## Packing of Russian Doll Clusters to Form a Nanometer-Scale CsCl-type Compound in a Cr-Zn-Sn Complex Metallic Alloy

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## **SUPPORTING INFORMATION**

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**Summary of synthetic procedures:** The schematic figure and table summarize the experimental procedures to prepare  $Cr_{22}Zn_{72}Sn_{24}$  carried out during this investigation. The loading compositions for the procedures I, III, IV, V, and VI are Cr: Zn: Sn= 1:6:2 (by moles). The loading composition for procedure II is Cr: Zn: Sn= 1:3:3. (Procedures I and II yielded crystals and procedure II has the highest crystal yield.)



Crystallographic unit cell: cubic (space group: $Fm-3c$ ), $a = b = c = 25.2562(1)$ Å, $Rp = 4.588$ , $Rwp = 5.084$ .							
Atom	Wyckoff	Occupancy	x	У	Ζ	U(eq) (Å)	
Cr1	8 <i>a</i>	1	1/4	1/4	1/4	0.042	
Cr2	8 <i>b</i>	1	0	0	0	0.045	
Cr3	64g	1	0.19441	0.19441	0.19441	0.054	
Cr4	96 <i>i</i>	1	0	0.10998	0.16782	0.045	
Zn5	192 <i>j</i>	1⁄4	0.13832	0.27097	0.26737	0.051	
Zn6	48 <i>e</i>	1	0.18984	0	0	0.054	
Zn7	96 <i>i</i>	1	0	0.05511	0.08944	0.046	
Zn8	96 <i>i</i>	1	0	0.21475	0.14173	0.045	
Zn9	96h	1	1/4	0.05701	0.05701	0.045	
Zn10	192 <i>j</i>	1	0.05372	0.14996	0.09111	0.048	
Sn11	192 <i>j</i>	1	0.08802	0.16267	0.20472	0.050	

**Table S1.** Powder X-ray diffraction refined parameters of using Rietveld analysis of  $Cr_{22}Zn_{72}Sn_{24}$ at 293K.

Specimen	Α	В	С		
Refined Formula	$Cr_{22}Zn_{72}Sn_{24}$	$Cr_{22}Zn_{72}Sn_{24}$	$Cr_{22}Zn_{72}Sn_{24}$		
Crystal Size	Irregular shape with radius of ca. 0.015mm				
F.W. (g/mol); F(000)	4349.60; 31104	4349.60; 31104	4349.60; 31104		
Space group; Z	<i>Fm</i> <sup>3</sup> <i>c</i> (No. 226); 16	<i>Fm</i> <sup>3</sup> <i>c</i> (No. 226); 16	<i>Fm</i> <sup>3</sup> <i>c</i> (No. 226); 16		
Lattice Parameter a (Å)	25.225(5)	25.179(7)	25.163(10)		
Volume (Å <sup>3</sup> )	16051(5)	15962(8)	15933(11)		
$d_{calc}$ (mg/m <sup>3</sup> )	7.200	7.240	7.253		
Absorption Correction	Multi-scan	Multi-scan	Multi-scan		
$\mu(mm^{-1})$	31.166	31.340	31.397		
θ range (°)	1.61 to 28.40	1.62 to 28.30	1.63 to 28.31		
hkl ranges	$-33 \le h, k, l \le 33$	$-32 \le h, k, l \le 32$	$-33 \le h, k, l \le 33$		
No. of reflections $[I > 2\sigma(I)]$	31824	30763	31563		
No. independent reflections; $R_{int}$	911; 0.0613	891; 0.1055	896; 0.1463		
No. parameters	60	60	60		
$R_1; wR_2 [I > 2\sigma(I)]$	0.0312; 0.0702	0.0342; 0.0613	0.0412;0.0808		
$R_1$ ; $wR_2$ (all $I$ )	0.0355; 0.0760	0.0398; 0.0616	0.0614; 0.0845		
Goodness of fit	1.072	1.110	1.078		
Diffraction peak and hole $(e^{-/}\text{Å}^3)$	4.166; -2.927	2.665; -2.775	2.239; -2.420		

**Table S2.** Selected crystallographic data at 293(2) K for additional crystalline specimens extracted from the Cr-Zn-Sn product.

