Electronic Supplementary Information (ESI) for the manuscript titled "Heteroatom functionalized activated porous biocarbons and their excellent performance for CO<sub>2</sub> capture at high pressure"

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**Fig. S1.** XRD patterns of **NDABn-T** (n=0-2, T= A) 500, B) 600 and C) 700 °C).

The XRD patterns of **NDABn-T** (n=2, T=500-700 °C) show the presence of two broad peaks centered on  $2\theta=23^{\circ}$  and  $43^{\circ}$ . This indicates that the samples are graphitic in nature. At any given temperature, the intensity of both the peaks increase slightly with increasing ratio of ZnCl<sub>2</sub> to *Arundo donax*/chitosan, which signifies the development of higher graphitization in the carbon structure.



**Fig. S2.** XPS survey spectra of A) **NDAB0-T** (T = 500-700 °C) and B) **NDAB3-T** (T = 500-700 °C).

The XPS survey spectra show three main peaks corresponding to the elements carbon, nitrogen and oxygen. The XPS results prove that the presence of chitosan during the carbonization can effectively incorporate the nitrogen in the carbon structure.



Fig. S3. High resolution XPS O1s spectra of NDAB3-T (T = A) 500, B) 600 and C) 700 °C).



**Fig. S4.** N<sub>2</sub> adsorption-desorption isotherms of **NDABn-T** (n=0-2, T = A) 500, B) 600 and C) 700 °C).

The isotherm curves correspond to the shape which belongs to type I as per the IUPAC classification of adsorption isotherms. In other words, the porous structure of these activated biocarbons is dominated by micropores followed by a lesser amount of meso and macropores in the structure. It can also be seen that when the impregnation amount of the activating agent is increased, there is a corresponding increasing the amount of  $N_2$  being adsorbed. This happens due to the fact that higher porosity is generated by the chemical reaction with activating agent.



**Fig. S5.** EDS analysis of **NDAB3-T** (T = 500-700 °C).

The result of EDS analysis further confirms the heteroatom functionalization of the porous activated biocarbons. The sample NDAB3-500 shows the highest amount of nitrogen content which is in consistency with the CNS and XPS analysis elemental composition. From these results, it can be concluded that the one step strategy of synthesizing heteroatom doped porous activated biocarbons has been successful.



Fig. S6. The relation between  $CO_2$  adsorption measured at different conditions with A) BET surface area and B) nitrogen content of the NDAB3-T materials. (Note: Error bars included for  $CO_2$  measurements recorded at 0°C)

Material	C (wt.%)	N (wt.%)	N/C	C (at.%)	N (at.%)	N/C
			(wt. ratio)			(at. ratio)
NDAB0-500	65	6.5	0.01	5.4	0.46	0.085
NDAB1-500	68	6.2	0.09	5.6	0.44	0.078
NDAB2-500	70	5.9	0.08	5.8	0.42	0.072
NDAB3-500	75	5.4	0.07	6.2	0.39	0.061
NDAB0-600	68	4.2	0.06	5.7	0.30	0.053
NDAB1-600	73	5.1	0.07	6.1	0.36	0.060
NDAB2-600	75	5.1	0.07	6.3	0.36	0.058
NDAB3-600	81	4.1	0.05	6.7	0.29	0.043
NDAB0-700	76	3.3	0.04	6.4	0.23	0.037
NDAB1-700	77	3.9	0.05	6.4	0.28	0.043
NDAB2-700	76	4.0	0.05	6.4	0.28	0.045
NDAB3-700	82	3.5	0.04	6.8	0.25	0.037

 Table S1. Carbon and nitrogen content in all NDABn-T (n=0-3, T=500-700 °C) materials.

The concentration of carbon and nitrogen in the activated materials is dependent upon the impregnation ratio of ZnCl<sub>2</sub> used and the carbonization temperature.

Sample	С	Ν	0	N/C	O/C
	(at.%)	(at. %)	(at. %)	(ratio)	(ratio)
NDAB0-500	89	4.0	6.7	0.045	0.076
NDAB0-600	88	2.4	4.5	0.027	0.051
NDAB0-700	92	2.0	4.7	0.022	0.051
NDAB3-500	90	4.4	5.4	0.049	0.060
NDAB3-600	92	3.0	4.1	0.032	0.044
NDAB3-700	94	1.9	3.7	0.021	0.040

**Table S2.** XPS surface elemental composition of **NDAB0-T** (T = 500-700 °C) and **NDAB3-T** (T = 500-700 °C).

Sample	С	Ν	0	N/C	O/C
	(at.%)	(at. %)	(at. %)	(ratio)	(ratio)
NDAB3-500	82	5.2	13.1	0.064	0.16
NDAB3-600	87	2.1	11.0	0.024	0.13
NDAB3-700	91	1.5	7.2	0.016	0.08

**Table S3. EDS** elemental composition of **NDAB3-T** (T = 500-700  $^{\circ}$ C).

Sample	<b>SA</b> <sub>BET</sub>	SA <sub>micro</sub> (%)	V <sub>total</sub>	V <sub>micro</sub> (%)	PW
	(m <sup>2</sup> g <sup>-1</sup> )	/SA <sub>meso</sub> (%)	(cm <sup>3</sup> g <sup>-1</sup> )	/V <sub>meso</sub> (%)	(nm)
NDAB0-500	3.0	-	-	-	-
NDAB1-500	1070	87/13	0.45	80/20	1.6
NDAB2-500	1559	80/20	0.67	72/18	1.6
NDAB0-600	4.1	-	-	-	-
NDAB1-600	1145	90/10	0.47	85/15	1.5
NDAB2-600	1328	80/20	0.66	76/24	1.6
NDAB0-700	3.7	-	-	-	-
NDAB1-700	1148	89/11	0.47	83/17	1.5
NDAB2-700	1335	75/25	0.63	71/29	1.5

**Table S4.** Textural properties of N-doped activated carbons derived from *Arundo donax* and chitosan under mild activation.

 $\overline{SA_{BET} - BET}$  surface area (Brunauer Emmett Teller),  $SA_{micro} - micropore$  surface area,  $SA_{meso} - mesopore$  surface area,  $PV_{total} - total$  pore volume,  $PV_{micro} - micropore$  volume,  $PV_{meso} - mesopore$  volume and PW - Pore Width calculated using DFT method

At any given temperature, a progressive increase in the surface area and pore volume is observed when the  $ZnCl_2$  impregnation ratio is increased from 0 to 2. At any given impregnation ratio, the highest textural parameters are observed at a temperature of 500 °C.