

Supporting Information

Colloidal synthesis of monodisperse trimetallic IrNiFe nanoparticles as highly active bifunctional electrocatalysts for acidic overall water splitting

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Supplementary Figures: Figure S1- S4

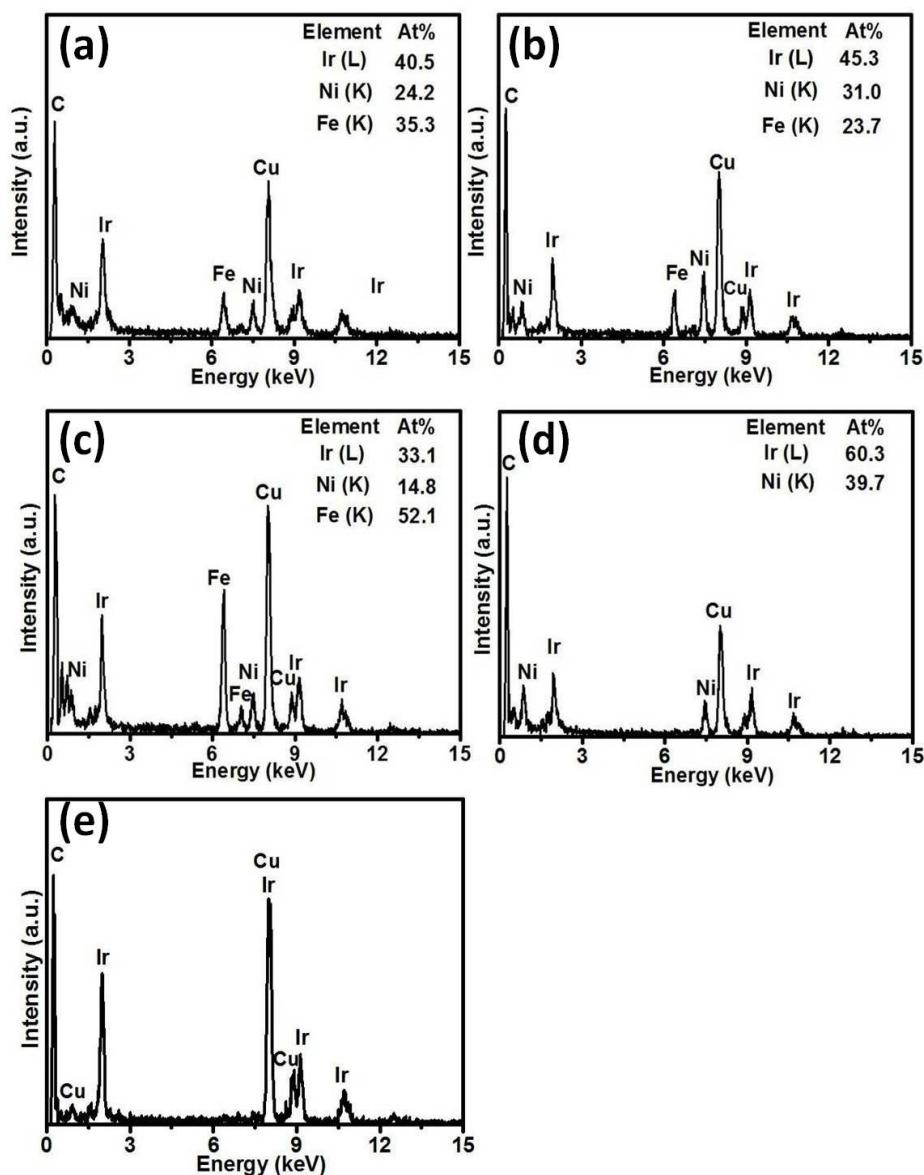


Fig. S1 EDX spectrum of IrNi_{0.57}Fe_{0.82} (a), IrNi_{0.70}Fe_{0.52} (b), IrNi_{0.46}Fe_{1.59} (c), IrNi_{0.68} (d), and Ir (e).

