

Supporting information

Linearly Polarized Electroluminescence from Ionic Iridium Complex-based Metallomesogens: The Effect of Aliphatic-Chain on Their Photophysical Properties

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Table S1. Electroluminescent data of EL devices

NMR spectra and MALDI-MS spectra

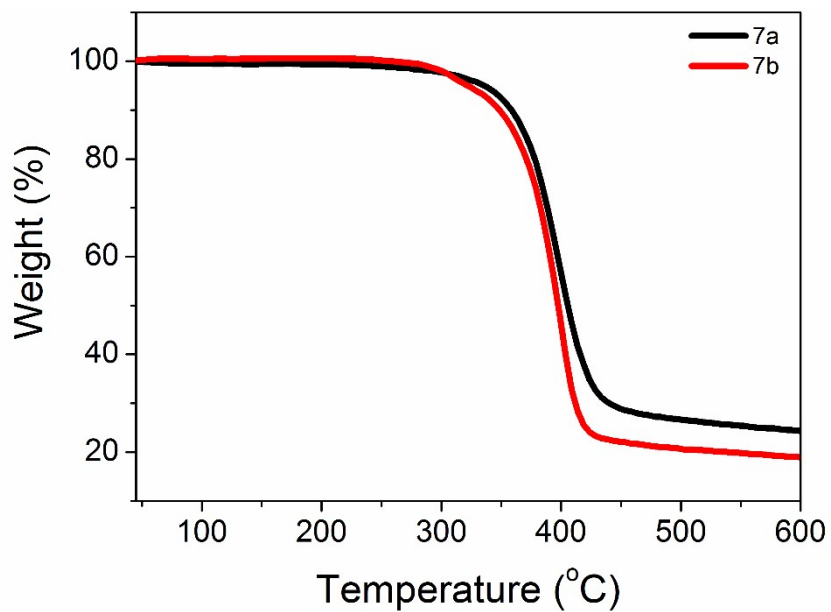


Figure S1. TGA curves of **7a** (black) and **7b** (red) measured at heating rate 10 °C/min under N₂

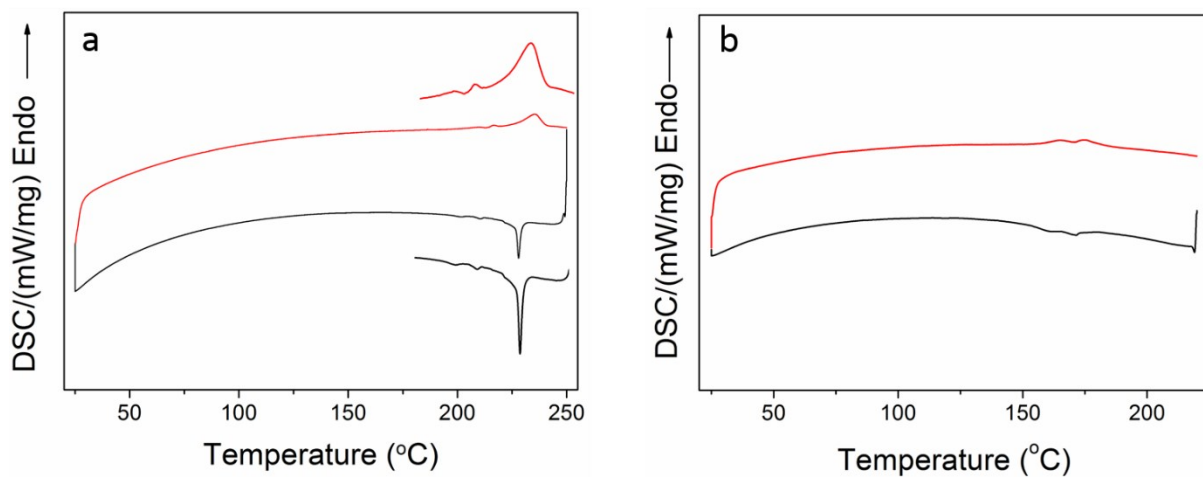


Figure S2. DSC traces of (a) **7a** and (b) **7b** (black: first cooling curve, scan: 10 °C/min; red: second heating curve, scan: 10 °C/min).

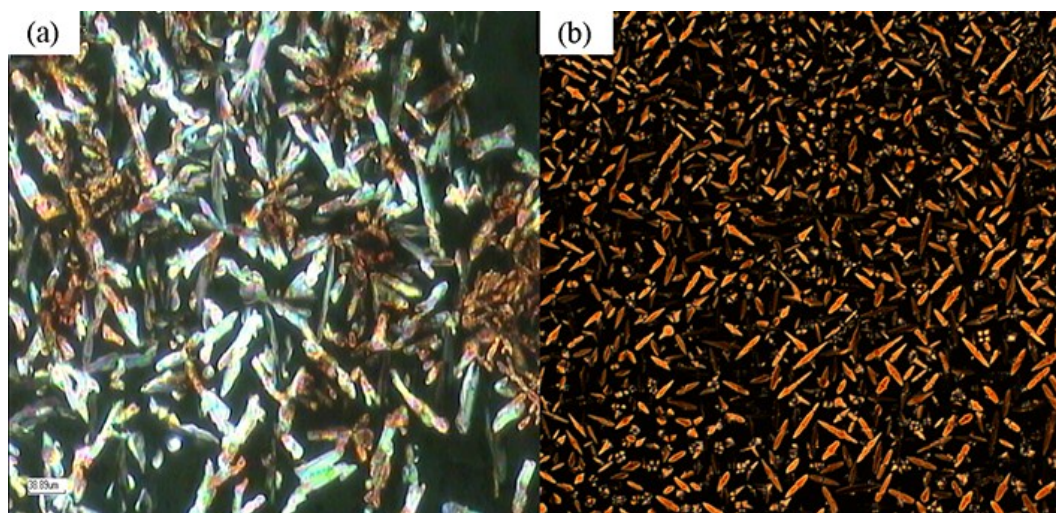


Figure S3. Photomicrographs of (a) **7a** at 225 °C on cooling, (b) **7b** at 174 °C on cooling.

