

Supplementary Information

Two Aggregation induced emission (AIE)-active Reaction-Type Probes: for Real-Time Detecting and Imaging Superoxide Anions

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Table of Contents

| | |
|--|----|
| Synthesis of TPA-CHO-1~3, TPA-DHP-1~3 and TPA-PPA-1~3..... | 2 |
| Scheme S1~2. Synthesis of TPA-CHO-1~3 | |
| Scheme S3~5. Synthesis of TPA-DHP-1~3 | |
| Scheme S5~8. Synthesis of TPA-PPA-1~3 | |
| Figure S1. The NMR data spectra of TPA-CHO-1..... | 9 |
| Figure S2. The NMR data spectra of TPA-CHO-2..... | 10 |
| Figure S3. The NMR data spectra of TPA-CHO-3..... | 11 |
| Figure S4. The NMR data spectra of TPA-DHP-1..... | 12 |
| Figure S5. The NMR data spectra of TPA-DHP-2..... | 13 |
| Figure S6. The NMR data spectra of TPA-DHP-3..... | 14 |
| Table S1. Photophysical Properties..... | 15 |
| Figure S7. PL spectra of solid condition..... | 15 |
| Figure S8. AIE performance of TPA-DHP-1~3..... | 16 |
| Figure S9. Solvent-induced effect of TPA-DHP-1~3..... | 17 |
| Figure S10. Theoretical calculations of all compounds..... | 18 |
| Figure S11. Reaction mechanism..... | 19 |
| Figure S12. Theoretical calculations before and after oxidation..... | 20 |
| Figure S13. Cyclic voltammograms..... | 21 |
| Figure S14. Structural characterization before and after oxidation..... | 21 |
| Figure S15. Cell viability of HeLa cells..... | 21 |

Synthesis and characterization of TPA-CHO-1, TPA-CHO-2, TPA-CHO-3.

Synthesis of 4-(diphenylamino)benzaldehyde (TPA-CHO-1).

Triphenylamine (4.91 g, 20 mmol) was dissolved in DMF (20 mL) and placed in a 100 mL flask. Phosphorous oxychloride (10 mL) was added dropwise in ice bath, and the reaction mixture was stirred for 10 min at 0 °C. And then the mixture was refluxed at 80 °C for 3 h under N₂ atmosphere. Then, reaction was quenched with cold water (300 mL) and yellow solid was precipitated. The crude product was purified over a silica gel column with mixture (ethyl acetate /petroleum ether, 1:50) as eluent to give TPA-CHO-1 as a yellow solid (yield: 85%). ¹H NMR (600 MHz, CDCl₃) δ (TMS, ppm): 9.80 (s, 1H), 7.67 (d, J = 8.8 Hz, 2H), 7.37 – 7.30 (m, 4H), 7.23 – 7.08 (m, 6H), 7.01 (d, J = 8.7 Hz, 2H). ¹³C NMR (150 MHz, CDCl₃) δ (TMS, ppm): 190.38 (s), 153.33 (s), 146.14 (s), 131.26 (s), 129.70 (s), 129.10 (s), 126.33 (s), 125.03 (s), 119.26 (s).

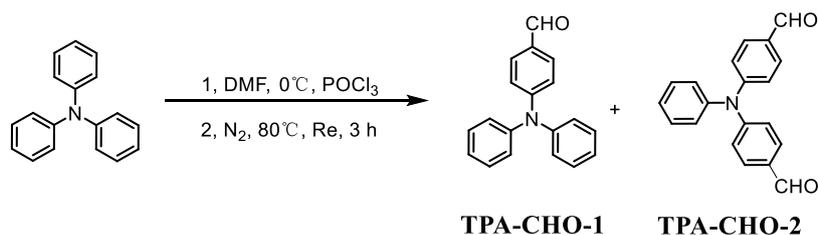
Synthesis of 4,4'-(phenylazanediyl)dibenzaldehyde (TPA-CHO-2).

