

Supporting Information

A novel strategy to significantly enhance initial voltage and suppress voltage fading of Li- and Mn-rich Layered Oxide Cathode Material for lithium-ion batteries†

Shiming Zhang,^a Jian Chen,^b Tian Tang,^a Yinzhu Jiang,^a Gairong Chen,^c Qinong Shao,^a Chenhui Yan,^a Tiejun Zhu,^a Mingxia Gao,^a Yongfeng Liu,^a Hongge Pan^{*,a}

[†]*State Key Laboratory of Silicon Materials, School of Materials Science and Engineering, Zhejiang University, Hangzhou, 310027, P.R. China.*

[‡]*School of Materials Science and Chemical Engineering, Xi'an Technological University, Xi'an, 710021, P.R. China.*

[§]*College of Chemistry and Chemical Engineering of Xinxiang University, Henan, 453003, P.R. China.*

XRD refinement process: In order to further explore the crystal structure of the $\text{Li}[\text{Li}_{0.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}]\text{O}_2\text{-xLiNiO}_2$ material with varying x, the Rietveld refinements of the XRD data of the $\text{LiNiO}_2\text{-0}$, $\text{LiNiO}_2\text{-10}$, $\text{LiNiO}_2\text{-20}$ and $\text{LiNiO}_2\text{-40}$ samples were carried out by a General Structure Analysis System (GSAS) software package. The Rietveld refinements of the XRD data are based on a three-phase model system with the trigonal layered $\text{LiMO}_2(\text{R-3m})$ phase, monoclinic layered $\text{Li}_2\text{MnO}_3(\text{C2/m})$ phase and spinel-like LiM_2O_4 phase. Due to the similar ionic radii of Ni^{2+} with the Li^+ , we performed the Ni^{2+} to occupy the Li^+ sites, and the goal of this refinement is in order to get possible sites of Ni^{2+} substituting the Li^+ sites and the corresponding qualitative value of Ni/Li occupancies in the Li slab and metal slab. This refinement on the Ni/Li occupancies was carried out for both the R-3m space group, C2/m space group and Fd-3m. Firstly, we set the theory occupancies as 0.000% of Ni^{2+} in the Li^+ 3b sites of R-3m space group, in the Li^+ 2b, Li^+ 2c and Li^+ 4h sites for the C2/m space group and in the Li^+ 8a sites of the Fd-3m space group. Then the real occupancies were refined step by step until they reached a reliable value.

