Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A. This journal is © The Royal Society of Chemistry 2017

## **Supporting Information for**

## Engineering SnS<sub>2</sub> Nanosheet Assemblies for Enhanced Electrochemical Lithium

## and Sodium Ion Storage

Yeyun Wang, Junhua Zhou, Jinghua Wu, Fengjiao Chen, Peirong Li, Na Han, Wenjing Huang,

Yuping Liu, Hualin Ye, Feipeng Zhao and Yanguang Li\*

Institute of Functional Nano and Soft Materials (FUNSOM), Jiangsu Key Laboratory for Carbon-

Based Functional Materials and Devices, Soochow University, Suzhou 215123, China

\*Correspondence to: <u>yanguang@suda.edu.cn</u>

Materials	Specific capacities	Cycle life	Reference
c-SnS <sub>2</sub> NSA	1200 mAh g <sup>-1</sup> at 0.2A g <sup>-1</sup> 620 mAh g <sup>-1</sup> at 10 A g <sup>-1</sup>	99.5% after 300 cycles at 1 A g <sup>-1</sup>	This study
SnS <sub>2</sub> /SnO	998 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 760 mAh g <sup>-1</sup> at 0.8 A g <sup>-1</sup>	66 % after 200 cycles at 0.2 A $\rm g^{\text{-}1}$	J. Mater. Chem. A 2017, <b>5</b> , 512
SnS <sub>2</sub> –NGS	1407 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 200 mAh g <sup>-1</sup> at 10 A g <sup>-1</sup>	73% after 150 cycles at 0.8 A $g^{\text{-1}}$	J. Mater. Chem. A 2016, <b>4</b> , 10719
SnS <sub>2</sub> @RGO	1278 mAh g <sup>-1</sup> at 0.065 A g <sup>-1</sup> 415 mAh g <sup>-1</sup> at 3.2 A g <sup>-1</sup>	81% after 200 cycles at 0. 065 A g <sup>-1</sup>	J. Mater. Chem. A 2013, <b>1</b> , 8658
SnS <sub>2</sub> NRGO	562 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 402 mAh g <sup>-1</sup> at 2 A g <sup>-1</sup>	97 % after 200 cycles at 0.2 A $g^{\text{-}1}$	ACS Nano 2016, <b>10,</b> 10778
$SnS_{2/}S/C$	875 mAh g <sup>-1</sup> at 0.84 A g <sup>-1</sup> 200 mAh g <sup>-1</sup> at 8.4 A g <sup>-1</sup>	75 % after 300 cycles at 0.84 A $g^{\text{-1}}$	ACS Appl. Mater. Interfaces 2016, <b>8</b> , 19550
CC-VN@SnS <sub>2</sub>	819 mAh g <sup>-1</sup> at 0.65 A g <sup>-1</sup> 349 mAh g <sup>-1</sup> at 13 A g <sup>-1</sup>	97 % after 100 cycles at 0.65 A g $^{-1}$	ACS Appl. Mater. Interfaces 2015, <b>7</b> , 23205
SnS <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub> / rGO	1284 mAh g <sup>-1</sup> at 0.1 A g <sup>-1</sup> 396 mAh g <sup>-1</sup> at 2 A g <sup>-1</sup>	100 % after 276 cycles at 0.1 A g $^{-1}$	RSC Adv. 2017, 7, 3125
SnS <sub>2</sub> /G-As	656 mAh g <sup>-1</sup> at 0.05 A g <sup>-1</sup> 240 mAh g <sup>-1</sup> at 1 A g <sup>-1</sup>	55 % after 30 cycles at 0.05 A $g^{\text{-1}}$	J. Power Sources 2013, <b>237</b> , 178

**Table S1**. Electrochemical performances of SnS<sub>2</sub>-based materials previously reported for lithiumion batteries.

Table S2. Electrochemical performances of SnS2-based materials previously reported for sodium-

ion batteries.

