

Investigation of Chloride Ion Adsorption onto Ti_2C MXene

Monolayers by First-principles Calculations

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Supporting Information

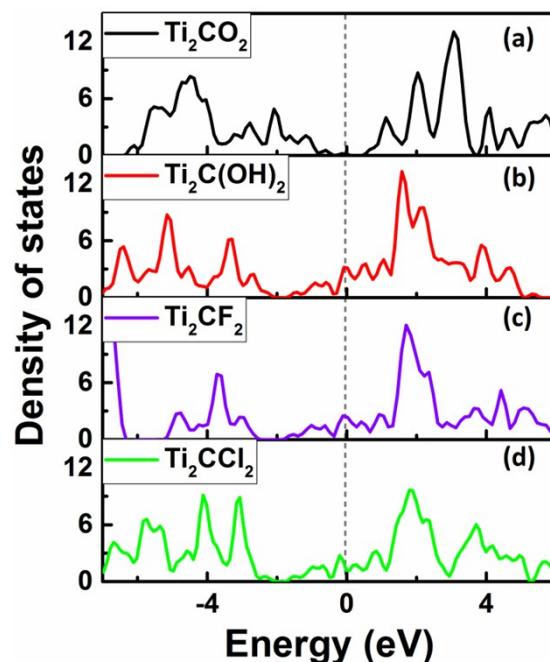


Fig. S1. Density of states (DOS) of (a) Ti_2CO_2 , (b) $\text{Ti}_2\text{C}(\text{OH})_2$, (c) Ti_2CF_2 , and (d) Ti_2CCl_2 monolayers, respectively.

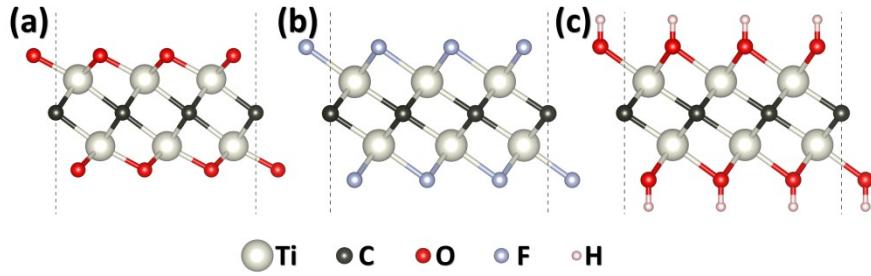


Fig. S2. Side views of (a) Ti_2CO_2 , (b) Ti_2CF_2 , and (a) $\text{Ti}_2\text{C}(\text{OH})_2$ monolayers.

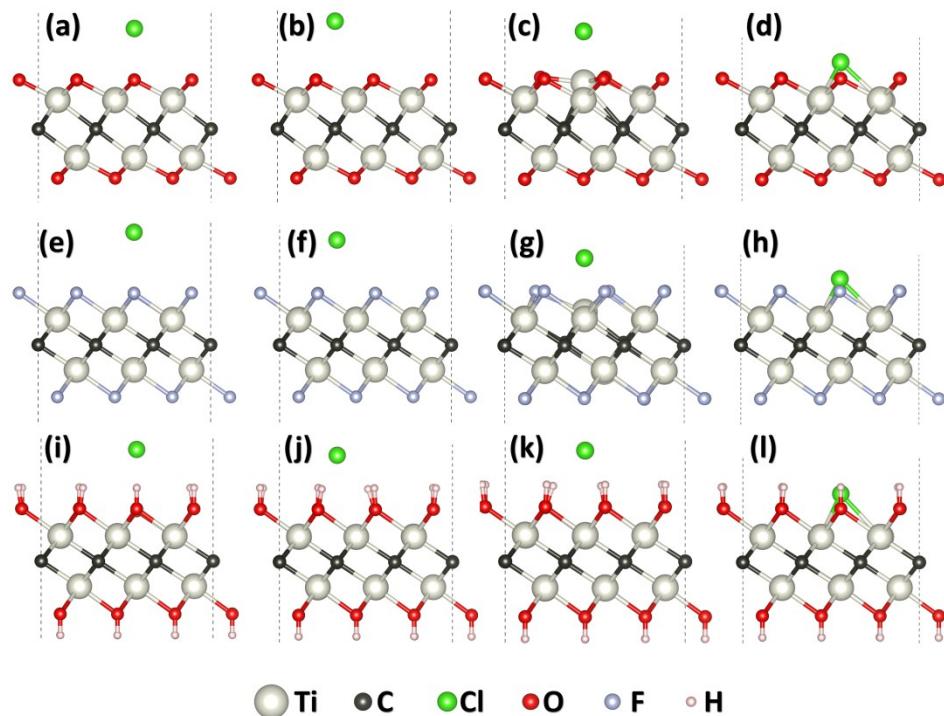


Fig. S3. Optimized Cl adsorption configurations for (a)-(d) Ti_2CO_2 , (e)-(h) Ti_2CF_2 , and (i)-(l) $\text{Ti}_2\text{C}(\text{OH})_2$ monolayers. (d), (h), and (l) correspond to the Cl adsorption configurations on Ti_2CT_2 monolayers with one T ($\text{T} = \text{O}$, F , or OH , respectively) vacancy.

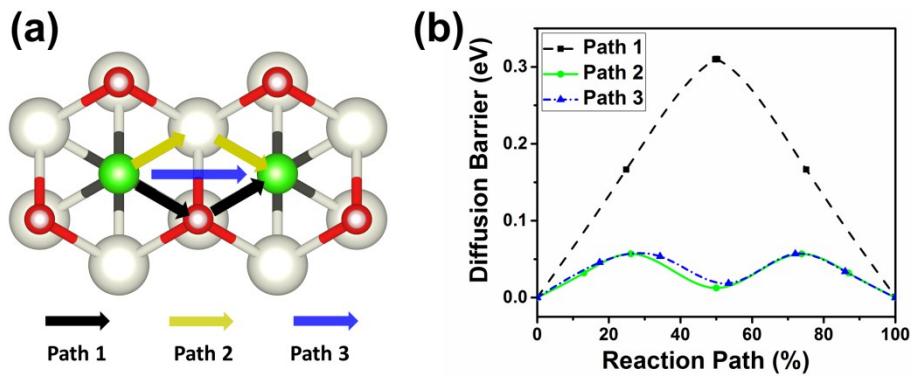


Fig. S4. (a) Possible Cl^- diffusion pathways on a $\text{Ti}_2\text{C}(\text{OH})_2$ monolayer. (b) Diffusion barrier profiles of Cl^- on a $\text{Ti}_2\text{C}(\text{OH})_2$ monolayer.