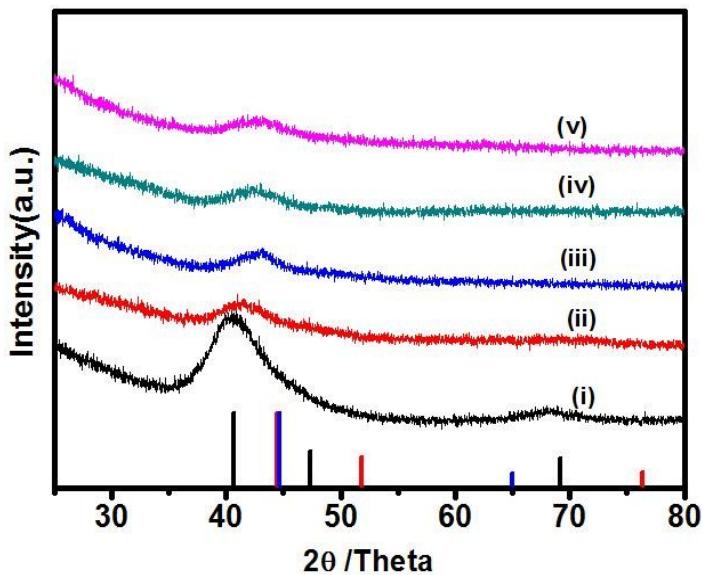


Fig. S2 XRD patterns for the as-prepared nanoparticles: (i) Ir, (ii) $\text{IrNi}_{0.68}$, (iii) $\text{IrNi}_{0.70}\text{Fe}_{0.52}$, (iv) $\text{IrNi}_{0.57}\text{Fe}_{0.82}$, (v) $\text{IrNi}_{0.46}\text{Fe}_{1.59}$. The bottoms are the corresponding standard values for Ir (black, PDF#06-0598), Ni (red, PDF#04-0850), and Fe (blue, PDF#06-0696), respectively.

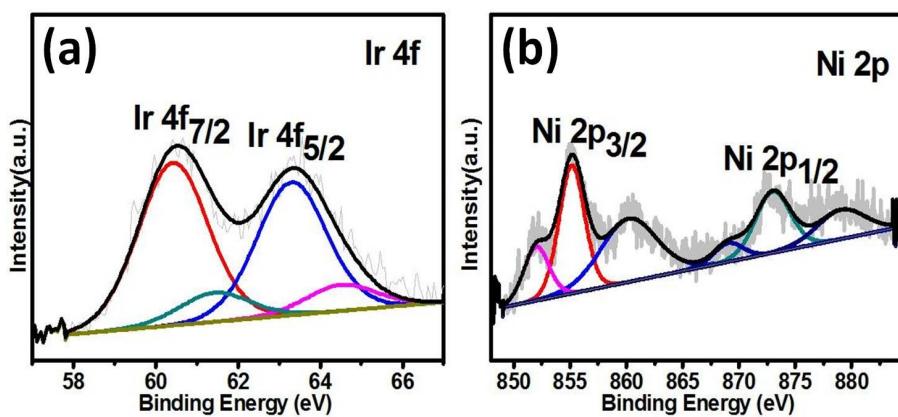


Fig. S3 The XPS spectra of (a) Ir 4f and (b) Ni 2p for $\text{IrNi}_{0.68}$.

