Synthesis and very large mechano- and thermo-hypsochromic luminescence of a new-type DPP-based derivative

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Synthesis of S1 (4-(1,3-dioxolan-2-yl)benzonitrile)



To a solution of 4-cyanobenzaldehyde (15.0 g, 114.4 mmol) in toluene (300 mL) was added ethylene glycol (11.5 mL, 206 mmol) followed by pyridinium-*p*-toluenesulfonic acid (1.5 g, 8.7 mmol) and then heated at 130 °C in Dean-Stark apparatus with continuous exclusion of water. After the formation of H₂O cease. Cooled the reaction solution to room temperature. The solution was washed with saturated aqueous NaHCO₃ (100 mL) and brine (50 mL × 3). The organic layer was dried over anhydrous MgSO₄, and then concentrated via rotary evaporation. The raw product was purified by column chromatography (silica gel; petroleum/ethyl acetate, 1/4, v/v)

to get 4-(1,3-dioxolan-2-yl)benzonitrile as an off white solid. (Yield: 18.53 g, 92%). ¹H NMR (500 MHz, Chloroform-d) δ 7.62 (ddt, J = 41.5, 7.4, 3.0 Hz, 4H), 5.89-5.77 (m, 1H), 4.15-3.98 (m, 4H).

S2. Supporting Figure



Figure S1. FTIR spectra of CODPP solids with different emission colors.

¹H NMR and ¹³C spectra of the compounds as follows











SI-3. Supporting Tables

Table S1 The solution fluorescence quantum yields (%) and the molar extinction coefficients (M^{-1} cm⁻¹) of CODPP in different solvents.

Solvents	PhMe	CHCl ₃	THF	DMF
Fluorescence	74 0	75 1	70.0	67.0
quantum yield (%)	74.0	75.1	12.5	07.0
The molar extinction	54000	50200	50000	52700
coefficients	34000	50300	39000	52700

Table S2 The solid fluorescence quantum yields (%) of CODPP in different states.

States	Pristine	Ground	Annealed
CODPP	15%	46%	95%