

## Supporting Information

### **A wide visible light driven complex perovskite $\text{Ba}(\text{Mg}_{1/3}\text{Ta}_{2/3})\text{O}_{3-x}\text{N}_y$ photocatalyst for water oxidation and reduction**

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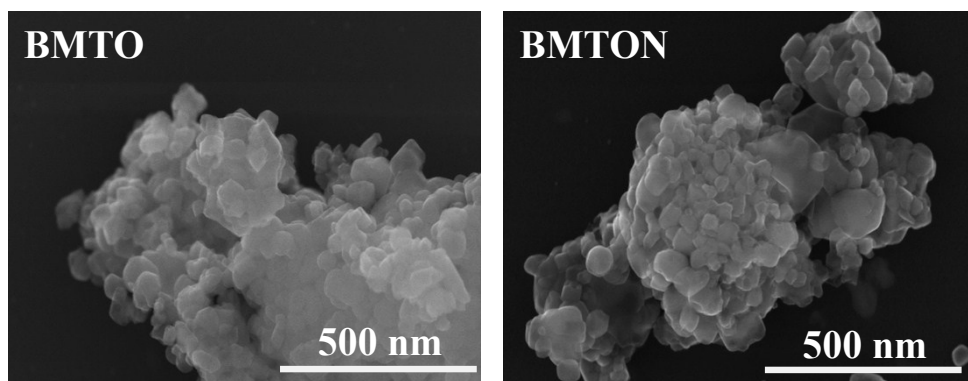
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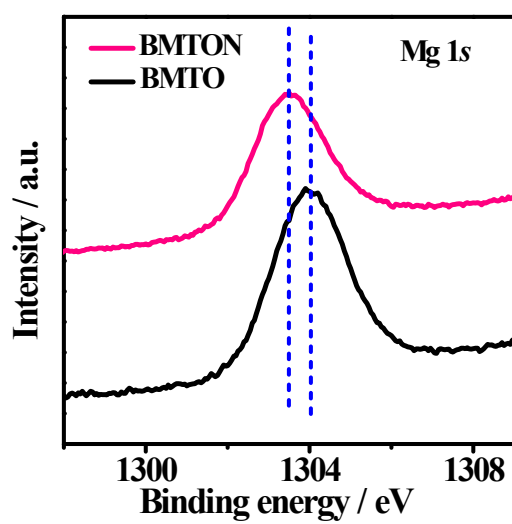
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**Fig S1.** Representative HRSEM images of BMTO and BMTON samples.



**Fig S2.** Mg 1s XPS spectra of the BMTO and BMTON samples.

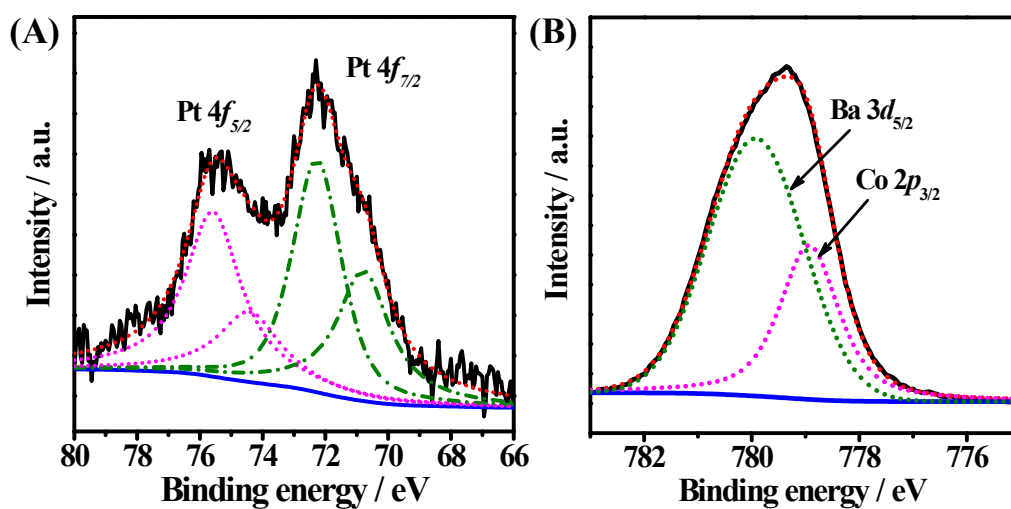


Fig S3. XPS spectra of BMTON samples: (A) Pt 4f; (B) Ba 3d<sub>5/2</sub> and Co 2p<sub>3/2</sub>.