Triphenylamine (4.91 g, 20 mmol) was dissolved in DMF (20 mL) and placed in a 100 mL flask. Phosphorous oxychloride (20 mL) was added dropwise in ice bath, and the reaction mixture was stirred for 10 min at 0 °C. And then the mixture was refluxed at 80 °C for 3 h under N₂ atmosphere. Then, reaction was quenched with cold water (300 mL) and yellow solid was precipitated. The crude product was purified over a silica gel column with mixture (ethyl acetate /petroleum ether, 1:20) as eluent to give TPA-CHO-2 as a yellow solid (yield: 76 %). ¹H NMR (600 MHz, CDCl₃) δ (TMS, ppm): 9.88 (s, 2H), 7.77 (d, J = 8.7 Hz, 4H), 7.39 (t, J = 7.9 Hz, 2H), 7.25 (t, J = 7.4 Hz, 1H), 7.17 (dd, J = 7.9, 5.9 Hz, 6H). ¹³C NMR (150 MHz, CDCl₃) δ (TMS, ppm): 190.46 (s), 151.98 (s), 145.48 (s), 131.27 (s), 130.12 (s), 127.03 (s), 126.23 (s), 122.74 (s).

Synthesis of 4,4',4''-nitriлотribenzaldehyde (TPA-CHO-3).

TPA-CHO-2 (2.87 g, 10 mmol) was dissolved in DMF (10 mL) and placed in a 100 mL flask. Phosphorous oxychloride (10 mL) was added dropwise in ice bath, and the reaction mixture was stirred for 10 min at 0 °C. And then the mixture was refluxed at 80 °C for 5 h under N₂ atmosphere. Then, reaction was quenched with cold water (300 mL) and yellow solid was precipitated. The crude product was purified over a silica gel column with mixture (ethyl acetate /petroleum ether, 1:10) as eluent to give TPA-CHO-3 as a yellow solid (yield: 73%). ¹H NMR (600 MHz, CDCl₃) δ (TMS, ppm): 9.94 (s, 3H), 7.83 (d, J = 8.6 Hz, 6H), 7.24 (d, J = 8.5 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ (TMS, ppm): 190.42, 151.14, 132.58, 131.41, 124.51

Synthesis of TPA-CHO-1, 2, 3



Scheme S1 The synthesis of TPA-CHO-1 and TPA-CHO-2



Scheme S2 The synthesis of TPA-CHO-3

Synthesis and characterization of TPA-DHP-1, TPA-DHP-2, TPA-DHP-3.

Synthesis of diethyl 4-(4-(diphenylamino)phenyl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate (TPA-DHP-1)

TPA-CHO-1 (2 mmol, 0.546 g), Ethyl acetoacetate (2.4 mmol 0.3 mL), $\text{NH}_3 \cdot \text{H}_2\text{O}$ (10 mmol 0.4 mL) were dissolved in the EtOH (30 mL). And then the mixture was refluxed at 80 °C for 8 h under N_2 atmosphere. The reaction mixture was concentrated by rotary evaporation. Then, the TPA-DHP-1 purified by column chromatography on silica gel (300-400 mesh) with a mixture of petroleum ether and ethyl acetate as eluent (20:1 by volume), obtaining a yellow solid (0.903 g, 91% yield). ^1H NMR (600 MHz, d_6 -DMSO) δ (TMS, ppm): 8.75 (s, 1H), 7.21 (t, $J = 7.8$ Hz, 4H), 7.05 (d, $J = 8.5$ Hz, 2H), 6.95 (t, $J = 7.4$ Hz, 2H), 6.89 (d, $J = 7.8$ Hz, 4H), 6.83 (d, $J = 8.5$ Hz, 2H), 4.81 (s, 1H), 4.04-3.92 (m, 4H), 2.23 (s, 6H), 1.09 (t, $J = 7.1$ Hz, 6H). ^{13}C NMR (150 MHz, d_6 -DMSO) δ (TMS, ppm): 167.41 (s), 147.81 (s), 145.72 (s), 145.33 (s), 143.62 (s), 129.79 (s), 128.93 (s), 124.10 (s), 123.75 (s), 122.83 (s), 102.22 (s), 59.38 (s), 18.70 (s), 14.61 (s).

