

Supplementary Information

Resolving the degradation pathways on O3-type layered oxides cathode surface through the nano-scale aluminum oxide coating for high-energy density sodium-ion batteries

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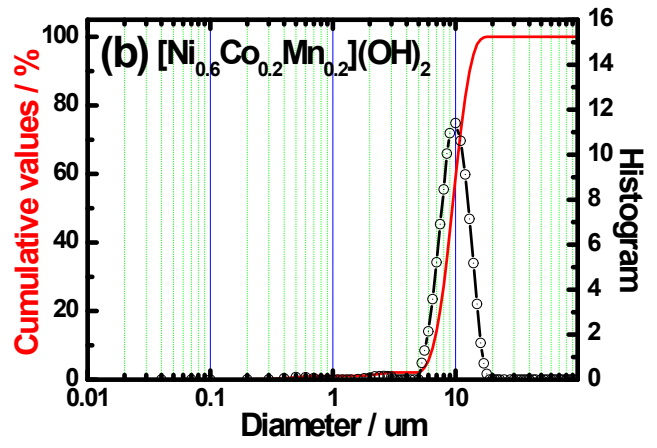
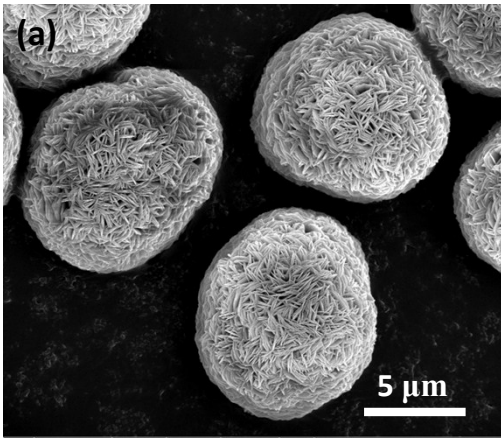


Fig. S1 (a) Scanning electron microscopy (SEM) image and (b) particle size analysis (PSA) of the $[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}](\text{OH})_2$ precursor.

Table S1. ICP-AES results for the $[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}](\text{OH})_2$ precursor.

Metal stoichiometry determined by ICP-AES			
Sample	Ni	Co	Mn
$[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}](\text{OH})_2$	0.61	0.21	0.19

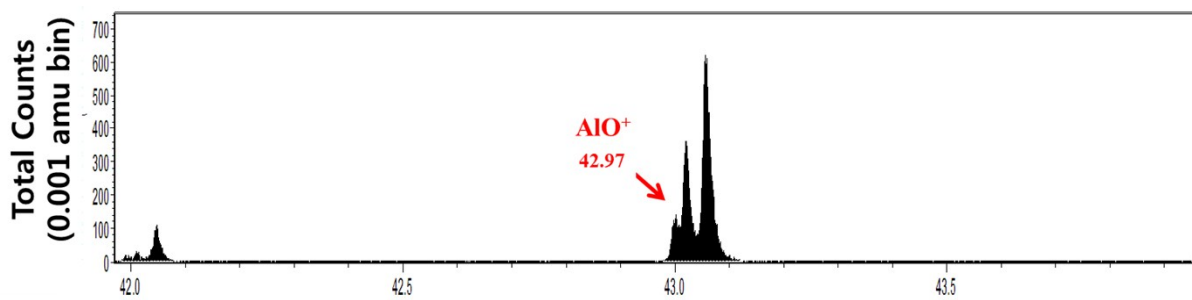


Fig. S2 ToF-SIMS spectra of the outmost surface of the Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ cathode.

