

Supporting Information

Synthesis of 2D/3D Carbon Hybrids by Heterogeneous Space-Confined Effect for Electrochemical Energy Storage[†]

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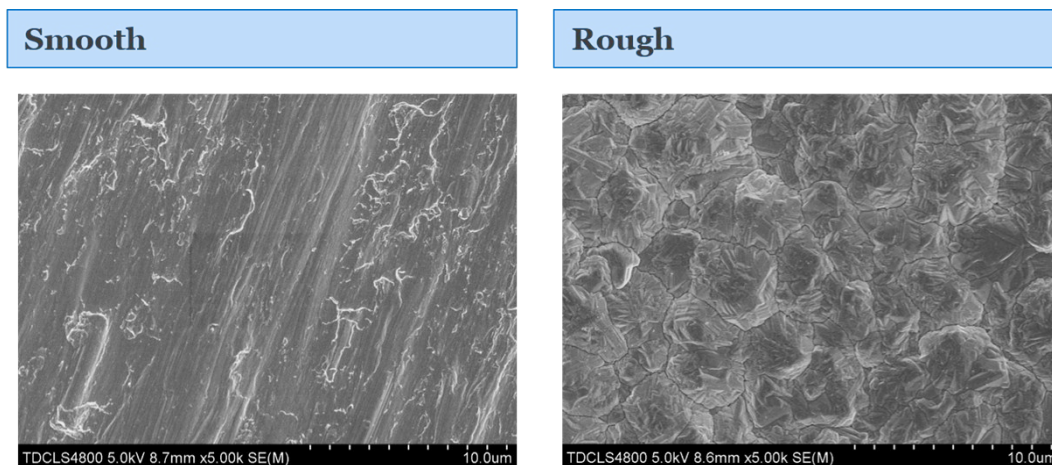


Figure S1. (a) The smooth surface of Cu foil, which can not remain the precursor solution. (b) The rough surface of Cu foil, which can be used for producing N-GPC.

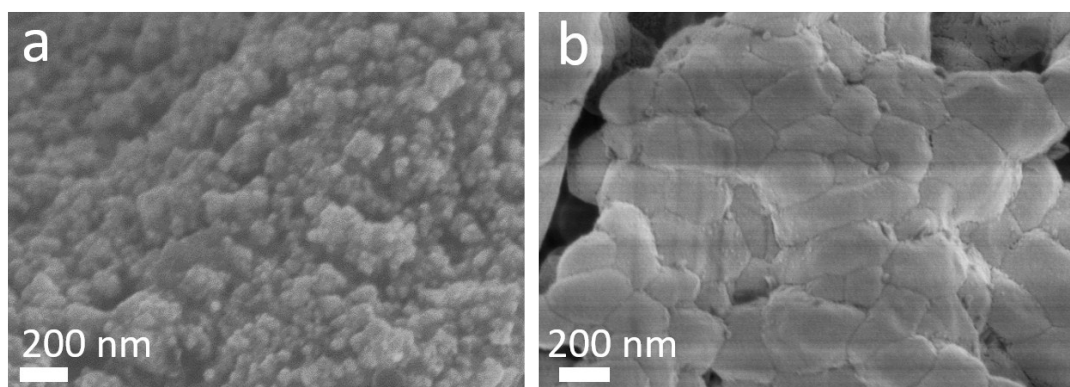


Figure S2. The carbon-coated salt particles of (a) N-GPC and (b) N-PC.

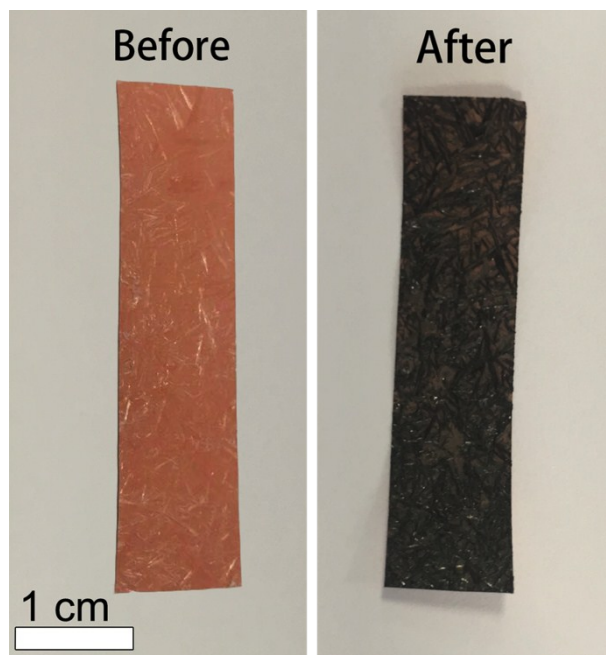


Figure S3. The optical images of the coated Cu foil ($4 \times 1 \text{ cm}^2$) before and after the calcination.

