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Electronic Supplementary Information to:

Toward Efficient Dual-Emissive Carbon Dots through Sulfur and

Nitrogen Co-Doped

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Figure S1. (a) TEM and HRTEM images of the of N-C-dots; (b) TEM and HRTEM images of the of S-C-dots; (c) AFM images of N -C-dots and height profiles analysis along the corresponding lines;(d) AFM images of S -C-dots and height profiles analysis along the corresponding lines.



Figure S2. (a)/(b) XPS survey of N-C-dots and S -C-dots. (c)-(h) High-resolution XPS spectra of N-C-dots and S -C-dots.



Figure S3. (a) FTIR spectra of S,N-C-dots, N-C-dots, S -C-dots and mPD.



Figure S4. The photograph of the three C-dots dispersed in aqueous solution under the irradiation of daylight (left) and 365 nm UV light (right). (a) S,N-C-dots; (b) N-C-dots; (c) S-C-dots.



Figure S5. PL spectra of N-C-dots (a) and S-C-dots (b) at different excitation wavelengths.



	S,N-C-dots					Rhodamine 6G				
Absorbance	0	0.014	0.023	0.035	0.047	0	0.009	0.016	0.028	0.043
Integrated PL	0	10470	19005	30508	40138	0	13050	25403	46020	68560
Slope	6.98×10 ⁵					1.603×10 ⁶				
QY(%)	41.37					95				

Figure S6. Plots of integrated PL intensity of S,N-C-dots and rhodamine 6G as a standard of optical absorbance at 460 nm and relevant data.



Figure S7. Photostability of S-C-dots, N-C-dots and S,N-C-dots in aqueous solutions (a-c) under continuous excitation at 360 nm for 3000s using a spectrofluorometer equipped with an xenon lamp (150 W).



Figure S8. PL decay spectra and fitting curves of S,N-C-dots.