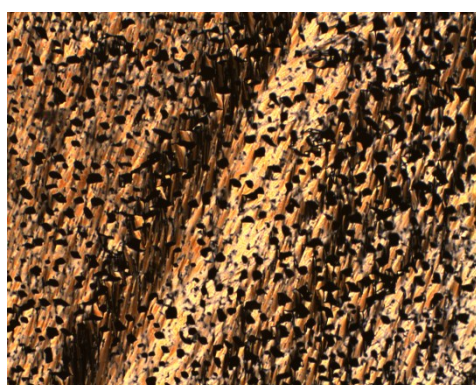


Figure S4. Photomicrographs of 7b at 171°C on cooling process after shearing

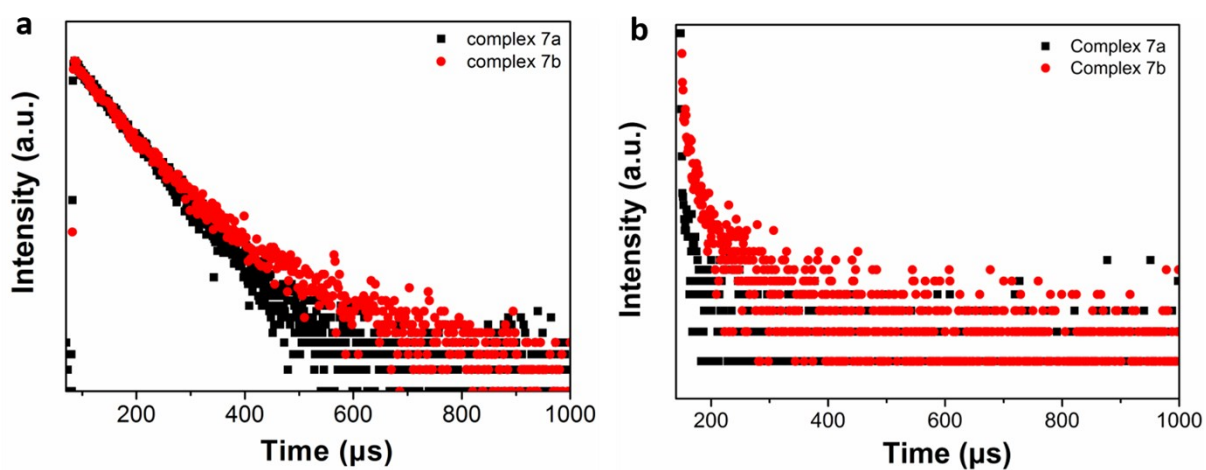


Figure S5. Luminescence decay time of complex in solution (a) and neat film (b) after degassed

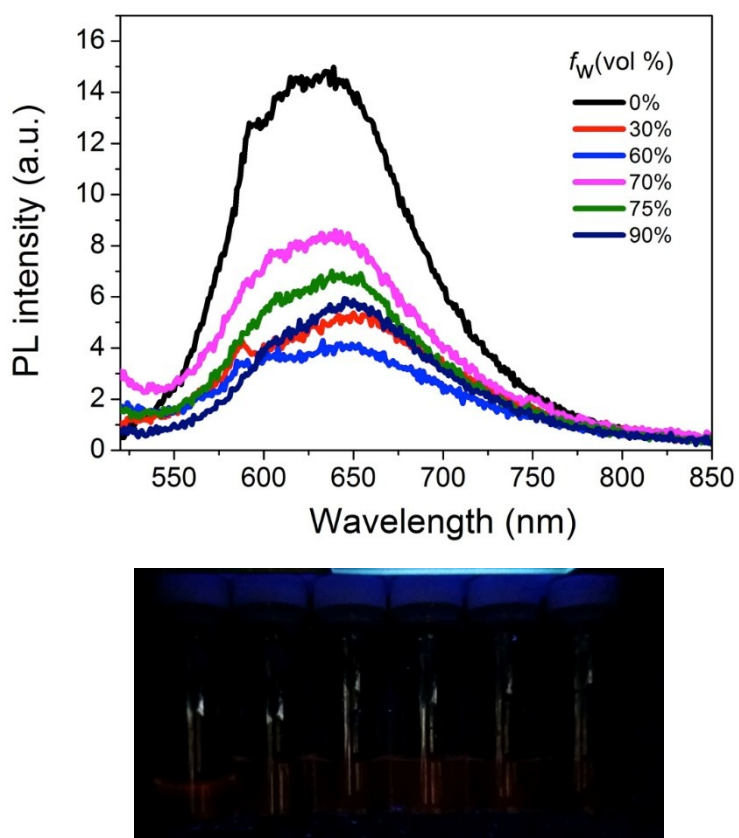


Figure S6. Emission spectra of **7b** in THF–water mixtures. The concentration of pure THF is 10^{-4} M. Excitation wavelength is 510 nm. Photographs of **7b** in THF–water mixtures with different f_w values (from left to right: 0%, 30%, 60%, 70%, 75% and 90%) taken under UV illumination at 365 nm.

