

Supporting Information

**Innovation and Challenges in Materials Design for Flexible Rechargeable Batteries: from 1D to 3D**

Yanghua He, Bryan Matthews, Jingyun Wang, Li Song, Xiaoxia Wang, and Gang Wu\*

Department of Chemical and Biological Engineering, University at Buffalo, The State University of New York, Buffalo, NY 14260, United States

\*Corresponding author: [gangwu@buffalo.edu](mailto:gangwu@buffalo.edu) (G.W.)

**Table S1 to S3**

**Table S1. Summary of 1D (wire-type) flexible batteries**

Battery type	Anode	Cathode	Electrolyte	Separator	Electrochemical performance		Ref
					Discharge capacity	Cycling performance	
All-solid-state Zn-air battery	Spiral zinc foil	Air cathode based on Fe/N/C catalyst	Gel polymer	/	0.9 mAh cm <sup>-2</sup> at 0.1 mA cm <sup>-2</sup>	/	1
Li-O <sub>2</sub> battery	Lithium metal rod	Super P coated on carbon textiles	Gel polymer	/	4800 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	90 cycles	2

Lithium battery	Twisted bundle of Li and Cu mesh	Carbon cloth/Al mesh	Organic liquid electrolyte	Polyethylene	$1.5 \text{ mAh cm}^{-2}$	50 cycles	3
Lithium -ion battery	Lithium wire	Spring-like fiber	Gel electrolyte	/	$92.4 \text{ mAh g}^{-1}$ at $0.1 \text{ mA cm}^{-2}$	100 cycles	4
Sodium-ion battery	Sodium foil	Prussian blue graphene composites coated Ni-coated cotton textile	/	Flexible membrane	$110 \text{ mAh g}^{-1}$ at $60 \text{ mA g}^{-1}$	1800 cycles at a high rate of 5 C	5
ZN-air battery	Zn plate	Strung Co <sub>4</sub> N and intertwined N-C fibers	6.0 M KOH with 0.2 M zinc acetate	/	774 and 701 $\text{mAh g}^{-1}$ at 10 and 50 $\text{mA cm}^{-2}$	136 h (up to 408 cycles)	6
Zn-air battery	Zinc spring	Aligned and cross-stacked carbon nanotube sheet and a RuO <sub>2</sub> -based catalyst	Hydrogel polymer	/	$1 \text{ A g}^{-1}$	30 cycles	7
Lithium -sulfur battery	Lithium wire	Sulfur-containing hybrid fiber	1 M Lithium bis-(trifluoromethanesulfonyl) imide (LiTFSI) and 1 wt% Lithium nitrate (LiNO <sub>3</sub> ) in a	Celgard 2400	$335 \text{ mAh g}^{-1}$ at $167.5 \text{ mA g}^{-1}$ (0.1 C)	100 cycles	8

			mixture of DOL (DOL), 1,2-dimethoxyethane (DME) (volume ratio 1:1)				
Lithium-sulfur battery	Li foil	Binder-free fibrous reduced graphene oxide/carbon nanotube/sulfur composite	1 m LiTFSI in DME and DOL mixture (1:1 v/v) with 1 wt% of $\text{LiNO}_3$	Celgard 2400	1255 mAh g <sup>-1</sup> ; 2.49 mAh cm <sup>-2</sup> at C/20	30 bending cycles	<sup>9</sup>
Li-O <sub>2</sub> battery	Lithium strip	RuO <sub>2</sub> /N-carbon nanotubes And N-carbon nanotubes loaded metal/cotton yarn	0.5 M LiTFSI in tetraethylene glycol Dimethoxyethane (TEGDME)	Celgard 3500	1981 mAh g <sup>-1</sup> at 320 mA g <sup>-1</sup>	100 cycles (more than 600 hours)	<sup>10</sup>
Lithium ion battery	Ni–S–N-coated Cu wires	An aluminum wire around the hollow-helix electrode	1 M lithium hexafluorophosphate (LiPF <sub>6</sub> ) in ethylene carbonate (EC) and propylene carbonate (PC) (1:1 by volume) containing 3 wt% vinylene carbonate (VC)	Modified poly(ethylene terephthalate) nonwoven support	1 mAh cm <sup>-2</sup>	slight loss of capacity after mechanical bending	<sup>7</sup>
Zinc–carbon battery	Zn powder/carbon fiber or Zn wire	MnO <sub>2</sub> /carbon fiber	The electrolyte consisted of NH <sub>4</sub> Cl, ZnCl <sub>2</sub> and deionized water, with a weight percentage ratio of	An insulated enameled wire	0.15, 0.14, 9.3*10 <sup>-2</sup> , and 7.2*10 <sup>-2</sup> mAh/cm at	/	<sup>11</sup>

Silver-zinc Battery	Plate zinc	Wrap silver electrode	Polyvinyl alcohol (PVA)-KOH polymer electrolyte	Cellophane film	0.15, 0.3, 0.6 and 1.5 mA 1.2 and 1.8 mAh cm <sup>-1</sup> at a 0.5C discharge rate	170 cycles	<sup>12</sup>
---------------------	------------	-----------------------	---	-----------------	--	------------	---------------

**Table S2. Summary of 2D (paper-type) flexible batteries**

Battery type	Anode	Cathode	Electrolyte	Separator	Electrochemical performance		Ref.
					Discharge capacity	Cycling performance	
Lithium ion battery	Graphene paper	Lithium metal foil	1.15 m LiPF <sub>6</sub> dissolved in a mixture of EC, ethyl methyl carbonate (EMC) , and dimethyl carbonate (DC) (3:2:5 volume ratio)	Porous polypropylene film	1300 mAh g <sup>-1</sup> at 60 mA g <sup>-1</sup>	No more than 10 cycle	<sup>13</sup>
Lithium rechargeable battery	Graphene papers	Lithium sheets	1 m LiPF <sub>6</sub> solution in a 1:1 (volume) mixture of EC and DMC	/	1350 mA h g <sup>-1</sup> at 50 mA g <sup>-1</sup>	100 cycles	<sup>14</sup>
Li-O <sub>2</sub> battery	Li metal	Graphene nanoplatelets/graphene oxides film	1.0 M LiNO <sub>3</sub> in dimethylacetamide (DMA)	Glass microfiber filter	9760 mAh/g at 9760 mAh/g	20 cycles	<sup>15</sup>

Lithium–sulfur battery	Lithium foil	Mesoporous graphene paper	1 m LiTFSI and 1 wt% LiNO <sub>3</sub> in DOL and DME (volume ratio 1 : 1)	/	1393 mA h g <sup>-1</sup>	50 cycles	<sup>16</sup>
Li-ion battery	Lithium metal	Exfoliated graphite attaching silicon nanoparticles	1.0 M LiPF <sub>6</sub> dissolved in a mixture of EC and DC (1: 1, v/v) with VC (5 v.% VC by volume)	Celgard 2400	902.8 mAh g <sup>-1</sup> at 200 mA g <sup>-1</sup>	40 cycles	<sup>17</sup>
Li–O <sub>2</sub> battery	Lithium foil	Graphene oxide paper	1 M LiPF <sub>6</sub> dissolved in TEGDME	Glass fibers	612 mAh g <sup>-1</sup> at 0.01 mA cm <sup>-2</sup>	10 cycles	<sup>18</sup>
Li–O <sub>2</sub> battery	Lithium belt	Paper-ink air cathode	1M LiCF <sub>3</sub> SO <sub>3</sub> in TEGDME electrolyte	Glass fiber	6500 mAh g <sup>-1</sup> at 200 mA g <sup>-1</sup>	1000 cycles	<sup>19</sup>
Lithium-ion battery	Additive-free thick graphene film	Lithium foil	1 M LiPF <sub>6</sub> mixed with EC and DC 1 : 1 by volume	Polypropylene film	350 mA h g <sup>-1</sup>	50 cycles	<sup>20</sup>
Lithium-ion battery	Lithium metal foil	Coral-like magnetite@cobalt(ii) oxide@graphene foam	1.0 M LiPF <sub>6</sub> solution in a mixture of EC and DEC (7:3 by volume)	/	1551.2 mA h g <sup>-1</sup> at 0.5 A g <sup>-1</sup>	400 cycles	<sup>21</sup>
Lithium-ion battery	CuO nanosheets/reduced-graphene oxide	Li metal foil	50:50 (w/w) EC and DC mixture solution with 1 M LiPF <sub>6</sub>	/	698.0 mA h g <sup>-1</sup> at 0.1C	50 cycles	<sup>22</sup>

