

Electronic Supplementary Information (ESI)

Oriented SnS nanoflakes bound on S-doped N-rich carbon nanosheets with rapid pseudocapacitive response as high-rate anodes for sodium-ion batteries

Jian Sheng,[†] Leping Yang,[†] Yuanen Zhu, Feng Li, Yue Zhang and Zhen Zhou*

School of Materials Science and Engineering, National Institute for Advanced Materials, Institute of New Energy
Material Chemistry, Tianjin Key Laboratory of Metal and Molecule Based Material Chemistry, Collaborative
Innovation Center of Chemical Science and Engineering (Tianjin), Nankai University, Tianjin 300350, China

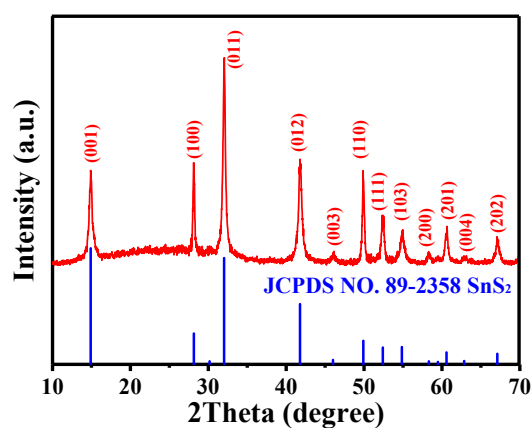


Fig. S1. XRD patterns of the intermediate products SnS₂/CN.

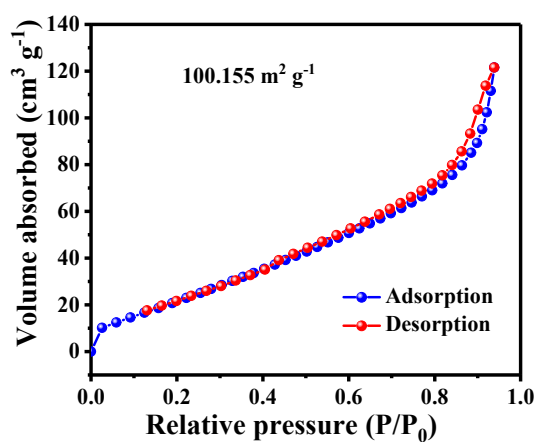


Fig. S2. N₂ adsorption/desorption isotherm of SnS/CNS composite.

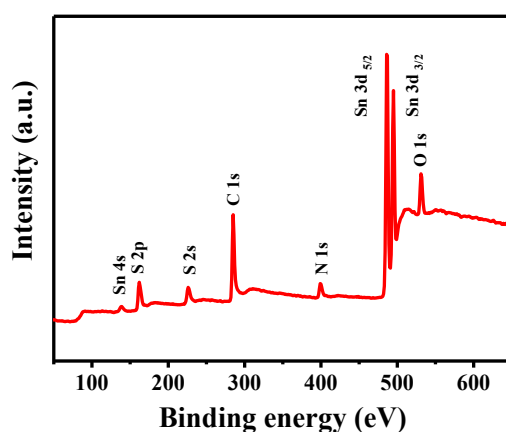


Fig. S3. XPS spectra of the SnS/CNS composite.

Table S1. Elemental composition of SnS/CNS are determined by CHONS Elemental Analyzer.