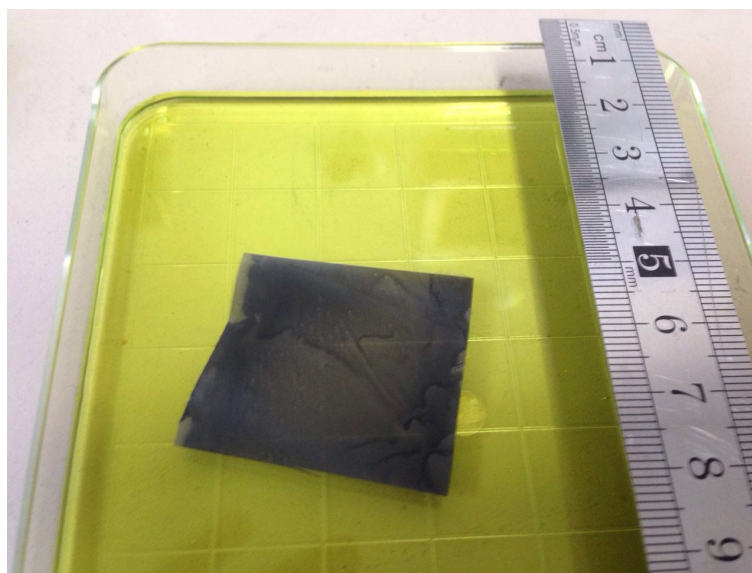


Figure S4. The optical image of free-standing N-GPC floating on the $\text{FeCl}_3\text{-HCl}$ solution.

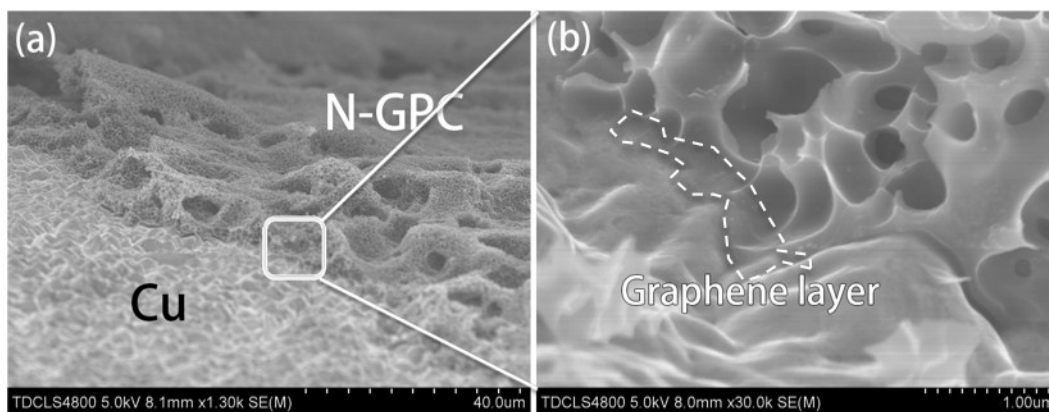


Figure S5. The SEM images: cross view of sandwich-like structure (N-GPC/Graphene/Cu).

