

Electronic Supplementary Information (ESI)

Continuous Amperometric Hydrogen Gas Sensing in Ionic Liquids†

Yongan Tang,^a Jianxin He,^b Xiaoli Gao,^c Tianbao Yang^b and Xiangqun Zeng^{a*}

^aDepartment of Chemistry, Oakland University, Rochester, Michigan 48309, USA

^bDepartment of Computer Science, The University of Iowa, Iowa City, IA 52242, USA

^cDepartment of Mathematics and Statistics, University of North Carolina Greensboro, NC 27412,
USA

Figure S1 summarized the IL-based amperometric hydrogen sensor responses to hydrogen when the gas sampling times is 10 s. **Table S1** describes the different analytical methods for quantitative analysis of hydrogen concentrations. **Figure S2~S4** detailed the three linear regression models based on the data sets of Figure 6, respectively. **Figure S5~S6** and **Table S2** summarized different hydrogen gas sampling times and the analytical figure of merits for hydrogen detection using the reported IL-based amperometric hydrogen sensor.

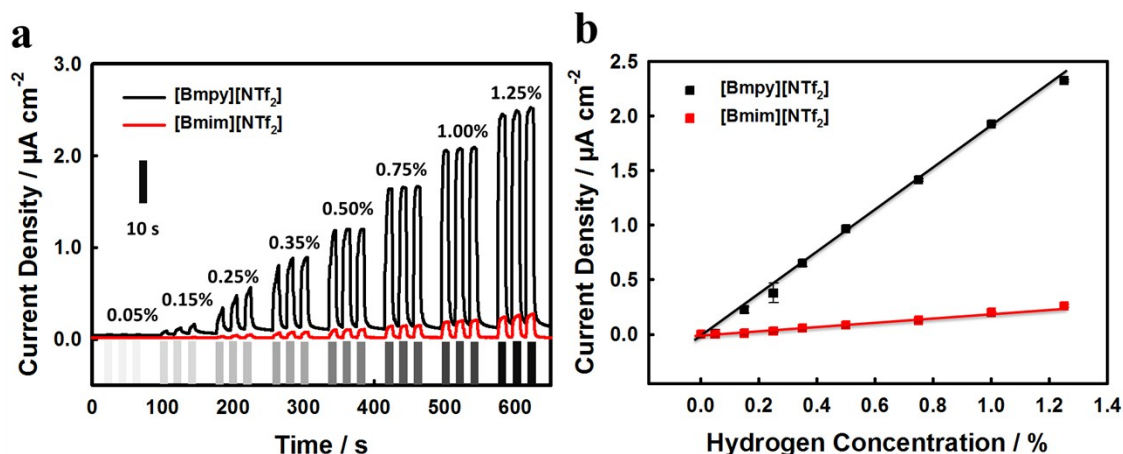


Figure S1A. (a) continuous sensing hydrogen using the amperometric hydrogen sensors in [Bmpy][NTf₂] (black) and [Bmim][NTf₂] (red), constant potential at $E=+0.4$ V vs. Fc⁺/Fc is the potential used for the hydrogen sensing, sampling time is 10 seconds for each measurement. (b) Calibration curves for concentration range in 0.05% ~ 1.25% hydrogen in two types of IL-based sensors at 10 second exposure time in Figure a.

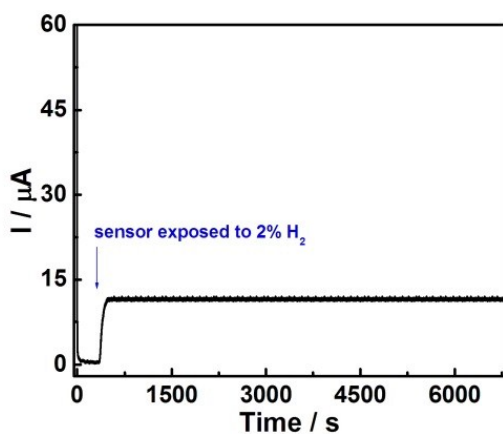


Figure S1B. Constant potential chronoamperometric recording of sensing response toward 2% hydrogen exposure starting at the 300th second (marked by the downward arrow).

Electrolyte: [Bmpy][NTf₂]; flowrate: 100 sccm.