Synthesis of tetraethyl 4,4'-((phenylazanediy)bis(4,1-phenylene))bis(2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate) (TPA-DHP-2).

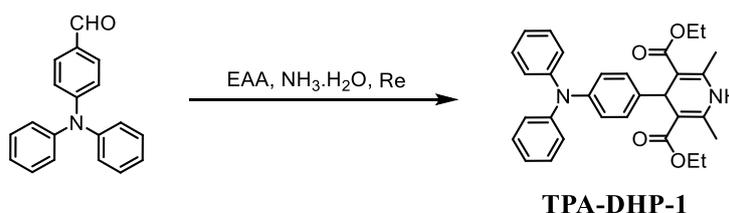
TPA-CHO-2 (2 mmol, 0.602 g), Ethyl acetoacetate (4.8 mmol 0.6 mL), $\text{NH}_3 \cdot \text{H}_2\text{O}$ (20 mmol 1 mL) were dissolved in the EtOH (30 mL). And then the mixture was refluxed at 80 °C for 10 h under N_2 atmosphere. The reaction mixture was concentrated by rotary evaporation. Then, the TPA-DHP-2 purified by column chromatography on silica gel (300-400 mesh) with a

mixture of petroleum ether and ethyl acetate as eluent (10:1 by volume), obtaining a yellow solid (1.227 g, 82% yield). ¹H NMR (600 MHz, *d*₆-DMSO) δ (TMS, ppm): 8.72 (s, 2H), 7.16 (t, J = 7.8 Hz, 2H), 6.99 (d, J = 8.4 Hz, 4H), 6.89 (t, J = 7.3 Hz, 1H), 6.81 (d, J = 7.9 Hz, 2H), 6.73 (d, J = 8.4 Hz, 4H), 4.79 (s, 2H), 3.97 (dd, J = 53.4, 10.8, 7.1 Hz, 8H), 2.22 (s, 12H), 1.08 (t, J = 7.1 Hz, 12H). ¹³C NMR (150 MHz, *d*₆-DMSO) δ (TMS, ppm): 167.40 (s), 147.99 (s), 145.68 (s), 145.42 (s), 143.27 (s), 129.63 (s), 128.81 (s), 123.80 (s), 123.17 (s), 122.39 (s), 102.21 (s), 59.35 (s), 18.71 (s), 14.57 (s).

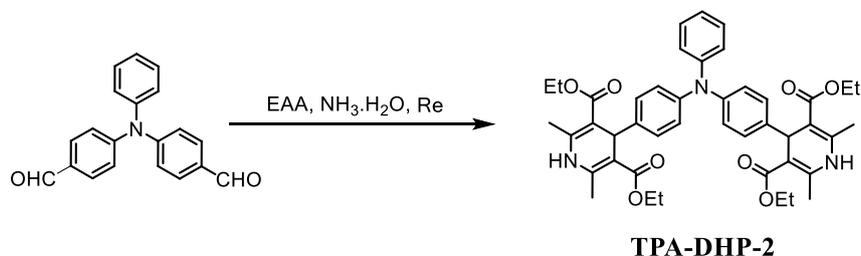
Synthesis of hexaethyl 4,4',4''-(nitrilotris(benzene-4,1-diyl)) tris(2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate) (TPA-DHP-3).