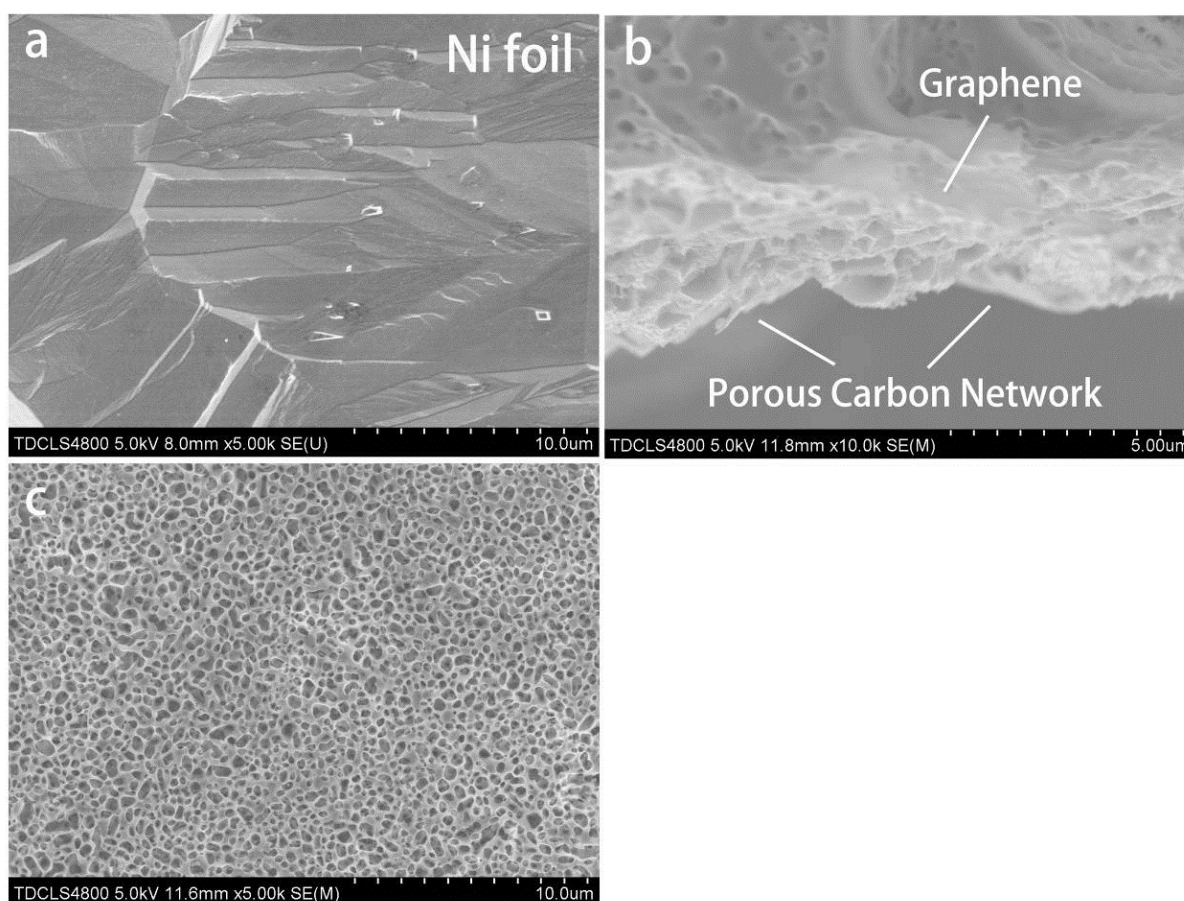


Figure S6. The SEM images of (a) Ni foil; (b) profile view and (c) top view of N-GPC based on Ni substrate.

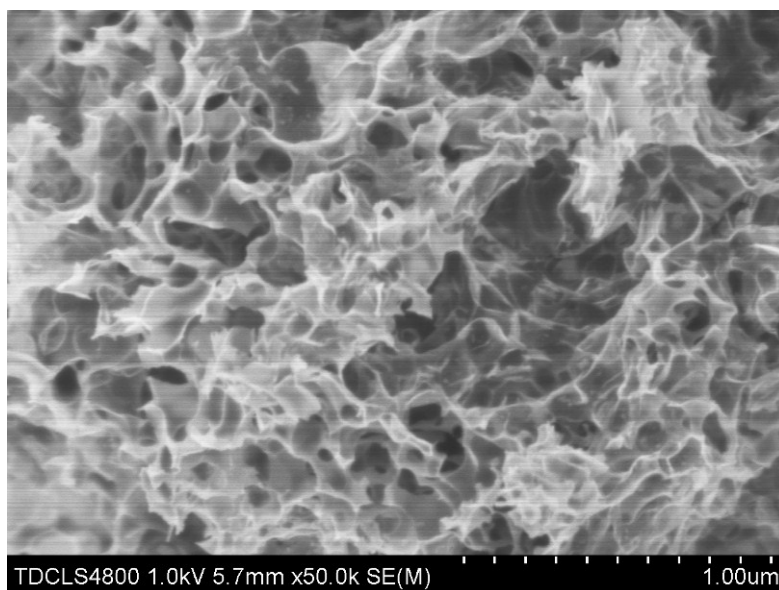


Figure S7. The SEM image of N-PC.