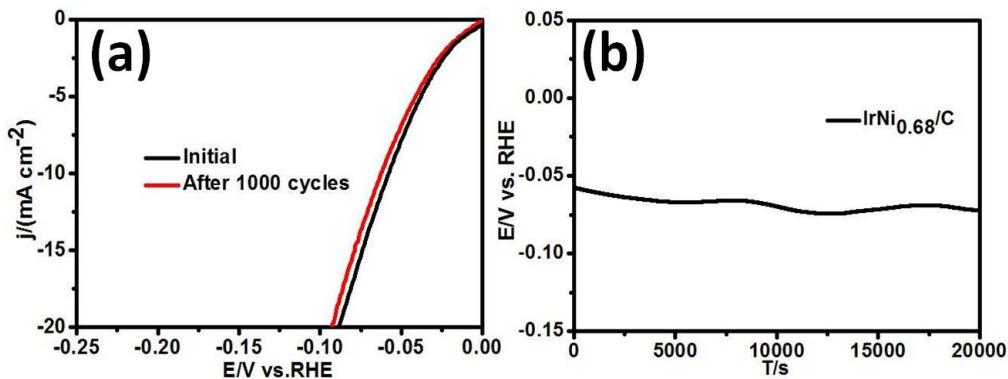


Fig. S4 (a) LSV curves of HER for $\text{IrNi}_{0.68}/\text{C}$ before and after 1000 cycles in 0.5 M HClO_4 solution; (b) Chronopotentiometry test of HER for $\text{IrNi}_{0.68}/\text{C}$ at 10 mA cm^{-2} in 0.5 M HClO_4 solution.

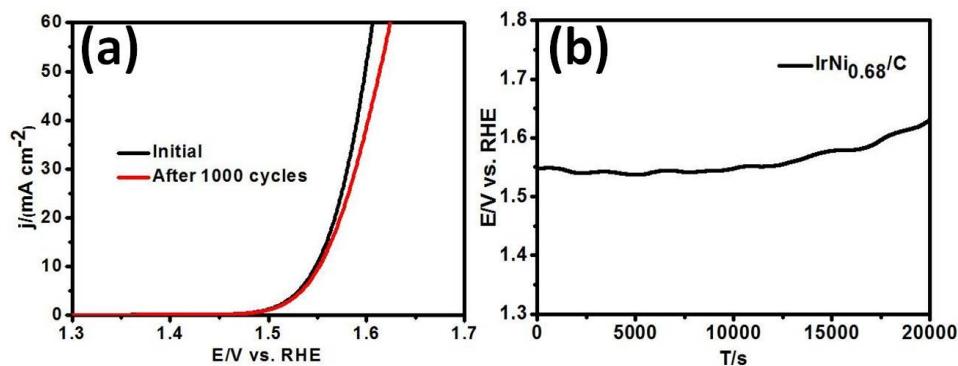


Fig. S5 (a) LSV curves of OER for $\text{IrNi}_{0.68}/\text{C}$ before and after 1000 cycles in 0.5 M HClO_4 solution; (b) Chronopotentiometry test of OER for $\text{IrNi}_{0.68}/\text{C}$ at 10 mA cm^{-2} in 0.5 M HClO_4 solution.

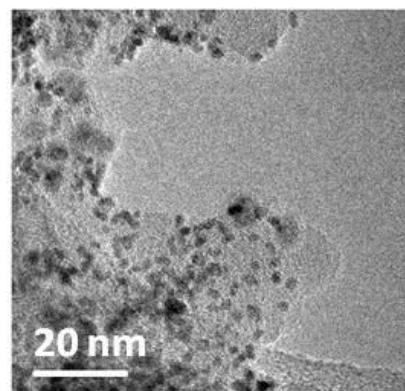


Fig. S6 TEM image of $\text{IrNi}_{0.57}\text{Fe}_{0.82}/\text{C}$ after the stability test.