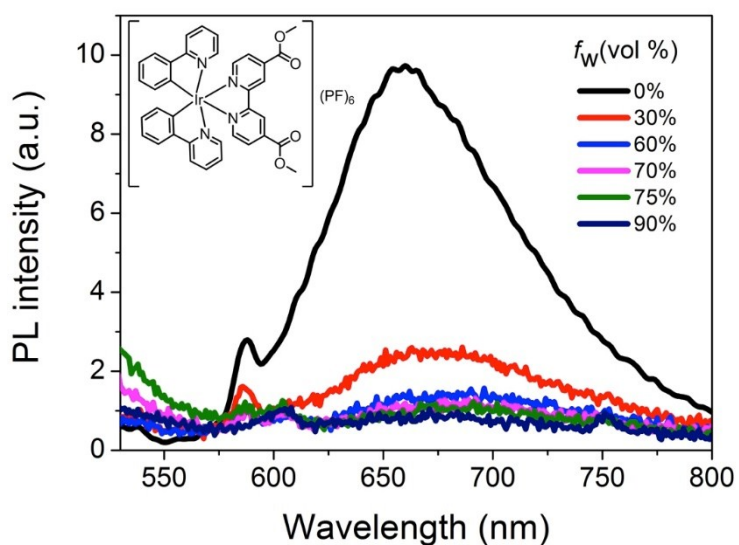


Figure S7. Emission spectra of $[\text{Ir}(\text{ppy})_2(\text{deeb})][\text{PF}_6]$ in THF–water mixtures. The concentration of pure THF is 10^{-4} M. Excitation wavelength is 510 nm.



Figure S8 The emission of complex **7a** under UV light (365 nm) at room temperature. The solution is the CH₂Cl₂:hexane (V/V = 1:1, 10⁻³ M). After the CH₂Cl₂ volatilized, the sample was self-assembled in hexane solution

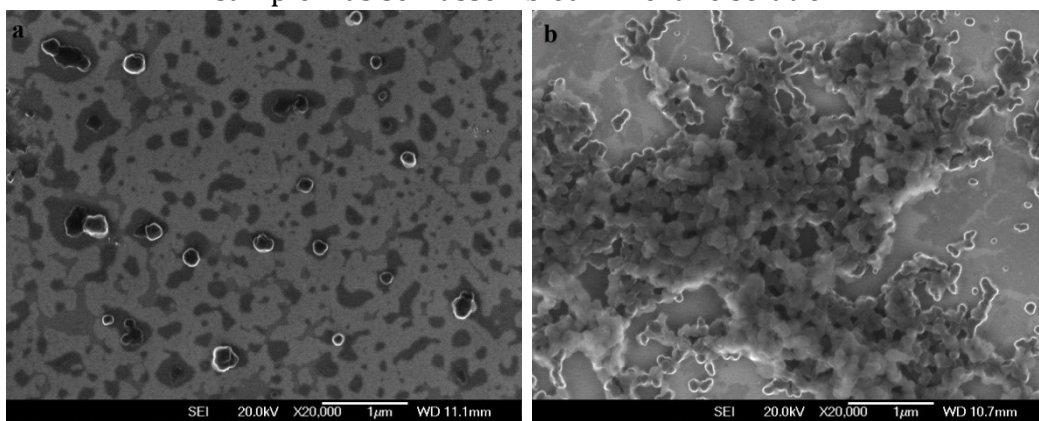


Figure S9. SEM images of **7a**, prepared in a THF–water mixture with (a) 30% and (b) 90% water content.

