

Theoretical investigations on crystal crosslinking in perovskite solar cells

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Electronic Supplementary Information

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Table S1 Change in Hirshfeld charge in the adsorbate (the crosslinking agent) and adsorbent (the halide perovskite substrate) upon molecular adsorption on the perovskite substrate

	7	8	9
Crosslinking Agent	-0.39	-0.54	-0.64
Perovskite Substrate	0.39	0.54	0.64

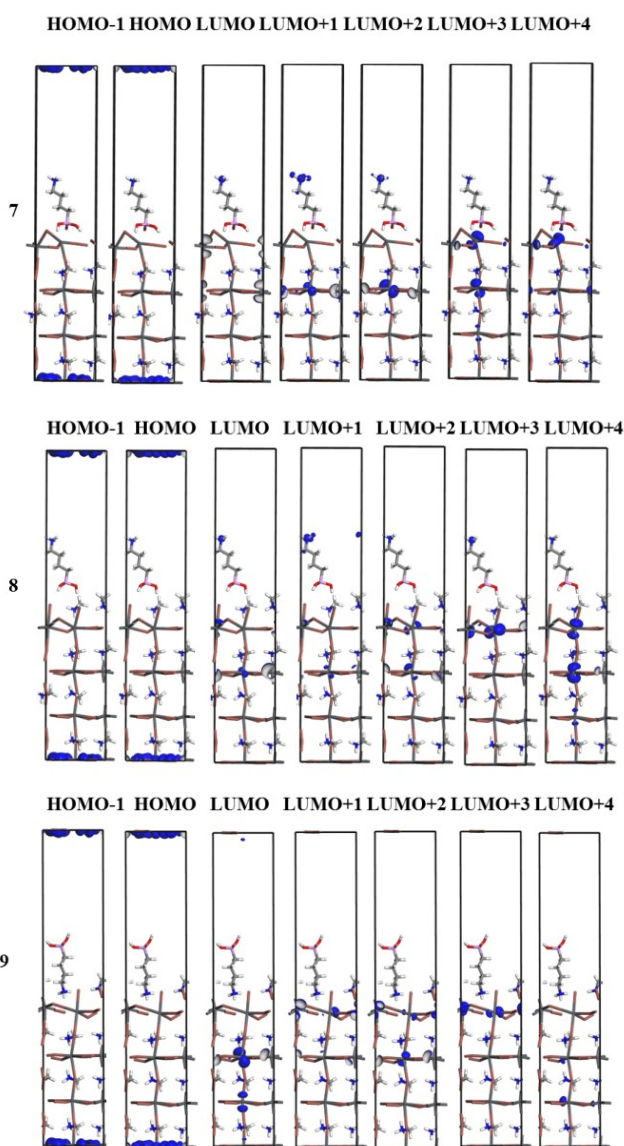


Figure S1 HOMO-1, HOMO, LUMO, LUMO+1, LUMO+2, LUMO+3 and LUMO+4 orbital distribution of 7-9