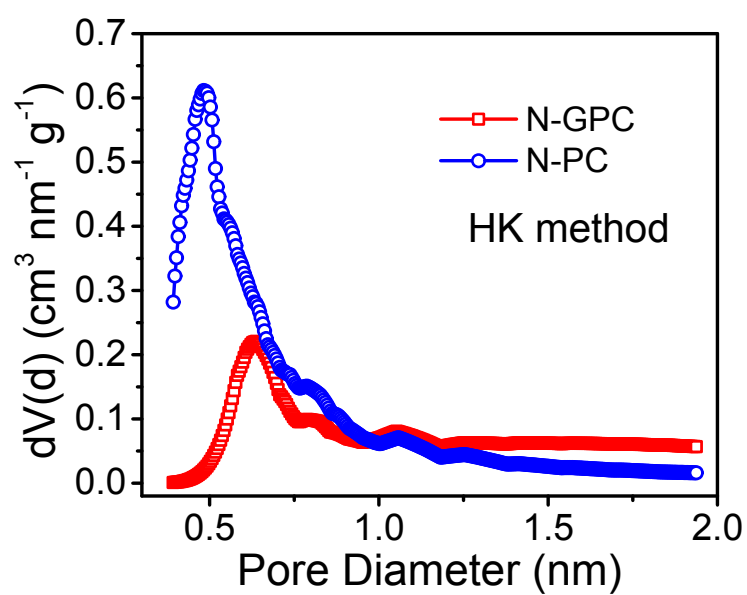


Figure S8. The micropores size distribution analysis based on HK method of N-GPC and N-PC.

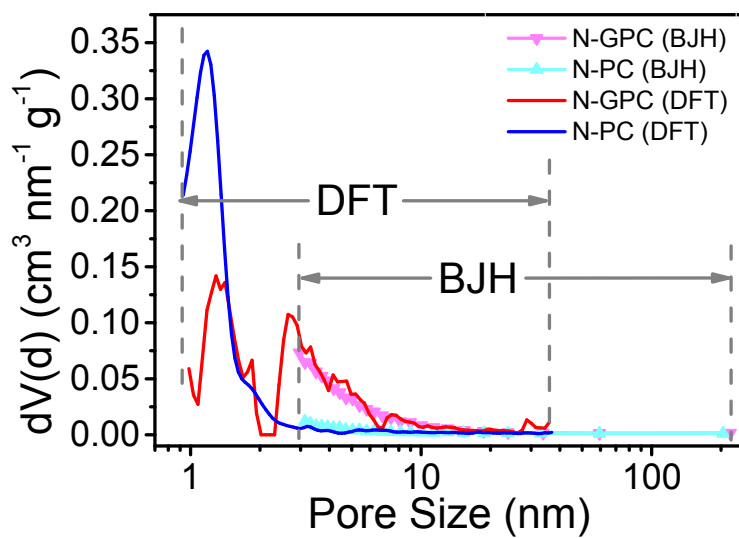


Figure S9. Pore size distribution analysis combining the DFT and BJH theories.

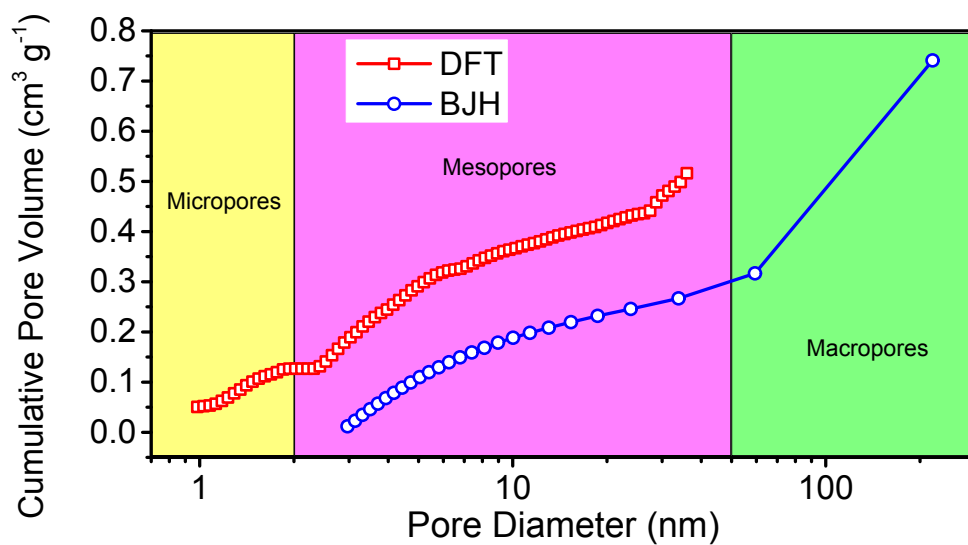


Figure S10. The cumulative pore volumes of N-GPC based on DFT and BJH method.

