## Supplementary Information

## Correlation between Small-molecule Dependent Nanomorphology and Device Performance of Organic Light-Emitting Diodes with Ternary Blend Emitting Layer

Odongo Francis Ngome Okello<sup>1</sup>, Young-Tae Kim<sup>1,\*</sup>, Hyeon-Dong Lee<sup>1</sup>,

Young-Hoon Kim<sup>2</sup>, Tae-Woo Lee<sup>2,3,4</sup>, and Chan-Gyung Park<sup>1,5,\*</sup>

<sup>1</sup>Department of Material Science and Engineering, Pohang University of Science and Technology (POSTECH), Pohang 790-784, South Korea

<sup>2</sup>Research Institute of Advanced Materials, Seoul National University, 1 Gwanak-ro,

Gwanak-gu, Seoul 08826, Republic of Korea

<sup>3</sup>BK21PLUS SNU Materials Division for Education Creative Global Leaders, Seoul National University, 1 Gwana-ro, Gwanak-gu, Seoul 08826, Republic of Korea

<sup>4</sup>Department of Material Science and Engineering, Seoul National University, 1 Gwanak-ro, Gwanak-gu, Seoul 08826, Republic of Korea

<sup>5</sup>National Institute for Nanomaterials Technology (NINT), Pohang University of Science and Technology (POSTECH), Pohang 790-784, South Korea

\*Correspondence and requests for materials should be addressed to Y.-T.Kim (email: <u>yt1001@</u>postech.ac.kr) or to C.G.P (email: <u>cgpark1@postech.ac.kr</u>)

Molecule	Chemical Structure	Electronic Property	Band gap (eV)	Homo (eV)	Lumo (eV)	Triplet Energy (eV)	Dipole moment (D)	Molecular weight (g/mol)
CBP	કેવ્વ્સ	HTM/ETL	3.1	6	2.9	2.6	~0.0	484.89
PBD	XCLOD	ET/HBL	3.8	6.2	2.4	2.6	~	298.34
ТСТА	de ofo	HTM	3.3	5.7	2.4	2.86	3.0	740.89
TPBI	9550 9550	ET/HBL	3.5	6.3	2.8	2.8	3.4	654.76
3TPYMB		ET/HBL	3.5	6.8	3.3	2.87	3.5	599.61
lr(ppy) <sub>3</sub>		Dopant	2.6	5.6	3.0	2.4	6.26	654.78

Table S1 Physical properties of materials used in the experiment<sup>3</sup>



**Fig. S1** Shows (a) TEM Cross-sectional image of TCTA:TPBI:Ir(ppy)<sub>3</sub> based device structure comprising of PEDOT:PSS:PIF/EML/TPBI/LiF/AI, (b) magnified image of the EML (c) energy band diagram of the green OLED.

Solvent Type	Boiling point (C)	Dipole moment (D)	Evaporation rate	Vapor Pressure 20(hPa)	HB (cal/cm³) <sup>1/2</sup>	HS (Mpa) <sup>1/2</sup>
Dichlorobenzene (DCB)	180	2.27	<1	1.33	10	20.5
Chlorobenzene (CB)	131.7	1.54	1.07	12	9.5	19.6
Toluene(T)	110.6	0.43	6.1	29	8.9	18.2

 Table S2 Physical properties of solvents in study<sup>57,58</sup>



**Fig. S2** TEM results: (a) BF image (b) HAADF image and (c) EDS mapping showing the  $Ir(ppy)_3$  needle-like aggregate observed in TCTA:TPBI:Ir(ppy)\_3 ternary blend emitting layer.



**Fig. S3** TEM BF image of (a) polymer-small molecule (PVK:3TPYMB:Ir(ppy)<sub>3</sub>), and (b) small molecule based (TCTA:3TPYMB:Ir(ppy)<sub>3</sub>) emitting layer. The ppy)<sub>3</sub> needle-like aggregates are more pronounced in the polymer-small molecule based EML compared to small molecule based EML.



**Fig. S4** Steady state Photoluminescence spectra of the TCTA:TPBI:Ir(ppy)<sub>3</sub> based EML spincoated in (a) Chlorobenzene (blue line), (b) 50CB:50T (pink line), and 100T (green line) solvents respectively. The blue-peak shift with increasing ratio of the toluene solvent indicates a decrease in the amount of Ir(ppy)<sub>3</sub> aggregates formed in the EML. The magnified inset image on the left is meant for clear observation.