TPA-CHO-3 (2 mmol, 0.758 g), Ethyl acetoacetate (7.2 mmol 0.9 mL), NH₃·H₂O (20 mmol 1 mL) were dissolved in the EtOH (30 mL). And then the mixture was refluxed at 80 °C for 10 h under N₂ atmosphere. The reaction mixture was concentrated by rotary evaporation. Then, the TPA-DHP-3 purified by column chromatography on silica gel (300-400 mesh) with a mixture of petroleum ether and ethyl acetate as eluent (5:1 by volume), obtaining a yellow solid (1.497 g, 75% yield). ¹H NMR (600 MHz, *d*₆-DMSO) δ (TMS, ppm): 8.78 (s, 3H), 7.61 (d, J = 8.8 Hz, 6H), 6.66 (d, J = 8.7 Hz, 6H), 4.83 (s, 3H), 3.97 (dd, J = 46.9, 7.1 Hz, 12H), 2.23 (s, 18H), 1.07 (t, J = 7.1 Hz, 18H). ¹³C NMR (150 MHz, *d*₆-DMSO) δ (TMS, ppm): 167.24 (s), 147.48 (s), 145.97 (s), 131.62 (s), 129.28 (s), 126.47 (s), 117.37 (s), 102.00 (s), 59.40 (s), 18.68 (s), 14.56 (s).

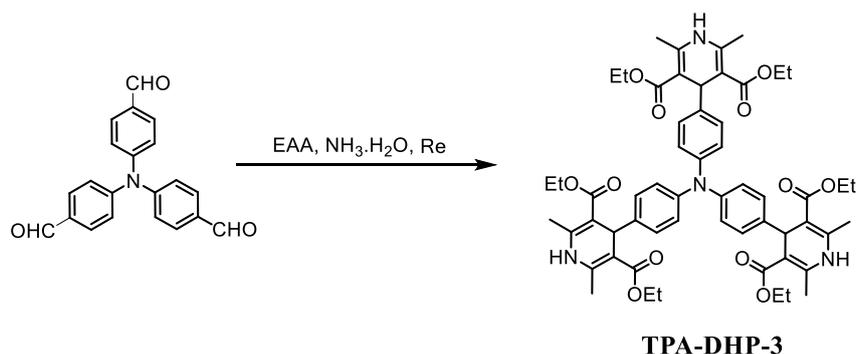
Synthesis of TPA-DHP-1, 2, 3



Scheme S3 The synthesis of TPA-DHP-1



Scheme S4 The synthesis of TPA-DHP-2



Scheme S5 The synthesis of TPA-DHP-3

Synthesis and characterization of TPA-PPA-1, TPA-PPA-2, TPA-PPA-3.

Synthesis and characterization of TPA-PPA-1.

4-bromotriphenylamine (4 mmol, 1.296 g), pyridine-4-boronic acid (4.2 mmol, 0.903 g), potassium carbonate (4 mmol, 0.552 g) and tetrakis(triphenylphosphine)palladium(0) (0.2 mmol, 5% eq., 0.231 g) were dissolved in the mixture of THF (30 mL) and MeOH (30 mL). And then the mixture was refluxed at 120 °C for 48 h under N₂ atmosphere. The reaction mixture was concentrated by rotary evaporation. Then, the N,N-diphenyl-4-(pyridin-4-yl)aniline purified by column chromatography on silica gel (300-400 mesh) with a mixture of petroleum ether and ethyl acetate as eluent (10:1 by volume), obtaining a white solid (1.1 g, 83% yield). Then, the white solid (2 mmol, 0.644 g), 4-bromomethylphenylboronic acid (2 mmol, 0.426 g) and THF (50 mL), were introduced into a clean round-bottom flask with a magnetic stirrer. Subsequently, the mixture was stirred at 90 °C for 48 h. After then, the reaction mixture was precipitated from THF, and washed using diethyl ether for several times. A yellow solid powder was obtained by filtration, and dried under vacuum at room temperature overnight. TPA-PPA-1 was a yellow solid (0.890 g, 87% yield). ¹H NMR (600 MHz, *d*₆-DMSO) δ (TMS, ppm): 9.03 (d, *J* = 6.9 Hz, 2H), 8.35 (d, *J* = 7.0 Hz, 2H), 7.96 (d, *J* = 8.9 Hz, 2H), 7.82 (d, *J* = 8.0 Hz, 2H), 7.45 (d, *J* = 7.9 Hz, 2H), 7.40 (d, *J* = 7.9 Hz, 4H), 7.21 (t, *J* = 7.4 Hz, 2H), 7.17 (d, *J* = 7.6 Hz, 4H), 6.93 (d, *J* = 8.9 Hz, 2H), 5.74 (s, 2H). ¹³C NMR (150 MHz, *d*₆-DMSO) δ (TMS, ppm): 154.45, 151.72, 146.00, 144.69, 136.69, 135.27, 130.46, 130.13, 127.93, 126.57, 125.75, 124.66, 123.28, 119.86, 62.36.

