Electronic Supplementary Information for

## Selectivity Enhancement of Amperometric Nitric Oxide Detection via Shape-Controlled

## **Electrodeposition of Platinum Nanostructures**

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Scheme S1 Illustration of feasible electrodeposition processes on an Au substrate electrode with relatively (A) lower and (B) higher concentrations of  $PtCl_4^{2-}$ .



Fig. S1 Water droplets (1  $\mu$ L) loaded on the Pt surfaces electrodeposited with (A) 5 mM K<sub>2</sub>PtCl<sub>4</sub>, (B) 15 mM K<sub>2</sub>PtCl<sub>4</sub>, (C) 40 mM K<sub>2</sub>PtCl<sub>4</sub>, (D) 75 mM K<sub>2</sub>PtCl<sub>4</sub> and (E) 75 mM H<sub>2</sub>PtCl<sub>6</sub>.

**Table S1** Statistical significances of the differences in the electrode sensitivities to NO (Figure 3B). p < 0.05; p < 0.01 (paired *t*-test, n = 5). In paired t-test, p-value less than 0.05 represents that the mean difference between the paired observations is significantly different from 0.

5 mM PtCl <sub>4</sub> <sup>2–</sup>	-		_		
15 mM PtCl <sub>4</sub> <sup>2–</sup>	*	-			
40 mM PtCl <sub>4</sub> <sup>2-</sup>	*	**	-		
75 mM PtCl <sub>4</sub> <sup>2–</sup>	**	**	•	-	
75 mM PtCl <sub>6</sub> <sup>2–</sup>	**	**	**	*	-
	5 mM PtCl <sub>4</sub> <sup>2–</sup>	15 mM PtCl <sub>4</sub> <sup>2–</sup>	40 mM PtCl <sub>4</sub> <sup>2–</sup>	75 mM PtCl <sub>4</sub> <sup>2–</sup>	75 mM PtCl <sub>6</sub> <sup>2-</sup>

**Table S2** Selectivity coefficients  $( \begin{array}{c} \log K_{NO,i}^{Amp} = \log \left( \frac{Current \ sensitivity \ to \ i}{Current \ sensitivity \ to \ NO} \right), i = interfering species) to NO against acetaminophen (AP, 5 <math>\mu$ M), L-ascorbic acid (AA, 5  $\mu$ M), nitrite (NO<sub>2</sub><sup>-</sup>, 5  $\mu$ M) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>, 5  $\mu$ M) for Pt electrodes variously deposited and a flat bare Pt disk electrode (2 mm in diameter).

	AP	AA	NO <sub>2</sub> -	H <sub>2</sub> O <sub>2</sub>
5 mM PtCl <sub>4</sub> <sup>2–</sup>	-1.879	-1.815	-2.641	-1.770
15 mM PtCl <sub>4</sub> <sup>2–</sup>	-1.822	-1.801	-2.594	-1.541
40 mM PtCl <sub>4</sub> <sup>2-</sup>	-1.483	-1.548	-2.208	-1.274
75 mM PtCl <sub>4</sub> <sup>2–</sup>	-1.191	-1.208	-2.093	-1.067
75 mM PtCl <sub>6</sub> <sup>2–</sup>	-0.6451	-0.6208	-1.412	-0.487
Flat bare Pt	-0.3779	-0.4878	-1.406	-0.446



**Fig. S2** Typical dynamic response curves to successive NO concentration changes obtained at a Pt electrode deposited with 5 mM PtCl<sub>4</sub><sup>2–</sup> with and without (A) human serum albumin (HSA, 10 mg mL<sup>-1</sup>) and (B)  $\gamma$ -globulin. (C) and (D) are the calibration curves corresponding to (A) and (B), respectively.