Electronic Supplementary Material (ESI) for Journal of Materials Chemistry B. This journal is © The Royal Society of Chemistry 2018

Impedance Spectroscopic Detection of Binding and Reactions in Acid-Labile Dielectric

Polymers for Biosensor Applications

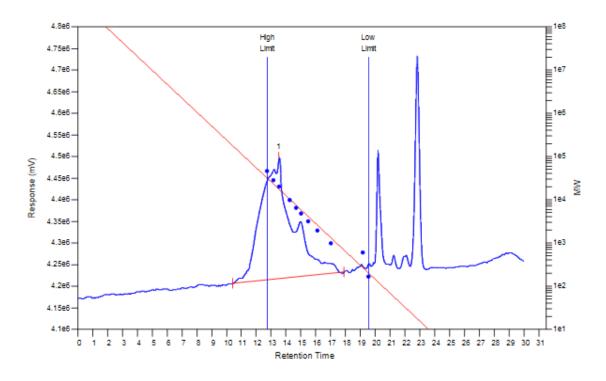
Jennifer Dailey, Michelangelo Fichera, Ellen Silbergeld, Howard Katz

Supplementary Information

SI 1: (a)-(d) GPC of all polymers and analysis 1mg/ml in THF.

(a) PolyBMMA

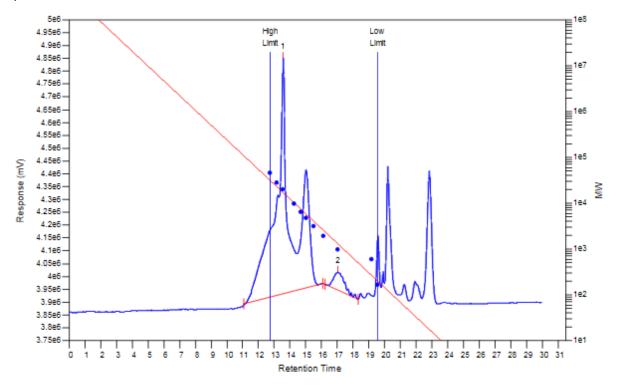
Processed Peaks



MW Averages									
Peak No	Mp	Mn	Mw	Mz	Mz+1	Mv	PD		
1	17728	10052	24194	40796	59341	22014	2.40688		

Peak No Name Start RT Max RT End RT Pk Height % Height % Area Area (mins) (mins) (mins) (mV) (mV.secs) 1 10.43 13.51 17.89 280609 4.20277e+007 100

(b) PolyBMTrM



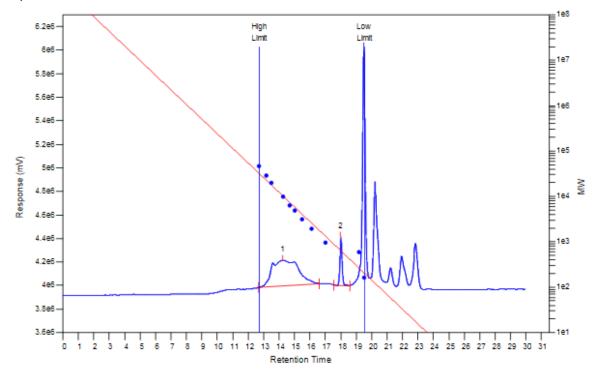
MW Averages

Peak No	Mp	Mn	Mw	Mz	Mz+1	Mv	PD
1	17467	12550	20666	31938	43905	19223	1.64669
2	1307	1152	1232	1303	1366	1221	1.06944

Processed Peaks

Peak No	Name				Pk Height (mV)	% Height	Area (mV.secs)	% Area
1 2		11.09 16.25	13.53 17.02	16.08 18.33	921581 70614.9		6.51088e+007 3.39346e+006	

(c) PolyBMMTrM



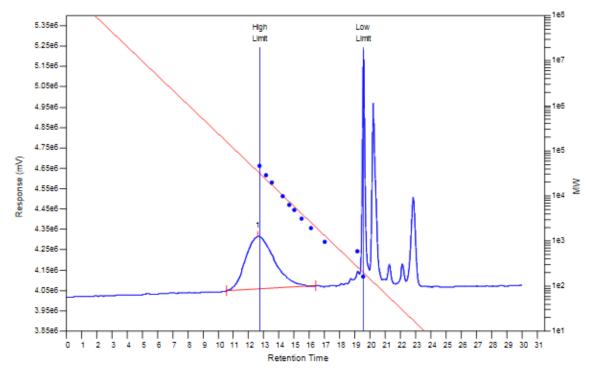
MW Averages

Peak No	Mp	Mn	Mw	Mz	Mz+1	Mv	PD
1	10493	7703	10104	12600	14845	9737	1.3117
2	641	635	637	639	641	637	1.00315

Processed Peaks

Peak No	Name			Pk Height (mV)	Area (mV.secs)	% Area
1 2		12.68 17.55	14.23 17.99	 	 2.61754e+007 4.30433e+006	