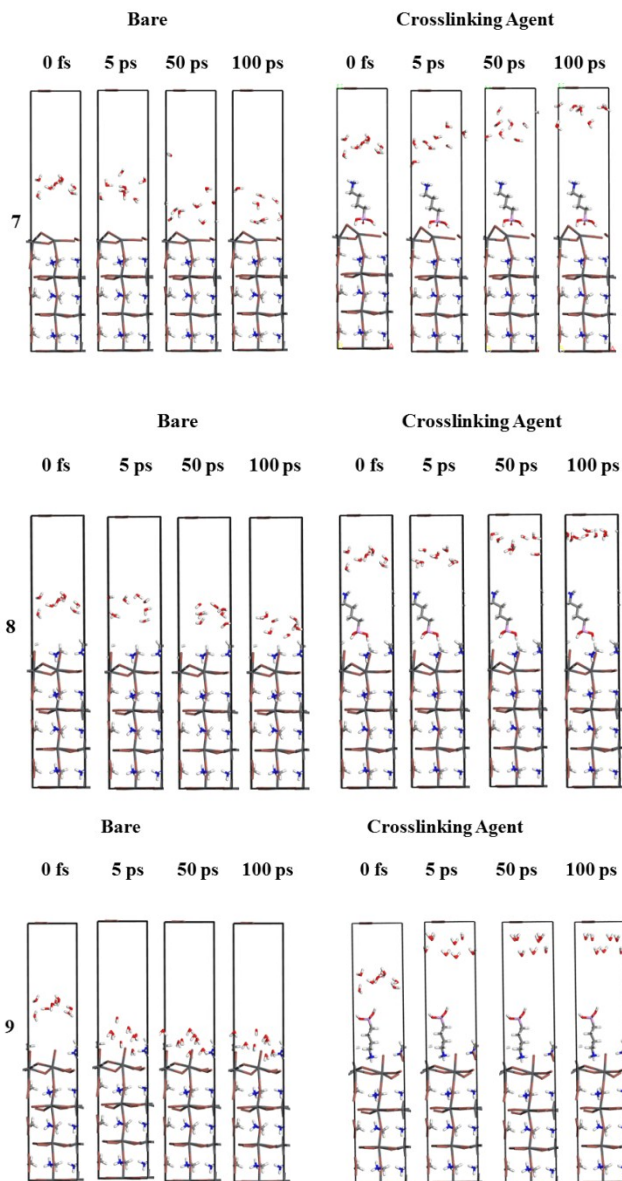
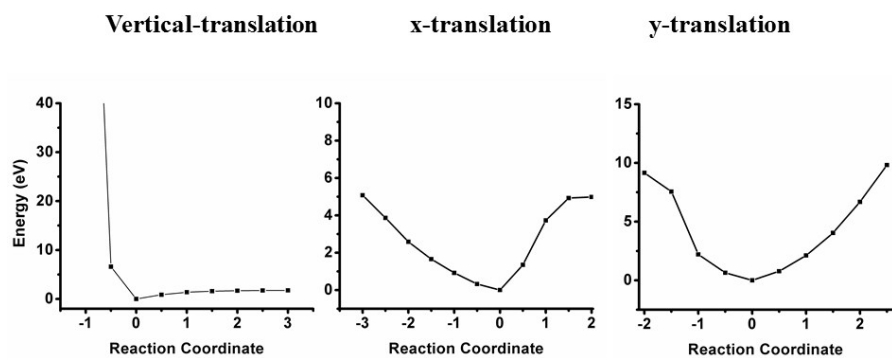
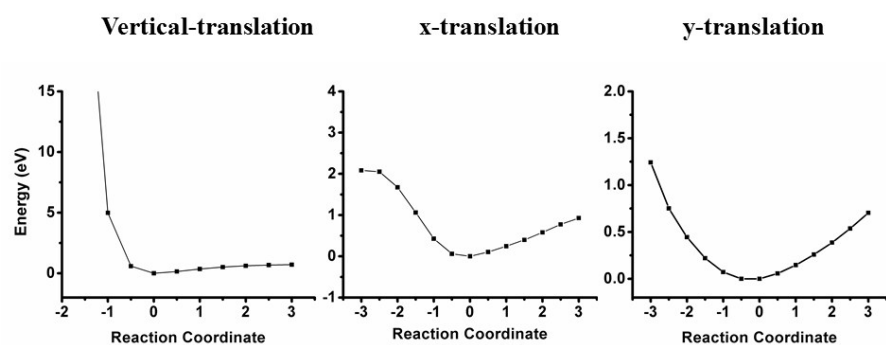


Figure S2 Molecular dynamics of the water interaction with the bare perovskite surface (left) and the perovskite surface capped by the crosslinking agents 7-9 (right). Snapshots at 0 fs, 4 ps, 40 ps and 100 ps are depicted in each case. Water molecules tend to approach the bare perovskite surface while they are expelled from the capped perovskite surface during the simulation time.

7



8



9

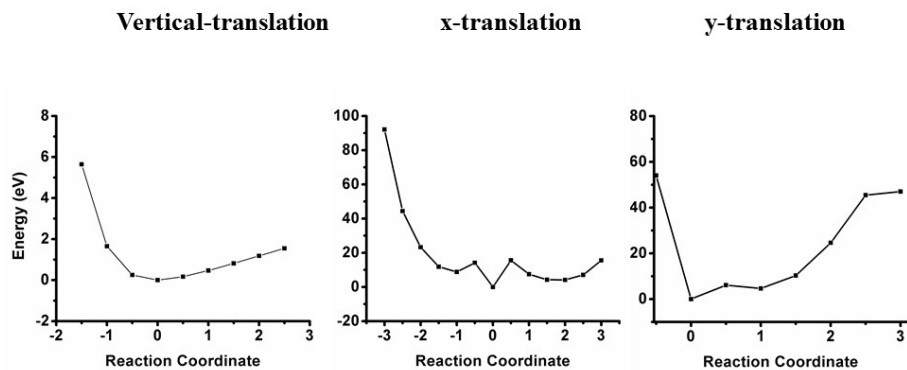


Figure S3 Relative energies along three reaction coordinates (vertical translation along z-axis, x-translation and y-translation of the adsorbate at the perovskite surface) for **7-9**. The optimized stable position is set to 0.

