

Supporting Information

Thiophene-*S,S*-dioxidized diarylethenes for light-starting irreversible thermosensors that can detect a rise in heat at low temperature

Daichi Kitagawa, Koki Tanaka and Seiya Kobatake*

Department of Applied Chemistry, Graduate School of Engineering, Osaka City University, 3-3-138 Sugimoto, Sumiyoshi-ku, Osaka 558-8585, Japan

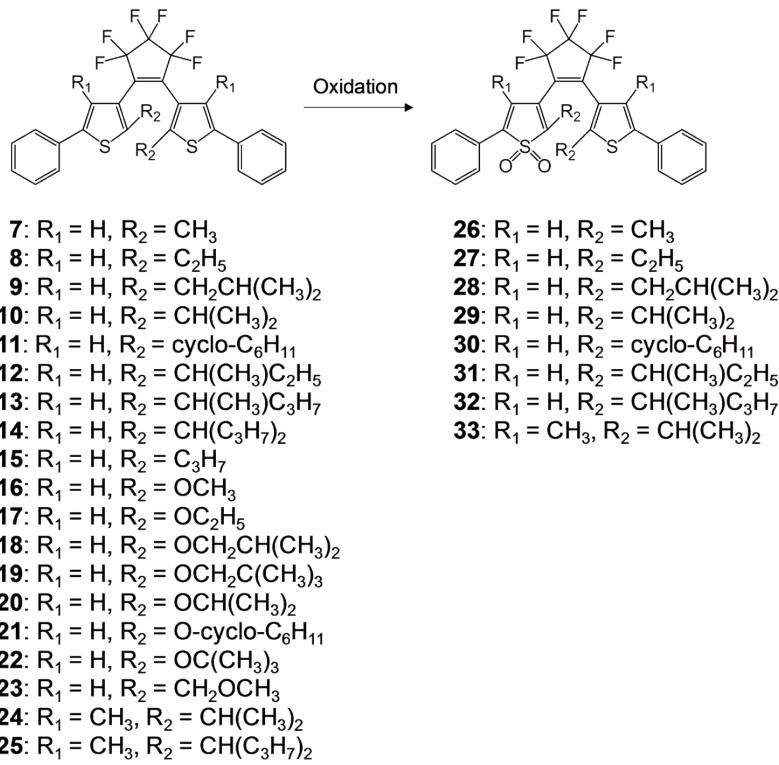
*E-mail: kobatake@a-chem.eng.osaka-cu.ac.jp

Contents

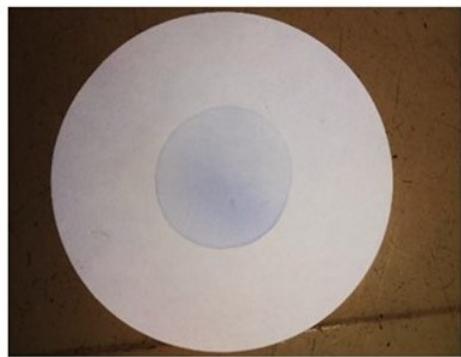
Scheme S1. Molecular structure of diarylethenes and thiophene-*S,S*-dioxidized diarylethenes reported previously.

Fig. S1. Thermal bleaching reaction in a filter paper at -10 °C.

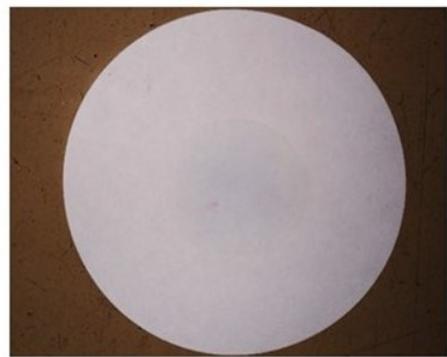
Table S1. Molar absorption coefficients of the open-ring isomer and the closed-ring isomer for diarylethenes and thiophene-*S,S*-dioxidized diarylethenes reported previously.



Scheme S1. Molecular structure of diarylethenes and thiophene-S,S-dioxidized diarylethenes reported previously.^{S1,S2}



Just after irradiation with UV light



after 20 min at $-10\text{ }^{\circ}\text{C}$

Fig. S1. Thermal bleaching behavior of **1b** in a filter paper at $-10\text{ }^{\circ}\text{C}$.

Table S1. Molar absorption coefficients of the open-ring isomer and the closed-ring isomer for diarylethenes and thiophene-*S,S*-dioxidized diarylethenes reported previously.^{S1, S2}

| Compound | open-ring isomer | closed-ring isomer |
|-----------|------------------------------------|------------------------------------|
| | $\epsilon_0/M^{-1} \text{cm}^{-1}$ | $\epsilon_c/M^{-1} \text{cm}^{-1}$ |
| 7 | 35600 | 15600 |
| 8 | 40000 | 17000 |
| 9 | 37200 | 14700 |
| 10 | 34900 | 14800 |
| 11 | 34700 | 13600 |
| 12 | 36200 | 14000 |
| 13 | 40600 | 14900 |
| 14 | 37600 | 12600 |
| 15 | 34100 | 14500 |
| 16 | 33000 | 15000 |
| 17 | 33000 | 13000 |
| 18 | 38700 | 13500 |
| 19 | 36800 | 14500 |
| 20 | 32000 | 13000 |
| 21 | 34000 | 13000 |
| 22 | 33000 | 10000 |
| 23 | 40000 | 16300 |
| 24 | 27100 | 9300 |
| 25 | 32000 | 8850 |
| 26 | 25000 | 19000 |
| 27 | 29200 | 20600 |
| 28 | 31500 | 19600 |
| 29 | 24100 | 17300 |
| 30 | 26000 | 17000 |
| 31 | 29700 | 16800 |
| 32 | 26200 | 16400 |
| 33 | 19400 | 15100 |

Reference

- S1. H. Shoji, D. Kitagawa and S. Kobatake, *New J. Chem.*, 2014, **38**, 933.
- S2. D. Kitagawa and S. Kobatake, *Chem. Rec.*, 2016, **16**, 2005.