(d) PolyBMDTrM



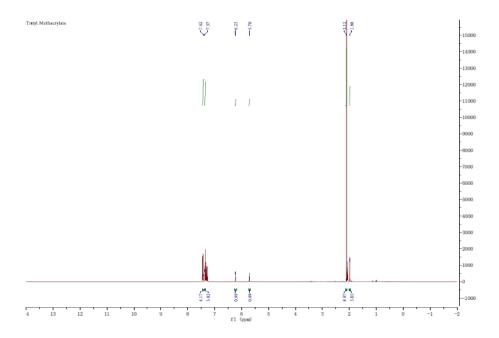
MW Averages

Peak No Mp Mn Mw Mz Mz+1 Mv PD 1 35089 23019 35722 49748 63798 33770 1.55185

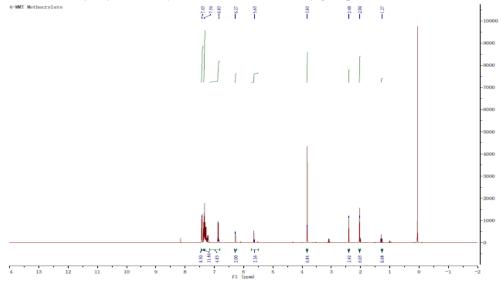
Processed Peaks

Peak No	Name				Pk Height (mV)	% Height	Area (mV.secs)	% Area
1		10.53	12.60	16.41	258997	0	3.41479e+007	100

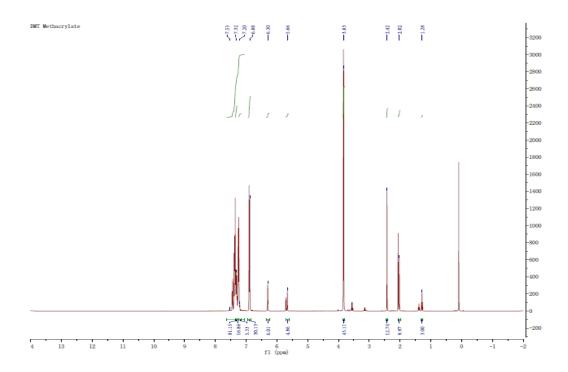
SI 2: (a)-(f) Proton NMR of each monomer and polymer. Trityl Methacrylate polymer



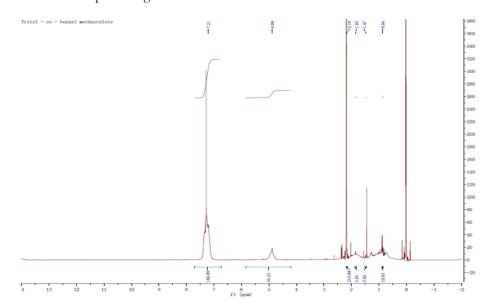
a.) Monomethoxytrityl Methacrylate- We see the methoxy group located near 3.8.



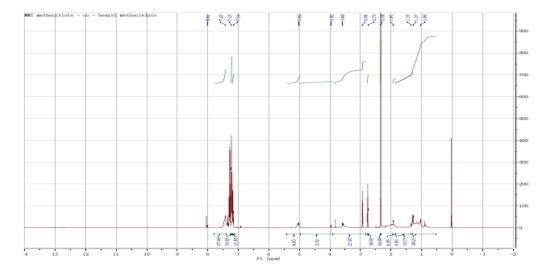
b.) Dimethoxytrityl Methacrylate- The methoxy group at 3.8 doubles due to there being twice as many groups



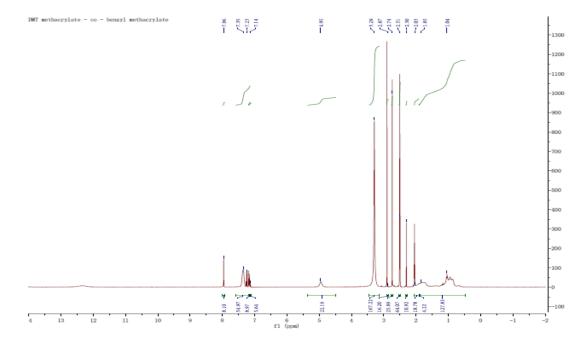
c.) PolyBMTrM- We compare the number of methyl groups to the quantity of trityl so that we can determine the percentage of each monomer.



d.) PolyBMMTrM- We can see the small 3.8 peak indicating the methoxy group, its size is correlated to the amount of monomer successfully included in the polymer.



e.) PolyBMDTrM- We can compare the methoxy group peak to the benzyl methacrylate to find the percentage of monomer incorporated.



SI 3 (Left) 3D image of DMT-based polymer exposed to pH 7.4 buffer. (Right) DIC image of same. Though not flat, this demonstrates that there is no pitting through the surface, but rather there are some small imperfections on top of the film.

