Supplementary Information

MOF derived Porous N-Co₃O₄@N-C Nanopolyhedra Wrapped With Reduced Graphene Oxide as High Capacity Cathodes for Lithium-Sulfur Batteries

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Figure S1 Chemical equation for the synthesis of ZIF-67.



Figure S2 (a, b) SEM image of ZIF-67



Figure S3 SEM images of (a, b) N-Co₃O₄@N-C.



Figure S4 Linear distributions of C, N, O and Co along the arrow line on the inset image of

N-Co₃O₄@N-C.



Figure S5 Co 2p spectrum of the $N-Co_3O_4$ (a) N-C composite.



Figure S6 SEM images of (a) N-Co₃O₄@N-C/GO and (b) N-Co₃O₄@N-C/rGO-S.



Figure S7 TEM images of (a) N-Co₃O₄@N-C/GO and (b) N-Co₃O₄@N-C/rGO-S.



Figure S8 (a) SEM image of N-Co₃O₄@N-C/rGO–S and the corresponding elemental mapping images of (b) Overall image (c) Carbon, (d) Sulfur, (e) Nitrogen, (f) Cobalt.



Figure S9 XRD patterns of N-Co₃O₄@N-C/rGO-S, pure sulfur.



Figure S10 Nitrogen adsorption-desorption isotherm curves of N-Co₃O₄@N-C/rGO-S.



Figure S11 XPS survey spectra of N-Co₃O₄@N-C/rGO-S, comfirming the existence of cobalt, oxygen, carbon, sulfur, nitrogen.



Figure S12 XPS survey spectra of N-Co₃O₄@N-C/rGO-S, comfirming the existence of cobalt, oxygen, carbon, sulfur, nitrogen.



Figure S13 The equivalent circuit of electrochemical impedance spectra in Figure 3b.

Electrode	R _e (ohm)	R _{ct} (ohm)	$Y1(\Omega^{-1}cm^{-2}s^{-n})$	nl
N-Co ₃ O ₄ @N-C/rGO-S	4.425	38.763	0.000035	0.79
N-Co ₃ O ₄ @N-C-S	4.578	41.687	0.000053	
rGO-S	4.08	52.732	0.000085	

Table S1 Parameters identification by modeling the impedance spectra in Figure 3b.



Figure S14 (a) The pressed N-Co₃O₄@N-C and Co₃O₄@C disks. (b) Top view of the asprepared disk. (c) Side view of the as-prepared disk. The cross-sectional view of (d) N-Co₃O₄@N-C disk and (e) Co₃O₄@C disks. (f ,g) Zhuhai Kaivo FP-001 Four-Point Probe Resistance Tester.

First, we pressed the N-Co₃O₄@N-C and Co₃O₄@C powder into disks, respectively. It should note that the Co₃O₄@C powder is made of commercial cobalt oxide mixed with the carbon black at the weight ratio of 9:1. We calculated the electronic conductivity of each sample based on the following equation: The electronic conductivity for the N-Co₃O₄@N-C disks:

$$\rho = R_{\Omega} \times W = R_x \times F(D/S) \times F(W/S) \times F_{sp} \times W$$

=0.1976 ×10³ Ω × F(14/1) × F(0.334/1) × 1.004 × 3.34×10⁻⁶ m

=0.1976 ×10³ Ω × 4.348 ×1 × 1.004 × 3.34×10⁻⁶ m

$=2.881 \times 10^{-3} \Omega m$

 $\sigma = 1/\rho = 1/(2.881 \times 10^{-3}) \Omega^{-1} m^{-1} = 347.1 S m^{-1}$

The electronic conductivity for the Co_3O_4 (*Q* C disks:

 $\rho = R_{\Omega} \times W = R_x \times F(D/S) \times F(W/S) \times F_{sp} \times W$

=0.3454 ×10³ Ω × F(14/1) × F(0.269/1) × 1.004 × 2.69×10⁻⁶ m

= $0.3454 \times 10^{3} \Omega \times 4.348 \times 1 \times 1.004 \times 2.69 \times 10^{-6} m$

 $=4.056 \times 10^{-3} \Omega m$

 $\sigma = 1/\rho = 1/(4.056 \times 10^{-3}) \Omega^{-1} m^{-1} = 246.5 S m^{-1}$

In the formular:

D: The diameter of the sample, in mm.

S: The distance between two neighbouring probes. The value S=1 mm for the four-point probe we use.

W: The thickness of the conductive layer, in mm.

 F_{sp} : The probe distance correction factor. The value F_{sp} = 1.004 for the four-point probe we use.

F(D/S): The diameter correction factor. This factor can be chosen from the work brochure (**Table S2**) according to the value of D/S.

F(W/S): The thickness correction factor. This factor can be chosen from the work brochure (**Table S3**) according to the value of W/S.

R_x: The testing result from the tester.

 ρ : The electrical resistivity.

 σ : The electronic conductivity.

Therefore, the electronic conductivity of N-Co₃O₄@N-C and Co₃O₄@C powder are 347.1 and 246.5 S m⁻¹, respectively. It also further confirms that nitrogen doping promotes electron conductivity of the material.