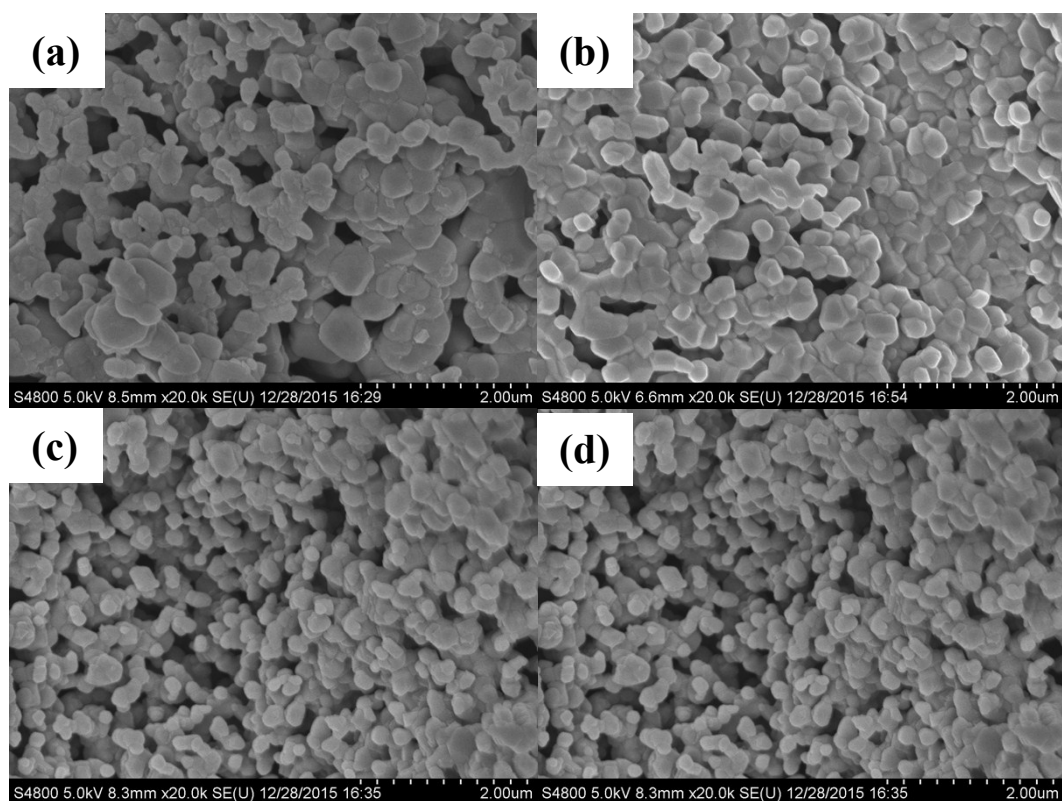


Fig. S1 SEM images of the LiNiO-0, LiNiO-10, LiNiO-20 and LiNiO-40 samples.

