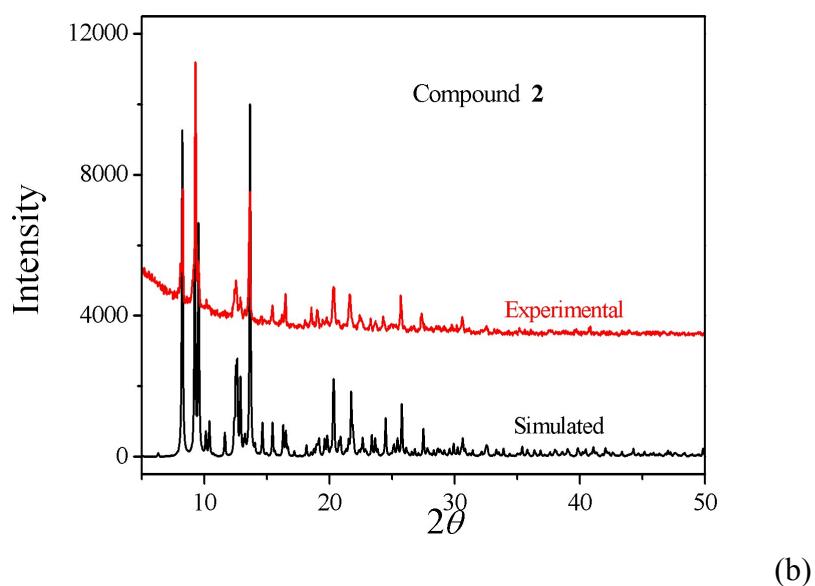
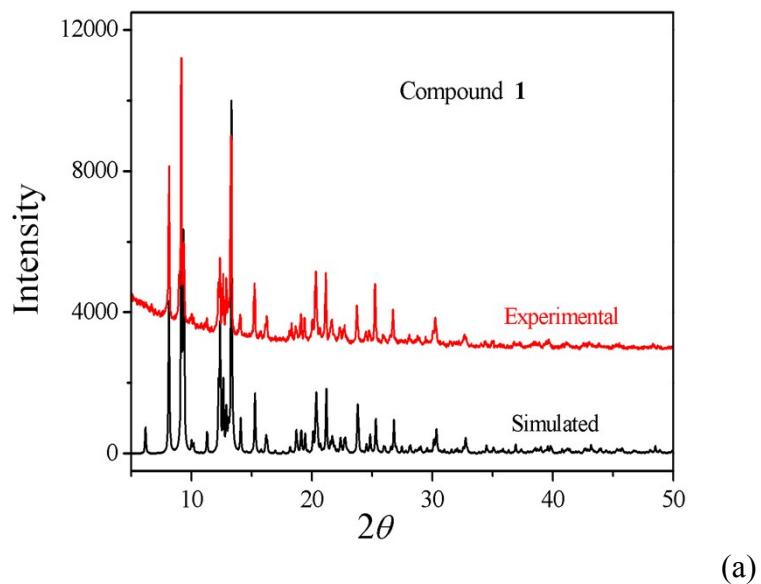
Fig. S5. Raman spectra of the crystal **1** and **1** treated TiO₂ electrode.

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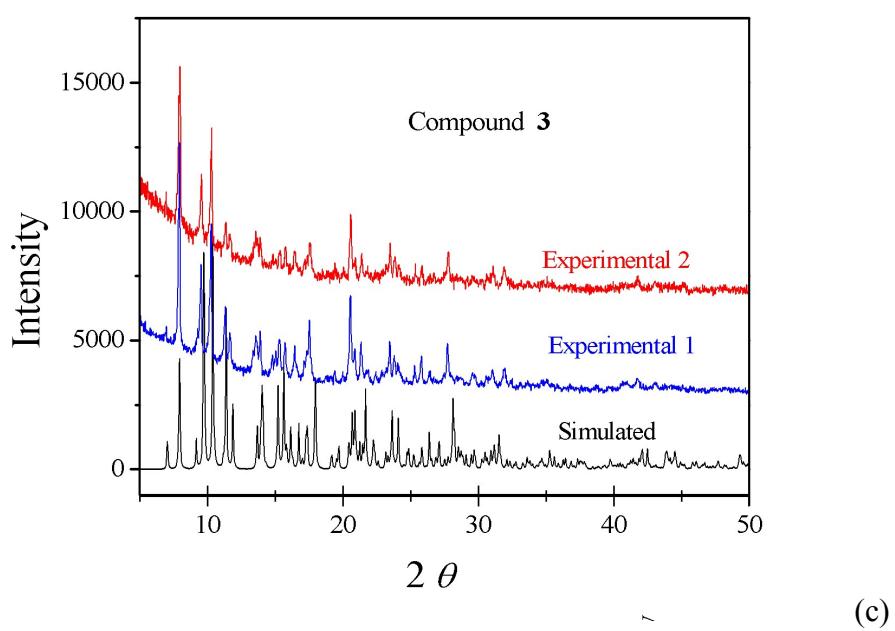


Fig. S1. Experimental XRD patterns of **1** (a), **2** (b) and **3** (c) along with the calculated patterns from the crystal data.