The testing position F (D/S) D/S Value	centre point	In the middle of radius	6mm from the edge
>200	4.532		and entry states
200	4.531	4.531	4.462
150	4.531	4.529	4.461
125	4.530	4.528	4.460
100	4.528	4.525	4.458
76	4.526	4.520	4.455
60	4.521	4.513	4.451
51	4.517	4.505	4.447
38	4.505	4.485	4.439
26	4.470	4.424	4.418
25	4.470	C S SUEN N'S N	
22.22	4.454		Contractor (Barris
20.00	4.436		
18.18	4.417		
16.67	4.395		
15.38	4.372		
14.28	4.348	Long La	Contraction of the local division of the loc
13.33	4.322	Long and	C. Station
12.50	4.294	Test and	
11.76	4.265	A REPAIR	
11.11	4.235		1000
10.52	4.204	1	
10.00	4.171		

Table S2 The relationship between F(D/S) and D/S.

W/S	F (W/S)	W/S	F (W/S)	W/S	F (W/S)	W/S	F (W/S)
<0.400	1.0000	0.530	0.9962	0.665	0.9858	0.800	0.9663
0.400	0.9997	0.535	0.9960	0.670	0.9853	0.805	0.9654
0.405	0.9996	0.540	0.9957	0.675	0.9847	0.810	0.9644
0.410	0.9996	0.545	0.9955	0.680	0.9841	0.815	0.9635
0.415	0.9995	0.550	0,9952	0.685	0.9835	0.820	0.9626
0.420	0.9994	0.555	0.9949	0.690	0.9829	0.825	0.9616
0.425	0.9993	0.560	0.9946	0.695	0.9823	0.830	0.9607
0.430	0.9993	0.565	0.9943	0.700	0.9817	0.835	0.9597
0.435	0.9992	0.570	0.9940	0.705	0.9810	0.840	0.9587
0.440	0.9991	0.575	0.9937	0.710	0.9804	0.845	0.9577
0.445	0.9990	0.580	0.9934	0.715	0.9797	0.850	0.9567
0.450	0.9989	0.585	0.9930	0.720	0.9790	0.855	0.9557
0.455	0.9988	0.590	0.9927	0.725	0.9783	0.860	0.9546
0.460	0.9987	0.595	0.9923	0.730	0.9776	0.865	0.9536
0.465	0.9985	0.600	0.9919	0.735	0.9769	0.870	0.9525
0.470	0.9984	0.605	0.9915	0.740	0.9761	0.875	0.9514
0.475	0.9983	0.610	0.9911	0.745	0.9754	0.880	0.9504
0.480	0.9981	0.615	0.9907	0.750	0.9746	0.885	0.9493
0.485	0.9980	0.620	0.9903	0.755	0.9738	0.890	0.9482
0.490	0.9978	0.625	0.9898	0.760	0.9731	0.895	0.9471
0.495	0.9976	0.630	0.9894	0.765	0.9723	0.900	0.9459
0.500	0.9975	0.635	0.9889	0.770	0.9714	0.905	0.9448
0.505	0.9973	0.640	0.9884	0.775	0.9706	0.910	0.9437
0.510	0.9971	0.645	0.9879	0.780	0.9698	0.915	0.9425
0.515	0.9969	0.650	0.9874	0.785	0.9689	0.920	0.9413
0.520	0.9967	0.655	0.9869	0.790	0.9680	0.925	0.9402
0.525	0.9965	0.660	0.9864	0.795	0.9672	0.930	0.9390
0.935	0.9378	1.10	0.8939	1.85	0.6718	2.60	0.5098
0.940	0.9366	1.15	0.8793	1.90	0.6588	2.65	0.5013
0.945	0.9354	1.20	0.8643	1.95	0.6460	2.70	0.4931
0.950	0.9342	1.25	0.8491	2.00	0.6337	2.75	0.4851
0.955	0.9329	1.30	0.8336	2.05	0.6216	2.80	0.4773
0.960	0.9317	1.35	0.8181	2.10	0.6099	2.85	0.4698
0.965	0.9304	1.40	0.8026	2.15	0.5986	2.90	0.4624
0.070	0.0202	1.45	0.7872	2.20	0.5875	2.05	0.4552
0.970	0.9292	1.50	0.7710	2.25	0.5767	2.95	0.4555
0.975	0.9279	1.50	0.7719	2.25	0.5767	3.00	0.4484
0.980	0.9267	1.55	0.7568	2.30	0.5663	3.2	0.422
0.985	0.9254	1.60	0.7419	2.35	0.5562	3.4	0.399
0.990	0.9241	1.65	0.7273	2.40	0.5464	3.6	0.378
0.995	0.9228	1.70	0.7130	2.45	0.5368	3.8	0.359
1.00	0.9215	1.75	0.6989	2.50	0.5275	4.0	0.342

Table S3 The relationship between F(W/S) and W/S.







Figure S16 The galvanostatic charge/discharge voltage profiles of N-Co₃O₄@N-C-S, rGO-S cathodes at various current densities from 0.1 C to 3 C.



Figure S17 Nyquist plots of N-Co₃O₄@N-C/rGO-S with the sulfur loading of 2.13 mg cm⁻², 5.89 mg cm⁻² cathode electrodes.



Figure S18 (a) Dissembled electrode of rGO-S after 50 cycles. (b) Dissembled electrode of $N-Co_3O_4@N-C/rGO-S$ after 50 cycles.