

Electronic Supplementary Information

Systematic Study on Effects of Precursors and Solvents for Optimization of Solution-Processed Oxide Semiconductor Thin- Film Transistors

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Fig. S1

The output characteristics of the (a) sample A, (b) sample B, (c) sample C, (d) sample D, (e) sample E, (f) sample F, and (g) sample G.

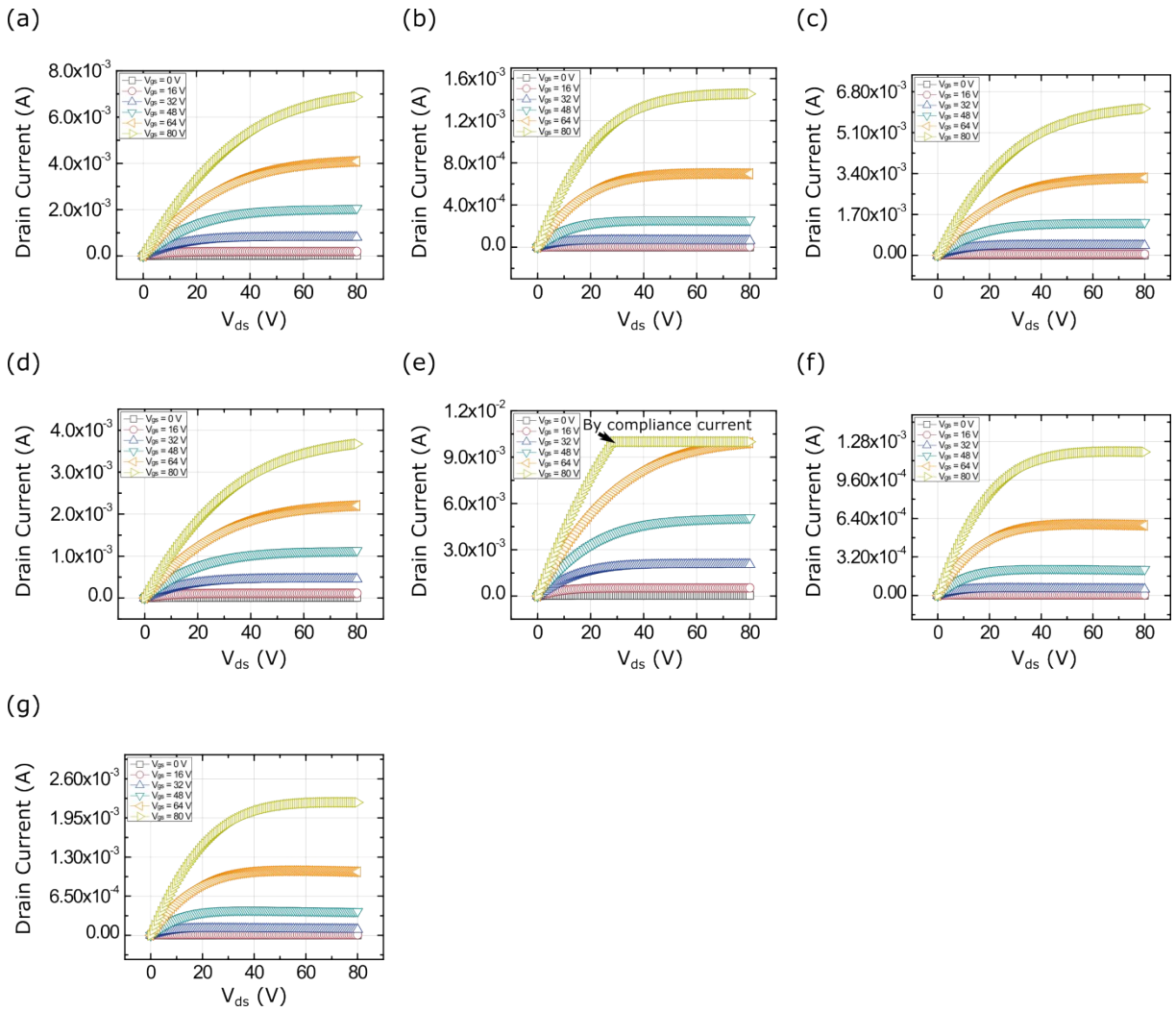


Fig. S2

(a) Schematic of device structure of sample A-G. The optical microscope images of (b) sample A, (c) sample B, (d) sample C, (e) sample D, (f) sample E, (g) sample F, and (h) sample G.