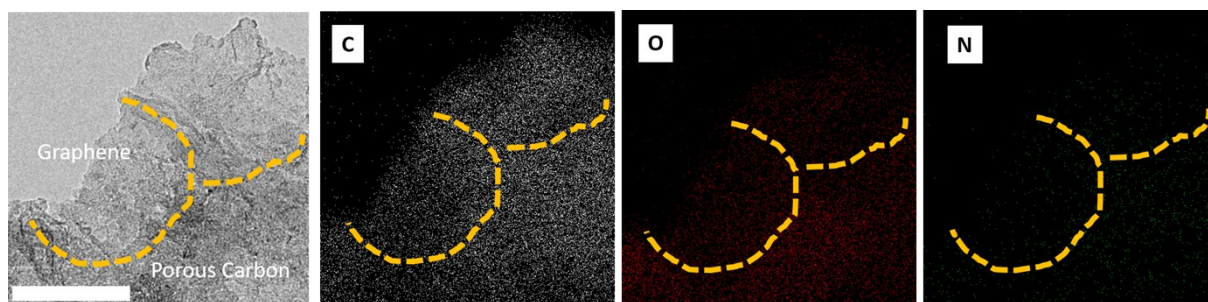


Figure S11. The EDS mapping of N-GPC. The orange line indicates the outline of 3D porous carbon.

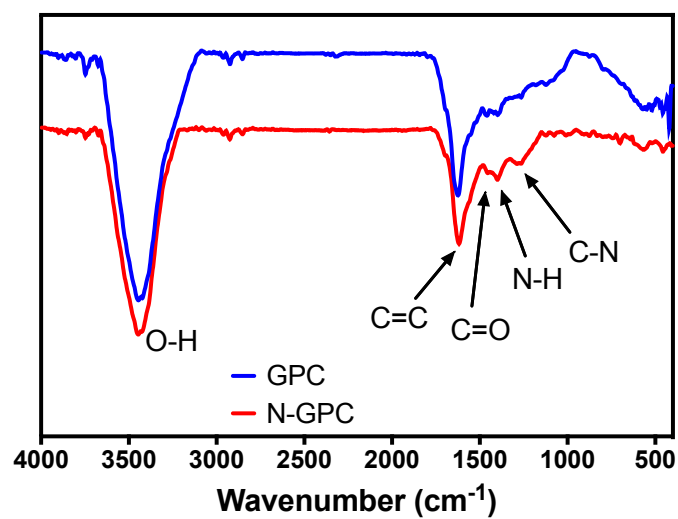


Figure S12. FTIR results of N-GPC and GPC.

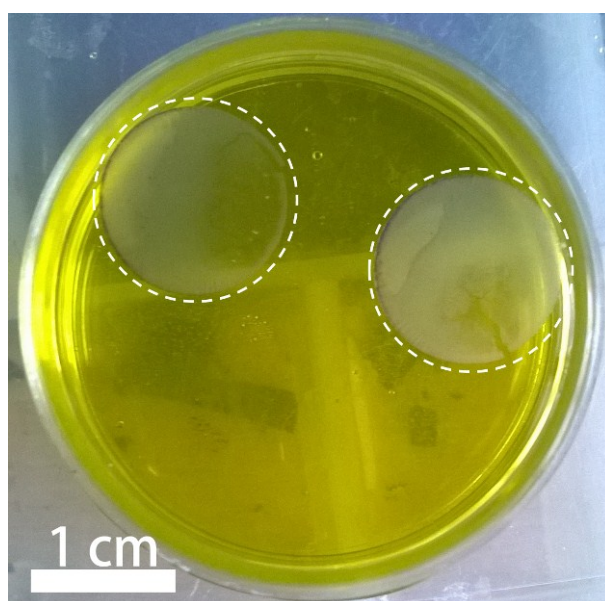


Figure S13. The optical image of wave-like graphene.

