

Electronic Supplementary Information

Ratiometric strategy based electrochemical sensing interface for imidacloprid sensitive and reliable detection

Xueyan Li, Xianwen Kan*

College of Chemistry and Materials Science, Anhui Normal University, Wuhu
241000, P.R. China; The Key Laboratory of Functional Molecular Solids, Ministry
of Education; Anhui Laboratory of Molecule-Based Materials, Anhui Key
Laboratory of Chemo-Biosensing.

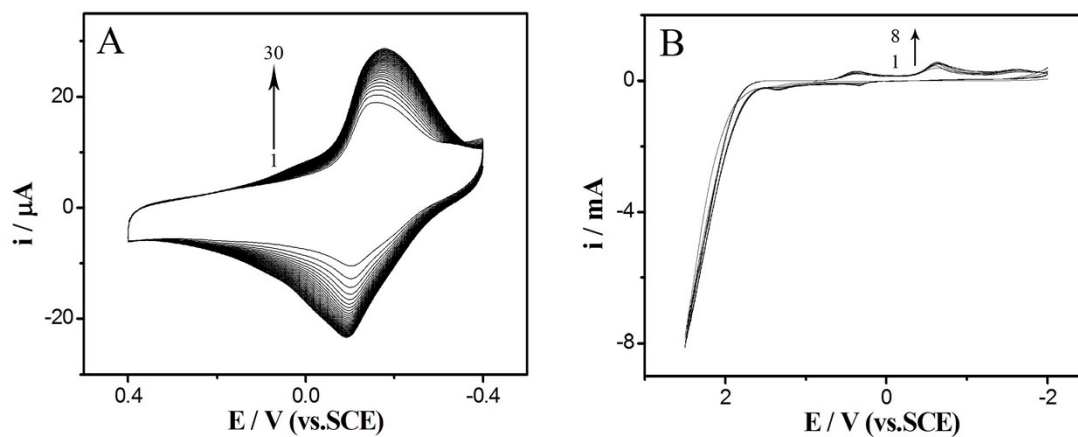


Fig. S1 Electropolymerization CV curves of THI (A) and CD (B).

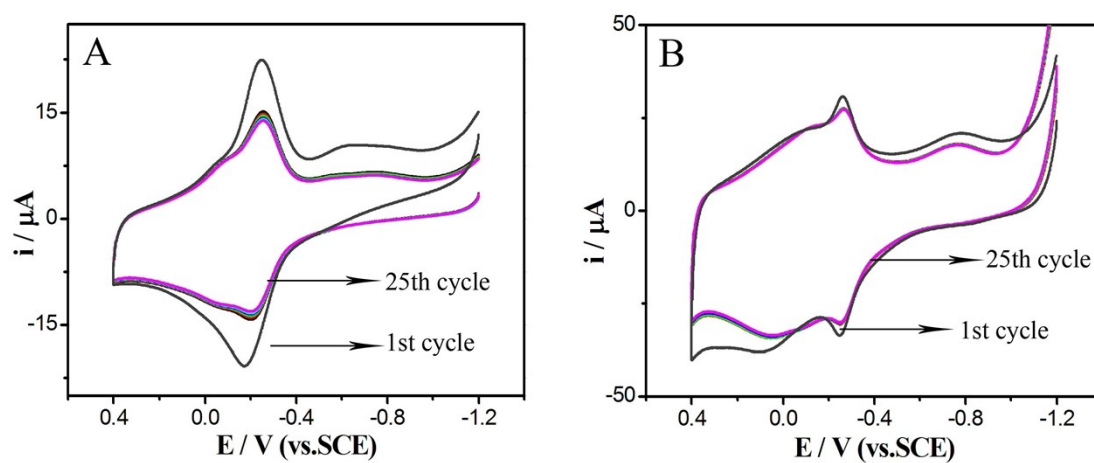


Fig. S2 30-cycle CV curves of pTHI/GCE (A) and pCD/pTHI/GCE (B) in 0.1 mol/L PBS (pH=6.0).

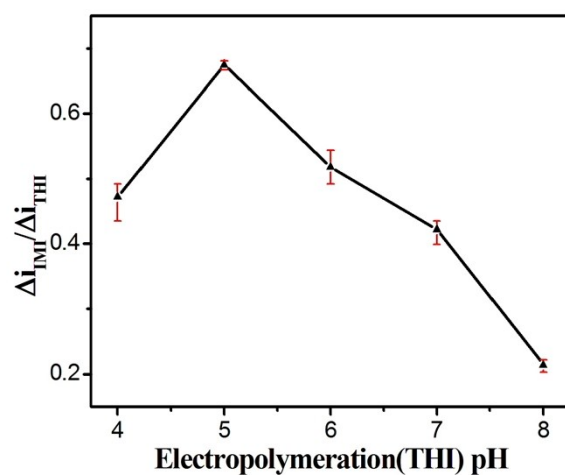


Fig. S3 The effect of pH value of electropolymerization solution.