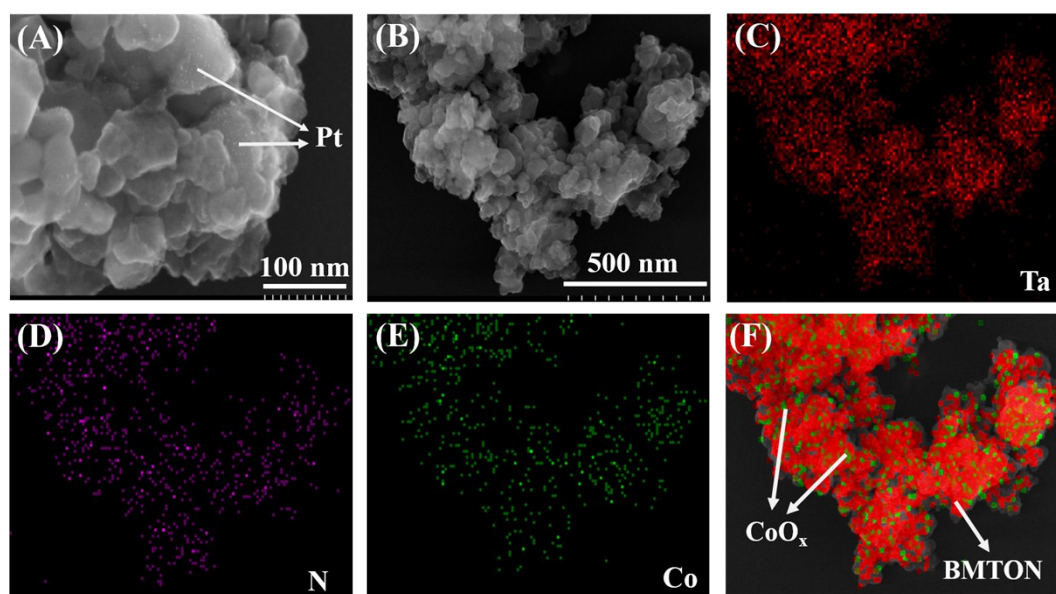


Fig S4. The HRSEM images of typical samples: (A) 0.5 wt% Pt/BMTON; (B) 1.0 wt% CoO<sub>x</sub>/BMTON, and the elemental mappings of CoO<sub>x</sub>/BMTON: (C) Ta element; (D) N element; (E) Co element; (F) simulated dispersion of CoO<sub>x</sub> and BMTON.

**Table S1.** Photocatalytic performances of typical samples under visible light irradiation ( $\lambda \geq 420$  nm).

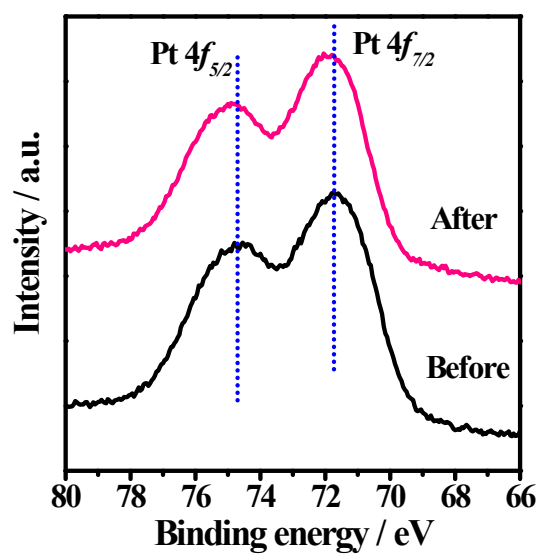
| Entry  | Sample                  | Sacrificial reagent | The rate of gases evolution           |
|--|-------------------------|---------------------|---------------------------------------|
| Water reduction half reaction ( $\text{H}_2$ ) | Pt                      | methanol            | <b>0</b> $\mu\text{mol h}^{-1}$       |
|  | BMTON                   | methanol            | <b>0</b> $\mu\text{mol h}^{-1}$       |
|  | Pt/BMTON                | methanol            | <b>0.9</b> $\mu\text{mol h}^{-1}$     |
| Water oxidation half reaction ( $\text{O}_2$ ) | $\text{Co}_3\text{O}_4$ | $\text{AgNO}_3$     | <b>0</b> $\mu\text{mol 0.5 h}^{-1}$   |
|  | BMTON                   | $\text{AgNO}_3$     | <b>1.4</b> $\mu\text{mol 0.5 h}^{-1}$ |
|  | $\text{CoO}_x$ /BMTON   | $\text{AgNO}_3$     | <b>3.5</b> $\mu\text{mol 0.5 h}^{-1}$ |