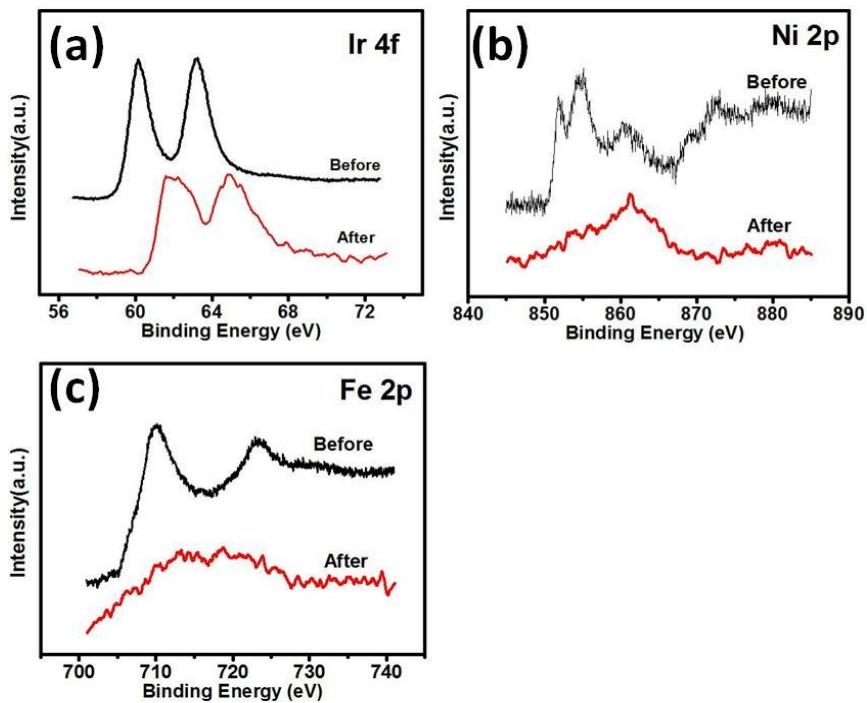


Fig. S7 The XPS spectra of (a) Ir 4f, (b) Ni 2p, and (c) Fe 2p for $\text{IrNi}_{0.57}\text{Fe}_{0.82}$ before and after OER.

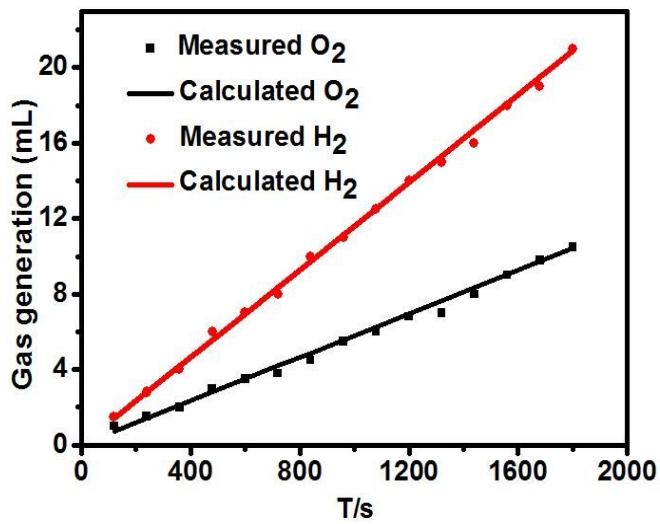


Fig. S8 Faraday efficiency of H_2 and O_2 production.

Table S1. Comparison of HER activity for different electrocatalysts in acidic electrolytes.

Catalysts	Electrolyte	Current density	η / mV	Reference
IrNiFe NPs	0.5 M HClO₄	10 mA cm⁻²	24	This work
IrNiN NPs	0.1 M HClO ₄	6 mA cm ⁻²	110	1
IrO ₂ –Fe ₂ O ₃	0.5 M H ₂ SO ₄	10 mA cm ⁻²	78	2
IrO ₂ –TiO ₂	0.5 M H ₂ SO ₄	10 mA cm ⁻²	112	3
Ru@C ₂ N	0.5 M H ₂ SO ₄	20 mA cm ⁻²	35	4
Ru/C ₃ N ₄ /C	0.5 M H ₂ SO ₄	10 mA cm ⁻²	70	5
Rh/Si	0.5 M H ₂ SO ₄	50 mA cm ⁻²	110	6
Rh-MoS ₂	0.5 M H ₂ SO ₄	10 mA cm ⁻²	47	7
Pt ₃ Ni ₃ NWs	0.5 M H ₂ SO ₄	10 mA cm ⁻²	30	8

References

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