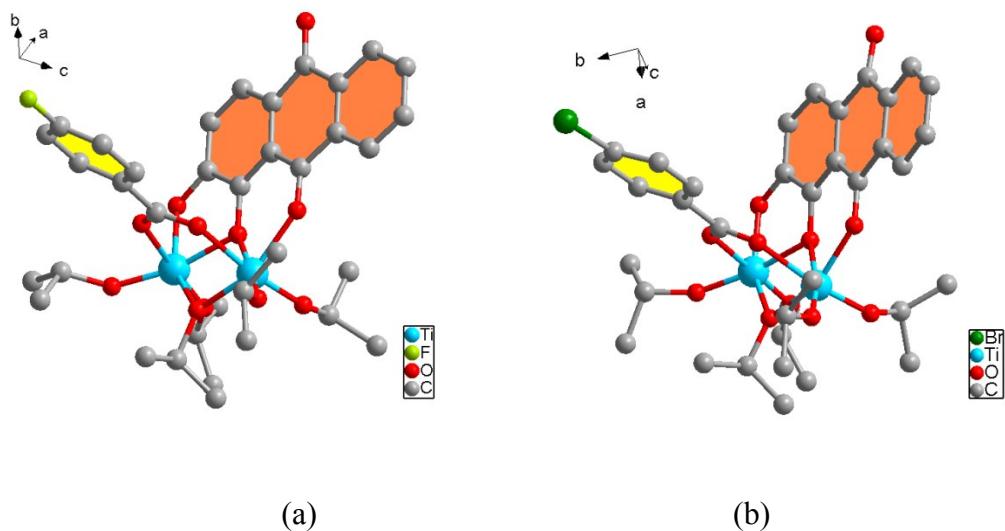


Fig. S2. Ball-stick plots of compound **2** (a), **3** (b), showing the molecular structure.

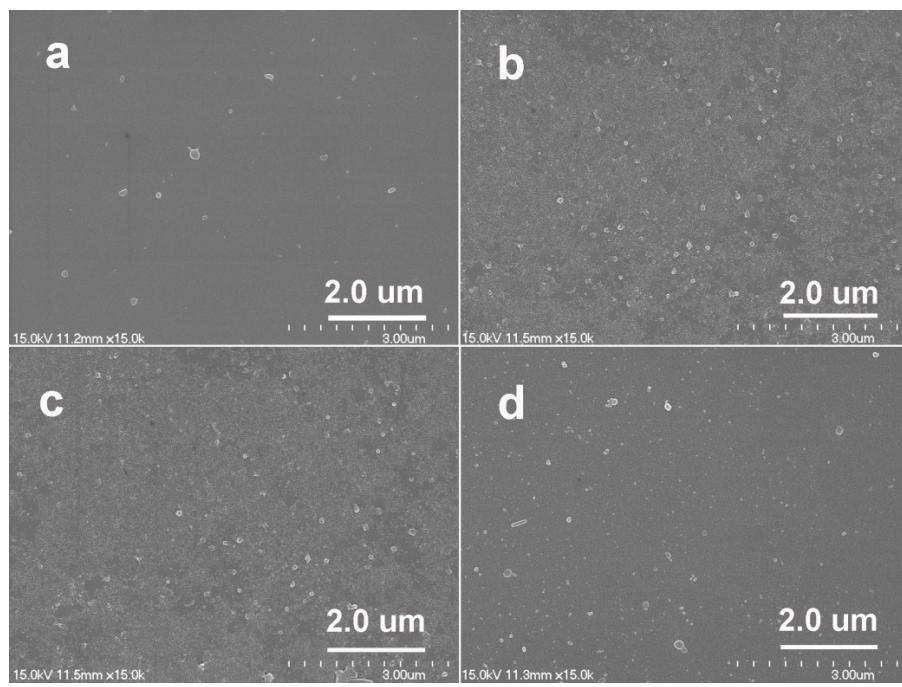


Fig. S3. The surface morphologies of the nano TiO₂ film (a) and films of **1** (b), **2** (c), **3** (d), showing the smooth and uniform surfaces.

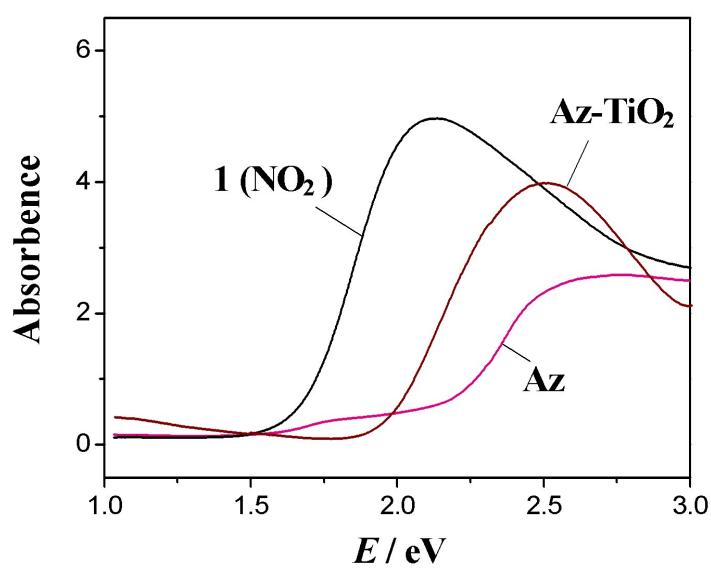


Fig. S4. The absorption spectra of **1**, dye Az and Az treated TiO₂ substrate.