	N content	S content	C content
SnS/CNS	6.24 wt.%	17.255 wt.%	16.53 wt.%

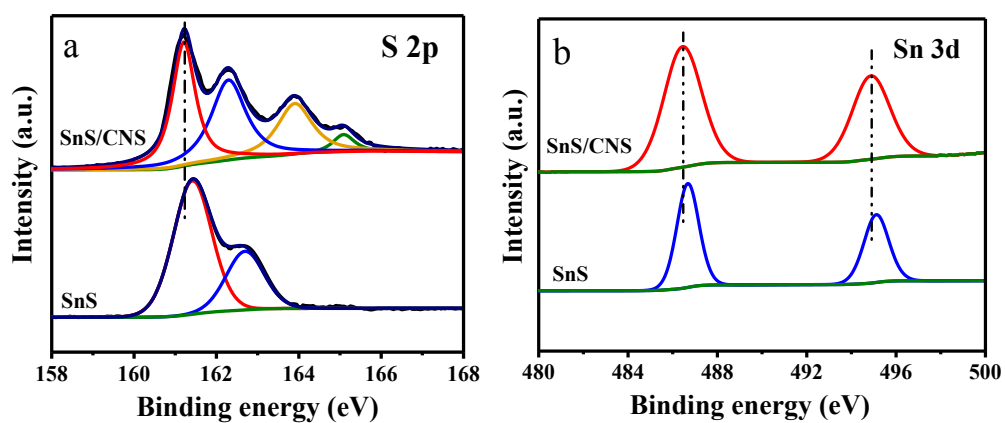


Fig. S4. (a) S2p and (b) Sn3d spectrum of SnS/CNS and SnS.

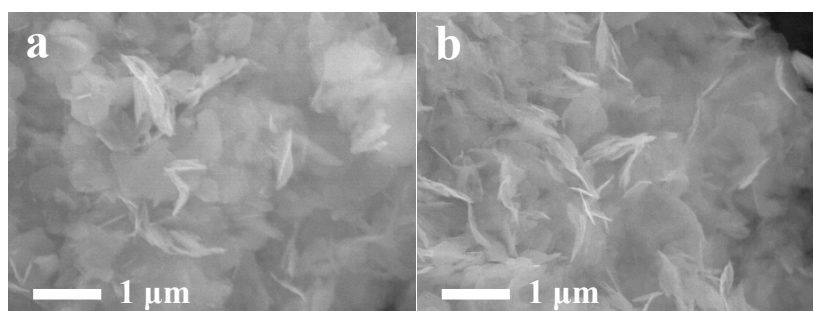


Fig. S5. SEM images of pristine SnS.

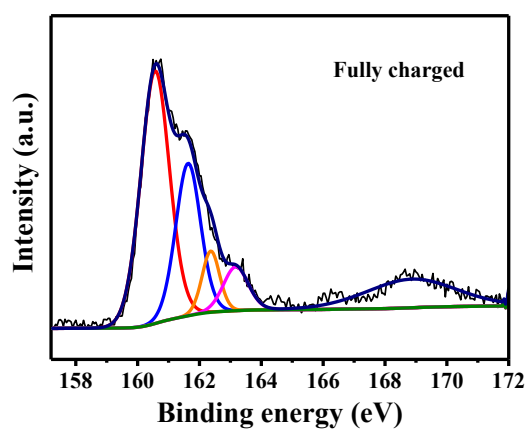


Fig. S6. XPS S2p of SnS/CNS at the fully charged state.

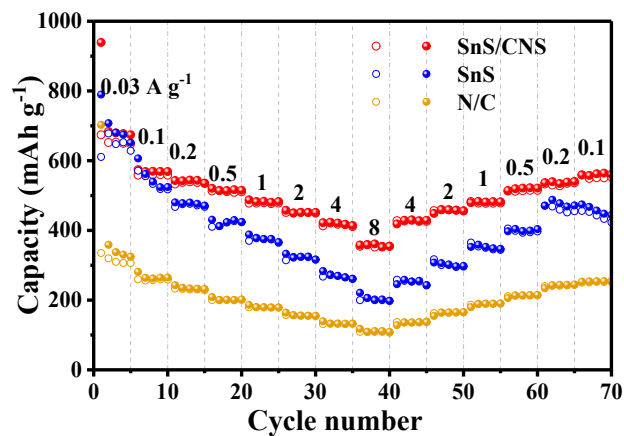


Fig. S7. Rate performance of SnS/CNS, SnS and N/C electrodes

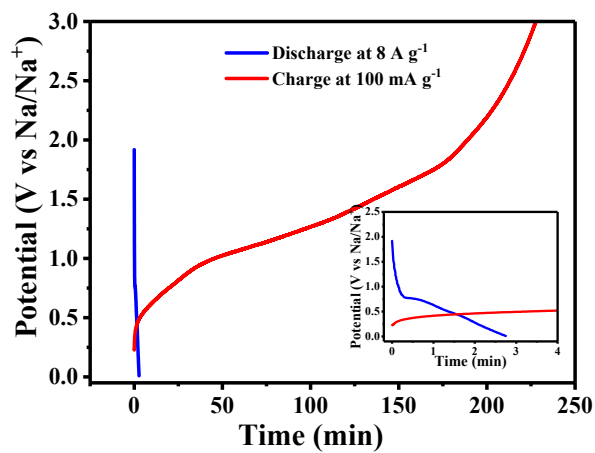


Fig. S8. Fast discharging (discharge at 8 A g⁻¹ in 3 min, charge at 100 mA g⁻¹ with 230 min)