Table S1. Parameters summary of different hydrogen sensors

Method	Detection range	Speed of respond	Sensitivity	LOD/LOQ	Ref
Solid electrolyte amperometric H ₂ sensor	10 ~ 40 ppm	< 1 min	N/A	10 ppm	21
Capacitive H ₂ sensor	40 ~ 400 ppm	N/A	N/A	40 ppm	23
Optical H ₂ sensor	0.5% ~ 4.0%	15 s	N/A	0.5%	22
Aqueous based electrochemical H ₂ sensor	0.4% ~ 4.0%	< 1 s	0.16 $\mu\text{A/ppm}$	0.4%	27
Ionic liquid based electrochemical H ₂ sensor	0.5% ~ 1.25%	< 1 s	3.08 $\mu\text{A cm}^{-2} \%^{-1}$	0.5%	Current work

Table S2. Data analysis approaches

Methods	Description	Training Data	Testing Data
Method 1	Train one model on Data1-10s and Data2-10s and test on Data3-10s	Data1-10s, Data2-10s	Data3-10s
Method 2	Train two models on Data1-10s and Data2-10s, compute a new model based on the difference between the two models and the model for Data2-10s.	Data1-10s, Data2-10s	Data3-10s
Method 3	Train one model on Data3-10s	Data3-10s	Data3-10s

Strategy 1: Predicting concentration based on summit of current

Strategy 2: Predicting the concentration based on current/time

Strategy 3: Predicting the concentration based on (current/square root of time)

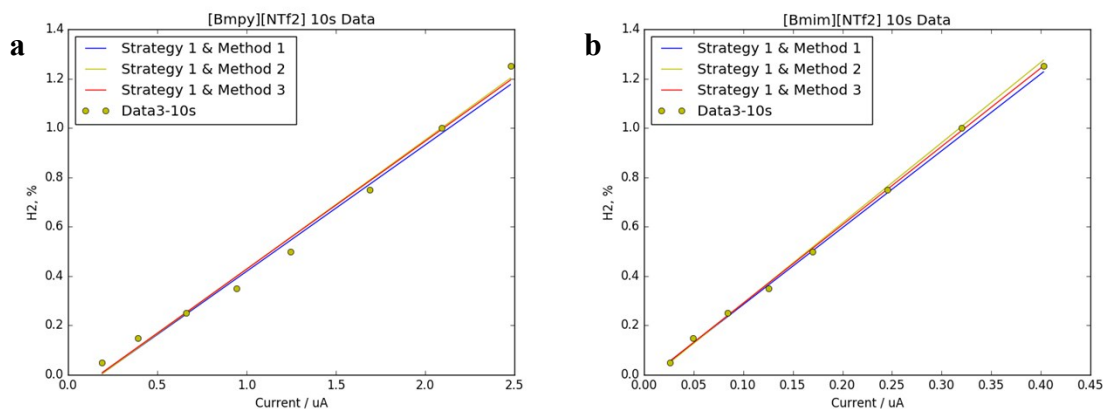


Figure S2. Curve for predicting hydrogen concentration based on summit of current by using strategy 1 in (a) [Bmpy][NTf₂], (b) [Bmim][NTf₂].