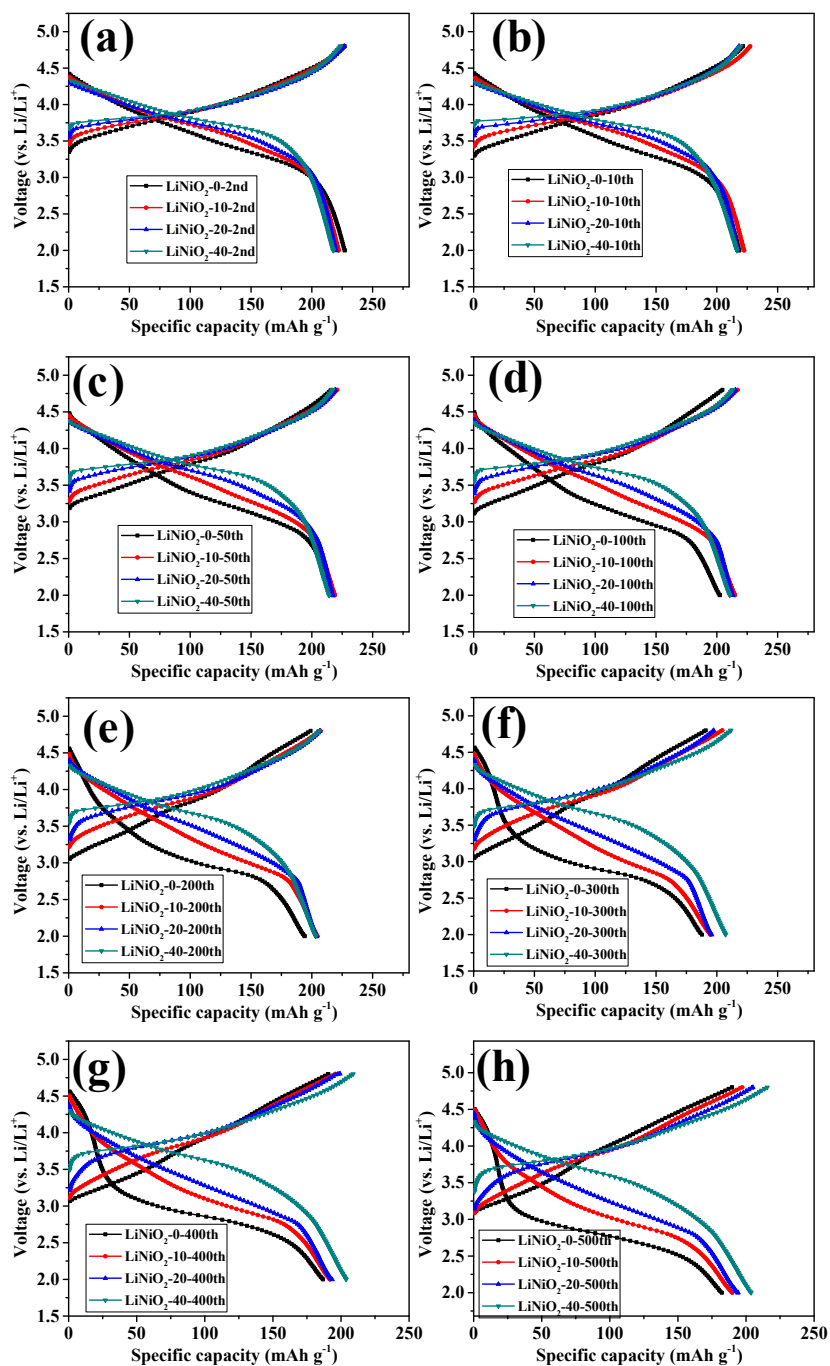


Fig. S2 Comparisons of charge and discharge curves of the LiNiO_2 -0, LiNiO_2 -10, LiNiO_2 -20 and LiNiO_2 -40 electrodes in the range of 2.0 to 4.8 V vs. Li/Li^+ at 200 mA g^{-1} under 30°C .