		composite						
All-solid-state zinc–air battery	Zinc foil	Bimetal FeCo nanoparticles encapsulated in in situ grown N-doped graphitic carbon nanotubes	Gel polymer (6 M KOH with 0.2 M zinc acetate)	/	872.2 mAh g <sup>-1</sup>	240 cycles	23	
Lithium ion battery	$\text{Li}_4\text{Ti}_5\text{O}_{12}$ nanosheets with a N-doped carbon	Lithium metal foil	1m LiPF <sub>6</sub> in EC and DC (1:1 vol)	Celgard 2400	170 mAh g <sup>-1</sup> at a rate of 1 C	100 cycles	24	
Lithium ion battery	$\text{Li}_4\text{Ti}_5\text{O}_{12}$ nanowire arrays on freestanding ultrathin graphite	Lithium metal foil	1 M LiPF <sub>6</sub> solution in a 1 : 1 (v/v) mixture of EC and DMC	Polyethylene film	154 mAh g <sup>-1</sup>	500 cycles	25	
Lithium-ion battery	MoS <sub>2</sub> on freestanding graphene films	Lithium metal foil	1 M LiPF <sub>6</sub> in EC/DEC (1:1 by volume)	Polypropylene film	580 mAh/g (@50 mA/g)	30 cycles	26	
Lithium-ion battery	MoS <sub>2</sub> –graphene hybrid paper	Lithium metal foil	1 M solution of LiPF <sub>6</sub> in EC–DCat a vol ratio of 1 : 1	Celgard 2400	1240 mAh g <sup>-1</sup>	100 cycles	27	

Sodium-ion battery	Sb/rGO Nanocomposites	$\text{Na}_3\text{V}_2(\text{PO}_4)_3$ /rGO nanocomposite s	A mixture of EC and DC1:1 (w/w) containing 1 M $\text{NaClO}_4$	Glass fiber	400 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	100 cycles	28
Lithium-ion battery	Ultrasmall $\text{SnS}_2$ nanocrystals decorated on flexible rGO	Lithium metal foil	A solution of 1 M $\text{LiPF}_6$ in EC–DMC–DC (1 : 1 : 1, by wt%)	Celgard 2400	1034 mAh g <sup>-1</sup>	450 cycles	29
Lithium-ion battery	Yolk–shell $\text{CoS}_2$ and N-doped graphene sheets	Lithium metal foil	1.0 M $\text{LiPF}_6$ solution in EC and DC(1:1 by volume)	Celgard 2400	992 mAh g <sup>-1</sup> 100 mAh g <sup>-1</sup>	150 cycles	30
Li-ion battery	Graphene/S composite	Lithium metal foil	1.0 M $\text{LiN}(\text{CF}_3\text{SO}_2)_2$ and 0.2 M $\text{LiNO}_3$ in a mixture of DME and DOL mixed in a 1 : 1 volumetric ratio	Celgard 2320	~1500 mAh g <sup>-1</sup>	250 cycles	31
Li–sulfur battery	Lithium-metal	Graphene-based porous carbon films	1 M LiTFSI and 0.2 M $\text{LiNO}_3$ in DOL and DME (the volume ratio was 1 : 1)	Glass fiber	1017 mAh g <sup>-1</sup> at 0.2C	300 cycles	32
Lithium-sulfur battery	Lithium foils	Reduced graphene oxide–sulfur composite	1.0 M LiTFSI in a mixture solution of DOL and DME (1:1, vol. Ratio) with 1 wt% $\text{LiNO}_3$ addition	Celgard 2320	1302 mAh g <sup>-1</sup> at 0.1 C	70 cycles	33
Lithium-sulfur	Lithium foil	Lithium foil	1 M LiTFSI and 0.25 M $\text{LiNO}_3$ additive in DOL:DME	Hybrid separator with n and s	1267 mAh g <sup>-1</sup> at 0.2C	500 cycles	34

battery			(1:1 v/v)	dual-doped mesoporous carbon			
Li-ion battery	Si/rGO hybrid film	Lithium foil	A solution of 1 M LiPF <sub>6</sub> in EC/DMC/DEC (1:1:1 in volume)	Celgard 2400	904 mAh g <sup>-1</sup> at 200 mAh g <sup>-1</sup>	150 cycles	<sup>35</sup>
Li-S battery	Lithium foil	Dual-confined flexible sulfur cathodes encapsulated in N-doped double-shelled hollow carbon spheres and wrapped with graphene	1.0 M LiCF <sub>3</sub> SO <sub>3</sub> , 1 M in DOL and DME (1:1 by volume) with 0.1 M LiNO <sub>3</sub> additive	Celgard 2500	1360 mA h g <sup>-1</sup> at 0.2C	200 cycles	<sup>36</sup>
Lithium-sulfur battery	Infiltrated porous polymer sheets	Pure Li foil	2.4 M LiTFSI salt solution in distilled DME:DOL (1:1, v:v) with 0.24 M LiI additive	Celgard 2400	1640 mAh g <sup>-1</sup> at C/20	200 cycles	<sup>37</sup>
Solid-state lithium battery	Lithium foil	LiFePO <sub>4</sub>	/	Composite membranes consisting of lithium garnet particles and Li-salt-free polyethylen	153.3 mAh g <sup>-1</sup> at 0.05C	200 cycles	<sup>38</sup>

			e oxides				
Li-ion battery	Lithium metal	Nano-fibrillated cellulose, LiFePO <sub>4</sub> and super-P carbon particles based paper	1M LiPF <sub>6</sub> in EC: DEC 1:1 by weight	Whatman Glass microfiber	151 mAh/g at C/10	50 cycles	<sup>39</sup>
Lithium-ion battery	Graphite	LiCoO <sub>2</sub>	1 M LiPF <sub>6</sub> in EC/DMC/DEC (1:1:1, v/v/v)	Compliant gel polymer electrolyte based on poly(methyl acrylate-co-acrylonitrile)/poly(vinyl alcohol)	140 mAh g <sup>-1</sup>	50 cycles	<sup>40</sup>
Lithium-sulfur battery	Lithium foil	Ultra-small sulfur nanoparticles configured inside a flexible organic mixed conducting network	1m LiTFSI in DOL : DME (1:1 v/v)	Whatman glass fiber	1669 mA h g <sup>-1</sup>	100 cycles	<sup>41</sup>
Lithium-ion	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	LiCoO <sub>2</sub>	Plastic crystal composite polymer electrolyte	Polyethylene	123 mA h g <sup>-1</sup> at 0.2C	40 cycles	<sup>42</sup>

battery								
Lithium-ion battery	Lithium metal foil	LiFePO <sub>4</sub> /C	1.0 M LiPF <sub>6</sub> solution in a 1:1 v/v mixture of EC and DMC	Zirconium oxide based-ceramic	123 mA h g <sup>-1</sup> at 0.1C	100 cycles		43
Lithium-ion battery	Li metal	LiMn <sub>2</sub> O <sub>4</sub>	1 m LiPF <sub>6</sub> in EC/DEC = 1/1 v/v	Flexible/functional porous ceramic membranes	~100 mA h g <sup>-1</sup>	100 cycles		44
Lithium ion battery	Nitridated hematite (α-Fe <sub>2</sub> O <sub>3</sub> ) nanorods	Li metal	1 m LiPF <sub>6</sub> in a mixture of EC and DMC (1:1 by volume)	Celgard 2400	1086 mAh g <sup>-1</sup> at 0.2C	200 cycles		45
Na-ion battery	Nanopore-structured g-Fe <sub>2</sub> O <sub>3</sub> film	Li metal	1 M NaCF <sub>3</sub> SO <sub>3</sub> in TEGME	Celgard 2300	450 mA h g <sup>-1</sup>	100 cycles		46
Li-ion battery	Li <sub>δ</sub> Ni <sub>0.75-z</sub> Co <sub>0.11</sub> Mn <sub>0.14</sub> V <sub>z</sub> O <sub>2</sub> doping layer	Multishelled LiNi <sub>0.75</sub> Co <sub>0.11</sub> Mn <sub>0.14</sub> O <sub>2</sub>	1.15 M LiPF <sub>6</sub> in EC/DMC/DC (3/4/3 vol %)	Microporous polyethylene	211 mAh g <sup>-1</sup>	200 cycles		47
Li-ion battery	Li metal	LiCoO <sub>2</sub> layer	/	/	106 μAh/cm <sup>2</sup>	100 cycles		48
Lithium-ion battery	Si nanopillar	Li foil	1M LiPF <sub>6</sub> in 1:1 EC: DC electrolyte	Celgard 2500	3000 mAh/g	40 cycles		49
Li-ion	Cellulose	Lithium foil	1.0 M LiPF <sub>6</sub> in a 1:1 mixture	/	350 mA h g <sup>-1</sup>	50 cycles		50