NMR spectra and MALDI-MS spectra

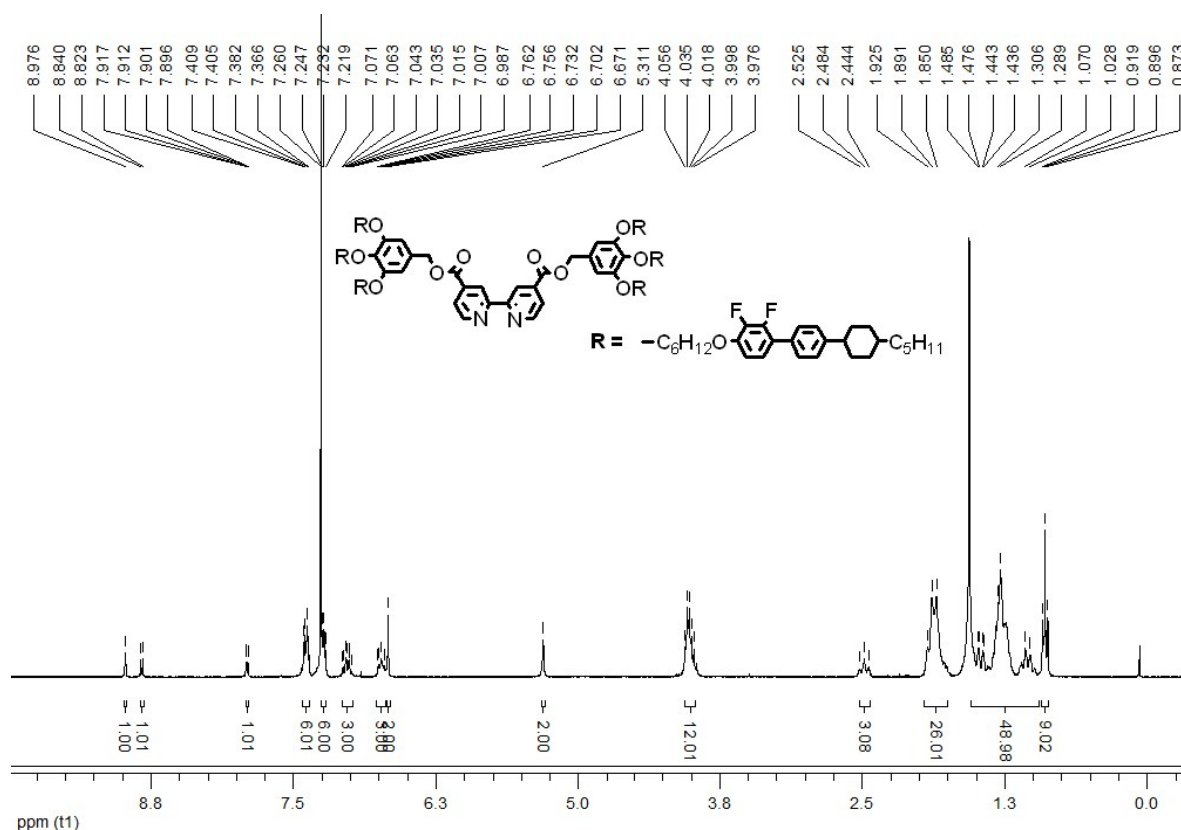


Figure S7. ¹H NMR spectra of **6a** in CDCl₃