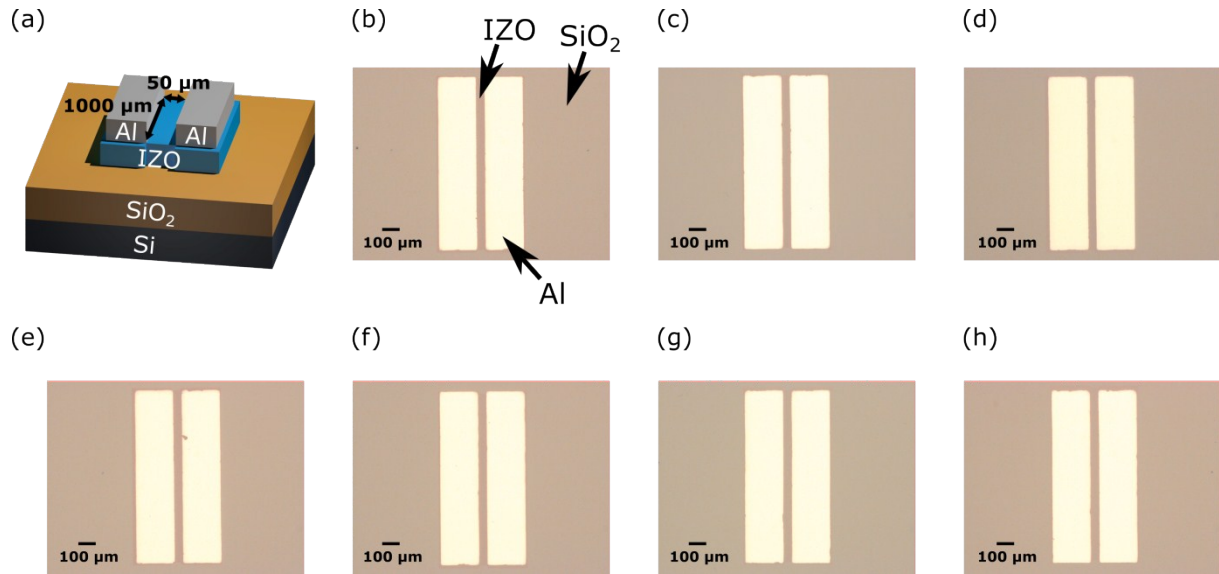


Fig. S3

At 193.15 K, 228.15 K, 263.15 K, and 298.15 K, the transfer characteristics of (a) sample A, (b) sample B, (c) sample C, (d) sample D, (e) sample E, (f) sample F, and (g) sample G.