performance of the SnS/CNS electrode

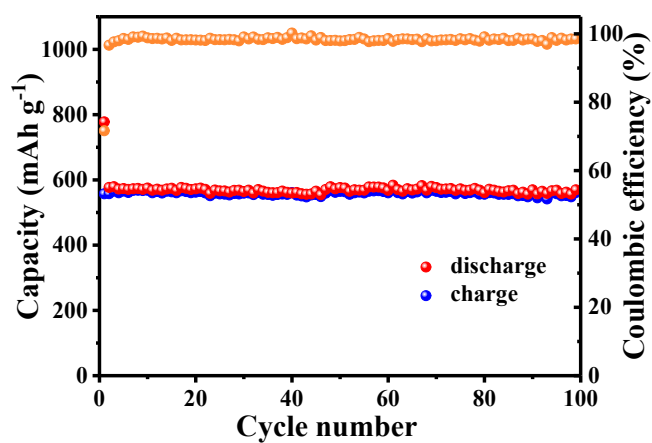


Fig. S9. Cycling performance of SnS/CNS at a current density of 100 mA g⁻¹ with coulombic efficiency.

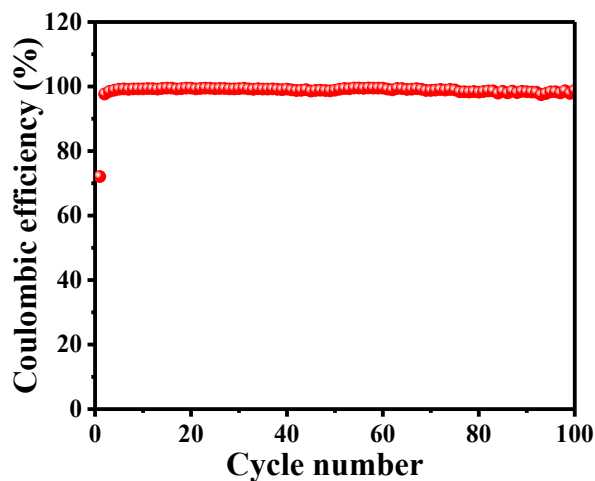


Fig. S10. Coulombic efficiency of the SnS/CNS electrode at 1 A g⁻¹ for 100 cycles.

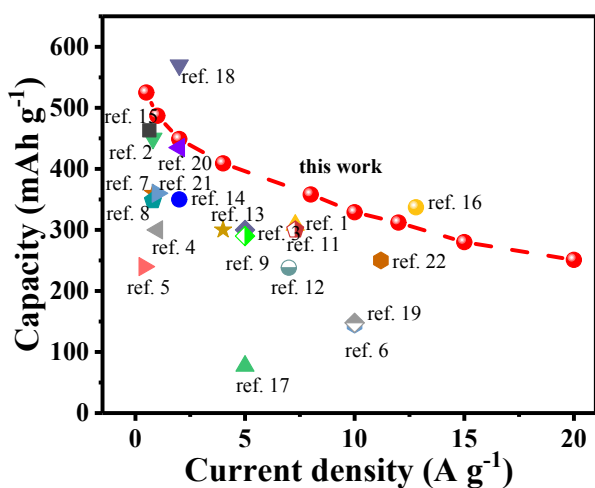


Fig. S11. Rate performance comparison of SnS/CNS and other tin sulfides.

Table S2. A survey of electrochemical performances of tin sulfide anodes in sodium ion batteries.