battery	based paper		of EC and DEC					
Li–S battery	Lithium metal foil	Sulfur-graphene-polypropylene separator	1.0 M LiTFSI in DOL and DME (1:1 by volume) with 1.0 wt% LiNO <sub>3</sub> additive	Celgard 2400	1278 mA h g <sup>-1</sup> at 0.3 A g <sup>-1</sup>	50 cycles		51
Lithium-ion battery	Lithium metal	In situ polymerized polyimide/single-wall carbon nanotube film	1 M LiN(CF <sub>3</sub> SO <sub>2</sub> ) <sub>2</sub> solution in a mixed solvent of DOL and DME (1:1, in weight)	Glass fiber	226 mA h g <sup>-1</sup> at 0.1C	200 cycles		52
Li-ion battery	Graphite	LiCoO <sub>2</sub>	Graphene-oxide embedded in solid polymer electrolyte (1 M LiPF <sub>6</sub> in EC and DC (1 : 1 vol/vol))	/	0.14 mAh cm <sup>-2</sup>	100 cycles		53
Lithium-ion battery	Polymer-carbon nanotube composite	Lithium metal	1 M of LiPF <sub>6</sub> in a 1:1 mixture of EC and DMC	Celgard 2400	190 mAh g <sup>-1</sup>	15 cycles		54
Zinc–air battery	LaNiO <sub>3</sub> /NC NT	Co <sub>3</sub> O <sub>4</sub>	/	PVA-gelled membrane	450 mAh g <sup>-1</sup> at 50 A kg <sup>-1</sup>	120 cycles		55
Lithium-ion battery	Lithium titanate/carbon on nanotube/cellulose nanofiber hybrid	Lithium metal	1 m LiPF <sub>6</sub> in a 1:1 (volume) mixture of EC and DEC	Celgard 2400	142 mAh g <sup>-1</sup> at 10C	500 cycles		56

		network film						
Li-ion battery	Carbon-coated lithium iron phosphate	Carbon-coated lithium iron phosphate	1 M LiPF <sub>6</sub> in EC:DEC 1 : 1 by weight with an addition of 2 wt% VC	/	146 mA h g <sup>-1</sup> at C/10	50 cycles		57
Lithium-ion battery	TiO <sub>2</sub> /graphene/PVDF films	Lithium metal	1 M LiPF <sub>6</sub> solution in EC+DMC (1:1, v/v)	Celgard 2400	202mAh g <sup>-1</sup> at 60mA g <sup>-1</sup>	40 cycles		58
Lithium ion battery	Lithium wafer	N-CNT film	Common electrolyte (Lb <sub>3</sub> O <sub>3</sub> )	Polyethylene	390 mAh g <sup>-1</sup> at 4C	200 cycles at 4C		59
Lithium-ion battery	Li-foil	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> , carbonized cellulose nanofiber and carbon nanotubes hybrid film	1 m LiPF <sub>6</sub> in a 1:1 (v:v) mixture of EC and DEC	Celgard 2400	160 mAh g <sup>-1</sup> at 0.5C	1000 cycles		60
Lithium-ion battery	Lithium foils	PCNF@MoS <sub>2</sub> core/sheath fiber	LiPF <sub>6</sub> (1 M) in a mixture of EC, EMC and DMC (1/1/1 in v/v/v)	Celgard-2400	475 mA h g <sup>-1</sup> at 1 A g <sup>-1</sup>	50 cycles		61
Lithium ion battery	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> /carbon nanotube	LiFePO <sub>4</sub> /carbon nanotube	LiPF <sub>6</sub> electrolyte	Celgard 2400	220 mAh cm <sup>-2</sup> at 1000 mA cm <sup>-2</sup>	50 cycles at 1000 mA cm <sup>-2</sup>		62
Lithium ion battery	Metallic Li	Carbon nanotubes /Fe <sub>3</sub> O <sub>4</sub>	1.0 M LiPF <sub>6</sub> in EC/DC (1:1 by volume) with additional VC (2% by volume)	Glass fiber papers	837 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	100 cycles		63

			composite electrodes					
Lithium ion battery	$\text{SnO}_{\text{x}}\text{-ZnO}/\text{carbon nanofiber composites}$	Metallic Li sheet	1 M $\text{LiPF}_6$ in a solvent mixture of EC, DMC, and EMC (1:1:1 by volume)	Celgard 2400	1910 $\text{mA}\cdot\text{h}\cdot\text{g}^{-1}$ at 100 $\text{mA}\cdot\text{g}^{-1}$	55 cycles at 100 $\text{mA}\cdot\text{g}^{-1}$		<sup>64</sup>
Lithium-ion battery	Lithium pellets	Flexiblecnts@ $\text{TiO}_2$ hybrid mesoporous nanocables	The solution of $\text{LiPF}_6$ (1M) in EC/DMC	/	$\sim 210 \text{mAhg}^{-1}$ at 20C (1C = 170 $\text{mA}\text{g}^{-1}$ ),	1000 cycles at 20C		<sup>65</sup>
Lithium ion battery	Aligned CNT-based silicon films	Lithium ribbon	1 m $\text{LiPF}_6$ /EC + DMC + DEC (1:1:1 by volume)	/	3322 $\text{mA g}^{-1}$ at 50 $\text{mA g}^{-1}$	30 cycles		<sup>66</sup>
Lithium-ion battery	$\text{SnO}_2/\text{N-doped carbon nanofiber films}$	Li foil	1 M $\text{LiPF}_6$ in EC and DC solvent (1:1 v/v)	Celgard 2325	890.5 $\text{mAh g}^{-1}$ at 100 $\text{mA g}^{-1}$	300 cycles at 1 A g <sup>-1</sup>		<sup>67</sup>
Lithium ion battery	$\text{Fe}_2\text{O}_3$ nanoshells coated on carbonized bacterial cellulose nanofibers	Lithium metal	1.0 M $\text{LiPF}_6$ in EC and DC mixture (1:1 by volume)	Celgard 2250 film	1135 $\text{mAh g}^{-1}$ at 400mA g <sup>-1</sup>	200 cycles at 400mA g <sup>-1</sup>		<sup>68</sup>
Lithium ion	Li film	$\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$ /carbon	1.2 M solution of $\text{LiPF}_6$ in EC and DMC (3:7)	/	135 mAh/g at C/2 (1C=140	100cycles at C/2 (1C=140		<sup>69</sup>

battery		nanotube film			mAh/g)	mAh/g)	
Sodium ion battery	Na foil	$\text{Na}_{2+2x}\text{Fe}_{2x}(\text{SO}_4)_3$ @porous carbon nanofiber hybrid films	1 M $\text{NaClO}_4$ dissolved in propylene carbonate (PC)	/	$\sim 100 \text{ mAh g}^{-1}$ at $0.03\text{C}$ ( $97 \text{ mA g}^{-1}$ )	500 cycles at alternate rates of $40\text{C}$ and $5\text{C}$	<sup>70</sup>
Li-S battery	Lithium metal foil	Sulphur–carbon nanotubes	1.0 M LiTFSI in DOL and DME (1 : 1 by volume) with 0.1 wt% $\text{LiNO}_3$ additive	/	$712 \text{ mA h/g}_{\text{sulphur}}^{-1}$ (23 wt% S)	100 cycles	<sup>71</sup>
Li-S battery	Lithium metal	Sulfur@graphene nanoscroll composite	1.0 M LiTFSI dissolved in a mixed solvent of DOL and DME (1:1 by volume) with 0.1 m $\text{LiNiO}_3$ additive	Celgard 2400	$1377 \text{ mAh g}^{-1}$ at $0.05\text{C}$ ( $1\text{C} = 1675 \text{ mA g}^{-1}$ )	100 cycles	<sup>72</sup>

