Supporting Information

Metal-enhanced fluorescence of gold nanoclusters adsorbed to Ag@SiO₂ core-shell nanoparticles

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Fig. S1 (a) MALDI-TOF MS of BSA (green) and BSA-capped Au NCs (red). (b) HRTEM image of BSA-capped Au₂₅ NCs.



Fig. S2 TEM images of $Ag@SiO_2(L)$ NPs, whose L values are indicated inside. Each scale bar indicates 50 nm.



Fig. S3 Extinction spectra Ag@SiO₂(L) NPs, whose L values are indicated inside.

Table S1 The fitting details of the PL enhancement factor of Au_{25} -adsorbed $Ag@SiO_2$ NPs as a function of separation between Au_{25} clusters and Ag NPs.

Separation region	Fitting equation	α	β
$d < d_c^{a}$	$\mathrm{EF}_1 = (\alpha + \beta \mathrm{d})^6$	1.3	7.5x10 ⁻³ nm ⁻¹
$d > d_c$	$EF_2 = \alpha + \beta/d^3$	3.0	$6.7 x 10^3 nm^3$

^a Maximum enhancement distance (d_c, 11.5 nm) when $EF_1 = EF_2 = maximum$ enhancement factor (EF_{max} , 7.4).



Fig. S4 TEM images of Ag NPs with diverse average diameters indicated inside. Each scale bar indicates 50 nm.

Table S2 Diameters and extinction λ_{max} values of Ag NPs prepared at various concentrations of sodium citrate and tannic acid.

Diameter/nm	$\lambda_{\text{max}}/\text{nm}$	Sodium citrate/mM	Tannic acid/mM
10±2	398	0.81	30.6
28±5	414	0.54	56.4
50±6	433	0.54	94.1
71±8	454	0.27	94.1
85±13	464	0.09	94.1



Fig. S5 Sensing selectivity of Au_{25} -adsorbed $Ag@SiO_2(0)$ NPs for Cu^{2+} over other heavy metal ions. The concentrations of the respective metal ions were 20 μ M, and samples were excited at 370 nm. The gray bars stand for the PL intensities of free Au_{25} clusters while the red bars stand for the PL intensities of Au_{25} -adsorbed $Ag@SiO_2(0)$ NPs.