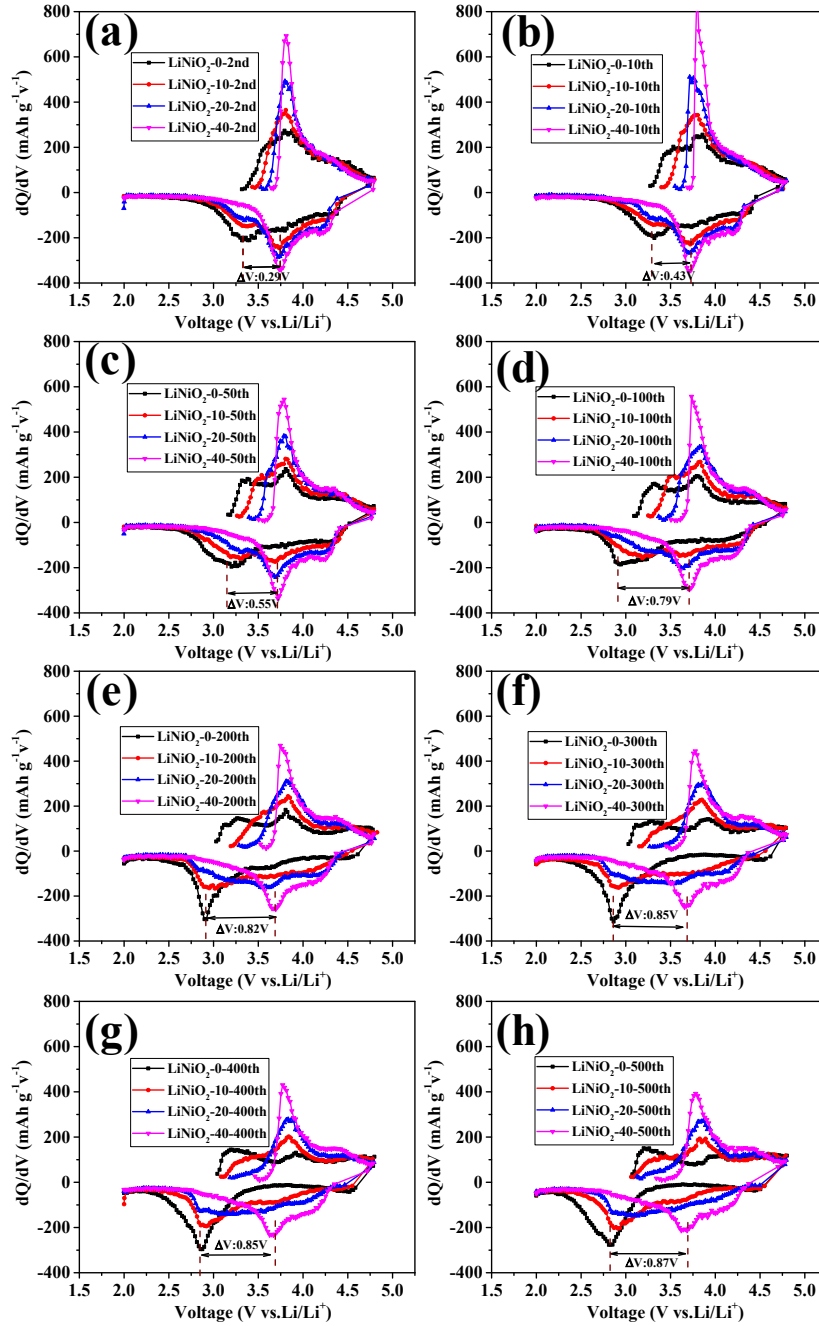


Fig. S3 Comparisons of the dQ/dV profiles for curves of the LiNiO₂-0, LiNiO₂-10, LiNiO₂-20 and LiNiO₂-40 electrodes in the range of 2.0 to 4.8 V vs. Li/Li⁺ at 200 mA g⁻¹ under 30°C.