**Table S3. Summary of 3D (porous-type) flexible Batteries**

Battery type	Anode	Cathode	Electrolyte	Separator	Electrochemical performance			Ref.
					Discharge capacity	Cycling performance		
Lithium ion battery	Carbon nanotube /Si hybrid electrodes	Lithium foil	1 M $\text{LiPF}_6$ in a mixture of EC and DC (volume ratio: 1/1)	/	$2562 \text{ mA h g}^{-1}$ at $1 \text{ A g}^{-1}$	1000 cycles at $5 \text{ A g}^{-1}$		<sup>73</sup>

Lithium-ion battery	$\text{Fe}_3\text{O}_4$ -bacterial cellulose-carbon nanofiber electrode	Metallic lithium	1 m LiPF <sub>6</sub> in EC-DMC-EMC (1 : 1 : 1 by volume)	Celgard 2400	754 mA h g <sup>-1</sup>	100 cycles at 100 mA g <sup>-1</sup>	74
Li-ion battery	3D MWCNTs grown on Cu mesh	Li metal foil	1 M solution of LiPF <sub>6</sub> salt in 1:1 (v/v) mixture solvent of EC and DEC	Separator 2400	$\sim$ 250 mAhg <sup>-1</sup>	50 cycles	75
Lithium ion battery	Graphene and carbon nanotube hybrid foams	Lithium metal foil	1 M LiPF <sub>6</sub> in EC/DMC electrolyte solution	Celgard battery separator	2640 mA h g <sup>-1</sup> at 0.186 A g <sup>-1</sup> and 236 mA h g <sup>-1</sup> at 27.9 A g <sup>-1</sup>	300 cycles	76
Lithium-ion battery	Nanostructured $\text{TiO}_2$ supported on N-doped carbon foams	Metallic lithium sheets	/	/	188 mA h g <sup>-1</sup> at 200 mA g <sup>-1</sup>	100 cycles	77
Lithium-ion battery	Three-dimensional graphene network-supported polyimide	Al foil	1M LiTFSI DOL/DME solution	Glass fiber	175 mA h g <sup>-1</sup> at 0.1 C	150 cycles	78
Lithium-sulfur	Lithium foil	Reduced graphene	1 M LiTFSI and 0.1 M LiNO <sub>3</sub> in 1:1 (v/v) mixture of DME	Borosilicate glass	1098 mA h g <sup>-1</sup>	500 cycles	79

battery	oxide/Mn <sub>3</sub> O <sub>4</sub> @ hybrid polyaniline/sodium alginate matrix		and DOL	microfiber filters	at 2 A g <sup>-1</sup>		
Li-ion battery	3D ordered macroporous MoS <sub>2</sub> @C on carbon cloth	Lithium foil	1 M LiPF <sub>6</sub> in EC/DMC (1:1 in volume)	Celgard 2400	3.802 mAh cm <sup>-2</sup> at 0.1 mA cm <sup>-2</sup>	100 cycles	80
Lithium-ion battery	3D flexible Si based-composite/carbon nanofibers electrode	Lithium foil	1.1 M LiPF <sub>6</sub> in EC: DMC (1:1)	Celgard 2400	665 mAh g <sup>-1</sup> at 10 A g <sup>-1</sup>	2000 cycles	81
Li-ion battery	3D N-doped graphene foam with encapsulated germanium/N -doped graphene yolk-shell nanoarchitecture	Lithium foil	1 M solution of LiPF <sub>6</sub> in a volumetric ratio of 1:1:1 mixture of EC and DME	Porous polypropylene film	1220mAhg <sup>-1</sup>	1000 cycles	82
Lithium-ion battery	NiCo <sub>2</sub> S <sub>4</sub> nanotube arrays grown on flexible N-	Lithium foil	1 M LiPF <sub>6</sub> dissolved in a 1 : 1 (v/v) mixture of EC and DMC	Celgard 2400	1721 mA h g <sup>-1</sup> at 500 mA g <sup>-1</sup>	100 cycles	83

	doped carbon foams							
Lithium ion battery	3D interconnected carbon nanotube conducting polymer hydrogel network	Lithium discs	1 M LiPF <sub>6</sub> in EC/DC/VC (1:1:0.02 v/v/v)	Celgard 2500	2.2 mAh cm <sup>-2</sup> or 2180 mAh g <sup>-1</sup> at 0.1C rate	100 cycles		84
Lithium ion battery	Graphene foam loaded with Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	Graphene foam loaded with LiFePO <sub>4</sub>	LiPF <sub>6</sub> in EC/DMC	Polypropylene separator	~143 mAh/g at 0.2C	100 cycles		85
Lithium and sodium battery	Lithium foil	3D graphene/polyimide composites	1M LiPF <sub>6</sub> in EC/DMC (1 : 1 by volume ratio)	Cellgard 2400	240 mA h g <sup>-1</sup> at 40 mA g <sup>-1</sup>	1000 cycles		86
Lithium-ion battery	Graphene/Sn sandwich nanosheets	Lithium metal	LiPF <sub>6</sub> (1 M) in EC/DC/DC (1 : 1 : 1 vol%)	/	1487 mA h g <sup>-1</sup> at 0.2 A g <sup>-1</sup>	500 cycles		87
Lithium-ion and sodium-ion battery	NiCo <sub>2</sub> O <sub>4</sub> nanowire arrays	Lithium-foil/sodium-foil	1 M LiPF <sub>6</sub> in a mixture of EC/DEC/EMC (1 : 1 : 1 by volume)	Celgard 2400	507 mA h g <sup>-1</sup> at 4000 mA g <sup>-1</sup>	100 cycles		88
Ni/Fe	3D graphene foam/carbon	Graphene foam/carbon	6 m KOH	/	120 mAh g <sup>-1</sup> at	1000 cycles		89

battery	nanotubes/Ni(OH) <sub>2</sub> hybrid	nanotubes/Fe <sub>2</sub> O <sub>3</sub> hybrid film		0.3 A g <sup>-1</sup>			
Lithium-ion battery	3D SnO <sub>2</sub> @TiO <sub>2</sub> double-shell nanotubes on carbon cloth	Lithium foil	1 M solution of LiPF <sub>6</sub> in EC and DEC (1:1 by volume)	Microporous polypropylene membrane	778.8 mA h g <sup>-1</sup> at 780 mA g <sup>-1</sup>	100 cycles	90
Li-ion battery	3D ordered macroporous TiO <sub>2</sub> electrode	Li metal	7:3 (v/v) EC and DC electrolyte solution containing 1M LiPF <sub>6</sub>	Celgard 2500	174 mAh g <sup>-1</sup> at 2 A g <sup>-1</sup>	1000 cycles	91
Lithium-ion battery	3D carbon nanotubes on graphene/PET film	Lithium metal	1M solution of LiPF <sub>6</sub> salt in a solvent mixture of EC and DMC (1:1 (v/v))	Polyester-cellulose blend paper	0.25 mAh cm <sup>-2</sup> /397 mAh g <sup>-1</sup> at 0.1C	80 cycles	92
Li-ion battery	TiO <sub>2</sub> coated 3D carbon nanostructures	Li foil	1 M LiPF <sub>6</sub> dissolved in EC and DC (1:1 by volume) containing 5.0 vol% fluoro EC	/	351mA h g <sup>-1</sup> at 0.2C	8000 cycles	93
Lithium-ion battery	Cloth substrate supported Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> nanowall arrays	Self-standing porous LiMn <sub>2</sub> O <sub>4</sub> nanowall arrays	1 M LiPF <sub>6</sub> in EC/DEC (v/v=1:1) solution	Celgard 2400	124.8mAhg <sup>-1</sup> at 1C and 92.1 mAhg <sup>-1</sup> at 20C	200 cycles	86
Lithium ion	Sn- and SnO <sub>2</sub> -graphene	Lithium metal foil	1.2 M LiPF <sub>6</sub> in EC and DC1 : 1 (v/v)	/	1010 mA h g <sup>-1</sup> at 0.05 A g <sup>-1</sup> and (470 mA h	50 cycles	94