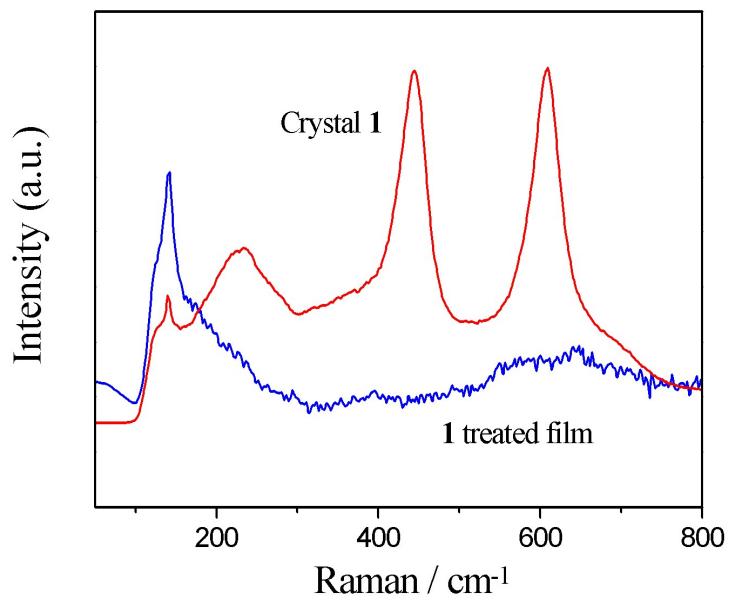


Fig. S5. Raman spectra of the crystal **1** and **1** treated TiO₂ electrode.

T Table S1 Crystal data and structural refinement parameters for **1–3**.

| | 1 | 2 | 3 |
|---|--|--|---|
| formula | C ₃₆ H ₄₅ NO ₁₃ Ti ₂ | C ₃₆ H ₄₅ FO ₁₁ Ti ₂ | C ₃₆ H ₄₅ BrO ₁₁ Ti ₂ |
| fw | 795.51 | 768.51 | 829.41 |
| cryst size (mm ³) | 0.35 × 0.20 × 0.10 | 0.50 × 0.40 × 0.30 | 0.26 × 0.24 × 0.20 |
| cryst syst | monoclinic | monoclinic | monoclinic |
| space group | P2 ₁ /c | P2 ₁ /c | P2 ₁ /c |
| <i>a</i> (Å) | 9.6859(3) | 9.6424(13) | 12.9644 (12) |
| <i>b</i> (Å) | 21.7550(6) | 21.475(3) | 13.1630(12) |
| <i>c</i> (Å) | 19.0222(5) | 18.640(2) | 23.030(2) |
| α (deg) | 90.00 | 90.00 | 90.00 |
| β (deg) | 93.686(3) | 96.6066(4) | 104.724(3) |
| γ (deg) | 90.00 | 90.00 | 90.00 |
| <i>V</i> (Å ³) | 4000.0(2) | 3834.2(9) | 3801.0(6) |
| Z | 4 | 4 | 4 |
| ρ_{calcd} (g cm ⁻³) | 1.321 | 1.331 | 1.449 |
| <i>F</i> (000) | 1664 | 1608 | 1712 |
| μ (mm ⁻¹) | 0.460 | 0.477 | 1.528 |
| <i>T</i> (K) | 293(2) | 293(2) | 120(2) |
| reflns collected | 31104 | 20324 | 60996 |
| unique reflns | 10589 | 6748 | 8698 |
| observed reflns | 6508 | 3688 | 5888 |
| no. params | 479 | 468 | 461 |
| GOF on <i>F</i> ² | 1.065 | 1.084 | 1.165 |
| <i>R</i> ₁ [<i>I</i> >2σ(<i>I</i>)] | 0.0665 | 0.0902 | 0.0543 |
| <i>wR</i> ₂ | 0.1442 | 0.1731 | 0.1345 |

Table S2 Results of TD-DFT calculation for compounds for **1**, **2** and **3**.

| | 1 | 2 | 3 |
|-----------------|----------------------------|----------------------------|----------------------------|
| Gap Energy (eV) | 2.127 | 2.120 | 2.112 |
| Absorption (nm) | 583 | 585 | 587 |
| <i>f</i> | 0.0996 | 0.1045 | 0.1024 |
| Transition | 198A → 199A 198B → 199B | 204A → 205A 204B → 205B | 191A → 192A 191B → 192B |
| Coefficient | 0.69933 | 0.69924 | 0.69942 |