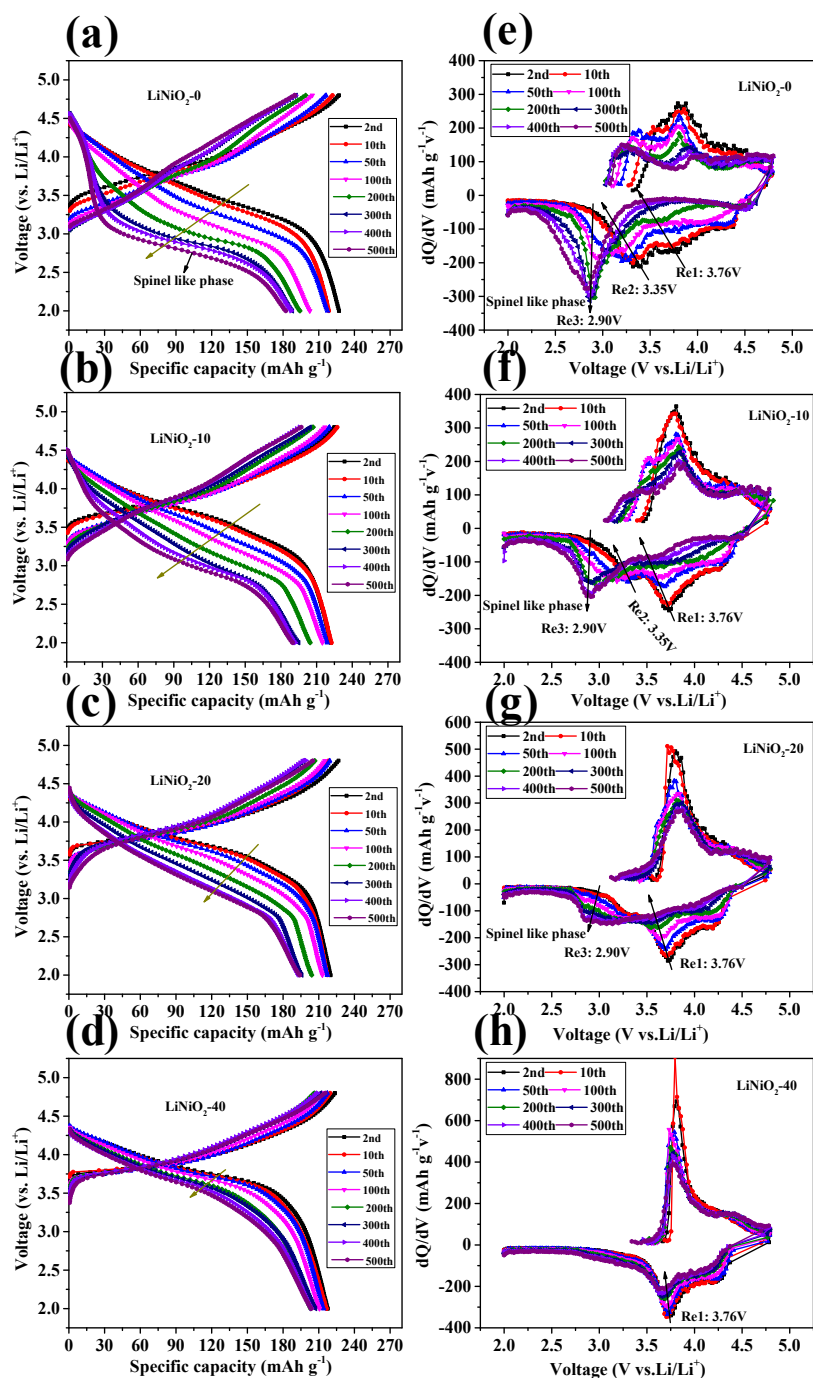


Fig. S4 (a-d) Charge and discharge curves of LiNiO_2 -0, LiNiO_2 -10, LiNiO_2 -20 and LiNiO_2 -40 electrodes in the range of 2.0 to 4.8 V vs. Li/Li^+ at 200 mA g^{-1} under 30°C. (e-h) dQ/dV profiles of LiNiO_2 -0, LiNiO_2 -10, LiNiO_2 -20 and LiNiO_2 -40 electrodes in the range of 2.0 to 4.8 V vs. Li/Li^+ at 200 mA g^{-1} under 30°C.