battery	flexible foams				$\text{g}^{-1}$ at 2 A $\text{g}^{-1}$ (~2 C)		
Solid-state zinc–air battery	Zinc powder	$\text{Co}_3\text{O}_4$ nanoparticles on carbon cloth	/	Conductive nanoporous cellulose membrane	$492 \text{ mA h g}^{-1}$ at $25 \text{ mA g}^{-1}$	over 2100 min	95
Solid lithium battery	Li metal foil	Aligned multiwalled carbon nanotubes and carbon nanohorns electrode	Poly(ethylene glycol) borate ester	/	352 mAh/g	/	11
Solid-state zinc–air battery	Zn electrode	N-doped $\text{Co}_3\text{O}_4$ mesoporous nanowire arrays	The gel electrolyte	NKK separator	$603.7 \text{ mAh g}^{-1}$ at $2.5 \text{ mA cm}^{-3}$	28 h	96
Lithium ion battery	$\text{SnO}_2$ nanosheets/Nickel/polyvinylidene fluoride ternary composite	Lithium metal foil	1 M $\text{LiPF}_6$ in EC and DEC (1:1 by volume) solution	Celgard 2400	$1533 \text{ mAh g}^{-1}$ at $200 \text{ mA g}^{-1}$	60 cycles	97
Li-ion battery	$\text{Li}_4\text{Ti}_5\text{O}_{12-\text{c}}$ nanotube arrays	$\text{LiCoO}_2$ on al foil	1 M $\text{LiPF}_6$ in EC/DEC (1:1 in volume)	/	$165 \text{ mAh g}^{-1}$ At 1C (1C = $175 \text{ mA g}^{-1}$ )	500 cycles at 10C	98

Sodium/sulfur battery	Sodium metal disc	Sulfurized polyacrylonitrile nanofiber web	1M NaPF <sub>6</sub> in EC and DEC (1:1 = v/v)	Glass fiber filter	604 mAh g <sup>-1</sup> at 0.01C	200 cycles at 0.1C	99
Lithium ion battery	Metallic lithium foils	Hybrid paper of Na <sub>2</sub> V <sub>6</sub> O <sub>16-x</sub> H <sub>2</sub> O nanowires and multi-wall carbon nanotubes or carbon black	1 MLiPF <sub>6</sub> in EC/DEC (1 : 1, volume ratio)	Celgard 2400	350 mA h g <sup>-1</sup> or 270 at mA h g <sup>-1</sup> 50 mA g <sup>-1</sup>	100 cycles at 150 mA g <sup>-1</sup>	100
Lithium-ion battery	Metal fibril mat-supported silicon	Li metal	1.15 M LiPF <sub>6</sub> in EC/EMC (30/70 by vol %) containing 5 wt % fluoroEC	Polyethylene	~3000 mAh g <sup>-1</sup> at 300 mA g <sup>-1</sup>	200 cycles at a 1C rate (2000 Ma g <sup>-1</sup> )	101
Lithium-ion battery	Lithium foil	V <sub>2</sub> O <sub>5</sub> nanobelt electrode	1 M LiPF <sub>6</sub> in EC-EMC-DMC (1:1:1,volumeratio)	Porous polypropylene film	127.4 mAhg <sup>-1</sup> at 60 mA g <sup>-1</sup>	200 cycles	102
Li-ion battery	MgCo <sub>2</sub> O <sub>4</sub> nanowires on Ni foam	Lithium foil	1 M LiPF <sub>6</sub> dissolved in a mixture of DMC, DC, and EC (1:1:1 by Volume)	Microporous polypropylene film	1357 mAh g <sup>-1</sup> at 2.5 A g <sup>-1</sup>	60 cycles at 2.5 A g <sup>-1</sup>	103
Li-ion battery	Carbon-Coated Germanium Nanowires on carbon	Li metal	1 M LiPF <sub>6</sub> in a mixture of EC and DEC (1:1 = w: w)	Celgard 2400	1440 mA h g <sup>-1</sup> at 0.1C	100 cycles at 0.1C	104

	nanofibers							
Li- Se/Na- Se battery	carbon- selenium composite nanofibers	Li (or Na) metal	1 M LiPF <sub>6</sub> /NaClO <sub>4</sub> in EC- DMC(1:1, in weight)	Glass fiber (Whatman)	1081 mAh g <sup>-1</sup> at 0.05 A g <sup>-1</sup>	500 cycles at 0.5 A g <sup>-1</sup> / 240 cycles at 0.5 A g <sup>-1</sup>		105
Li-ion battery	Carbon nanotubes – LiNBO <sub>3</sub> nanoplate – polypyrrole hybrid	Lithium metal	1 M LiPF <sub>6</sub> in 1:1 EC/DMC	Celgard 3501	300 mA h g <sup>-1</sup> at 0.4 A g <sup>-1</sup>	500 cycles		106
Lithium ion battery	Carbon nanotubes/Ti O <sub>2</sub> nanofibres composite electrodes	Lithium foil	1 M LiPF <sub>6</sub> in EC and DC (1:1 ratio)	/	200 mAh g <sup>-1</sup> at 16.7 mA/g	100 cycles		107
Lithium- ion battery	CoO– graphene– carbon nanofiber mats	Lithium foil	1 M LiPF <sub>6</sub> in EC–DC (1 : 1 by volume)	Celgard 2400	400 mA h g <sup>-1</sup> at 2 A g <sup>-1</sup>	352 cycles at 500 mA g <sup>-1</sup>		108
Lithium- ion battery	Lithium foil	Polyimide/SW CNT nanocable composite	1 M LiN(CF <sub>3</sub> SO <sub>2</sub> ) <sub>2</sub> (LiTFSI) solution in a mixed solvent of DOL and DME (1:1, in weight)	Celgard 2325	160 mAh g <sup>-1</sup> at 0.1 C	at 0.5 C for 200 cycles		109
Lithium ion	Lithium foil	Polyimide- single-walled carbon	1 m LiN(CF <sub>3</sub> SO <sub>2</sub> ) <sub>2</sub> (LiTFSI) solution dissolved in a mixed solvent of DOL and DME (1 :	Celgard 2325	190 mA h g <sup>-1</sup> at 20C	/		110

battery		nanotubes film	1, in weight)					
Lithium-ion battery	Tin nanoparticles embedded in carbon nanotubes on carbon cloth	Lithium metal	1M solution of LiPF <sub>6</sub> in EC and DEC (1:1 by vol.)	Polypropylene	2.15 mAhcm <sup>-2</sup> at 0.1 mAcm <sup>-2</sup>	80 cycles		<sup>111</sup>
Lithium-ion battery	Aligned CuO nanosheets on conductive carbon cloth	Li foil	1M LiPF <sub>6</sub> in a 50:50 w/w mixture of EC and DEC	Celgard 2400	711.2mAh g <sup>-1</sup> at 500mA g <sup>-1</sup>	100 cycles		<sup>112</sup>
Na-ion battery	Na <sub>2</sub> FeP <sub>2</sub> O <sub>7</sub> nanoparticles on porous carbon cloth	Na-metal foil	1 M NaClO <sub>4</sub> dissolved in a solution of EC and PC (volume ratio of 1 : 1)	Glass microfiber	95 mA h g <sup>-1</sup>	10 000 cycles		<sup>113</sup>
Lithium-ion battery	Carbon-coated silicon nanowires on carbon fabric	Lithium foils	1 M LiPF <sub>6</sub> in a 1:1 (v/v) mixture of EC and DMC with 5 wt% fluorinated EC	Celgard 2500	~3500 mA h g <sup>-1</sup> at 100 mA g <sup>-1</sup>	500 cycles at 1.0 A g <sup>-1</sup>		<sup>114</sup>
Na-O <sub>2</sub> battery	Co <sub>3</sub> O <sub>4</sub> nanowire arrays vertically grown onto carbon textiles	Sodium metal foil	0.5 M NaCF <sub>3</sub> SO <sub>3</sub> in TEGDME	Glass fiber	4687 mAh/g	62 cycles		<sup>38</sup>
Lithium-ion	Lithium metal	Activated cotton textile	1 M LiPF <sub>6</sub> in EC+DMC+DEC organic solvent (1:1:1 in	Celgard	~1710 mAh g <sup>-1</sup> at 1 C after			<sup>115</sup>

