

## Supporting Information

# Boosting molecular oxygen activation of SrTiO<sub>3</sub> by engineering exposed facet for high efficient photocatalytic oxidation

Xiaoyong Wu <sup>a, ‡</sup>, Xiaoyang Wang <sup>a, ‡</sup>, Jun Li <sup>a</sup>, Gaoke Zhang <sup>a, b\*</sup>

<sup>a</sup> Hubei Key Laboratory of Mineral Resources Processing and Environment, Hubei

Provincial Collaborative Innovation Center for High Efficient Utilization of

Vanadium Resources, School of Resources and Environmental Engineering, Wuhan

University of Technology, 122 Luoshi Road, Wuhan 430070, China

<sup>b</sup> State Key Laboratory of Silicate Materials for Architectures, Wuhan University of

Technology, Wuhan 430070, China

\*Corresponding author. Tel: 86-27-87651816; fax: 86-27-87887445.

E-mail: gkzhang@whut.edu.cn

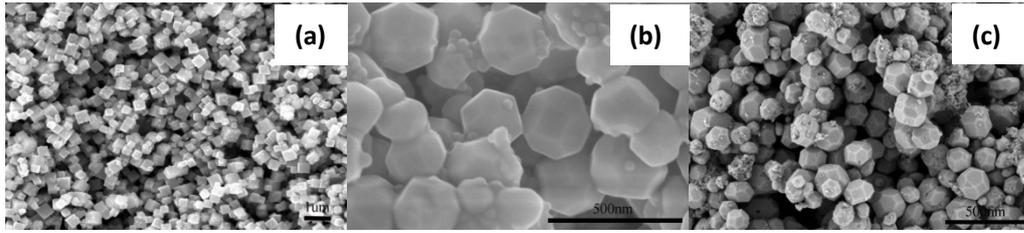


Figure S1. The SEM images of samples SrTiO<sub>3</sub>-001 (a), SrTiO<sub>3</sub>-110-1 (b) and SrTiO<sub>3</sub>-110-2 (c).

Table S1. The fraction of exposed facets in the samples.

SrTiO <sub>3</sub>	Facet fraction	
	001 (%)	110 (%)
SrTiO <sub>3</sub> -001	100	0
SrTiO <sub>3</sub> -110-1	34.6	65.4
SrTiO <sub>3</sub> -110-2	12.5	87.5