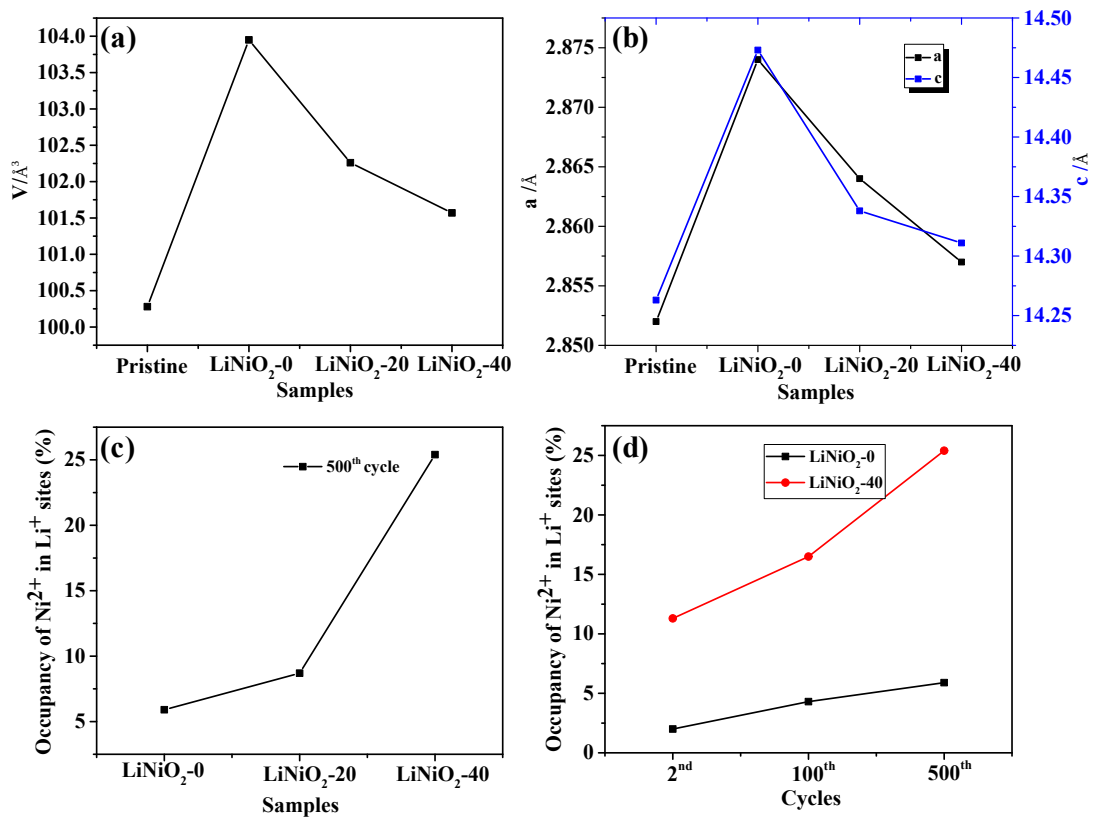


Fig. S5 (a, b) Refined lattice parameters of the LiNiO_2 -0, LiNiO_2 -20 and LiNiO_2 -40 electrodes after 500 cycles. (c, d) The occupancy of the Ni^{2+} ions in the Li^+ ions sites of the LiNiO_2 -0, LiNiO_2 -20 and LiNiO_2 -40 electrodes at different cycles.

Table S1 Experimental and theoretical results of Li/Ni/Co/Mn mole ratios of the $\text{Li}[\text{Li}_{0.2}\text{Ni}_{0.13}\text{Co}_{0.13}\text{Mn}_{0.54}]\text{O}_2\text{-xLiNiO}_2$ materials.

Samples	Li:Ni:Co:Mn	
	Theoretical	Experimental
LiNiO₂-0	1.20: 0.13:0.13:0.54	1.1950:0.1312:0.1291:0.5384
LiNiO₂-10	1.18: 0.217:0.117:0.486	1.1780:0.2190:0.1184:0.4832
LiNiO₂-20	1.16: 0.304:0.104:0.432	1.1540: 0.3071:0.1011:0.4301
LiNiO₂-40	1.12: 0.478:0.078:0.324	1.1151: 0.4802:0.0791:0.3239

Table S2 Crystallographic parameters obtained from the results of the Rietveld refinement: (a) for the LiMO_2 (R-3m) phase and (b) for the Li_2MO_3 (C2/m) phase of the LiNiO_2 -0 sample.

(a)LiNiO ₂ -0 (54.9 wt.%)		<i>R-3m</i>		$\chi^2=1.085$ $R_{wp}=0.0133$ $R_p=0.0099$		
<i>a</i> = 2.852 Å, <i>c</i> = 14.237 Å,						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	3a	0	0	0	1.0000	0.9933
Ni	3a	0	0	0	0.0000	0.0067
Mn	3b	0	0	0.5	0.5400	0.5400
Ni	3b	0	0	0.5	0.1330	0.1267
Co	3b	0	0	0.5	0.1330	0.1330
Li	3b	0	0	0.5	0.2000	0.2033
O	6c	0	0	0.243	1.0000	1.0000
(b)LiNiO ₂ -0 (45.1 wt.%)		<i>C2/m</i>		$\chi^2=1.085$ $R_{wp}=0.0133$ $R_p=0.0099$		
<i>a</i> = 4.94 Å, <i>b</i> = 8.52 Å, <i>c</i> = 5.022 Å, β = 108.99°						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Mn	4g	0	0.173	0	0.6760	0.6760
Ni	4g	0	0.173	0	0.1620	0.1547
Co	4g	0	0.173	0	0.1620	0.1620
Li	4g	0	0.173	0	0.0000	0.0073
Ni	2b	0	0.5	0	0.0000	0.1056
Li	2b	0	0.5	0	1.0000	0.8944
Li	2c	0	0	0.5	1.0000	0.9803
Ni	2c	0	0	0.5	0.0000	0.0197
Ni	4h	0	0.715	0.5	0.0000	0.0912
Li	4h	0	0.715	0.5	1.0000	0.9088
O	4i	0.238	0	0.219	1.0000	1.0000
O	8j	0.253	0.320	0.258	1.0000	1.0000