battery	with porous tubular fibers embedded with $\text{NiS}_2$ nanobowls and wrapped with conductive graphene sheets	volume)	2400	$^1$ at 0.01 C	100 cycles	
Lithium ion battery	$\text{Fe}_2\text{N}$ nanoparticles on carbon textile	Lithium foil	1M $\text{LiPF}_6$ in 1:1 by volume of EC/DMC	Celgard 2400	900mAh/g	300 cycles at 6000mA/g <sup>116</sup>
Lithium ion battery	Red P/hollow carbon cloth composite	Lithium metal foil	1M solution of $\text{LiPF}_6$ in EC and DC (1:1 in volume)	/	1063 mAh g <sup>-1</sup>	200 cycles <sup>117</sup>
Lithium-oxygen battery	Lithium belt	$\text{TiO}_2$ nanowire arrays grown onto carbon textiles	1M $\text{LiCF}_3\text{SO}_3$ in TEGDME electrolyte	Glass fibre	3,000 mAhg <sup>-1</sup> at 100 mA $\text{g}^{-1}$	1,000 cycles <sup>118</sup>
Lithium-ion battery	$\text{ZnCo}_2\text{O}_4$ nanowire arrays/carbon cloth	Lithium foil	1 M $\text{LiPF}_6$ in EC And DEC (1:1 by volume)	Celgard 2400	1300-1400 mAh g <sup>-1</sup>	160 cycles <sup>119</sup>
Lithium-oxygen battery	Li foil	Needle-like $\text{NiCo}_2\text{O}_4$ nanowire arrays	1 M LiTFSI in TEGDME electrolyte	Glass-fiber	4221 mAh g <sup>-1</sup>	200 cycles <sup>120</sup>

Lithium ion battery	$Zn_3P_2$ nanowire arrays grafted on carbon fabrics	Li foil	1 M LiPF <sub>6</sub> dissolved in the mixed solvents of EC and DEC (volume Ratio = 1 : 1)	Polymer–aluminum membranes	1200 mA h g <sup>-1</sup>	200 cycles	121
---------------------	--	---------	--	----------------------------	---------------------------	------------	-----

## References

1. J. Park, M. Park, G. Nam, J. s. Lee and J. Cho, *Adv. Mater.*, 2015, **27**, 1396-1401.
2. T. Liu, Q. C. Liu, J. J. Xu and X. B. Zhang, *Small*, 2016, **12**, 3101-3105.
3. M. Park, D. S. Shin, J. Ryu, M. Choi, N. Park, S. Y. Hong and J. Cho, *Adv. Mater.*, 2015, **27**, 5141-5146.
4. Y. Zhang, W. Bai, X. Cheng, J. Ren, W. Weng, P. Chen, X. Fang, Z. Zhang and H. Peng, *Angew. Chem.*, 2014, **126**, 14792-14796.
5. Y. h. Zhu, S. Yuan, D. Bao, Y. b. Yin, H. x. Zhong, X. b. Zhang, J. m. Yan and Q. Jiang, *Adv. Mater.*, 2017, **29**.
6. F. Meng, H. Zhong, D. Bao, J. Yan and X. Zhang, *J. Am. Chem. Soc.*, 2016, **138**, 10226-10231.
7. Y. H. Kwon, S. W. Woo, H. R. Jung, H. K. Yu, K. Kim, B. H. Oh, S. Ahn, S. Y. Lee, S. W. Song and J. Cho, *Adv. Mater.*, 2012, **24**, 5192-5197.
8. R. Liu, Y. Liu, J. Chen, Q. Kang, L. Wang, W. Zhou, Z. Huang, X. Lin, Y. Li and P. Li, *Nano Energy*, 2017.
9. W. G. Chong, J. Q. Huang, Z. L. Xu, X. Qin, X. Wang and J. K. Kim, *Adv. Funct. Mater.*, 2016.
10. X. Lin, Q. Kang, Z. Zhang, R. Liu, Y. Li, Z. Huang, X. Feng, Y. Ma and W. Huang, *J. Mater. Chem. A*, 2017, **5**, 3638-3644.
11. X. Yu, Y. Fu, X. Cai, H. Kafafy, H. Wu, M. Peng, S. Hou, Z. Lv, S. Ye and D. Zou, *Nano Energy*, 2013, **2**, 1242-1248.
12. A. M. Zamarayeva, A. M. Gaikwad, I. Deckman, M. Wang, B. Khau, D. A. Steingart and A. C. Arias, *Adv. Electron. Mater.*, 2016, **2**.
13. K. M. Oh, S.-W. Cho, G.-O. Kim, K.-S. Ryu and H. M. Jeong, *Electrochim. Acta*, 2014, **135**, 478-486.
14. G. Ning, C. Xu, Y. Cao, X. Zhu, Z. Jiang, Z. Fan, W. Qian, F. Wei and J. Gao, *J. Mater. Chem. A*, 2013, **1**, 408-414.
15. D. Y. Kim, M. Kim, D. W. Kim, J. Suk, O. O. Park and Y. Kang, *Carbon*, 2015, **93**, 625-635.
16. X. Huang, B. Sun, K. Li, S. Chen and G. Wang, *J. Mater. Chem. A*, 2013, **1**, 13484-13489.
17. C. Ma, C. Ma, J. Wang, H. Wang, J. Shi, Y. Song, Q. Guo and L. Liu, *Carbon*, 2014, **72**, 38-46.
18. T. Cetinkaya, S. Ozcan, M. Uysal, M. O. Guler and H. Akbulut, *J. Power Sources*, 2014, **267**, 140-147.
19. Q. C. Liu, L. Li, J. J. Xu, Z. W. Chang, D. Xu, Y. B. Yin, X. Y. Yang, T. Liu, Y. S. Jiang and J. M. Yan, *Adv. Mater.*, 2015, **27**, 8095-8101.
20. K. Rana, S. D. Kim and J.-H. Ahn, *Nanoscale*, 2015, **7**, 7065-7071.
21. Y. Sun, C. Wang, Y. Xue, Q. Zhang, R. G. Mendes, L. Chen, T. Zhang, T. Gemming, M. H. Rümmeli and X. Ai, *ACS Appl. Mater. Interfaces*, 2016, **8**, 9185-9193.
22. Y. Liu, W. Wang, L. Gu, Y. Wang, Y. Ying, Y. Mao, L. Sun and X. Peng, *ACS Appl. Mater. Interfaces*, 2013, **5**, 9850-9855.
23. C. Y. Su, H. Cheng, W. Li, Z. Q. Liu, N. Li, Z. Hou, F. Q. Bai, H. X. Zhang and T. Y. Ma, *Adv. Energy Mater.*, 2017.
24. N. Li, G. Zhou, F. Li, L. Wen and H. M. Cheng, *Adv. Funct. Mater.*, 2013, **23**, 5429-5435.