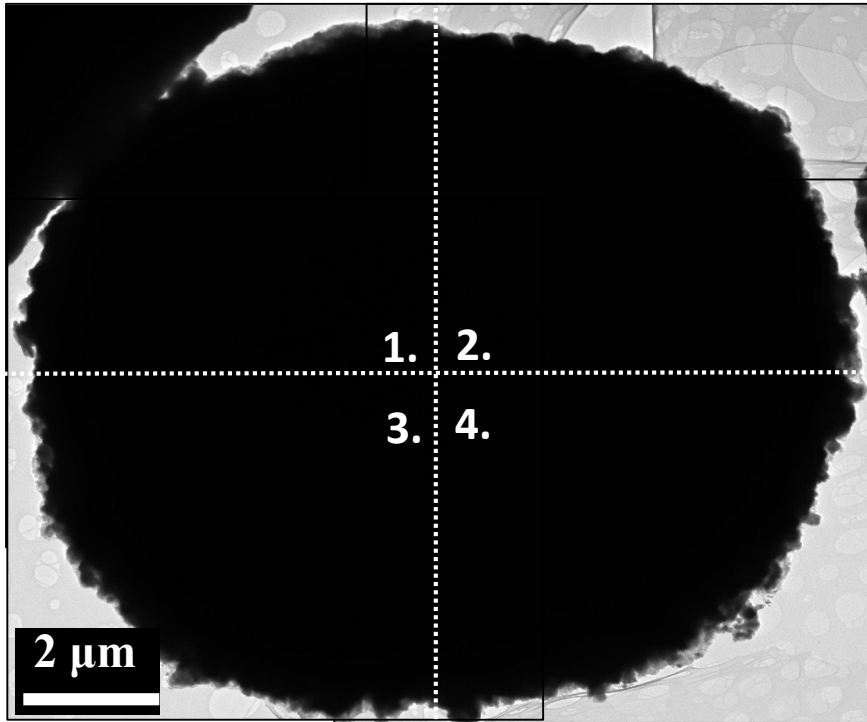


Fig. S3 Mosaic TEM image of Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ particle.

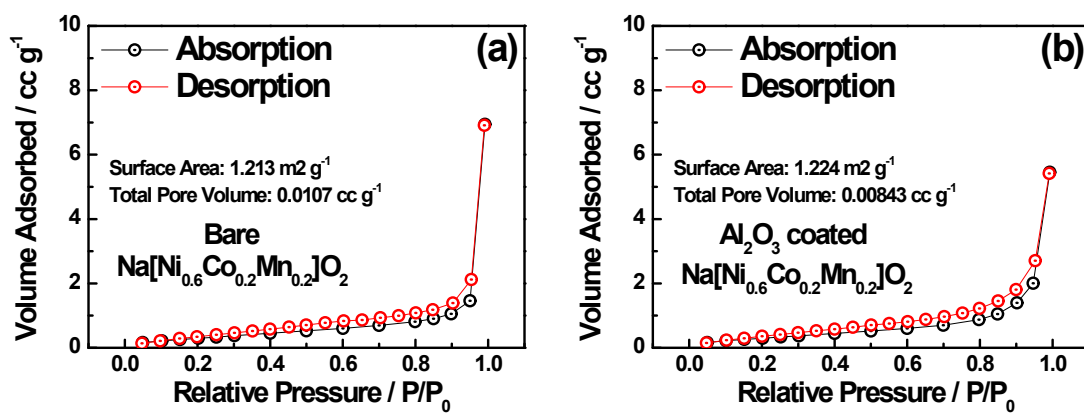


Fig. S4 Isotherm profiles with surface area and total pore volume of bare $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ and Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ cathodes.