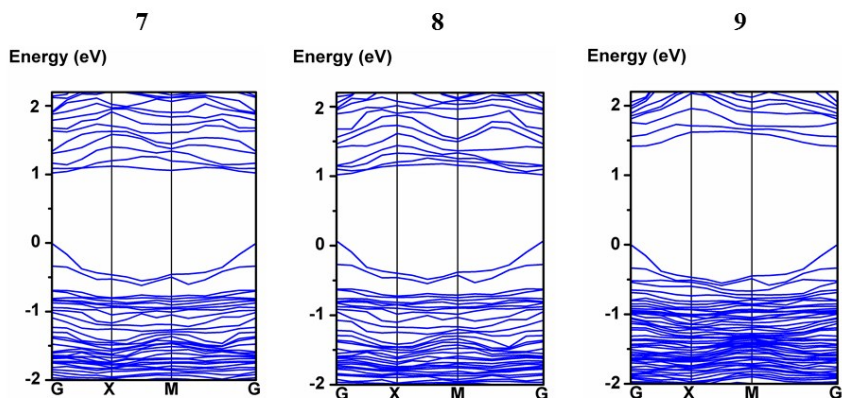


Figure S4 The band structures of **7-9** along lines connecting high symmetry points G (0, 0, 0), X (0, 0.5, 0), M (0.5, 0.5, 0) and G (0, 0, 0). The maximum energy of valence band is set to 0.

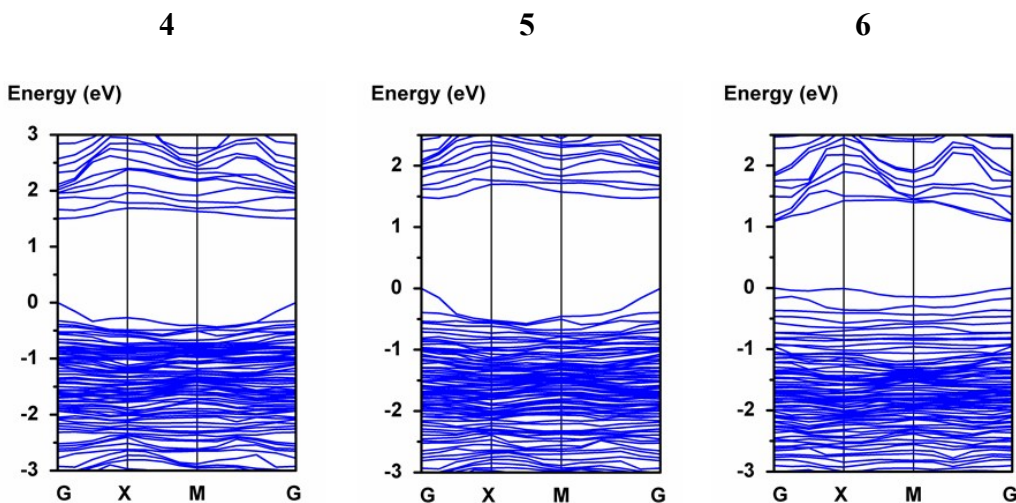


Figure S5 Band structures of **4-6**. The band gaps are 1.51 eV in **4** (1.51 eV from reference ¹), 1.47 eV (1.36 eV from reference ²) in **5**, and 1.09 eV in **6**. The Experimental values are 1.55 – 1.70 eV.²⁻⁴

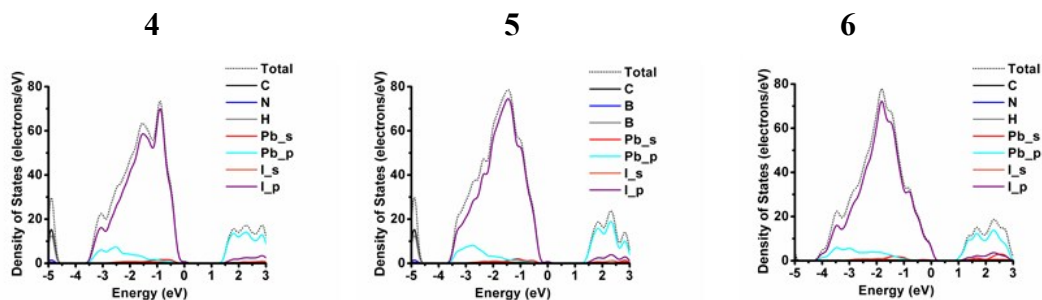


Figure S6 PDOS of **4-6** from -5 eV to 3 eV. In PDOS, the valence band is mainly contributed by the p orbital of I atoms and slightly by the s orbital of Pb atoms, while the conduction band is mainly contributed by the p orbital of Pb atoms and slightly by the s orbital of I atoms. These are consistent with the literature.^{1,5-7}

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