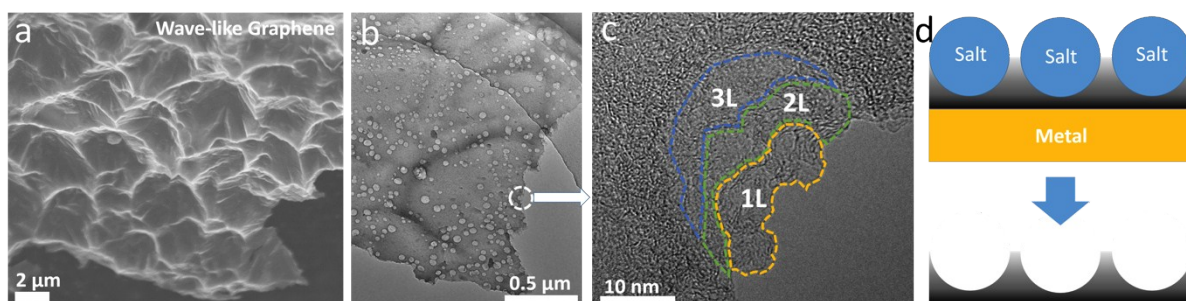


Figure S14. (a) SEM and (b) TEM image and (c) enlarged edge part of wave-like graphene; (d) schematic illustration of heterogeneous space-confined effect for fabricating of wave-like graphene.

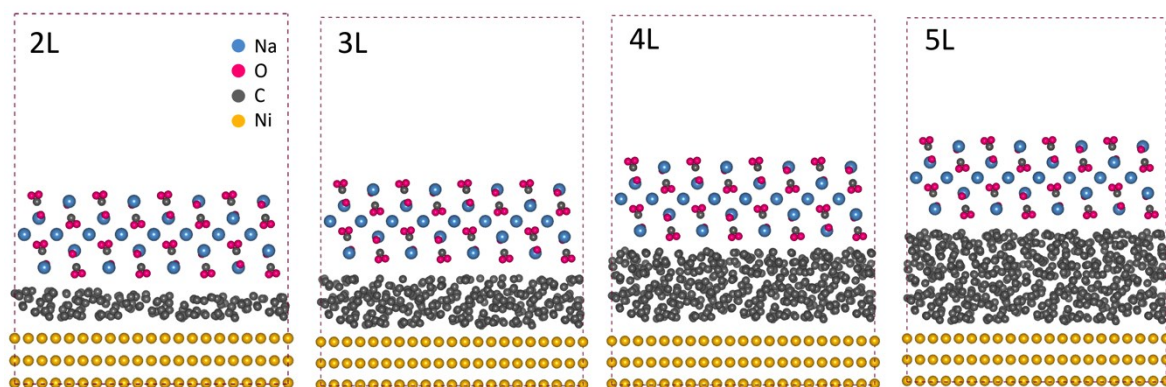


Figure S15. The models of 2, 3, 4, 5 layers of carbon with the Na_2CO_3 confine.

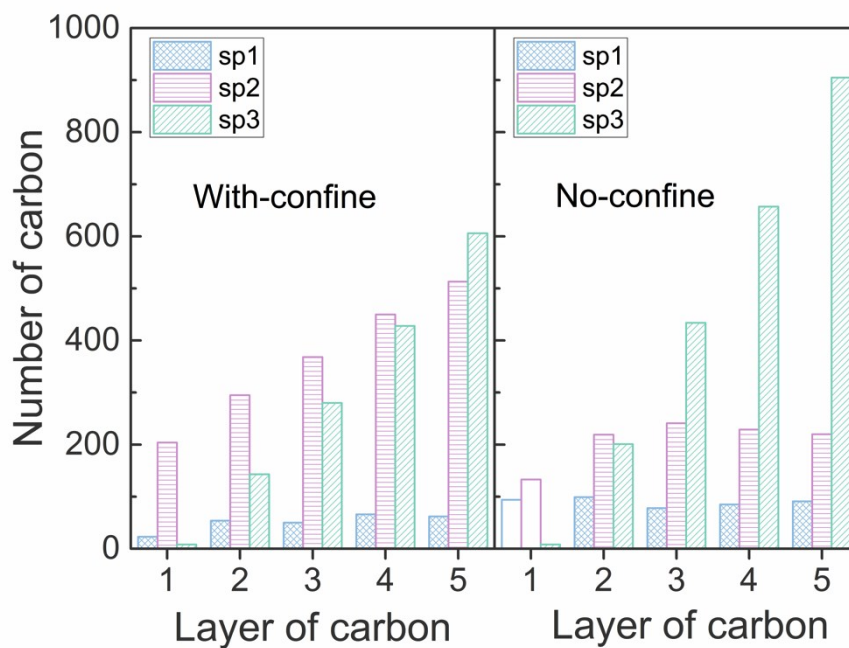


Figure S16. The number of sp¹, sp² and sp³ type carbon for 1-5 layers.