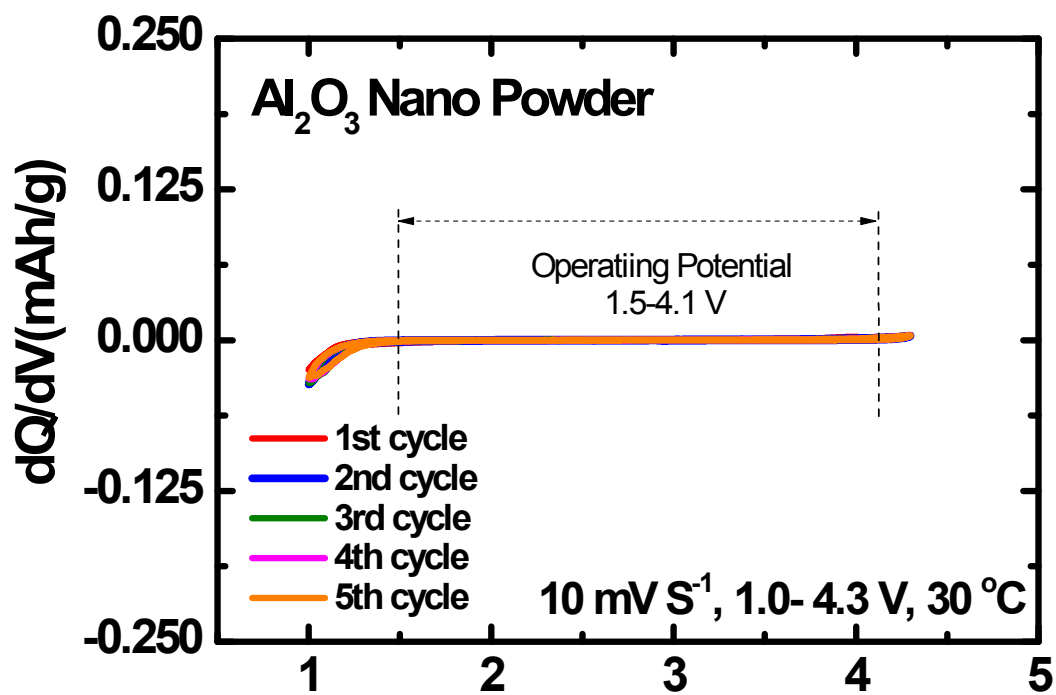


Fig. S5 Cyclic voltammetry (CV) curves of Al₂O₃ nano powder.

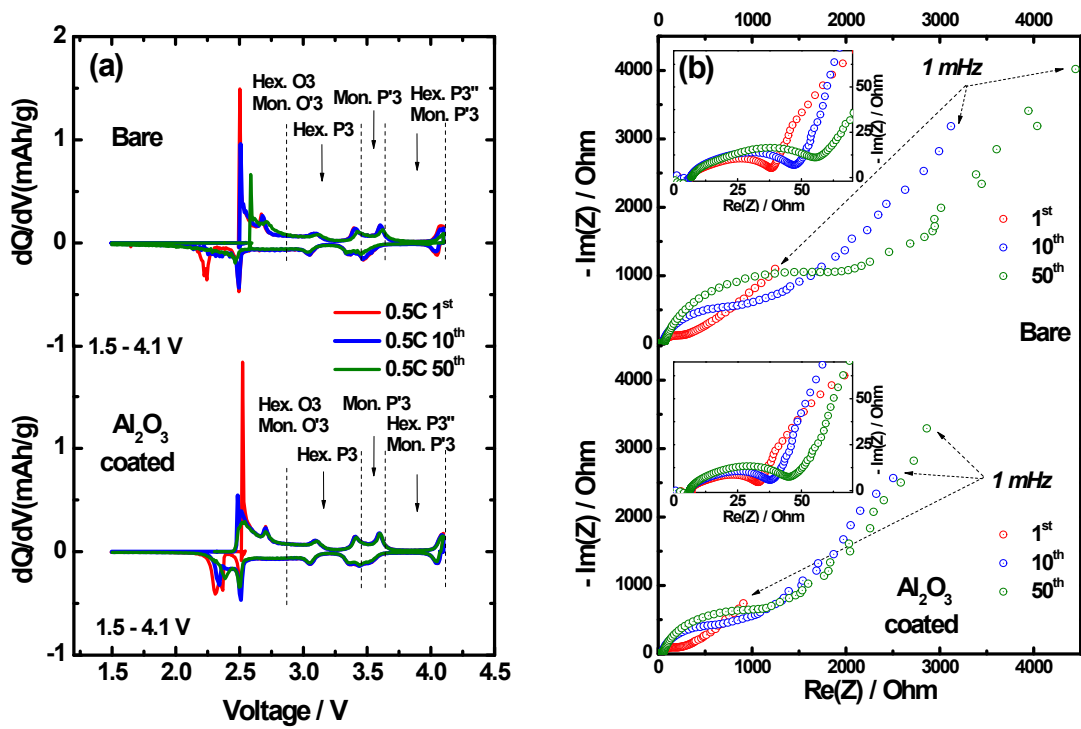


Fig. S6 (a) $dQ dV^{-1}$ (mAh g^{-1}) curves and (b) electrochemical impedance spectroscopy (EIS) results for bare $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ and Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ cathodes during cycling at a 0.5 C rate (75 mA g^{-1}) in a half cell.