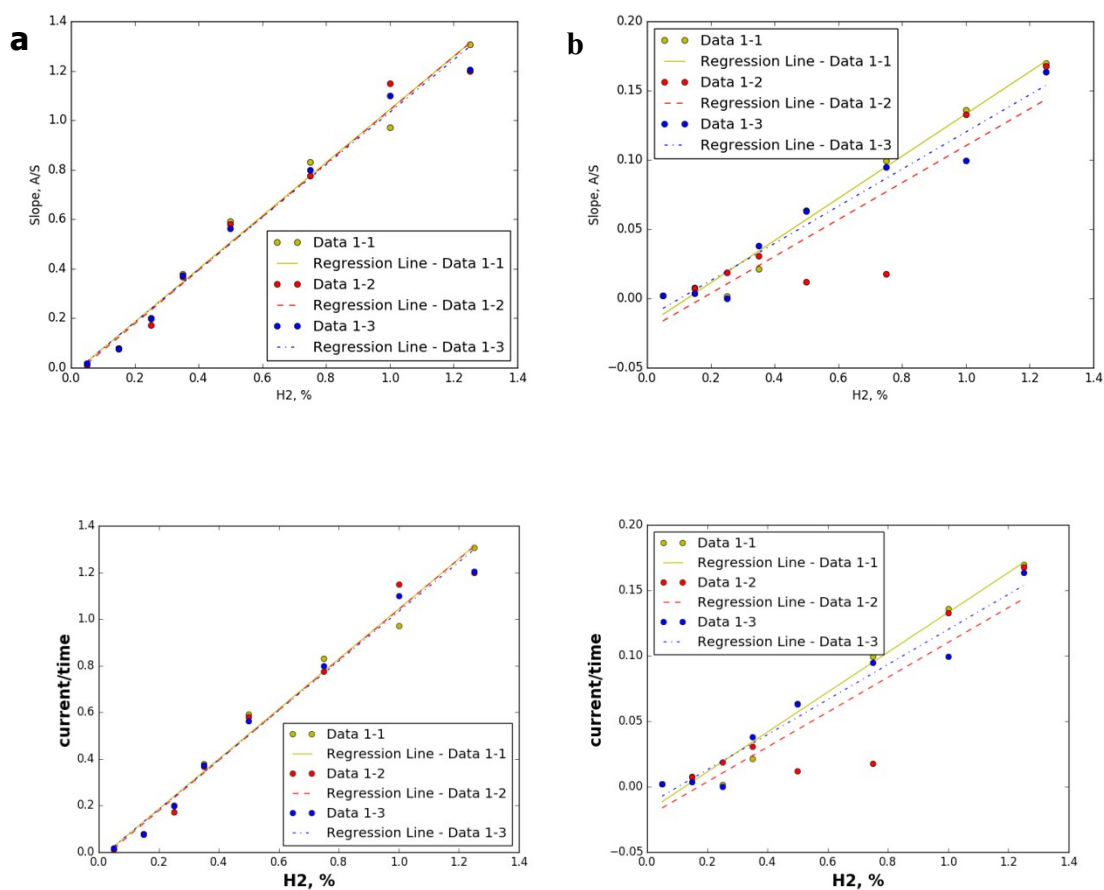


Figure S3. Curve for predicting hydrogen concentration based on current/time by using strategy 2 in (a) [Bmpy][NTf₂], (b) [Bmim][NTf₂].

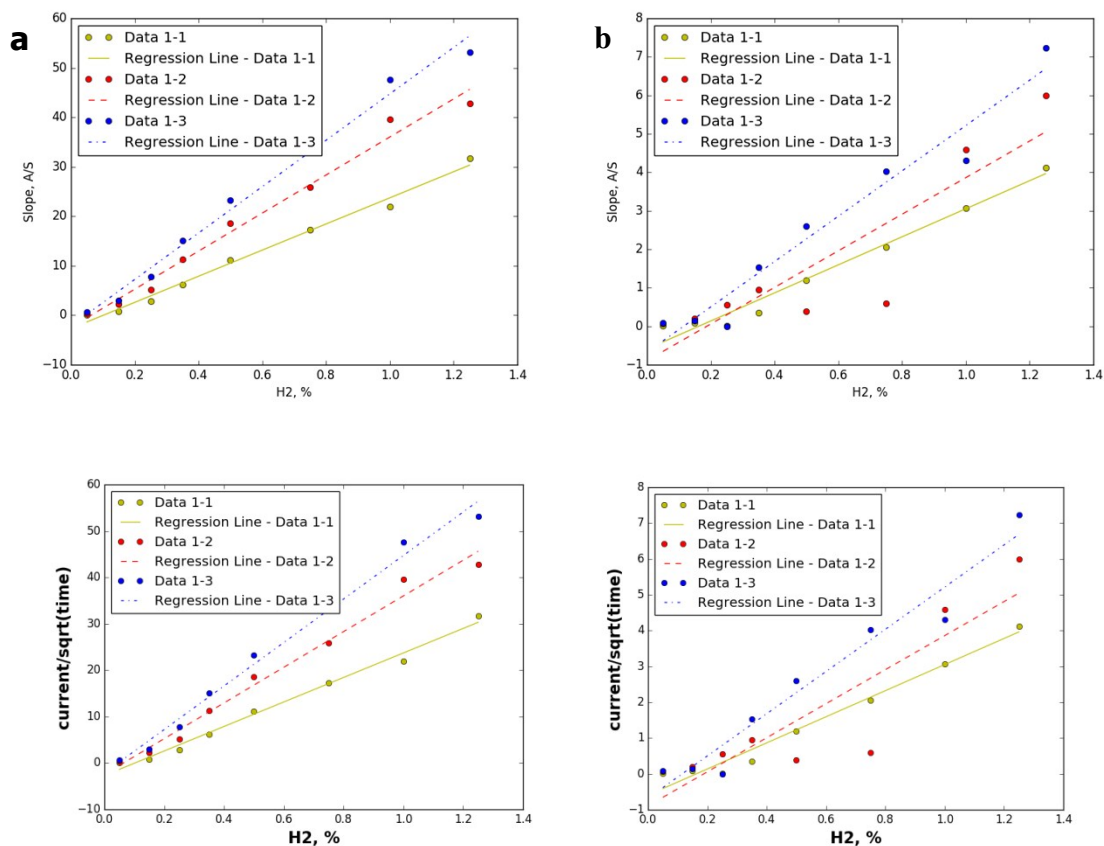


Figure S4. Curve for predicting hydrogen concentration based on (current/square root of time) by using strategy 3 in (a) [Bmpy][NTf₂]; (b) [Bmim][NTf₂].

Reference

1. Z. Wang, M. Guo, G. A. Baker, J. R. Stetter, L. Lin, A. J. Mason and X. Zeng, *Analyst*, 2014, **139**, 5140-5147.
2. Y. A. Tang, X. W. Chi, S. Z. Zoub and X. Q. Zeng, *Nanoscale*, 2016, **8**, 5771-5779.