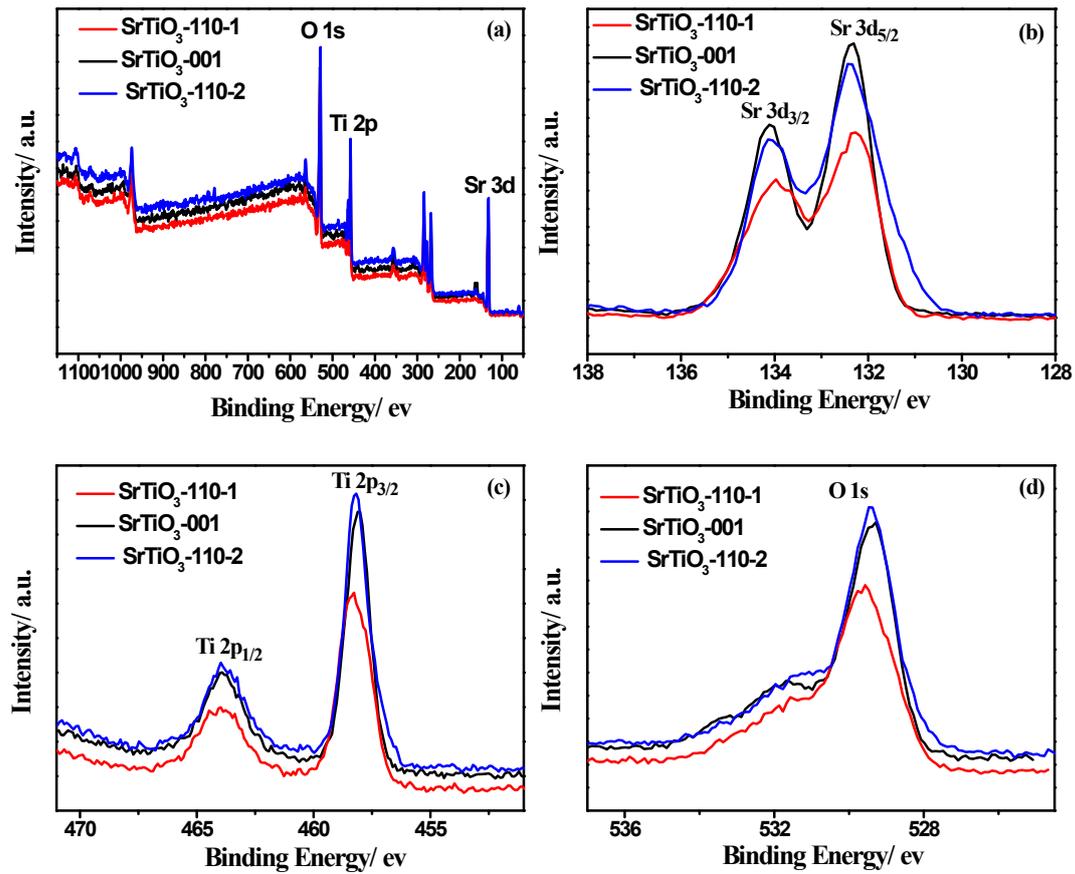


Figure S2. XPS spectra of samples: survey (a), Sr (b), Ti (c) and O (d).

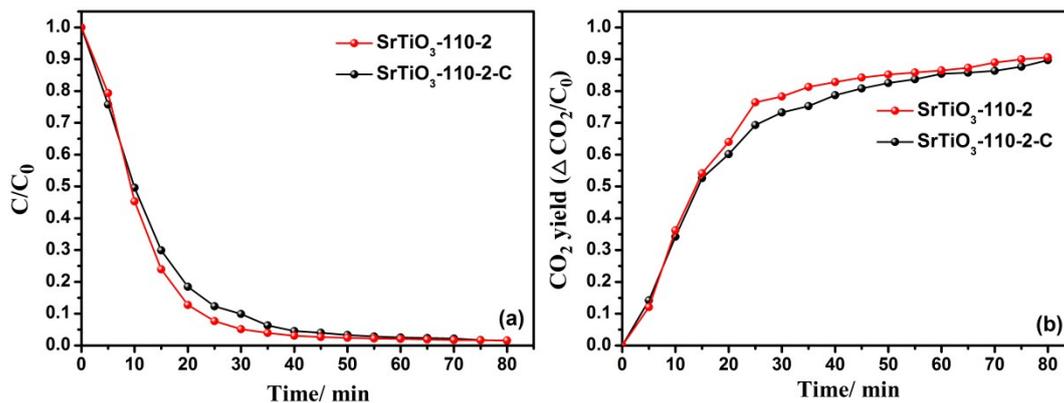


Figure S3. HCHO photocatalytic oxidation (a), the corresponding CO<sub>2</sub> yield (b) of samples under the irradiation of Xe lamp.

