Supporting Information

Ratiometric fluorescent imaging for endogenous selenocysteine in cancer cell matrix

Yong Tian, Fangyun Xin, Congcong Gao, Jing Jing*, Xiaoling Zhang*

Key Laboratory of Cluster Science of Ministry of Education, Beijing Key Laboratory of Photoelectronic/Electrophotonic Conversion Materials, School of Chemistry and Chemical Engineering, Beijing Institute of Technology, Beijing 100081, PR China

* Corresponding authors: zhangxl@bit.edu.cn, hellojane@bit.edu.cn

Contents

- 1. Comparison of previous probes for selenocysteine detection
- 2. Synthesis of Rat-Sec
- **3.** Characterization of Rat-Sec
- 4. Measurement of quantum yield for Rat-Sec and Nap-OH
- 5. UV-vis responses of Rat-Sec toward various concentration of Sec
- 6. HPLC chromatogram of Rat-Sec and Nap-OH
- 7. ESI-MS spectra characterization of Rat-Sec reaction with Sec
- 8. Photos of Rat-Sec toward various analytes

9. Semiquantitative determination of exogenous and endogenous Sec in living cells

10. Calibration curve of Sec and quantification of Sec in cells

1. Comparison of previous probes for selenocysteine detection

Probe	Detection condition	Detection limit	Response time	$\lambda_{ex}/\lambda_{em}$	Reference
	pH 5.8 PBS/0.1% EtOH	4 nM	10 min	450/580 nm	Angew. Chem. Int. Ed., 2006, 45, 1810–1813
$O_2 N - \left(\begin{array}{c} O \\ H \\ S \\ H \\ N \\ O_2 \end{array} \right) \left(\begin{array}{c} O \\ S \\ H \\ O \\ O \\ O \end{array} \right) \left(\begin{array}{c} O \\ S \\ O \\$	pH 7.4 PBS/1% DMSO	62 nM	5 min	370/502 nm	J. Am. Chem. Soc., 2015, 137, 757–769
N ⁺	pH 7.4 PBS/1% DMSO	7 nM	30 min	460/580 nm	Chem. Commun., 2015, 51, 3102– 3105
$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	pH 7.4 PBS	1.5×10 ⁻⁷ M	5 min	360/550 nm	Anal. Chem., 2016, 88, 7259–7267
This work	pH 7.4 PBS/5% DMSO	12 nM	1 min	380/420, 546 nm	

Table S1. Comparison of previous probes for selenocysteine detection

2. Synthesis of Rat-Sec

Nap-OH (269.3 mg, 1 mmol) and triethylamine (101.2 mg, 0.3 mmol) were dissolved in 10 mL anhydrous dichloromethane, and then acryloyl chloride (135.8 mg, 1.5 mmol) was added into the solution, the system was stirred at room temperature for 6h under argon protection. After reaction was over, the solvent was removed under reduced pressure, and the residues were purified by silica gel chromatography with dichloromethane to obtain the desired product **Rat-Sec** (171.3 mg, 0.53 mmol), a white solid. Yield was 53%. ¹H NMR (400 MHz, CDCl₃, ppm) δ : 0.98 (t, 3H), 1.41-1.50 (m, 2H), 1.65-1.78 (m, 2H), 4.18 (t, 2H), 6.20 (d, 1H), 6.50 (dd, 1H), 6.79 (d, 1H), 7.59 (d, 1H), 7.76 (t, 1H), 8.24 (d, 1H), 8.57-8.667 (m, 2H). ESI-MS calcd. for C19H18NO4 [M+H⁺]: 324.1158; found: 324.1283.



Scheme S1 Synthetic route of Rat-Sec

3. Characterization of Rat-Sec



Fig. S1 ¹H NMR spectrum of Rat-Sec



Fig. S2 HRMS spectrum of Rat-Sec

4. Measurement of quantum yield for Rat-Sec and Nap-OH

The quantum yield of Rat-Sec and Nap-OH were detected in PBS/5% DMSO solution. The standard samples of quinine sulfate (Φ =0.54, in 0.5 M H₂SO₄) and rhodamine 6G (Φ =0.94, in ethanol) were used as reference for Rat-Sec and Nap-OH respectively. The quantum yield can be calculated via the following equation:

$$\Phi_x = \Phi_s \times (G_x/G_s) \times (\eta_x^2/\eta_s^2)$$

Where the subscripts x and s denote test and standard sample respectively, Φ is the fluorescence quantum yield, G the gradient from the plot of integrated fluorescence intensity vs absorbance, and η the refractive index of the solvent.

The plot of integrated fluorescence intensity vs absorbance is given below, then the values of quantum yield are 0.077 (Rat-Sec) and 0.32 (Nap-OH).



Fig. S3 The linear plots of integrated fluorescence intensity vs absorbance for quinine sulfate and Rat-Sec (a); Rhodamine 6G and Nap-OH (b).

5. UV-vis responses of Rat-Sec toward various concentration of Sec



Fig. S4 Absorption spectra of Rat-Sec (5 μ M) upon addition of Sec (0–50 μ M).

6. HPLC chromatogram of Rat-Sec and Nap-OH



Fig. S5 HPLC chromatogram of Rat-Sec before and after 5 min reaction with Sec, and Nap-OH. Eluent solvent: methanol/H₂O (v/v, 8/2), flow rate = 1 mL min⁻¹, detection wavelength at 380 nm.

7. ESI-MS spectra characterization of Rat-Sec reaction with Sec



Fig. S6 ESI-MS spectra characterization of Rat-Sec reaction with Sec.

8. Photos of Rat-Sec toward various analytes



Fig. S7 Photos of Rat-Sec toward various analytes under daylight.

9. Semiquantitative determination of exogenous and endogenous Sec in living cells



Fig. S8 Semiquantitative determination of endogenous Sec in Hela cells according to the ratio of averaged fluorescence intensity of green-yellow channel (500–600 nm) to blue channel (410–460 nm).

10. Calibration curve of Sec and quantification of Sec in cells



Fig. S9 Calibration curve for quantification of Sec concentration from the ratio value. Ratio value is from signals between green-yellow channel (500–600 nm) and blue channel (410–460 nm).