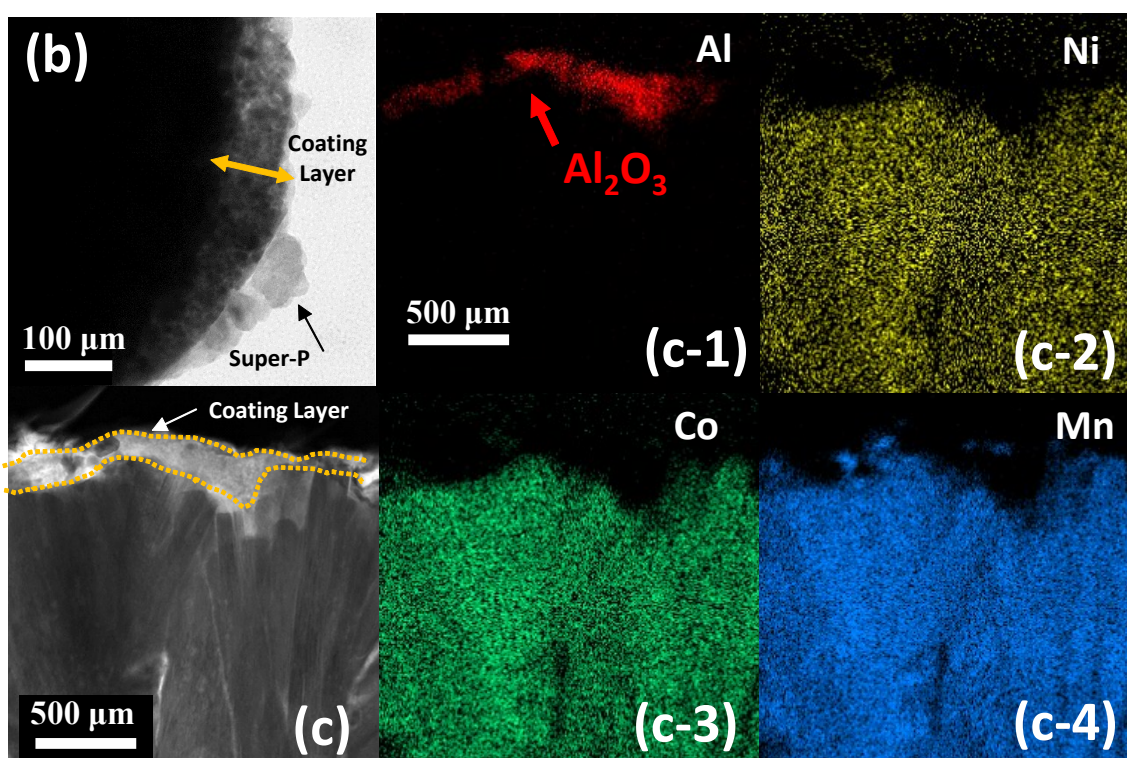
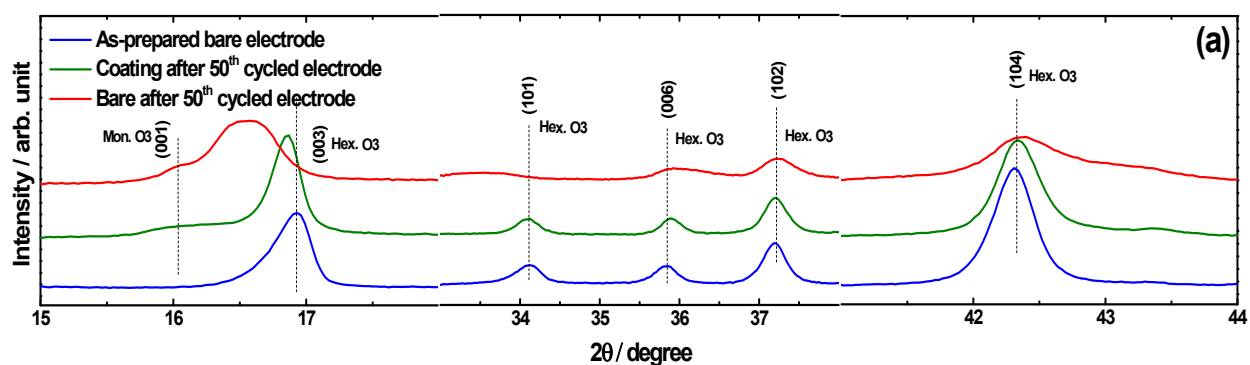


Fig. S7 Post-cycling study: (a) ex-situ XRD results before and after cycling for bare $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ and Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ electrodes. (b) Bright field TEM image and (c) cross sectional TEM image with (c-1 to c-4) corresponding EDX mapping results for cycled Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2$ cathode in a half cell.

Table S2. Electrochemical impedance spectroscopy (EIS) results for pouch-type bare Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂/hard carbon full cell and Al₂O₃ coated Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂/hard carbon full cell at the 1st and 300th cycles at a 0.5 C rate (75 mA g⁻¹).