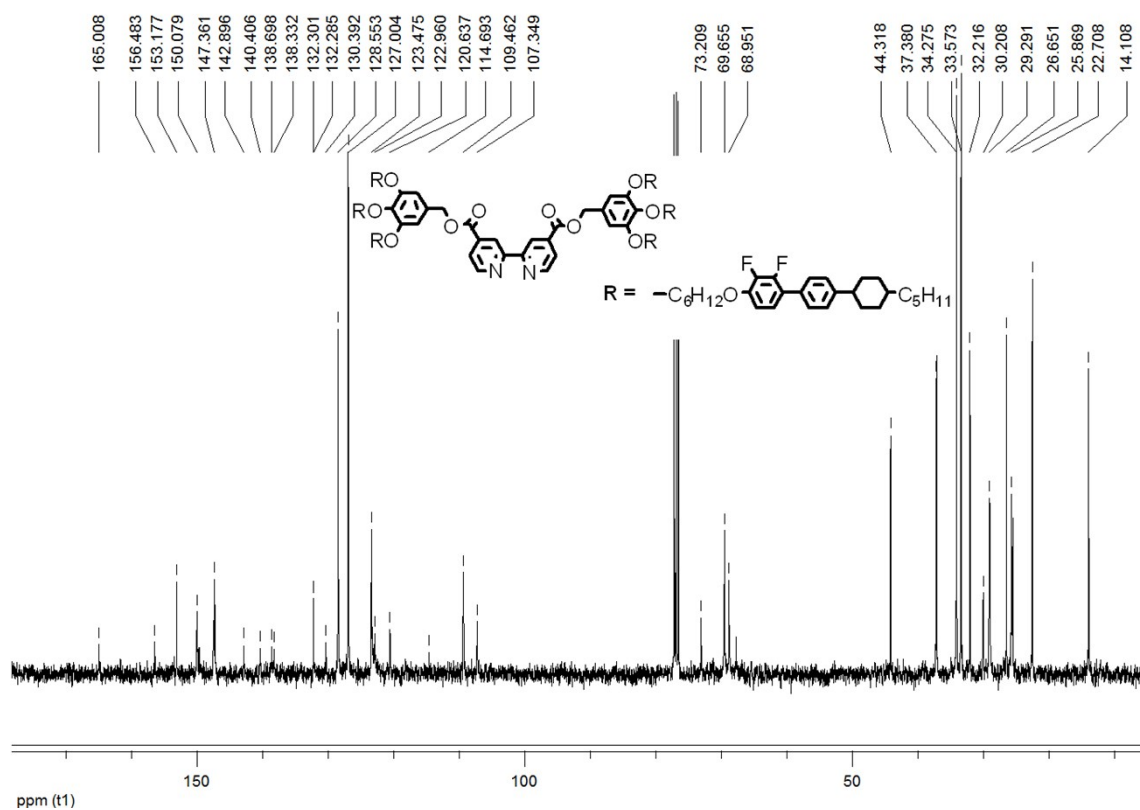
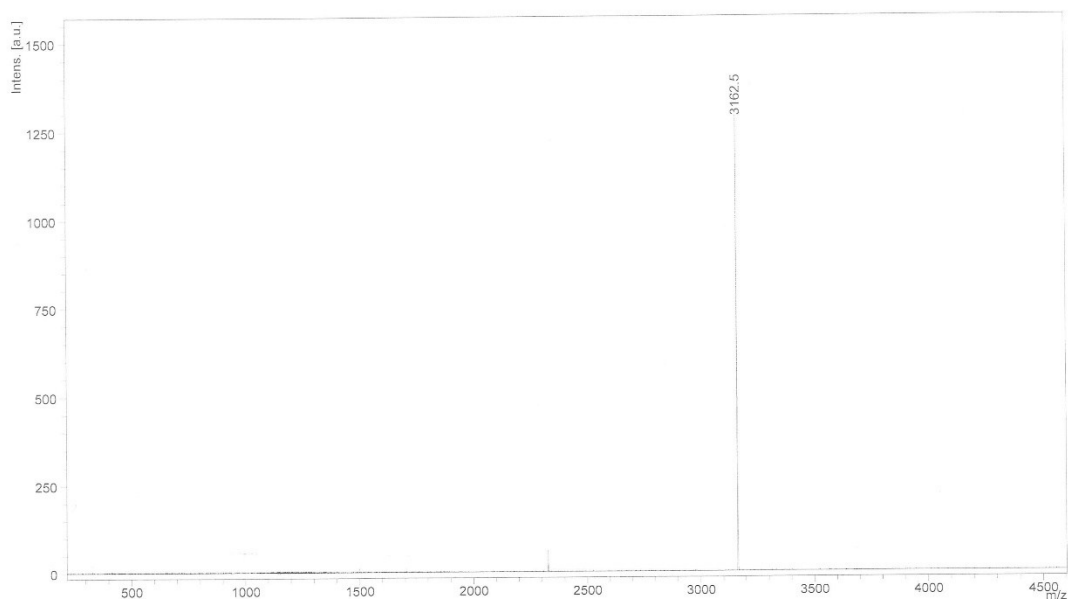
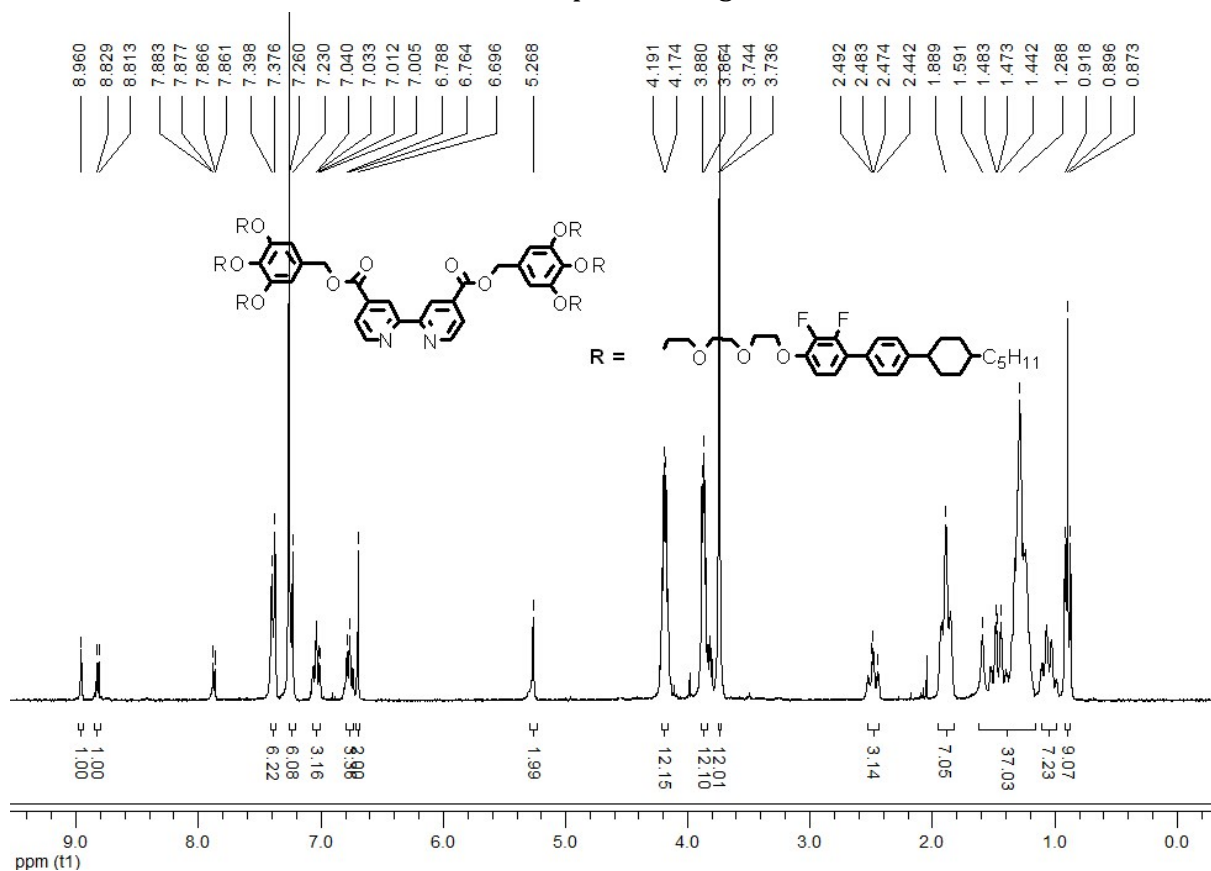


