Titanium alkylphosphate functionalised mesoporous silica for enhanced uptake of rare-earth ions

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Electronic Supplementary Information (ESI)



Figure S1. Nitrogen adsorption-desorption isotherms (*a*) and respective NLDFT pore-size distribution curves (*b*) of the synthesised materials.



Figure S2. ATR-FTIR spectra of the synthesised materials (*a*) with an enlarged view on the wavenumber region from 700 to 1400 cm⁻¹ (*b*).



Figure S3. TG/DSC–MS results of the M41-TiEtP material. The top graph shows the thermogravimetrical curve while the remaining four graphs indicate the corresponding evolved gas captured by MS detector.



Figure S4. TG/DSC–MS results of the M41-TiPrP material. The top graph shows the thermogravimetrical curve, while the remaining four graphs indicate the corresponding evolved gas captured by MS detector.



Figure S5. TG/DSC–MS results of the M41-TiBuP material. The top graph shows the thermogravimetrical curve while the remaining four graphs indicate the corresponding evolved gas captured by MS detector.



Figure S6. Desorption efficiencies using water, 0.01 M and 0.1 M nitric acid on M41-TiPrP material. Initial sorption solution composition: 1 mM Nd, 1 mM Dy and 5 M NH_4NO_3 , sorption solid-to-liquid ratio: 50 mg/ 20 mL and equilibrium pH ca. 2. Desorption solid-to-liquid ratio: 50 mg/ 20 mL.



Figure S7. Cyclic sorption-desorption tests on M41-TiPrP material: (*a*) sorption and desorption ratios of Nd and Dy normalised to the first sorption cycle uptake, and (*b*) SF(Dy/Nd) values for each sorption cycle. Initial sorption solution composition: 1 mM Nd, 1 mM Dy and 5 M NH₄NO₃, sorption solid-to-liquid ratio: 50 mg/ 20 mL and equilibrium pH ca. 2. Desorption solution: 0.1 M HNO₃ and desorption solid-to-liquid ratio: 50 mg/ 20 mL.