Atom	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
Cr1	0.005(1)	0.005(1)	0.005(1)	0	0	0
Cr2	0.007(1)	0.008(1)	0.006(1)	-0.000(1)	0	0
Zn3	0.011(1)	0.015(1)	0.009(1)	0	0	0
Zn4	0.010(1)	0.010(1)	0.008(1)	-0.001(1)	0	0
Zn5	0.017(1)	0.013(1)	0.019(1)	0.001(1)	0	0
Zn6	0.009(1)	0.014(1)	0.014(1)	0.002(1)	0.001(1)	-0.001(1)
Zn7	0.015(1)	0.011(1)	0.010(1)	0.002(1)	0.002(1)	0.001(1)
Sn8	0.015(1)	0.020(1)	0.018(1)	0.001(1)	-0.004(1)	-0.006(1)
Cr9	0.005(1)	0.005(1)	0.005(1)	0	0	0
Cr10	0.047(1)	0.047(1)	0.047(1)	-0.018(1)	-0.018(1)	-0.018(1)
Zn11	0.047(3)	0.077(7)	0.012(7)	0.076(6)	0.095(7)	0.050(4)

Table S3. Anisotropic temperature factors for  $Cr_{22}Zn_{72}Sn_{24}$  at 293(2) K.

CMA:	$Cr_{11}Zn_{36}Sn_{12}$ (216 $e^{-}$ )		$\frac{Mo_7 Zn_{38.7} Sn_{12}^{[7]}}{(224.4 \ e^{-})}$		Ru <sub>13</sub> Zn <sub>33.5</sub> Sb <sub>12</sub> <sup>[32]</sup> ("CA"; 278.0 <i>e</i> <sup>-</sup> )		$V_7 Z n_{39} G a_{12}^{[31]}$ (209 $e^-$ )		
Pearson Symbol	cF944		cF922.88		<i>cF</i> 936.64		cF928	cF928	
a (Å)	25.184		25.447		25.098		24.857		
Wyckoff Site in $Fm^3c$	Atom	R (Å)	Atom	<i>R</i> (Å)	Atom	$R(\text{\AA})$	Atom	R (Å)	
			/1	3-Cluster-					
8 <i>b</i>	Cr		Mo		Ru		V		
96 <i>i</i>	Zn	2.586	Zn	2.626	Zn	2.555	Zn	2.646	
48 <i>e</i>	Zn	4.478	Zn	4.519	Zn	4.524	Zn	4.469	
192 <i>j</i>	Zn	4.560	Zn	4.611	Zn	4.653	Zn	4.679	
96 <i>i</i>	Cr	5.222	Мо	5.292	Ru	4.989	V	5.215	
96h	Zn	6.608	Zn	6.674	Zn	6.610	Zn	6.567	
96 <i>i</i>	Zn	6.659	Zn	6.714	Ru	7.973	Zn	6.763	
192 <i>j</i>	Sn	7.032	Sn	7.065	Sb	6.921	Ga	6.913	
			bc	c Cluster					
8 <i>a</i>	Cr		Mo		Ru		V		
64 <i>g</i>	Cr	2.600	0.91 Zn	2.649	0.24 Zn	2.627			
192 <i>j  </i> 48 <i>f</i>	Zn	2.513	0.68 Zn	2.804	0.34 Zn	2.549			
96h							Zn	2.724	
24 <i>d</i>					Zn	6.275			

**Table S4.** Structural features of three cubic CMAs related to  $Cr_{22}Zn_{72}Sn_{24}$ . *R* designates the distance of the atomic shell from the corresponding center of either the 8*b* (center of *I*13-cluster) or the 8*a* (bcc cluster) sites.

Figure S1. TEM diffraction patterns of the crystalline [001] axes for  $Cr_{22}Zn_{72}Sn_{24}$ .





**Figure S2.** Electronic density distribution in  $Cr_{22}Zn_{72}Sn_{24}$  without the Zn11 (192*j*) site along the *x*-axis.



*x*=0.158