Figure S8. ¹³C NMR spectra of **6a** in CDCl₃

MALDI-TOF, CCA, 6A-C6, 20180124



MALDI-MS spectra of ligand 6a

Figure S9. ¹H NMR spectra of **6b** in CDCl₃



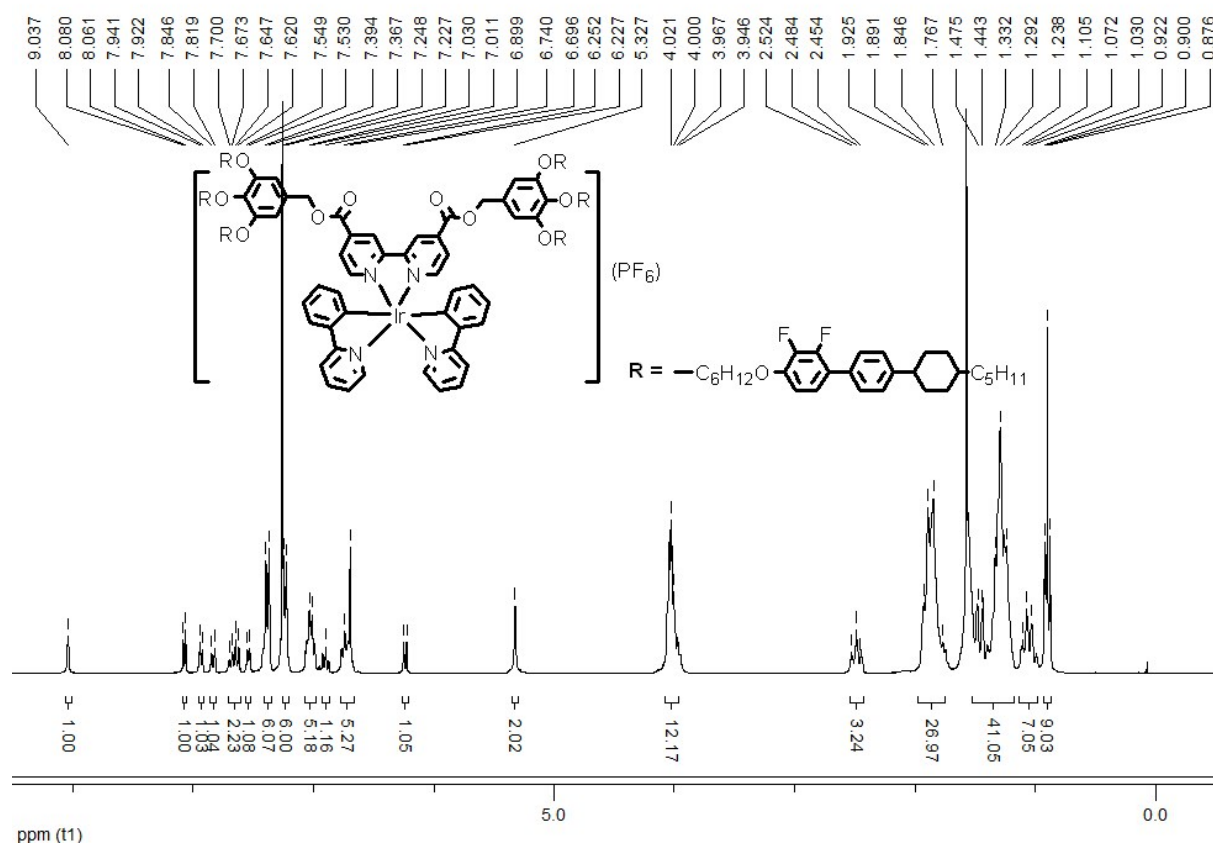


Figure S11. ¹H NMR spectra of **7a** in CDCl₃

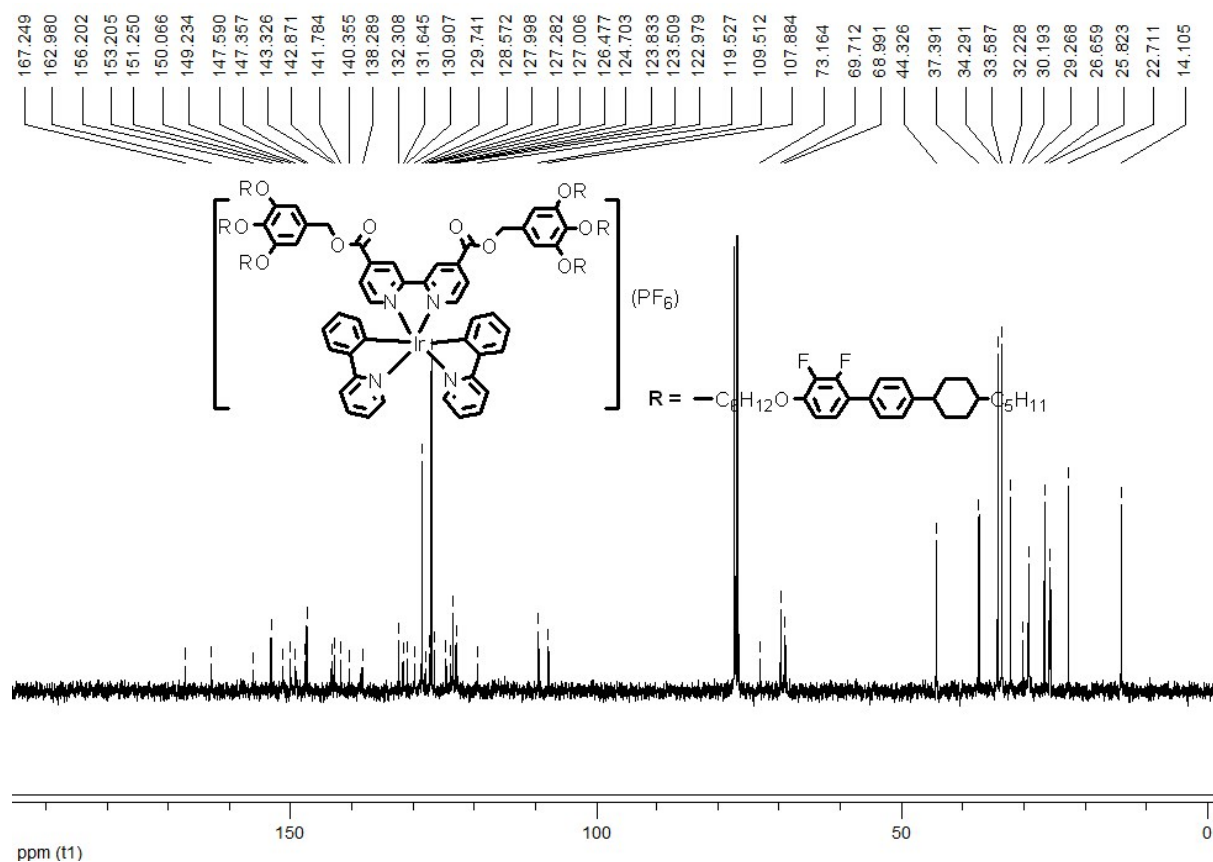


Figure S12. ¹³C NMR spectra of **7a** in CDCl₃

MALDI-MS spectra of ligand 7a