Synthesis and characterization of TPA-PPA-2.

Bis(4-bromophenyl)amine (3 mmol, 0.978 g), iodobenzene (3.3 mmol, 0.609 g), 1,10-phenanthroline (0.9 mmol, 0.178 g), CuI (0.3 mmol, 0.057 g), and potassium hydroxide (3 mmol, 0.168 g) were introduced to a 100 mL round-bottomed flask containing 60 mL of toluene under N₂ atmosphere. The reaction mixture was rapidly heated to the reflux temperature of 120 °C for 3 h. Then, the reaction mixture was cooled to 75 °C and extracted

by 200 mL of toluene and 150 mL of deionized water, respectively. And the organic phase was decolorized by activated carbon. The adsorbents were removed by hot filtration, and the solvent was removed by rotary evaporation. The product was subjected to further purified by column chromatography on silica gel (300-400 mesh) with a mixture of petroleum ether and ethyl acetate as eluent (5:1 by volume), obtaining a white solid (0.916 g, 76% yield). Then, the white solid (2 mmol, 0.804 g) was reacted with pyridine-4-boronic acid (4.4 mmol, 0.946 g) through Suzuki reaction. Therefore, the pure product was obtained (0.630 g, 79% yield). Finally, it was reacted with 4-bromomethylphenylboronic acid through salt-forming reaction. Reaction condition was similar to Synthesis of TPA-PP. TPA-PPA-2 was an orange solid (0.721 g, 87% yield). ¹H NMR (600 MHz, *d*₆-DMSO) (TMS, ppm): δ 9.23 – 9.04 (m, 4H), 8.53 - 8.39 (m, 4H), 8.14 (d, J = 18.1 Hz, 4H), 8.09 (t, J = 8.8 Hz, 4H), 7.84 (t, J = 7.0 Hz, 4H), 7.49 (d, J = 8.7 Hz, 4H), 7.48-7.40 (m, 2H), 7.33 (d, J = 12.9, 5.5 Hz, 1H), 7.27 (d, J = 6.2, 4.1 Hz, 2H), 7.22 (d, J = 8.8 Hz, 4H), 5.94-5.60 (m, 4H). ¹³C NMR (150 MHz, *d*₆-DMSO) δ (TMS, ppm): 154.36, 150.24, 144.96, 136.61, 135.27, 130.80, 130.33, 128.97, 127.97, 127.36, 126.95, 124.11, 123.48, 121.56, 116.31, 62.59.

Synthesis and characterization of TPA-PPA-3.

Tris(4-iodophenyl)amine (2 mmol, 1.246 g), pyridine-4-boronic acid (6.3 mmol, 1.354 g), potassium carbonate (6 mmol, 0.828 g) and tetrakis(triphenylphosphine)palladium(0) (0.3 mmol, 5% eq., 0.346 g) were dissolved in the mixture of THF (30 mL) and MeOH (30 mL). And then the mixture was refluxed at 120 °C for 48 h under N₂ atmosphere. The reaction mixture was concentrated by rotary evaporation. Then, the crude product purified by column chromatography on silica gel (300-400 mesh) with a mixture of petroleum ether and ethyl acetate as eluent (1:1 by volume), obtaining a white solid (0.599 g, 63% yield). Then, the white solid (1 mmol, 0.476 g), 4-bromomethylphenylboronic acid (3 mmol, 0.629 g) and THF (50 mL), were introduced into a clean round-bottom flask with a magnetic stirrer. Subsequently, the mixture was stirred at 90 °C for 48 h. After then, the reaction mixture was precipitated from THF, and washed using diethyl ether for several times. An orange powder was obtained by filtration, and dried under vacuum at room temperature overnight. TPA-PPA-3 was an orange solid (0.953 g, 85% yield). ¹H NMR (600 MHz, *d*₆-DMSO) δ (TMS, ppm): 9.16 (d, J = 6.9 Hz, 6H), 8.49 (d, J = 6.9 Hz, 6H), 8.12 (d, J = 8.5 Hz, 6H), 8.12 (s, 1H), 7.83 (d, J = 7.8 Hz, 6H), 7.48 (d, J = 4.7 Hz, 6H), 7.31 (d, J = 11.6 Hz, 6H), 5.81 (d, J = 9.1 Hz, 6H). ¹³C NMR (150 MHz, *d*₆-DMSO) δ (TMS, ppm): 154.33, 149.90, 145.02, 136.60, 135.28, 130.46, 129.24, 127.98, 125.32, 124.53, 124.32, 62.64.