Table S3 Crystallographic parameters obtained from the results of the Rietveld refinement: (a) for the LiMO_2 (R-3m) phase and (b) for the Li_2MO_3 (C2/m) phase of the LiNiO_2 -10 sample.

(a) LiNiO ₂ -10 (60.2 wt.%)		<i>R-3m</i>			$\chi^2=1.313$ $R_{wp}=0.0154$ $R_p=0.0115$	
<i>a</i> = <i>b</i> = 2.858 Å, <i>c</i> = 14.240 Å,						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	3a	0	0	0	1.0000	0.9888
Ni	3a	0	0	0	0.0000	0.0112
Mn	3b	0	0	0.5	0.4900	0.4900
Ni	3b	0	0	0.5	0.2091	0.1949
Co	3b	0	0	0.5	0.1181	0.1181
Li	3b	0	0	0.5	0.1800	0.1942
O	6c	0	0	0.244	1.0000	1.0000
(b) LiNiO ₂ -10 (39.8 wt.%)		<i>C2/m</i>			$\chi^2=1.313$ $R_{wp}=0.0154$ $R_p=0.0115$	
<i>a</i> = 4.974 Å, <i>b</i> = 8.499 Å, <i>c</i> = 5.034 Å, β = 109.340°						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Mn	4g	0	0.1784	0	0.5970	0.5970
Ni	4g	0	0.1784	0	0.2590	0.2583
Co	4g	0	0.1784	0	0.1440	0.1440
Li	4g	0	0.1784	0	0.0000	0.0007
Ni	2b	0	0.5	0	0.0000	0.1142
Li	2b	0	0.5	0	1.0000	0.8858
Li	2c	0	0	0.5	1.0000	0.9596
Ni	2c	0	0	0.5	0.0000	0.0404
Ni	4h	0	0.702	0.5	0.0000	0.1035
Li	4h	0	0.702	0.5	1.0000	0.8965
O	4i	0.200	0	0.210	1.0000	1.0000
O	8j	0.261	0.312	0.269	1.0000	1.0000

Table S4 Crystallographic parameters obtained from the results of the Rietveld refinement: (a) for the LiMO_2 (R-3m) phase, (b) for the Li_2MO_3 (C2/m) phase and (c) for the $\text{LiNi}_y\text{Mn}_{2-y}\text{O}_4$ (Fd-3m) phase of the LiNiO_2 -20 sample.