Materials	Reversible capacity (mAh g ⁻¹)	Cycling stability (%)	ICE (%)	High rate capability (mAh g ⁻¹)	Ref.
SnS/CNS	654 mAh g ⁻¹ (0.03 A g ⁻¹) 483 mAh g ⁻¹ (1 A g ⁻¹)	98% (100 cycles, 1A g ⁻¹)	72%	360 mAh g ⁻¹ (8 A g ⁻¹) 250.2 mAh g ⁻¹ (20 A g ⁻¹)	This work
SnS-G	500 mAh g ⁻¹ (0.81 A g ⁻¹)	94% (250 cycles, 0.81 A g ⁻¹)	70%	308 mAh g ⁻¹ (7.29 A g ⁻¹)	1
SnS-C	544 mAh g ⁻¹ (0.1 A g ⁻¹)	98% (80 cycles, 0.1 A g ⁻¹)	66%	450 mAh g ⁻¹ (0.8 A g ⁻¹)	2

SnS-C	490 mAh g ⁻¹ (0.5 A g ⁻¹)	88 % (50 cycles, 0.5 A g ⁻¹)	63%	300 mAh g ⁻¹ (5 A g ⁻¹)	3
SnS	520 mAh g ⁻¹ (0.125 A g ⁻¹)	65 % (30 cycles, 0.125 A g ⁻¹)	60%	300 mAh g ⁻¹ (1 A g ⁻¹)	4
SnS-rGO	457 mAh g ⁻¹ (0.02 A g ⁻¹)	94% (100 cycles, 0.1 A g ⁻¹)	56%	240 mAh g ⁻¹ (0.4 A g ⁻¹)	5
SnS/C	415 mAh g ⁻¹ (0.1 A g ⁻¹)	80% (300 cycles, 1 A g ⁻¹)	79%	145 mAh g ⁻¹ (10 A g ⁻¹)	6
SnS 3D flowers	455 mAh g ⁻¹ (0.03 A g ⁻¹)	64% (50 cycles, 0.15 A g ⁻¹)	none	360 mAh g ⁻¹ (0.8 A g ⁻¹)	7
SnS-Sn-C	450 mAh g ⁻¹ (0.1 A g ⁻¹)	87% (150 cycles, 0.1 A g ⁻¹)	59%	348 mAh g ⁻¹ (0.8 A g ⁻¹)	8
SnS nanotube	520 mAh g ⁻¹ (0.05 A g ⁻¹)	95% (100 cycles, 0.2 A g ⁻¹)	76%	290 mAh g ⁻¹ (5 A g ⁻¹)	9
GF-SnS NH	1100 mAh g ⁻¹ (0.03 A g ⁻¹)	92% (200 cycles, 0.03 A g ⁻¹)	81%	420 mAh g ⁻¹ (30 A g ⁻¹)	10
SnS/SnO ₂ Heterostr- uctures	729 mAh g ⁻¹ (0.03 A g ⁻¹)	73% (500 cycles, 0.81 A g ⁻¹)	74%	300 mAh g ⁻¹ (7.29 A g ⁻¹)	11
SnS-MoS ₂	455 mAh g ⁻¹ (0.5 A g ⁻¹)	89% (100 cycles, 0.5 A g ⁻¹)	81%	238 mAh g ⁻¹ (7 A g ⁻¹)	12
SnS ₂ -G	610 mAh g ⁻¹ (0.2 A g ⁻¹)	63 % (100 cycles, 0.2 A g ⁻¹)	67%	300 mAh g ⁻¹ (4 A g ⁻¹)	13
SnS ₂ -GO	580 mAh g ⁻¹ (0.05 A g ⁻¹)	95 % (50 cycles, 0.05 A g ⁻¹)	67%	350 mAh g ⁻¹ (2 A g ⁻¹)	14
SnS ₂ -G	725 mAh g ⁻¹ (0.02 A g ⁻¹)	89% (60 cycles, 0.02 A g ⁻¹)	52%	463 mAh g ⁻¹ (0.64 A g ⁻¹)	15
SnS ₂ -rGO	649 mAh g ⁻¹ (0.1 A g ⁻¹)	89% (400 cycles, 0.8 A g ⁻¹)	64%	337 mAh g ⁻¹ (12.8 A g ⁻¹)	16
SnS ₂ nanoplates	349 mAh g ⁻¹ (0.1 A g ⁻¹)	57 % (80 cycles, 0.5 A g ⁻¹)	38%	77 mAh g ⁻¹ (5 A g ⁻¹)	17
SnS ₂ -rGO	550 mAh g ⁻¹ (1 A g ⁻¹)	91 % (400 cycles, 1 A g ⁻¹)	79%	570 mAh g ⁻¹ (2 A g ⁻¹)	18
SnS ₂ -NGS	608 mAh g ⁻¹ (0.2 A g ⁻¹)	71% (100 cycles, 0.2 A g ⁻¹)	66%	148 mAh g ⁻¹ (10 A g ⁻¹)	19
SnS ₂	733 mAh g ⁻¹ (0.1 A g ⁻¹)	88% (50 cycles, 0.1 A g ⁻¹)	59%	435 mAh g ⁻¹ (2 A g ⁻¹)	20
SnS ₂ -C	660 mAh g ⁻¹ (0.05 A g ⁻¹)	86% (100 cycles, 0.05 A g ⁻¹)	60%	360 mAh g ⁻¹ (1 A g ⁻¹)	21
SnS ₂ NC/EDA- RGO	749 mAh g ⁻¹ (0.2 A g ⁻¹) 480 mAh g ⁻¹ (1 A g ⁻¹)	90% (100 cycles, 0.2 A g ⁻¹)	73%	250 mAh g ⁻¹ (11.2 A g ⁻¹)	22

