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## **Supporting Information**

## Few-layer MoS<sub>2</sub> anchored at nitrogen-doped carbon ribbons for sodium-ion battery anode with high rate performance

Yuanchao Pang, Shuyang Zhang, Limin Liu, Jin Liang, Zongjie Sun, Yuankun Wang, Chunhui Xiao, Shujiang Ding\*



Figure S1. SEM images of wild raupos (a), and AMCRs (b).



Figure S2. The excellent dispersion of AMCRs in distilled water. Moderate AMCRs in a vial (a), the AMCRs dispersing in equivalent amount of water relative with synthetic procedure (b), and Tyndall effect under laser irradiation (c), the dispersion retention of AMCRs in water after a few hours.



Figure S3. Zeta potential pattern of AMCRs (a), XPS survey spectrum of AMCRs (b).



Figure S4. Electron microscopic characterization of  $MoS_2$  aggregations. SEM morphologys (a, b) in different magnifications, TEM image (c), SAED pattern (d), and HRTEM images (e, f).



Figure S5. SEM images of MoS<sub>2</sub>@AMCRs at different scales (a, b, c, d).



Figure S6. XPS survey spectrum of MoS<sub>2</sub>@AMCRs composite.



Figure S7. Thermogravimetric analysis of self-assembly  $MoS_2$  and  $MoS_2@AMCRs$ . Let the weight percentage of  $MoS_2$  in the  $MoS_2@AMCRs$  to be x. Assuming AMCRs are completely removed after combustion, one has 0.745x = 0.576. Therefore x = 0.773.



Figure S8. Postmortern morphologys of anodes after 10 charge-discharge cycles: CMC as the binder (a, b), and PVDF as the binder (c, d) in different magnifications.

## Table S1. Literature analysis of battery performance including capacity, cycling stability, and rate capability.

	Cycle performance					
Mophology	Initial discharge capacity [mA h g <sup>-1</sup> ]	Cycle number	Final capacity [mA h g <sup>-1</sup> ]	Current density [A g <sup>-1</sup> ]	Rate performance	Ref.
MoS <sub>2</sub> @AMCRs	582	300	305	1	495mA h g <sup>-1</sup> @0.05 A g <sup>-1</sup> ; 456@0.1; 425@0.2; 381@0.5; 366@1; 302@2.	This work
Ultrathin MoS <sub>2</sub>	800	100	251	0.32	500mA h g <sup>-1</sup> @0.04 A g <sup>-1</sup> ; 410@0.08; 400@0.16; 220@0.32.	36a
MoS <sub>2</sub> /Graphene	600	100	380	0.02	520mA h g <sup>-1</sup> @0.02 A g <sup>-1</sup> ; 500@0.04; 450@0.08; 400@0.16; 350@0.32.	36b
MoS <sub>2</sub> @Graphen e paper	338	20	218	0.025	240 mA h g <sup>-1</sup> @0.025 A g <sup>-1</sup> ; 223@0.05; 214@0.1; 160@0.2.	36c
wormlike MoS <sub>2</sub>	670	80	410.5	0.06	450 mA h g <sup>-1</sup> @0.06 A g <sup>-1</sup> ; 380@0.1; 160@0.25; 80@0.6.	36d
MoS <sub>2</sub> @PEO	242	70	119	0.05	185 mA h g <sup>-1</sup> @0.05 A g <sup>-1</sup> ; 162@0.1; 143@0.25; 127@0.5; 112@1.	36e
TiO <sub>2</sub> @MoS <sub>2</sub> @Ca rbon cloth	460	200	157	0.5	650 mA h g <sup>-1</sup> @0.1 A g <sup>-1</sup> ; 550@0.2; 450@0.4; 380@0.8.	36f
MoS₂@PAN	300	100	130	0.3	350 mA h g <sup>-1</sup> @0.02 A g <sup>-1</sup> ; 200@0.1; 150@0.3; 100@1.	36g
MoS₂@Graphe ne paper	470	300	76.8	0.1	264 mA h g <sup>-1</sup> @0.1 A g <sup>-1</sup> ; 200@0.2; 180@0.4; 150@1.6; 80@3.2.	36h
MoS2@Nitroge n-doped carbon	810	150	340	0.15	510 mAh g <sup>-1</sup> @ 0.150 A g <sup>-1</sup> , 400@ 0.3, 310@0.6, 240@1.5, 180@3A g <sup>-1</sup>	36i