(a)LiNiO ₂ -20 (74.5 wt.%)		R-3m		χ ² =1.528 R _{wp} =0.0164 R _p =0.0123		
a = b= 2.861 Å, c = 14.242 Å,						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	3a	0	0	0	1.0000	0.9861
Ni	3a	0	0	0	0.0000	0.0139
Mn	3b	0	0	0.5	0.4500	0.4500
Ni	3b	0	0	0.5	0.2750	0.2611
Co	3b	0	0	0.5	0.1080	0.1080
Li	3b	0	0	0.5	0.1700	0.1839
O	6c	0	0	0.243	1.0000	1.0000
(b)LiNiO ₂ -20 (23.1 wt%)		C2/m		χ ² =1.528 R _{wp} =0.0164 R _p =0.0123		
a = 4.961 Å, b = 8.516 Å, c = 5.0211 Å, β = 109.014°						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Mn	4g	0	0.175	0	0.5422	0.5422
Ni	4g	0	0.175	0	0.3310	0.3302
Co	4g	0	0.175	0	0.1301	0.1301
Li	4g	0	0.175	0	0.0000	0.0008
Ni	2b	0	0.5	0	0.0000	0.1160
Li	2b	0	0.5	0	1.0000	0.884
Li	2c	0	0	0.5	1.0000	0.9262
Ni	2c	0	0	0.5	0.0000	0.0738
Ni	4h	0	0.698	0.5	0.0000	0.1230
Li	4h	0	0.698	0.5	1.0000	0.8770
O	4i	0.239	0	0.213	1.0000	1.0000
O	8j	0.250	0.321	0.265	1.0000	1.0000
(c)LiNiO ₂ -20 (2.4 wt.%)		Fd-3m		χ ² =1.528 R _{wp} =0.0164 R _p =0.0123		
a = b = c= 7.74Å						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	8a	0	0	0	1.0000	0.9069
Ni	8a	0.625	0.625	0.625	0.00	0.0931
Mn	16d	0.625	0.625	0.625	0.7500	0.7500
Ni	16d	0.625	0.625	0.625	0.2500	0.2210
Li	16d	0.625	0.625	0.625	0.0000	0.0290
O	32e	0.388	0.388	0.388	1.0000	1.0000

Table S5 Crystallographic parameters obtained from the results of the Rietveld refinement: (a) for the LiMO_2 (R-3m) phase, (b) for the Li_2MO_3 (C2/m) phase and (c) for the $\text{LiNi}_y\text{Mn}_{2-y}\text{O}_4$ (Fd-3m) phase of the LiNiO_2 -40 sample.

(a) LiNiO ₂ -40 (78.7 wt.%)		R-3m		χ ² =1.769 R _{wp} =0.0168 R _p =0.0109		
a = b = 2.868 Å, c = 14.253 Å,						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	3a	0	0	0	1.0000	0.9770
Ni	3a	0	0	0	0.0000	0.0230
Mn	3b	0	0	0.5	0.3860	0.3860
Ni	3b	0	0	0.5	0.3790	0.3560
Co	3b	0	0	0.5	0.0950	0.0950
Li	3b	0	0	0.5	0.1400	0.1430
O	6c	0	0	0.243	1.0000	1.0000
(b) LiNiO ₂ -40 (15.2 wt.%)		C2/m		χ ² =1.769 R _{wp} =0.0168 R _p =0.0109		
a = 5.021 Å, b = 8.374 Å, c = 4.962 Å, β = 108.703°						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Mn	4g	0	0.168	0	0.4488	0.4488
Ni	4g	0	0.168	0	0.4406	0.4397
Co	4g	0	0.168	0	0.1106	0.1106
Li	4g	0	0.168	0	0.0000	0.0009
Ni	2b	0	0.5	0	0.0000	0.1298
Li	2b	0	0.5	0	1.0000	0.8944
Li	2c	0	0	0.5	1.0000	0.8901
Ni	2c	0	0	0.5	0.0000	0.1099
Ni	4h	0	0.687	0.5	0.0000	0.1783
Li	4h	0	0.687	0.5	1.0000	0.8217
O	4i	0.241	0	0.229	1.0000	1.0000
O	8j	0.274	0.325	0.285	1.0000	1.0000
(c) LiNiO ₂ -40 (6.1 wt.%)		Fd-3m		χ ² =1.769 R _{wp} =0.0168 R _p =0.0109		
a = b = c = 7.74 Å						
atom	site	x	y	z	Occ. _{theo}	Occ. _{refined}
Li	8a	0	0	0	1.0000	0.8575
Ni	8a	0.625	0.625	0.625	0.00	0.1425
Mn	16d	0.625	0.625	0.625	0.7500	0.7500
Ni	16d	0.625	0.625	0.625	0.2500	0.2330
Li	16d	0.625	0.625	0.625	0.0000	0.0170
O	32e	0.388	0.388	0.388	1.0000	1.0000