Cathode materials	Cycles	R_{sf} (Ω)	R_{ct} (Ω)
Bare	1st	3.7	12.8
Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂	300th	11.7	220.2
Al₂O₃ coated	1st	3.2	10.7
Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂	300th	3.1	77.5

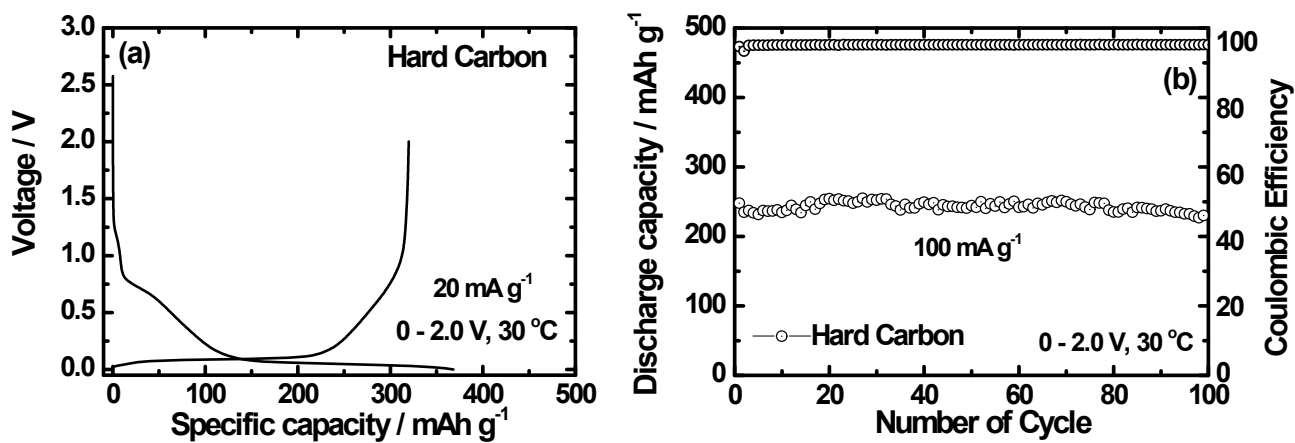


Fig. S8 Electrochemical performance of hard carbon electrode: (a) 1st discharge-charge curves at 20 mA g⁻¹ and (b) cycle life test at 100 mA g⁻¹ in the voltage range of 0–2.0 V.

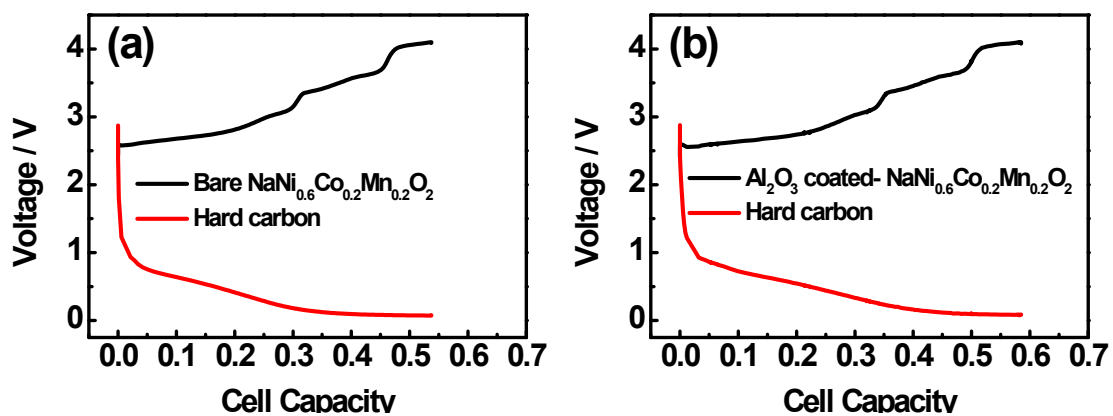


Fig. S9 Galvanostatic voltage profiles of (a) bare $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2/\text{hard carbon}$ and (b) Al_2O_3 coated $\text{Na}[\text{Ni}_{0.6}\text{Co}_{0.2}\text{Mn}_{0.2}]\text{O}_2/\text{hard carbon}$ cells upon charge process.