25. S. D. Kim, K. Rana and J.-H. Ahn, *J. Mater. Chem. A*, 2016, **4**, 19197-19206.
26. K. Rana, J. Singh, J.-T. Lee, J. H. Park and J.-H. Ahn, *ACS Appl. Mater. Interfaces*, 2014, **6**, 11158-11166.
27. Y.-T. Liu, X.-D. Zhu, Z.-Q. Duan and X.-M. Xie, *Chem. Commun.*, 2013, **49**, 10305-10307.
28. W. Zhang, Y. Liu, C. Chen, Z. Li, Y. Huang and X. Hu, *Small*, 2015, **11**, 3822-3829.
29. L. Mei, C. Xu, T. Yang, J. Ma, L. Chen, Q. Li and T. Wang, *J. Mater. Chem. A*, 2013, **1**, 8658-8664.
30. W. Qiu, J. Jiao, J. Xia, H. Zhong and L. Chen, *Chem-Eur. J.*, 2015, **21**, 4359-4367.
31. P. Kumar, F.-Y. Wu, L.-H. Hu, S. A. Abbas, J. Ming, C.-N. Lin, J. Fang, C.-W. Chu and L.-J. Li, *Nanoscale*, 2015, **7**, 8093-8100.
32. C. Wu, L. Fu, J. Maier and Y. Yu, *J. Mater. Chem. A*, 2015, **3**, 9438-9445.
33. J. Cao, C. Chen, Q. Zhao, N. Zhang, Q. Lu, X. Wang, Z. Niu and J. Chen, *Adv. Mater.*, 2016, **28**, 9629-9636.
34. J. Balach, H. K. Singh, S. Gomoll, T. Jaumann, M. Klose, S. Oswald, M. Richter, J. r. Eckert and L. Giebelser, *ACS Appl. Mater. Interfaces*, 2016, **8**, 14586-14595.
35. H. Zhang, S. Jing, Y. Hu, H. Jiang and C. Li, *J. Power Sources*, 2016, **307**, 214-219.
36. G. Zhou, Y. Zhao and A. Manthiram, *Adv. Energy Mater.*, 2015, **5**.
37. F. Wu, E. Zhao, D. Gordon, Y. Xiao, C. Hu and G. Yushin, *Adv. Mater.*, 2016, **28**, 6365-6371.
38. J. Zhang, N. Zhao, M. Zhang, Y. Li, P. K. Chu, X. Guo, Z. Di, X. Wang and H. Li, *Nano Energy*, 2016, **28**, 447-454.
39. S. Leijonmarck, A. Cornell, G. Lindbergh and L. Wågberg, *Nano Energy*, 2013, **2**, 794-800.
40. X. Ma, X. Huang, J. Gao, S. Zhang, Z. Deng and J. Suo, *Electrochim. Acta*, 2014, **115**, 216-222.
41. S. Sen, D. Dutta and A. J. Bhattacharyya, *J. Mater. Chem. A*, 2015, **3**, 20958-20965.
42. S.-H. Kim, K.-H. Choi, S.-J. Cho, J.-S. Park, K. Y. Cho, C. K. Lee, S. B. Lee, J. K. Shim and S.-Y. Lee, *J. Mater. Chem. A*, 2014, **2**, 10854-10861.
43. S. Suriyakumar, M. Raja, N. Angulakshmi, K. S. Nahm and A. M. Stephan, *RSC Adv.*, 2016, **6**, 92020-92027.
44. J. H. Kim, J. H. Kim, J. M. Kim, Y. G. Lee and S. Y. Lee, *Adv. Energy Mater.*, 2015, **5**.
45. M.-S. Balogun, Z. Wu, Y. Luo, W. Qiu, X. Fan, B. Long, M. Huang, P. Liu and Y. Tong, *J. Power Sources*, 2016, **308**, 7-17.
46. B. Sun, S. J. Bao, J. Le Xie and C. M. Li, *RSC Adv.*, 2014, **4**, 36815-36820.
47. M.-H. Park, M. Noh, S. Lee, M. Ko, S. Chae, S. Sim, S. Choi, H. Kim, H. Nam and S. Park, *Nano Lett.*, 2014, **14**, 4083-4089.
48. M. Koo, K.-I. Park, S. H. Lee, M. Suh, D. Y. Jeon, J. W. Choi, K. Kang and K. J. Lee, *Nano Lett.*, 2012, **12**, 4810-4816.
49. E. Mills, J. Cannarella, Q. Zhang, S. Bhadra, C. B. Arnold and S. Y. Chou, *Journal of Vacuum Science & Technology B, Nanotechnology and Microelectronics: Materials, Processing, Measurement, and Phenomena*, 2014, **32**, 06FG10.
50. L. Jabbour, M. Destro, C. Gerbaldi, D. Chaussy, N. Penazzi and D. Beneventi, *J. Mater. Chem.*, 2012, **22**, 3227-3233.
51. G. Zhou, L. Li, D. W. Wang, X. y. Shan, S. Pei, F. Li and H. M. Cheng, *Adv. Mater.*, 2015, **27**, 641-647.
52. H. Wu, S. A. Shevlin, Q. Meng, W. Guo, Y. Meng, K. Lu, Z. Wei and Z. Guo, *Adv. Mater.*, 2014, **26**, 3338-3343.
53. M. Kammoun, S. Berg and H. Ardebili, *Nanoscale*, 2015, **7**, 17516-17522.
54. H. Lee, J. K. Yoo, J. H. Park, J. H. Kim, K. Kang and Y. S. Jung, *Adv. Energy Mater.*, 2012, **2**, 976-982.
55. J. Fu, D. U. Lee, F. M. Hassan, L. Yang, Z. Bai, M. G. Park and Z. Chen, *Adv. Mater.*, 2015, **27**, 5617-5622.
56. S. Cao, X. Feng, Y. Song, X. Xue, H. Liu, M. Miao, J. Fang and L. Shi, *ACS Appl. Mater. Interfaces*, 2015, **7**, 10695-10701.

57. S. Leijonmarck, A. Cornell, G. Lindbergh and L. Wågberg, *J. Mater. Chem. A*, 2013, **1**, 4671-4677.
58. H. Ren, Y. Ding, F. Chang, X. He, J. Feng, C. Wang, Y. Jiang and P. Zhang, *Appl. Surf. Sci.*, 2012, **263**, 54-57.
59. Z. Pan, J. Ren, G. Guan, X. Fang, B. Wang, S. G. Doo, I. H. Son, X. Huang and H. Peng, *Adv. Energy Mater.*, 2016, **6**.
60. S. Cao, X. Feng, Y. Song, H. Liu, M. Miao, J. Fang and L. Shi, *ACS Appl. Mater. Interfaces*, 2016, **8**, 1073-1079.
61. Y.-E. Miao, Y. Huang, L. Zhang, W. Fan, F. Lai and T. Liu, *Nanoscale*, 2015, **7**, 11093-11101.
62. Y. Wu, H. Wu, S. Luo, K. Wang, F. Zhao, Y. Wei, P. Liu, K. Jiang, J. Wang and S. Fan, *RSC Adv.*, 2014, **4**, 20010-20016.
63. Y. Cheng, Z. Chen, M. Zhu and Y. Lu, *Adv. Energy Mater.*, 2015, **5**.
64. B. N. Joshi, S. An, H. S. Jo, K. Y. Song, H. G. Park, S. Hwang, S. S. Al-Deyab, W. Y. Yoon and S. S. Yoon, *ACS Appl. Mater. Interfaces*, 2016, **8**, 9446-9453.
65. Y. Liu, A. A. Elzatahry, W. Luo, K. Lan, P. Zhang, J. Fan, Y. Wei, C. Wang, Y. Deng and G. Zheng, *Nano Energy*, 2016, **25**, 80-90.
66. K. Fu, O. Yildiz, H. Bhanushali, Y. Wang, K. Stano, L. Xue, X. Zhang and P. D. Bradford, *Adv. Mater.*, 2013, **25**, 5109-5114.
67. L. Xia, S. Wang, G. Liu, L. Ding, D. Li, H. Wang and S. Qiao, *Small*, 2016, **12**, 853-859.
68. Y. Huang, Z. Lin, M. Zheng, T. Wang, J. Yang, F. Yuan, X. Lu, L. Liu and D. Sun, *J. Power Sources*, 2016, **307**, 649-656.
69. X. Fang, C. Shen, M. Ge, J. Rong, Y. Liu, A. Zhang, F. Wei and C. Zhou, *Nano Energy*, 2015, **12**, 43-51.
70. T. Yu, B. Lin, Q. Li, X. Wang, W. Qu, S. Zhang and C. Deng, *Phys. Chem. Chem. Phys.*, 2016, **18**, 26933-26941.
71. G. Zhou, D.-W. Wang, F. Li, P.-X. Hou, L. Yin, C. Liu, G. Q. M. Lu, I. R. Gentle and H.-M. Cheng, *Energy Environ. Sci.*, 2012, **5**, 8901-8906.
72. Y. Guo, G. Zhao, N. Wu, Y. Zhang, M. Xiang, B. Wang, H. Liu and H. Wu, *ACS Appl. Mater. Interfaces*, 2016.
73. W. Weng, H. Lin, X. Chen, J. Ren, Z. Zhang, L. Qiu, G. Guan and H. Peng, *J. Mater. Chem. A*, 2014, **2**, 9306-9312.
74. Y. Wan, Z. Yang, G. Xiong and H. Luo, *J. Mater. Chem. A*, 2015, **3**, 15386-15393.
75. C. Kang, R. Baskaran, J. Hwang, B.-C. Ku and W. Choi, *Carbon*, 2014, **68**, 493-500.
76. A. P. Cohn, L. Oakes, R. Carter, S. Chatterjee, A. S. Westover, K. Share and C. L. Pint, *Nanoscale*, 2014, **6**, 4669-4675.
77. S. Chu, Y. Zhong, R. Cai, Z. Zhang, S. Wei and Z. Shao, *Small*, 2016, **12**, 6724-6734.
78. Y. Meng, H. Wu, Y. Zhang and Z. Wei, *J. Mater. Chem. A*, 2014, **2**, 10842-10846.
79. A. Ghosh, R. Manjunatha, R. Kumar and S. Mitra, *ACS Appl. Mater. Interfaces*, 2016, **8**, 33775-33785.
80. Z. Deng, H. Jiang, Y. Hu, Y. Liu, L. Zhang, H. Liu and C. Li, *Adv. Mater.*, 2017.
81. S.-J. Kim, M.-C. Kim, S.-B. Han, G.-H. Lee, H.-S. Choe, D.-H. Kwak, S.-Y. Choi, B.-G. Son, M.-S. Shin and K.-W. Park, *Nano Energy*, 2016, **27**, 545-553.
82. R. Mo, D. Rooney, K. Sun and H. Y. Yang, *Nat. Commun.*, 2017, **8**.
83. X. Wu, S. Li, B. Wang, J. Liu and M. Yu, *Phys. Chem. Chem. Phys.*, 2016, **18**, 4505-4512.
84. G. Lui, G. Li, X. Wang, G. Jiang, E. Lin, M. Fowler, A. Yu and Z. Chen, *Nano Energy*, 2016, **24**, 72-77.
85. C. Kang, E. Cha, R. Baskaran and W. Choi, *Nanotechnology*, 2016, **27**, 105402.
86. X. H. Wang, C. Guan, L. M. Sun, R. A. Susantyoko, H. J. Fan and Q. Zhang, *J. Mater. Chem. A*, 2015, **3**, 15394-15398.
87. H. Xia, Q. Xia, B. Lin, J. Zhu, J. K. Seo and Y. S. Meng, *Nano Energy*, 2016, **22**, 475-482.
88. C. Botas, D. Carriazo, G. Singh and T. Rojo, *J. Mater. Chem. A*, 2015, **3**, 13402-13410.

