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Supporting Information

Directly coupled dual emitting core based molecular design of thermally activated delayed fluorescent emitters

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Figure S1. Cyclic voltammograms of mCBPTrz-1 and mCBPTrz-2.



Figure S2. TGA and DSC curves of mCBPTrz-1 and mCBPTrz-2.



Figure S3. Energy level diagram and chemical structure of materials.



Figure S4. Power and current efficiencies (a), (b) for mCBPTrz-1 and (c), (d) for mCBPTrz-2.



Figure S5. PL decay curves of mCBPTrz-1 and mCBPTrz-2 measured at various temperatures.



Figure S6. The PL emission spectra at prompt and delayed time.

(left: mCBPTrz-1, right: mCBPTrz-2)



Figure S7. ¹H and ¹³C NMR spectra of A1.



Figure S8. ¹H and ¹³C NMR spectra of A2.



Figure S9. ¹H and ¹³C NMR spectra of A3.















Figure S12. ¹H and ¹³C NMR spectra of B2.





Figure S13. ¹H and ¹³C NMR spectra of B3.



Figure S14. ¹H NMR spectrum of B4.



Figure S15. ¹H NMR spectrum of mCBPTrz-1.



Figure S16. ¹H NMR spectrum of mCBPTrz-2.



Figure S17. High resolution mass spectrometry (HRMS) data of mCBPTrz-1 and mCBPTrz-2.

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Operator ID: SNU-EA2000 Company name: Thermo Fisher					
				(Unit: wt%)	
Sample name	Nitrogen	Carbon	Hydrogen		
mCBP-Trz1	11.8677	83.7546	4.3393		
mCBP-Trz2	9.5487	84.0477	6.2947		

Figure S18. Elemental analysis data of mCBPTrz-1 and mCBPTrz-2.

$$\tau_{\rm p} = 1/k_{\rm p}$$

$$\tau_{\rm d} = 1/k_{\rm d}$$

$$k_{\rm ISC} = (1-\Phi_{\rm F}) \times k_{\rm p}$$

$$k_{\rm RISC} = (k_{\rm p}k_{\rm d}/k_{\rm ISC}) \times (\Phi_{\rm TADF}/\Phi_{\rm F})$$

$$k_{\rm r}^{\rm S} = k_{\rm p}\Phi_{\rm F}$$

$$k_{\rm nr}^{\rm T} = k_{\rm d} - k_{\rm RISC}\Phi_{\rm F}$$

Equations for the calculation of rate constants