Materials	Specific capacities	Cycle life	Reference
c-SnS <sub>2</sub> NSA	600 mAh g <sup>-1</sup> at 0.5A g <sup>-1</sup> 140 mAh g <sup>-1</sup> at 5 A g <sup>-1</sup>	71% after 100 cycles at 0.5 A g <sup>-1</sup>	This study
SnS <sub>2</sub> NGS	608 mAh g <sup>-1</sup> at 0.2A g <sup>-1</sup> 148 mAh g <sup>-1</sup> at 10 A g <sup>-1</sup>	74% after 100 cycles at 0.2 A g <sup>-1</sup>	J. Mater. Chem. A 2016, <b>4</b> , 10719
SnS <sub>2</sub> /GO	610 mAh g <sup>-1</sup> at 0.05A g <sup>-1</sup> 320 mAh g <sup>-1</sup> at 2 A g <sup>-1</sup>	99.6 % after 150 cycles at 0.05 A g <sup>-1</sup>	J. Mater. Chem. A 2014, <b>2</b> , 8431
SnS <sub>2</sub> /rGO	469 mAh g <sup>-1</sup> at 0.8A g <sup>-1</sup> 337 mAh g <sup>-1</sup> at 12.8 A g <sup>-1</sup>	61 % after 1000 cycles at 0.8A g <sup>-1</sup>	Adv. Funct. Mater. 2015, <b>25</b> , 481
SnS <sub>2</sub> /G	650 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 326 mAh g <sup>-1</sup> at 4 A g <sup>-1</sup>	94 % after 300 cycles at 0.2 A $\rm g^{-1}$	Nanoscale 2015, 7, 1325
SnS <sub>2</sub> /RGO	630 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 544 mAh g <sup>-1</sup> at 2 A g <sup>-1</sup>	95 % after 400 cycles at 1 A g <sup>-1</sup>	Adv. Mater. 2014, <b>26</b> , 3854
B-SnS <sub>2</sub>	900 mAh g <sup>-1</sup> at 0.2 A g <sup>-1</sup> 400 mAh g <sup>-1</sup> at 10 A g <sup>-1</sup>	99 % after 40 cycles at 1 A g <sup>-1</sup>	ACS Nano 2016, <b>10</b> , 10211
SnS <sub>2</sub> NWA	576 mAh g <sup>-1</sup> at 0.5 A g <sup>-1</sup> 370 mAh g <sup>-1</sup> at 5 A g <sup>-1</sup>	64 % after 100 cycles at 0.5 A $\rm g^{-1}$	ACS Appl. Mater. Interfaces 2017, <b>10</b> , 1021
SnS <sub>2</sub> /C	660 mAh g <sup>-1</sup> at 0.05 A g <sup>-1</sup> 360 mAh g <sup>-1</sup> at 1 A g <sup>-1</sup>	86 % after 100 cycles at 0.05 A $\rm g^{-1}$	ACS Appl. Mater. Interfaces 2015, <b>7</b> , 11476
FL-SnS <sub>2</sub> /RGO	843 mAh g <sup>-1</sup> at 0.1A g <sup>-1</sup> 335 mAh g <sup>-1</sup> at 8.4 A g <sup>-1</sup>	98 % after 100 cycles at 0.1 A $\rm g^{-1}$	J. Phys. Chem. 2017, <b>121</b> , 3261



Figure S1. FT-IR spectrum of as-prepared and annealed SnS<sub>2</sub> NSA.



**Figure S2.** SEM images of products collected after the solvothermal reaction at 180°C for (a) 1 h, (b) 2 h, (c) 4 h and (d) 6 h.



**Figure S3.** SEM images of products prepared using (a) water and (b) ethanol as the reaction solvent under otherwise identical conditions.



Figure S4. (a) XRD pattern of  $c-SnS_2$  NSA. (b) XPS spectra of  $c-SnS_2$  NSA and  $SnS_2$  NSA. (c) TGA curve of  $c-SnS_2$  NSA in air.



**Figure S5.** (a) CV curves of  $c-SnS_2$  NSA at diffident scan rates as indicated. (b) Galvanostatic charge and discharge curves of  $c-SnS_2$  NSA at 200 mA g<sup>-1</sup> for the first three cycles.



**Figure S6**. Electrochemical impedance spectroscopy (EIS) analysis of (a)  $SnS_2 NSA$  and (b) c- $SnS_2 NSA$  at the 1st and 30th cycles. c- $SnS_2 NSA$  consistently showed smaller charge transfer resistance.



**Figure S7.** SEM images of (a,b) c-SnS<sub>2</sub> NSA and (c,d) SnS<sub>2</sub> NSA before (a,c) and after (b,d) cycling. c-SnS<sub>2</sub> NSA was obviously more resistant to pulverization.



**Figure S8.** Galvanostatic charge and discharge voltage profiles of c-SnS<sub>2</sub> NSA for (a) LIBs and (b) SIBs under different specific currents as noted.