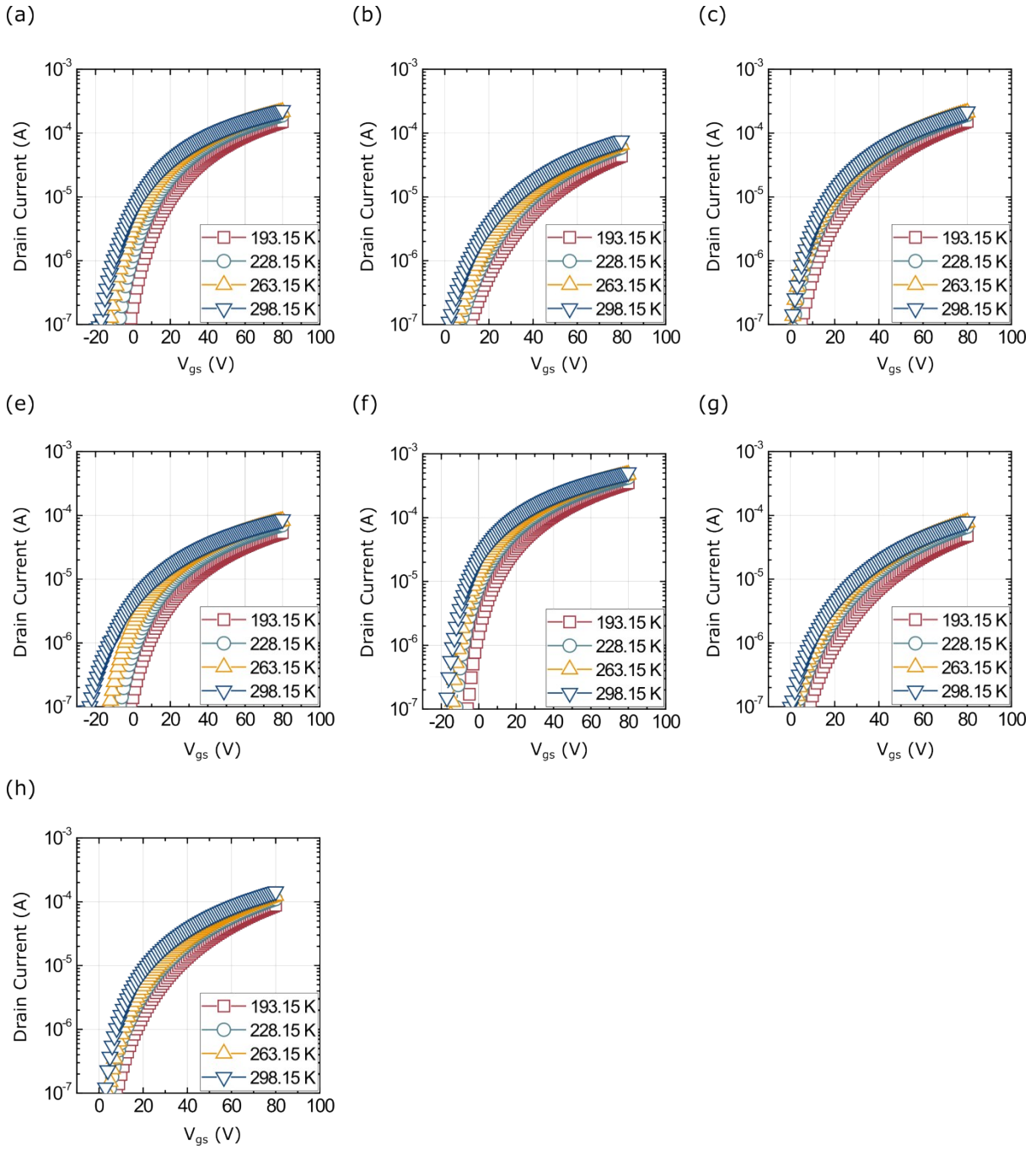


Fig. S4

From the theoretical formula with physical parameters obtained by physical-based analytical TFT model, the free charge carrier concentration (n_b), the trapped charge concentration (n_t), and the total charge carrier concentration (n_T) of (a) sample A, (b) sample B, (c) sample C, (d) sample D, (e) sample E, (f) sample F, and (g) sample G.