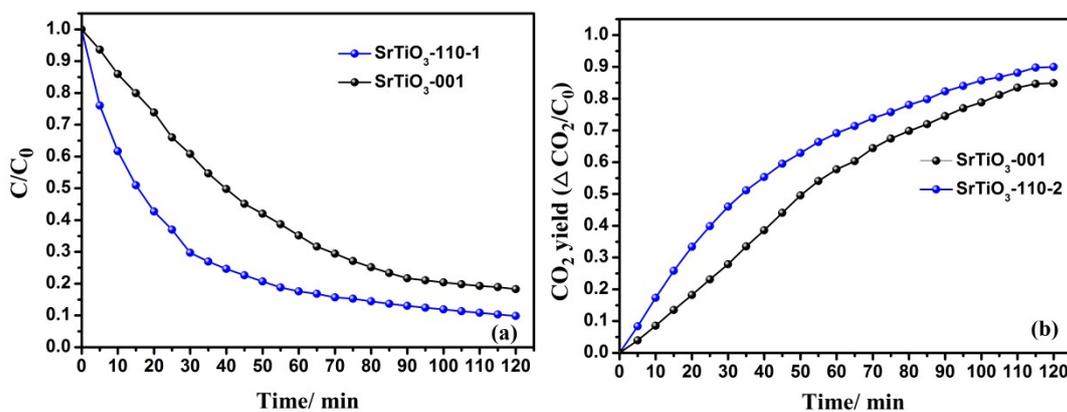


Figure S4. HCHO photocatalytic oxidation (a), the corresponding CO<sub>2</sub> yield (b) of samples under the irradiation of monochromatic light (365 nm).

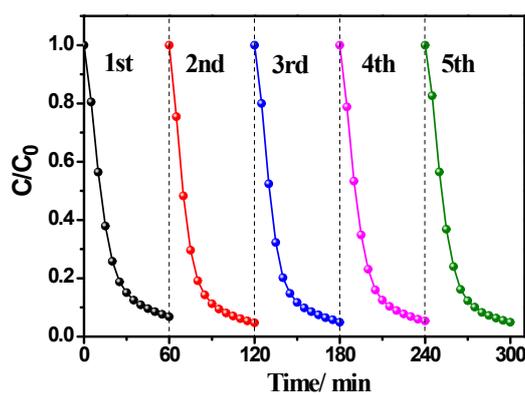


Figure S5. The stability of the sample SrTiO<sub>3</sub>-110-2 for photocatalytic degradation of HCHO.

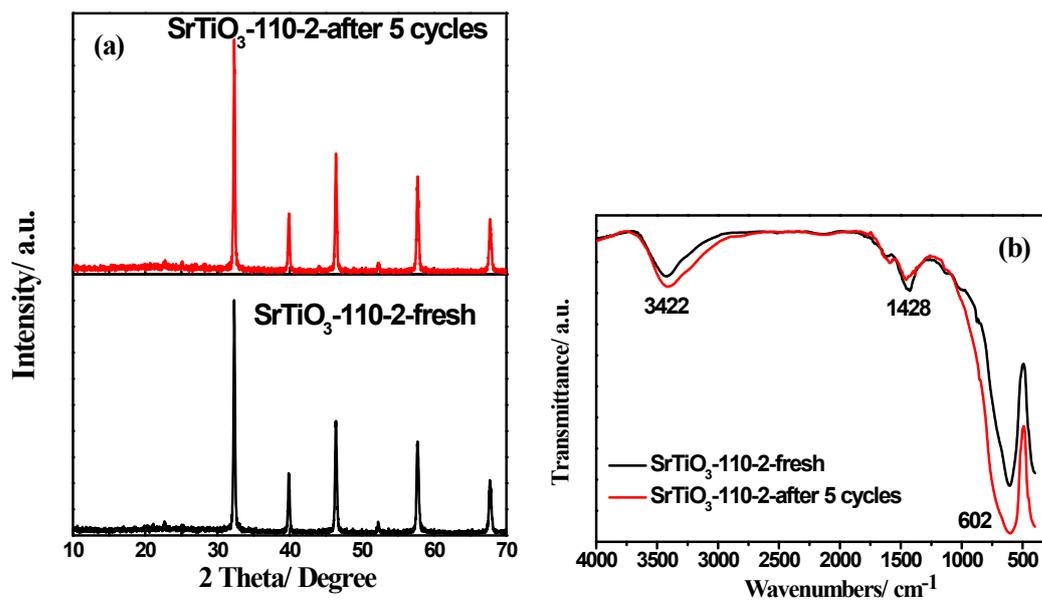


Figure S6. XRD patterns (a) and FTIR spectra (b) of samples SrTiO<sub>3</sub>-110-2 fresh one without photocatalysis and the used one after 5 runs photocatalysis.