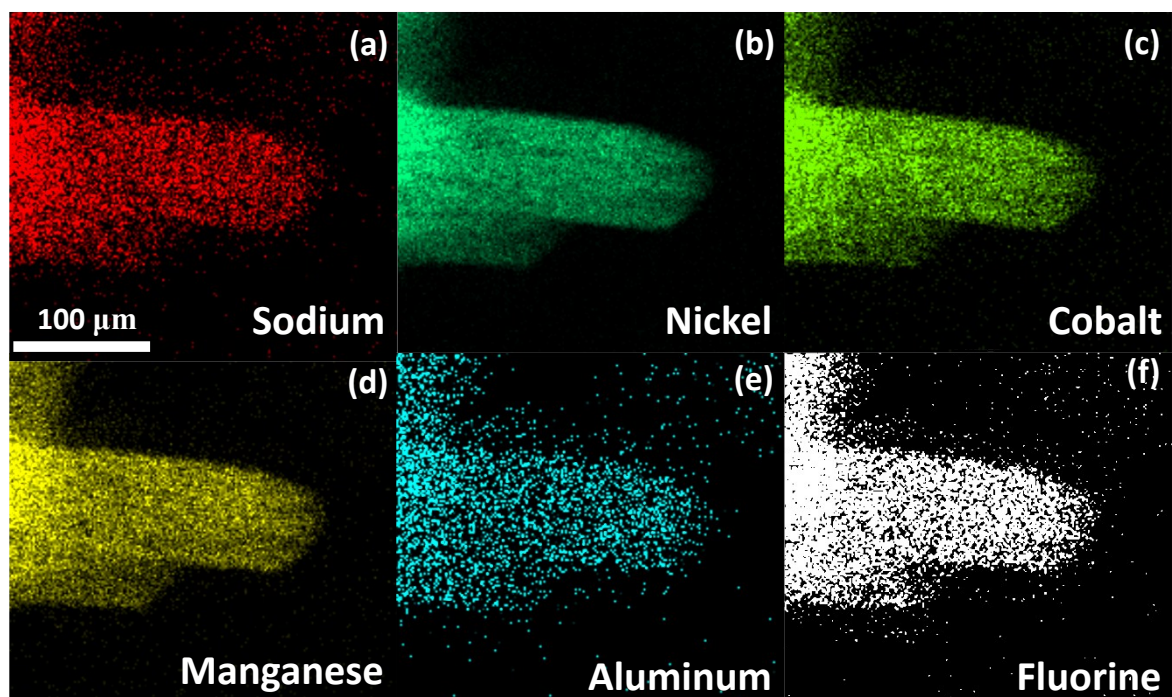


Fig. S10 TEM EDX mapping data of cycled Al_2O_3 electrode collected from pouch-type full cell: (a) sodium, (b) nickel, (c) cobalt, (d) manganese, (e) aluminum and (f) fluorine.

Table S3. HF amounts in bare Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂ and Al₂O₃ coated Na[Ni_{0.6}Co_{0.2}Mn_{0.2}]O₂ cathodes collected from pouch-type full cells after 300 cycles.

Sample	HF Amount (ppm)
Blank	50.02526
Bare Na[Ni _{0.6} Co _{0.2} Mn _{0.2}]O ₂ cathode	349.6795
Al ₂ O ₃ coated Na[Ni _{0.6} Co _{0.2} Mn _{0.2}]O ₂ cathode	83.33222

Table S4. Estimated energy density calculation and individual mass fraction based on pouch cell parameters of traditional LIB cells.

Energy Density Calculation (Wh/kg)								
		Energy Density Calculation (Wh/kg)			Value	Quantity	weight	
N/P ratio	1.15							
Cathode	Cathode	Capacity (mAh/g)				151		
		Working potential (vs Na ⁺)				3		
		Composition of Active material				0.95		
		Mass Loading of active materials (mg/cm ²)				25	7.5	7.894737
		Length (mm)				50		
		Width (mm)				30		
		Area capacity (mAh/cm ²)				3.75		
	Al foil	Thickness (mm)				0.012	10	0.48
	Cathode tab						1	0.04
	Anode	Capacity (mAh/g)				300		
		Working potential (vs Na ⁺)				0		
		Composition of Active material				0.95		
		Mass Loading of active materials (mg/cm ²)				14.375	4.543	4.784

		Length (mm)				51		
		Width (mm)				31		
		Area capacity (mAh/cm ²)				4.3125		
	Cu foil	Thickness (mm)				0.008	11	1.265
	Anode tab						1	0.13
	Packing foil	Thickness (mm)				0.086	1	0.6
	Electrolyte							4
	Separator	Thickness (mm)				0.011		0.33
	Sealant						1	0.01
					Cell weight (g)		19.534	

Cell Size (mm x mn)		Layers	Cell weight (g)	Cell capacity (mAh)	Voltage (V)	Energy density (Wh/kg)
35	57.5	20	19.53	843.75	3	130.23