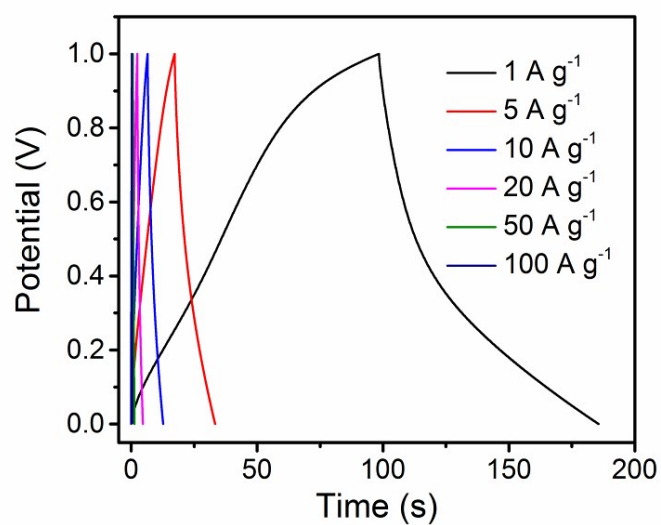


Figure S17. The galvanostatic charging/discharging curves of N-PC powder at various current densities.

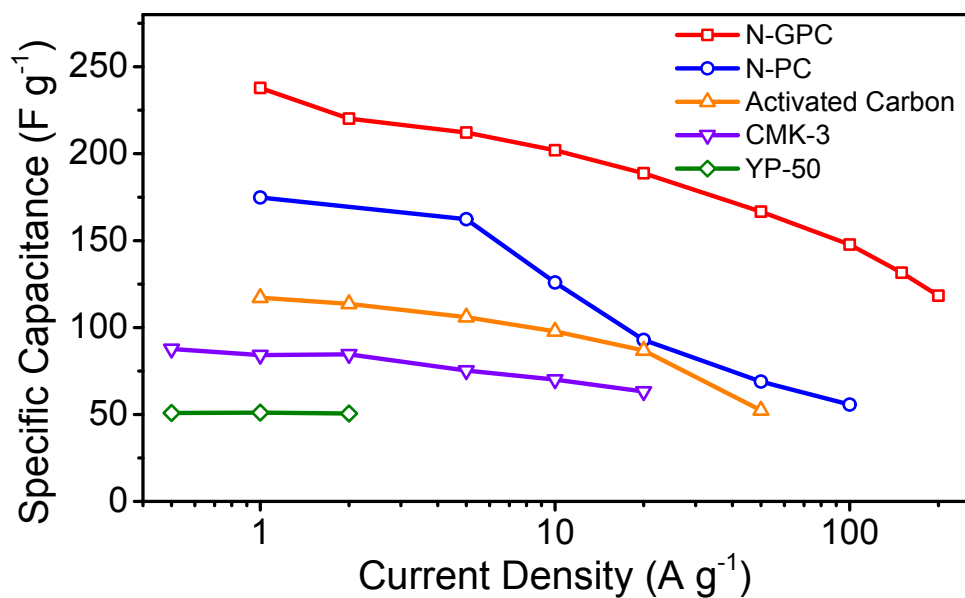


Figure S18. The capacitive performances of various carbon materials. The reference samples are the typical commercial carbon, including YP-50F (Kuraray Chemical. Co., Japan), CMK-3 (Nanjing XFNANO Materials Tech Co., China) and Activated carbon (Lision Technology Inc., China), which are treated as the process of N-PC to act as the electrode of supercapacitor.

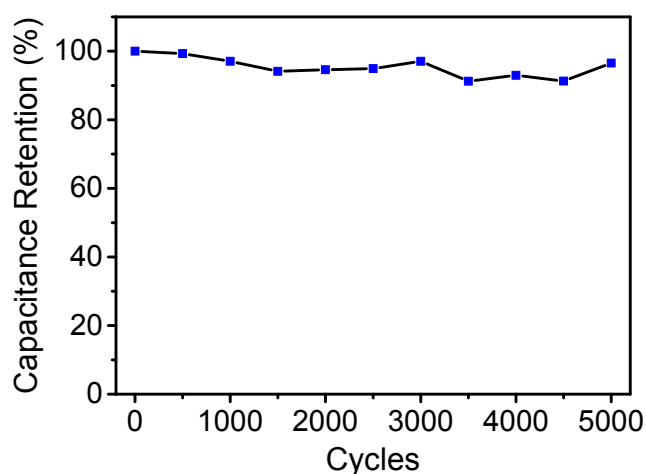


Figure S19. The cycling stability of N-GPC, which is tested at 5 A g⁻¹ for 5000 cycles.