Reaction conditions: 150 mL of 20 v% methanol aqueous solution (for  $\text{H}_2$  evolution half reaction) or 0.01 M  $\text{AgNO}_3$  aqueous solution (for  $\text{O}_2$  evolution half reaction) with 0.15 g of the sample; 0.15 g of  $\text{La}_2\text{O}_3$ ; 300 W xenon lamp ( $\lambda \geq 420$  nm).

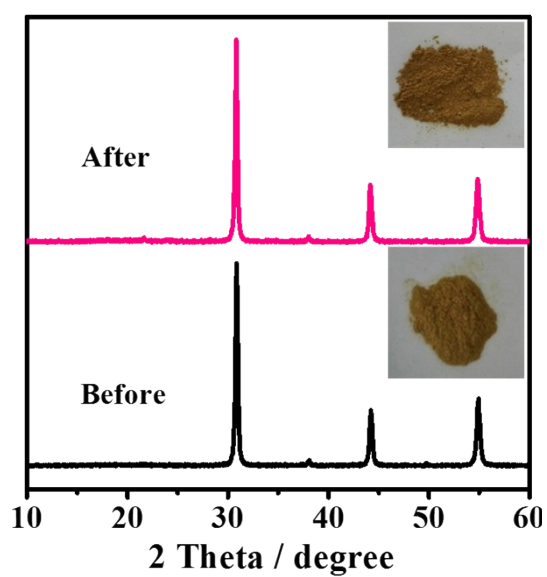
**Table S2.** Photocatalytic performances of typical photocatalysts under visible light irradiation ( $\lambda \geq 420$  nm).

| Entry  | The nitrated samples at different temperature | Cocatalyst     | Sacrificial reagent | The rate of gases evolution           |
|--|---|----------------|---------------------|---------------------------------------|
| Water reduction half reaction ( $\text{H}_2$ ) | 923 K   | Pt             | methanol            | <b>trace</b>                          |
|  | 1023 K  | Pt             | methanol            | <b>0.2</b> $\mu\text{mol h}^{-1}$     |
|  | 1123 K  | Pt             | methanol            | <b>0.3</b> $\mu\text{mol h}^{-1}$     |
|  | 1223 K  | Pt             | methanol            | <b>0.9</b> $\mu\text{mol h}^{-1}$     |
| Water oxidation half reaction ( $\text{O}_2$ ) | 923 K   | $\text{CoO}_x$ | $\text{AgNO}_3$     | <b>trace</b>                          |
|  | 1023 K  | $\text{CoO}_x$ | $\text{AgNO}_3$     | <b>1.9</b> $\mu\text{mol 0.5 h}^{-1}$ |
|  | 1123 K  | $\text{CoO}_x$ | $\text{AgNO}_3$     | <b>2.2</b> $\mu\text{mol 0.5 h}^{-1}$ |
|  | 1223 K  | $\text{CoO}_x$ | $\text{AgNO}_3$     | <b>3.5</b> $\mu\text{mol 0.5 h}^{-1}$ |

Reaction conditions: 150 mL of 20 v% methanol aqueous solution (for  $\text{H}_2$  evolution half reaction) or 0.01 M  $\text{AgNO}_3$  aqueous solution (for  $\text{O}_2$  evolution half reaction) with 0.15 g of the photocatalyst; 0.15 g of  $\text{La}_2\text{O}_3$ ; 300 W xenon lamp ( $\lambda \geq 420$  nm).



**Fig S5.** Pt 4f XPS spectra of 0.5 wt% Pt/BMTON samples before and after photocatalytic hydrogen evolution half reaction.



**Fig S6.** XRD patterns of BMTON samples before and after photocatalytic hydrogen evolution reaction.