## **Supporting Information**

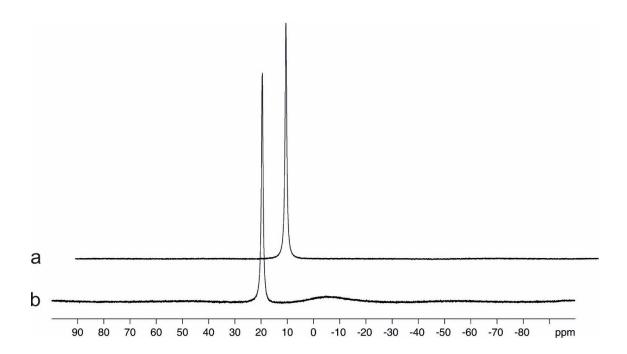
Pushing the frontiers: Boron-11 NMR as a Method for Quantitative Boron Analysis and its Application to Determine Boric Acid in Commercial Biocides

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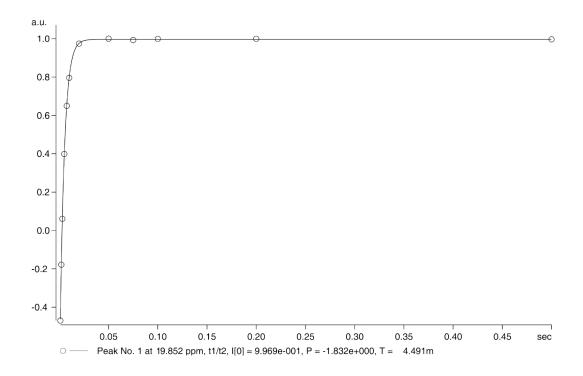
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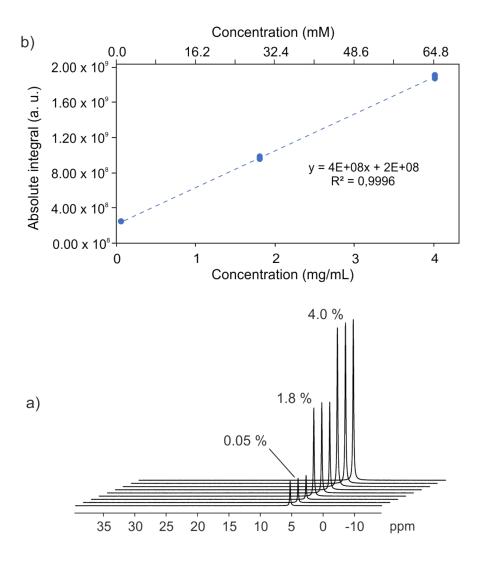
Figure S1. <sup>11</sup> B qNMR spectra (96.294 MHz) measured in D <sub>2</sub> O of a commercial biocide using a
quartz and b) standard 5 mm NMR tubes
<b>Figure S2</b> . <sup>11</sup> B T1 inversion-recovery experiment (294 K, 96.294 MHz) of boric acid in $D_2O$ at a concentration of 1.0 % (w/w).
Figure S3. a) <sup>11</sup> B NMR spectra (96.294 MHz) of borax solutions in D2O at 294 K at pH 13.
Experimental time of each spectrum of 50 seconds; b) Calibration curve stack plot of borax
solutions in pure D2O at pH 13. Values coming from three replicates are shown.



**Figure S1**.  $^{11}$ B qNMR spectra (96.294 MHz) measured in  $D_2$ O of a commercial biocide using a) quartz and b) standard 5 mm NMR tubes.



**Figure S2**. <sup>11</sup>B T1 inversion-recovery experiment (294 K, 96.294 MHz) of boric acid in  $D_2O$  at a concentration of 1.0 % (w/w).



**Figure S3**. a) <sup>11</sup>B NMR spectra (96.294 MHz) of borax solutions in D2O at 294 K at pH 13. Experimental time of each spectrum of 50 seconds; b) Calibration curve stack plot of borax solutions in pure D2O at pH 13. Values coming from three replicates are shown.