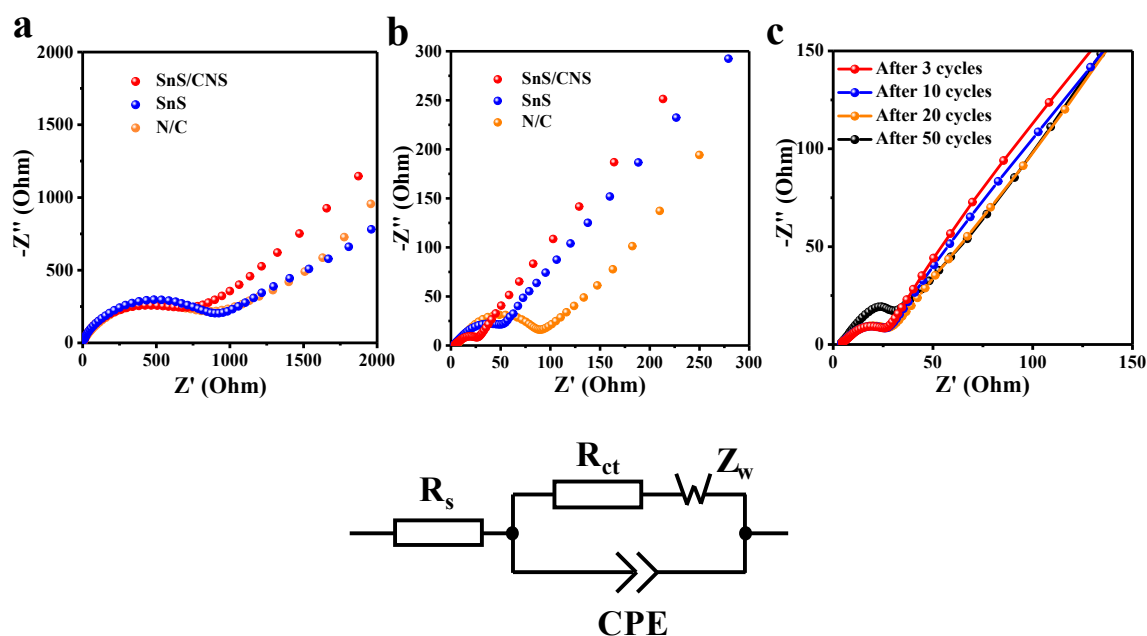


Fig. S12. Electrochemical impedance spectra and equivalent circuit of the SnS/CNS electrode. (a) Nyquist plots of SnS/CNS, pristine SnS and N/C electrodes before cycling and (b) after 10 cycles. (c) Nyquist plots of SnS/CNS electrode after different cycles.

Table S3. R_s and R_{ct} values of SnS/CNS electrodes after different cycles.

	R_s (Ω)	R_{ct} (Ω)
After 3 cycles	4.037	25.48
After 10 cycles	3.989	27.08
After 20 cycles	4.012	29.17
After 50 cycles	3.917	34.39

Supplementary References

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