Supporting Information

Improved Efficiency of Blue Polymer Light-Emitting Diodes Utilizing A Hole Transport Material

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Figure S1. The CV characteristics of BCFN. The observed onset of the oxidation potential (E_{OX}) is 0.75 V, and the redox of ferrocene/ferrocenium is estimated as 0.25 V under exactly the same condition. According to the equation of HOMO = $-e(E_{OX} + 4.80 - 0.25)$ eV, the HOMO is calculated to be -5.30 eV. The lowest unoccupied molecular orbital (LUMO) energy level of BCFN is estimated by adding the optical bandgap ($E_g^{opt} = 3.06$ eV, estimated from the onset of the absorption of BCFN, as shown in Figure 2a) to the HOMO, which is determined to be -2.24 eV.



Figure S2. The electroluminescent spectra (a), luminous efficiency–current density (b) and luminance–voltage (c) characteristics of devices.



Figure S3. The PL spectra of PFSO (80 nm) films processed on quartz/PEDOT:PSS with or without 20 nm PVK or PVK:BCFN = 3:7 as hole transport layer.

Table S1.	. The hole	mobility	of the HT	L computed	with t	the field	dependent	space-
charge lin	nited curren	nt (SCLC)) model.					

HTL	Hole mobility ^a
PVK	2.09×10^{-9}
PVK:BCFN = 9:1	1.03×10^{-8}
PVK:BCFN = 7:3	9.86×10^{-7}
PVK:BCFN = 5:5	1.15×10^{-5}
PVK:BCFN = 4:6	2.14×10^{-5}
PVK:BCFN = 3:7	2.89×10^{-5}
PVK:BCFN = 2:8	3.32×10^{-5}
BCFN	5.43×10^{-5}

^a The hole mobility is calculated in the electric field of $5 \times 10^5 \, V \, cm^{-1}$.