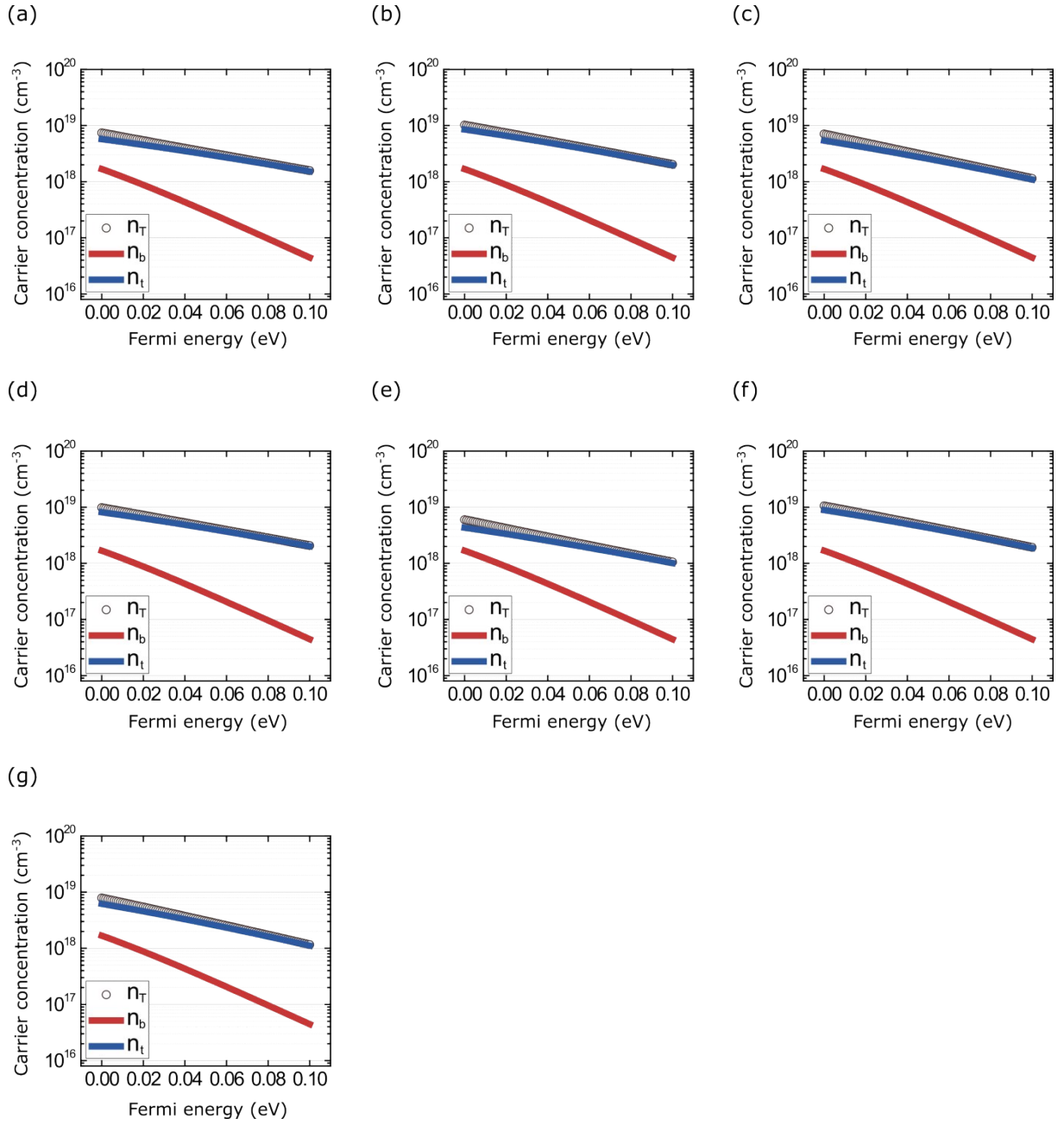


Fig. S5

The simple calculation with ions and atoms volume of decomposed impurities in metal chloride, metal nitrate, and metal acetate.



The volume of chloride impurities in metal chloride (1 chloride ion) : $2.5 * 10^{-2} \text{ nm}^3$

The volume of nitrate impurities in metal nitrate (1 nitrogen ion) : $6.5 * 10^{-5} \text{ nm}^3$

The volume of acetate impurities in metal acetate (2 carbon atoms) $2.8 * 10^{-3} \text{ nm}^3$

Table S1

The composition ratio of In and Zn obtained XPS analysis.

	In / Zn ratio
Sample A	1.52
Sample B	1.55
Sample C	1.55
Sample D	1.56
Sample E	1.53
Sample F	1.51
Sample G	1.54

Table S2

The ratio of O-H bond area to M-O bond area and O_{vac} bond area to M-O bond area.

	Area _{O-H} / Area _{M-O}	Area _{O-VAC} / Area _{M-O}
Sample A	0.41	0.42
Sample B	0.46	0.48
Sample C	0.40	0.42
Sample D	0.43	0.45
Sample E	0.31	0.41
Sample F	0.49	0.57
Sample G	0.46	0.71