89. K. K. Fu, Y. Gong, J. Dai, A. Gong, X. Han, Y. Yao, C. Wang, Y. Wang, Y. Chen and C. Yan, *Proceedings of the National Academy of Sciences*, 2016, **113**, 7094-7099.
90. W. Li, L. Yang, J. Wang, B. Xiang and Y. Yu, *ACS Appl. Mater. Interfaces*, 2015, **7**, 5629-5633.
91. Z. Chen, J. W. To, C. Wang, Z. Lu, N. Liu, A. Chortos, L. Pan, F. Wei, Y. Cui and Z. Bao, *Adv. Energy Mater.*, 2014, **4**.
92. N. Li, Z. Chen, W. Ren, F. Li and H.-M. Cheng, *Proceedings of the National Academy of Sciences*, 2012, **109**, 17360-17365.
93. Y. Huang, K. Li, J. Liu, X. Zhong, X. Duan, I. Shakir and Y. Xu, *J. Mater. Chem. A*, 2017.
94. Y. Mo, Q. Ru, J. Chen, X. Song, L. Guo, S. Hu and S. Peng, *J. Mater. Chem. A*, 2015, **3**, 19765-19773.
95. J. Fu, J. Zhang, X. Song, H. Zarrin, X. Tian, J. Qiao, L. Rasen, K. Li and Z. Chen, *Energy Environ. Sci.*, 2016, **9**, 663-670.
96. M. Yu, Z. Wang, C. Hou, Z. Wang, C. Liang, C. Zhao, Y. Tong, X. Lu and S. Yang, *Adv. Mater.*, 2017, **29**.
97. Y. Zhang, Q. Xiao, G. Lei, Z. Li and X. Li, *Electrochim. Acta*, 2015, **178**, 336-343.
98. J. Liu, K. Song, P. A. van Aken, J. Maier and Y. Yu, *Nano Lett.*, 2014, **14**, 2597-2603.
99. I. Kim, C. H. Kim, S. hwa Choi, J.-P. Ahn, J.-H. Ahn, K.-W. Kim, E. J. Cairns and H.-J. Ahn, *J. Power Sources*, 2016, **307**, 31-37.
100. W. Zhang, G. Xu, L. Yang and J. Ding, *RSC Adv.*, 2016, **6**, 5161-5168.
101. S. Song, S. W. Kim, D. J. Lee, Y.-G. Lee, K. M. Kim, C.-H. Kim, J.-K. Park, Y. M. Lee and K. Y. Cho, *ACS Appl. Mater. Interfaces*, 2014, **6**, 11544-11549.
102. P.-P. Wang, Y.-X. Yao, C.-Y. Xu, L. Wang, W. He and L. Zhen, *Ceram. Int.*, 2016, **42**, 14595-14600.
103. X. Wang, G. Zhai and H. Wang, *J. Nanopart. Res.*, 2015, **17**, 339.
104. W. Li, M. Li, Z. Yang, J. Xu, X. Zhong, J. Wang, L. Zeng, X. Liu, Y. Jiang and X. Wei, *Small*, 2015, **11**, 2762-2767.
105. L. Zeng, X. Wei, J. Wang, Y. Jiang, W. Li and Y. Yu, *J. Power Sources*, 2015, **281**, 461-469.
106. Q. Fan, L. Lei, G. Yin and Y. Sun, *Chem. Commun.*, 2014, **50**, 2370-2373.
107. P. Zhang, J. Qiu, Z. Zheng, G. Liu, M. Ling, W. Martens, H. Wang, H. Zhao and S. Zhang, *Electrochim. Acta*, 2013, **104**, 41-47.
108. M. Zhang, F. Yan, X. Tang, Q. Li, T. Wang and G. Cao, *J. Mater. Chem. A*, 2014, **2**, 5890-5897.
109. H. Wu, Q. Meng, Q. Yang, M. Zhang, K. Lu and Z. Wei, *Adv. Mater.*, 2015, **27**, 6504-6510.
110. H. Wu, Q. Yang, Q. Meng, A. Ahmad, M. Zhang, L. Zhu, Y. Liu and Z. Wei, *J. Mater. Chem. A*, 2016, **4**, 2115-2121.
111. W. Ren, D. Kong and C. Cheng, *ChemElectroChem*, 2014, **1**, 2064-2069.
112. S. Cheng, T. Shi, X. Tao, Y. Zhong, Y. Huang, J. Li, G. Liao and Z. Tang, *Electrochim. Acta*, 2016, **212**, 492-499.
113. H. J. Song, D.-S. Kim, J.-C. Kim, S.-H. Hong and D.-W. Kim, *J. Mater. Chem. A*, 2017, **5**, 5502-5510.
114. X. Wang, G. Li, M. H. Seo, G. Lui, F. M. Hassan, K. Feng, X. Xiao and Z. Chen, *ACS Appl. Mater. Interfaces*, 2017, **9**, 9551-9558.
115. Z. Gao, N. Song, Y. Zhang and X. Li, *Nano Lett.*, 2015, **15**, 8194-8203.
116. M.-S. Balogun, M. Yu, Y. Huang, C. Li, P. Fang, Y. Liu, X. Lu and Y. Tong, *Nano Energy*, 2015, **11**, 348-355.
117. Y. Du, Y. Tang and C. Chang, *J. Electrochem. Soc.*, 2016, **163**, A2938-A2942.
118. Q.-C. Liu, J.-J. Xu, D. Xu and X.-B. Zhang, *Nat. Commun.*, 2015, **6**.
119. B. Liu, J. Zhang, X. Wang, G. Chen, D. Chen, C. Zhou and G. Shen, *Nano Lett.*, 2012, **12**, 3005-3011.
120. H. Xue, S. Wu, J. Tang, H. Gong, P. He, J. He and H. Zhou, *ACS Appl. Mater. Interfaces*, 2016, **8**, 8427-8435.

121. W. Li, L. Gan, K. Guo, L. Ke, Y. Wei, H. Li, G. Shen and T. Zhai, *Nanoscale*, 2016, **8**, 8666-8672.