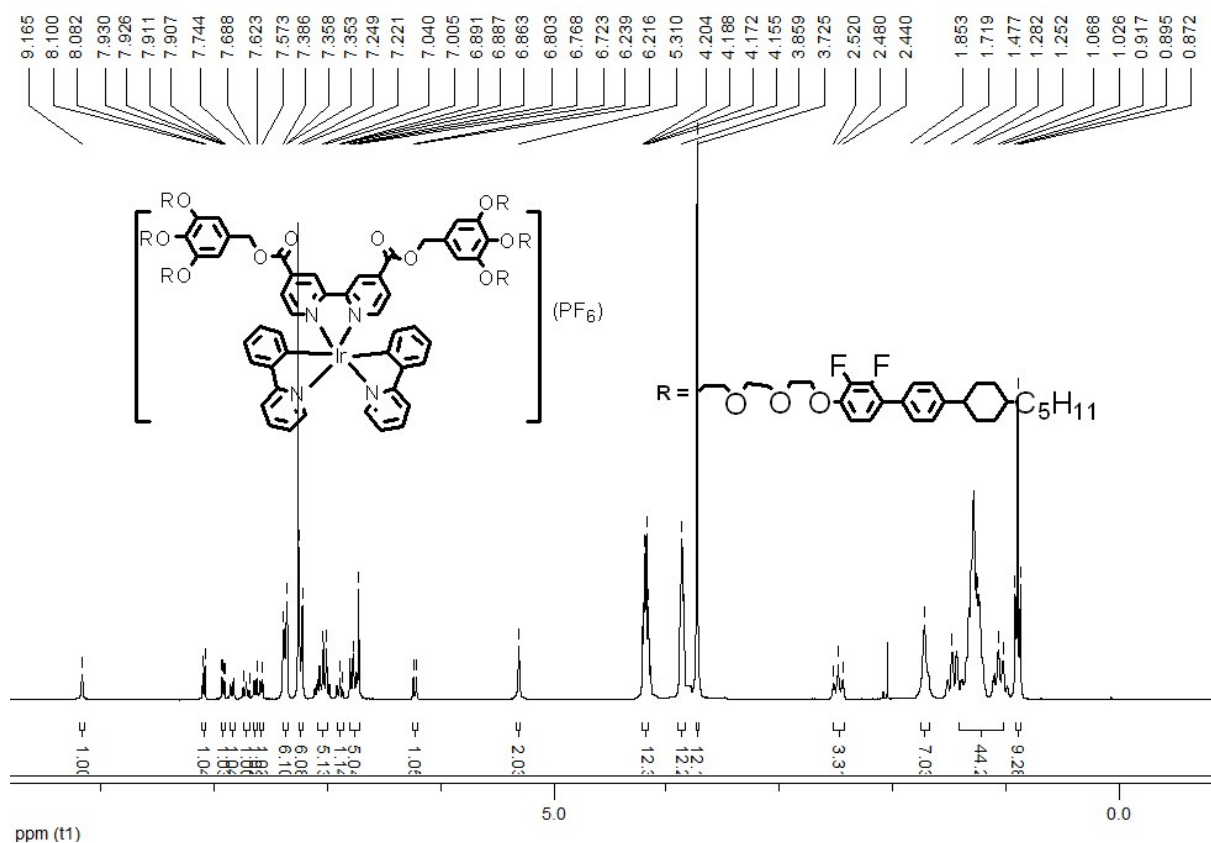


Figure S13. ^1H NMR spectra of **7b** in CDCl_3

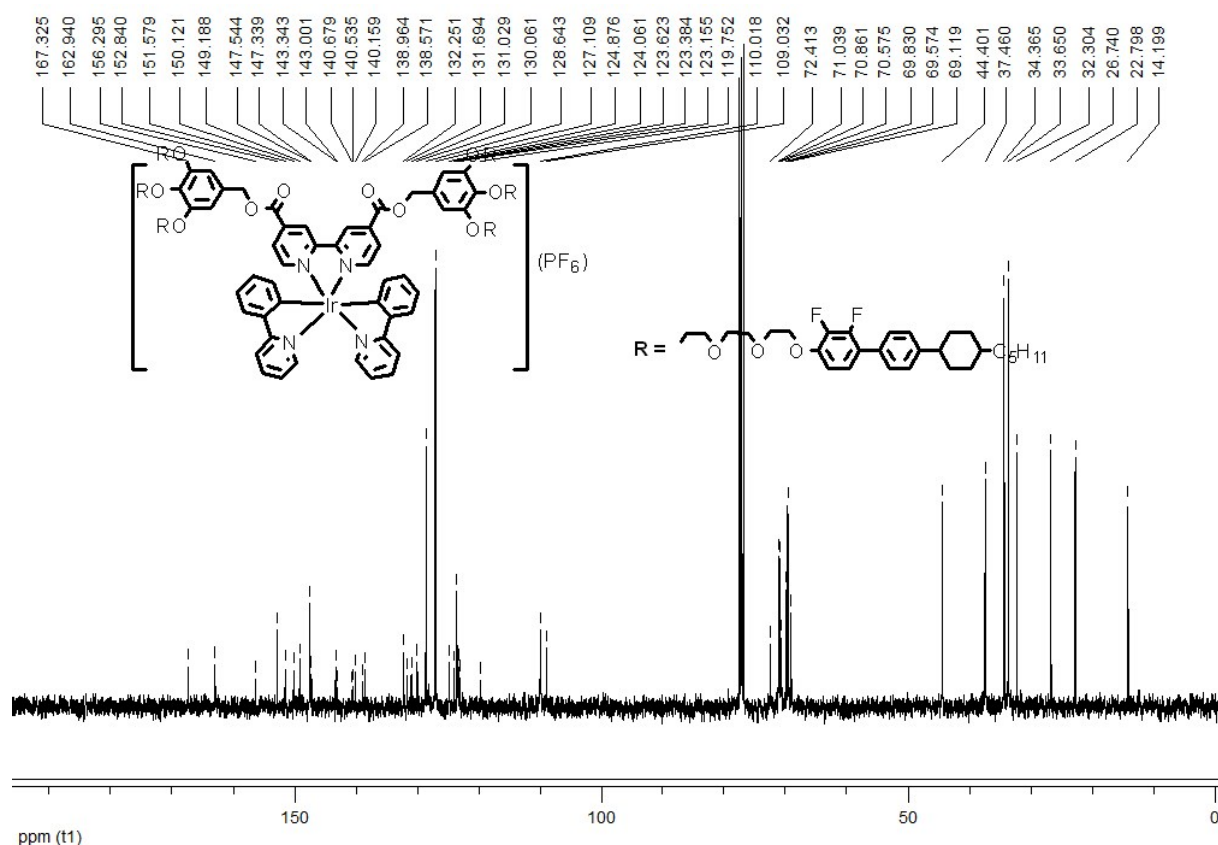
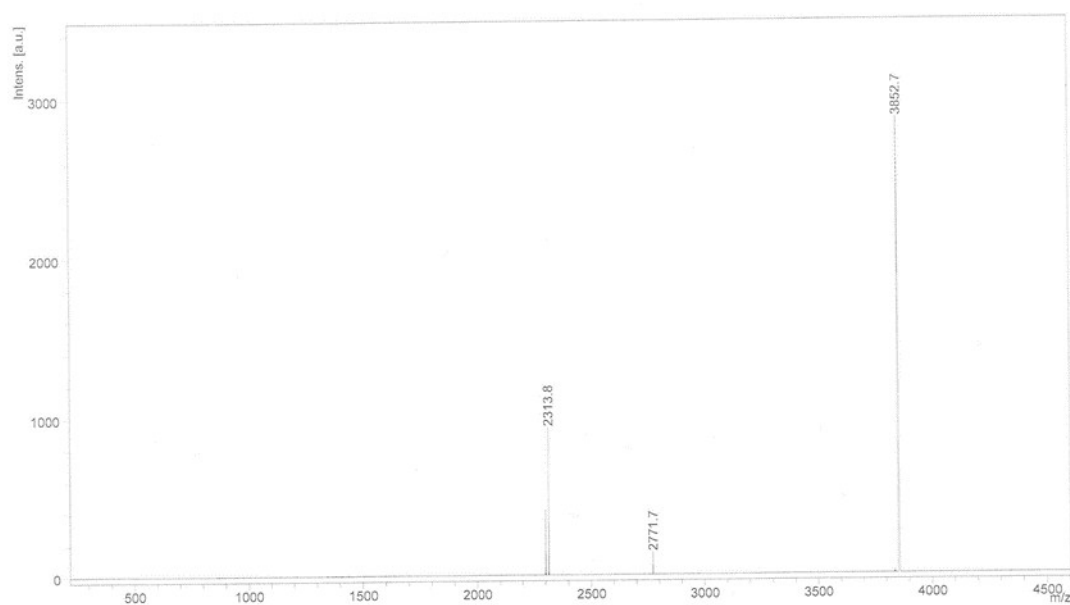


Figure S14. ^{13}C NMR spectra of **7b** in CDCl_3

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MALDI-TOF,CCA,7B-C6,20180124



MALDI-MS spectra of ligand **7b**

Table S1. Electroluminescent data of EL devices.

Devices	Complex	Emission /nm	<i>R</i>	Luminance /cd m ⁻²	Current efficiency /cd A ⁻¹	EQE %	<i>V</i> _{turn-on} V
I	7a	640	N/A	2.5	0.01	0.03	8.8
	7b	647	1.1	2	0.04	0.01	9.2
II	7a	661	1.4	20	0.57	0.85	6
	7b	707	1.1	2	0.1	0.36	12
III	7a	641	1.2	5	1.1	1.1	8.5
	7b	642	N/A	34	0.1	0.09	5
IV	7a	640	1.1	4	0.03	0.04	12
	7b	678/660	4	1.2	0.01	0.01	5.5