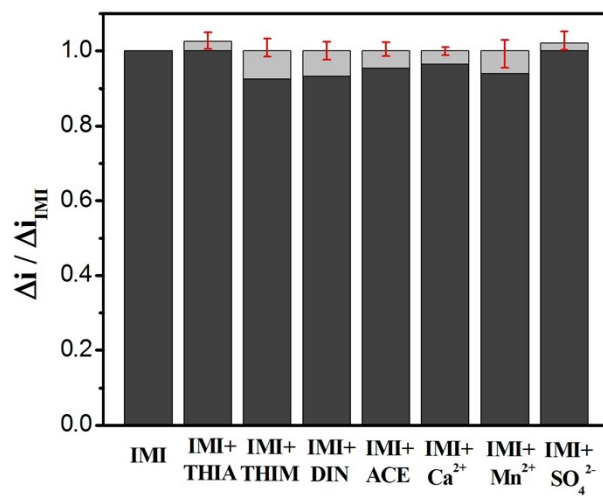


Fig. S4 Selectivity of the sensor. Current responses of 5.0×10^{-6} mo/L THIA, 5.0×10^{-6} mo/L THIM, 5.0×10^{-6} mo/L DIN, 5.0×10^{-6} mo/L ACE, 5.0×10^{-4} mo/L Ca²⁺, 5.0×10^{-4} mo/L Mn²⁺ or 5.0×10^{-4} mo/L SO₄²⁻, the ratios of current responses on the sensor calculated by current recorded in 5.0×10^{-6} mo/L IMI.

Table S1 Comparison of other electrochemical methods for IMI detection with our work.

Modified electrode	linear range (mol/L)	LOD (mol/L)	Reference
Graphene modified GCE	8.0×10^{-7} - 1.0×10^{-5}	3.6×10^{-7}	1
Electrochemically pretreated boron-doped diamond electrode	3.0×10^{-5} - 2.0×10^{-4}	8.6×10^{-6}	2
Graphene/imprinted polymer modified GCE	5.0×10^{-7} - 1.5×10^{-5}	1.0×10^{-7}	3
Nanosilver Nafion/nanoTiO ₂ Nafion modified GCE	5.0×10^{-7} - 3.5×10^{-6}	2.5×10^{-7}	4
β -CD polymer/ reduced graphene oxide modified GCE	5.0×10^{-8} - 1.5×10^{-5}	2.0×10^{-8}	5
Poly(carbazole)/reduced graphene oxide modified GCE	4.4×10^{-7} - 1.5×10^{-6}	2.2×10^{-7}	6
Nitrogen-doped grapheme modified GCE	4.0×10^{-6} - 1.0×10^{-4}	5.5×10^{-7}	7
Ag nanodendrimers/grapheme modified GCE	1.0×10^{-6} - 1.0×10^{-4}	8.1×10^{-7}	8
Pt-In catalytic nanoparticles /Bromophenol blue doped imprinted polymer modified GCE	2.0×10^{-10} - 5.0×10^{-8}	1.2×10^{-11}	9
Ionic liquid modified carbon-ceramic electrode	5.0×10^{-8} - 7.0×10^{-6}	3.1×10^{-8}	10
Graphene oxide modified GCE	1.0×10^{-5} - 2.0×10^{-4}	7.9×10^{-6}	11
DMIP/PTH/MWNTs/GCE modified GCE	1.0×10^{-7} - 1.0×10^{-4}	6.5×10^{-8}	12
Ionic liquid functionalized gold nanoparticles probe	1.0×10^{-6} - 1.0×10^{-5}	5.0×10^{-7}	13
Molecularly imprinted sensor	7.5×10^{-7} - 7.0×10^{-5}	4.0×10^{-7}	14
MIP/GN modified GCE	5.0×10^{-6} - 1.5×10^{-5}	1.0×10^{-7}	15
pCD/pTHI/GCE	4.0×10^{-8} - 1.0×10^{-5}	1.7×10^{-8}	This work

References

- 1 W. Lei, Z. Han, W. Si, Q. Hao, Y. Zhang, M. Xia, F Wang, ChemElectroChem, 2014, 1, 1063.
- 2 M.B. Brahim, H.B. Ammar, R. Abdelhédi, Y. Samet, Chin. Chem. Lett., 2016, 27, 666.
- 3 M. Zhang, H.T. Zhao, T.J. Xie, X. Yang, A.J. Dong, H. Zhang, J. Wang, Z.Y. Wang,

- Sens. Actuators B, 2017, 252, 991.
- 4 A. Kumaravel, M. Chandrasekaran, Sens. Actuators B, 2011, 158, 319.
 - 5 M. Chen, Y. Meng, W. Zhang, J. Zhou, J. Xie, G. Diao, Electrochim. Acta, 2013, 108, 1.
 - 6 W. Lei, Q. Wu, W. Si, Z. Gu, Y. Zhang, J. Deng, Q. Hao, Sens. Actuators B, 2013, 183, 102.
 - 7 W. Si, W. Lei, Q. Hao, X. Xia, H. Zhang, J. Li, Q. Li, R. Cong, Electrochim. Acta, 2016, 212, 784.
 - 8 M.R. Majidi, S. Ghaderi, J. Electroanal. Chem., 2017, 792, 46.
 - 9 S. Li, C. Liu, G. Yin, J. Luo, Z. Zhang, Microchim. Acta, 2016, 183, 3109.
 - 10 M.R. Majidi, R.F.B. Baj, M. Bamorowat, Measurement, 2016, 93, 29.
 - 11 V. Urbanová, A. Bakandritsos, P. Jakubec, Biosens. Bioelectron., 2017, 89, 532.
 - 12 Y. Dai, X. Kan, Chem. Commun., 2017, 53, 11755.
 - 13 X. Zhang, Z. Sun, Z. Cui, H. Li, Sens. Actuators B, 2014, 191, 313.
 - 14 L. Kong, X. Jiang, Y. Zeng, T. Zhou, G. Shi, Sens. Actuators B, 2013, 185, 424.
 - 15 M. Zhang, H.T. Zhao, T.J. Xie, Sens. Actuators B, 2017, 252, 991.