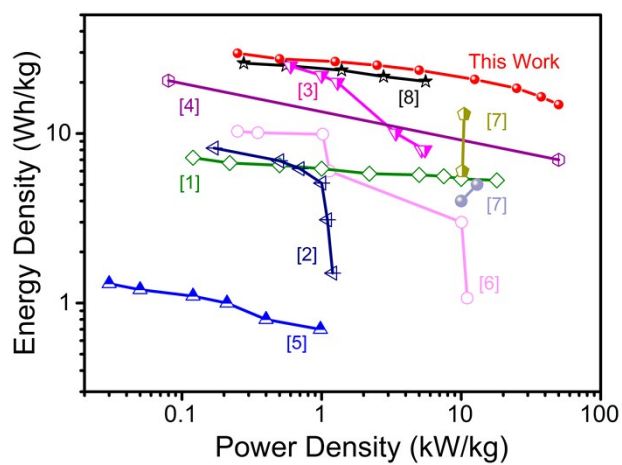


Figure S20. Ragone plot of N-GPC and other reported materials.

Table S1. The SSA, micropores, mesopores and macropores volumes of N-GPC and N-PC

| | SSA (m ² g ⁻¹) | Micropores Volume (cm ³ g ⁻¹) | Mesopores Volume (cm ³ g ⁻¹) | | Macropores Volume (cm ³ g ⁻¹) |
|-------|--|--|--|-------------|--|
| N-GPC | 498.5 | 0.127 | 0.389 (DFT) | 0.316 (BJH) | 0.425 |
| N-PC | 665.0 | 0.230 | 0.113 (DFT) | 0.121 (BJH) | 0.283 |

#The mesopores volumes are calculated by both of DFT and BJH methods.

Table S2. The XPS details of N-GPC, N-PC and GPC

| | C (at. %) | O (at. %) | N (at. %) | Other# (at. %) |
|-------|--------------|--------------|--------------|-------------------|
| N-GPC | 79.3 | 15.93 | 4.11 | 0.65 |
| N-PC | 81.56 | 11.88 | 5.82 | 0.73 |
| GPC | 92.21 | 7.45 | 0 | 0.34 |

#The other elements include Na, Fe and Cl, which are derived from the salt template or the residual etching solution.

Table S3. The performances of several carbon-based electrode materials

| Sample | Lithium ion battery | | Supercapacitor | | Ref |
|--|---------------------------------------|-------------------------|-------------------------------------|--------------------------|------|
| | Capacitance (mAh g ⁻¹) | Test Condition | Capacitance (F g ⁻¹) | Test Condition | |
| This work | | | 238 | 1 A g ⁻¹ | |
| N-GPC | 775 | 0.1 A g ⁻¹ | 118 | 200 A g ⁻¹ | |
| Graphene-CNTs hybrid materials | 458 | 74.4 mA g ⁻¹ | | | [9] |
| Graphene-CNTs hybrid papers | 330 | 100 mA g ⁻¹ | | | [10] |
| Graphitized boron-doped carbon foam | 310 | 20 cycles | | | [11] |
| Hybrid carbon nanotube and graphene nanostructures | 370 | 1.5 A g ⁻¹ | | | [12] |
| MOF-derived N-doped porous carbon | 2163 | 0.1 A g ⁻¹ | | | [13] |
| N-doping graphene | 798 | 0.5 A g ⁻¹ | | | [14] |
| Graphene-carbon nanotube hybrid materials | | | 230 | 0.01 Hz | [15] |
| Graphene-single-walled carbon nanotube hybrid | | | 350 | 0.1 A g ⁻¹ | [16] |
| 3D graphene and carbon nanotube foam | | | 286 | 1.78 mA cm ⁻² | [17] |
| Graphenated carbon nanotube | | | ~100 | 1 mA cm ⁻² | [18] |
| Graphene/PANI | | | 409 | 2 A g ⁻¹ | [19] |
| Edge-enriched graphene quantum dots | | | 236 | 0.5 A g ⁻¹ | [20] |

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