^1H NMR (600 MHz, CDCl_3) δ (TMS, ppm): 9.80 (s, 1H), 7.67 (d, $J = 8.8$ Hz, 2H), 7.37 – 7.30 (m, 4H), 7.23 – 7.08 (m, 6H), 7.01 (d, $J = 8.7$ Hz, 2H).

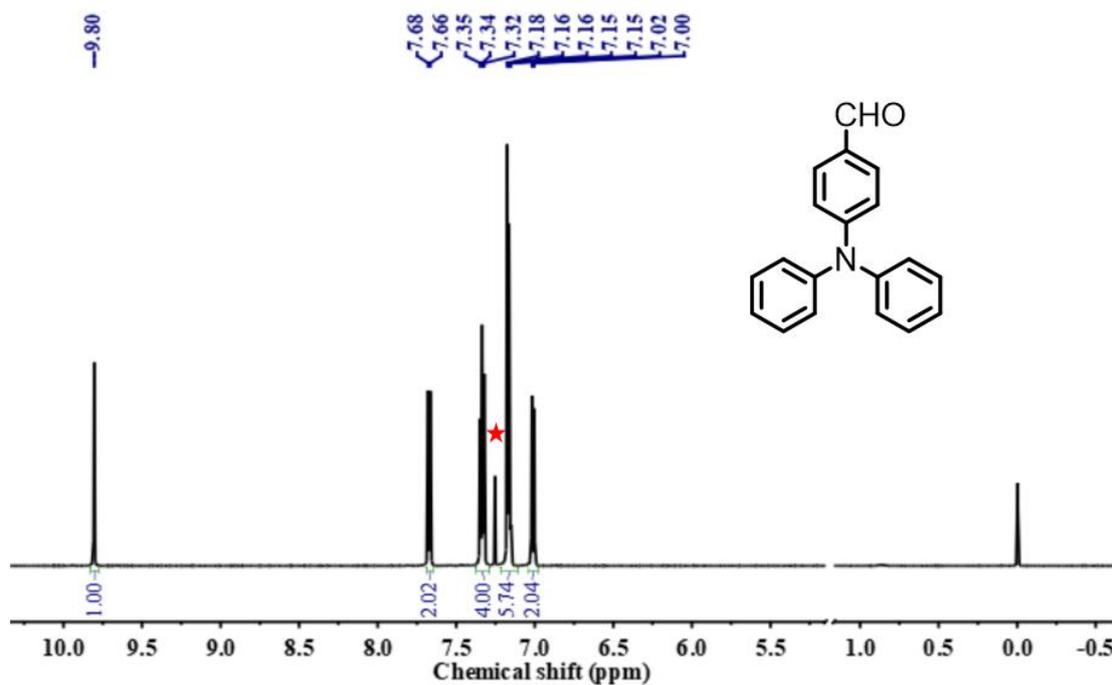


Figure S1-a. The ^1H NMR data spectrum of TPA-CHO-1

^{13}C NMR (150 MHz, CDCl_3) δ 190.38 (s), 153.33 (s), 146.14 (s), 131.26 (s), 129.70 (s), 129.10 (s), 126.33 (s), 125.03 (s), 119.26 (s).

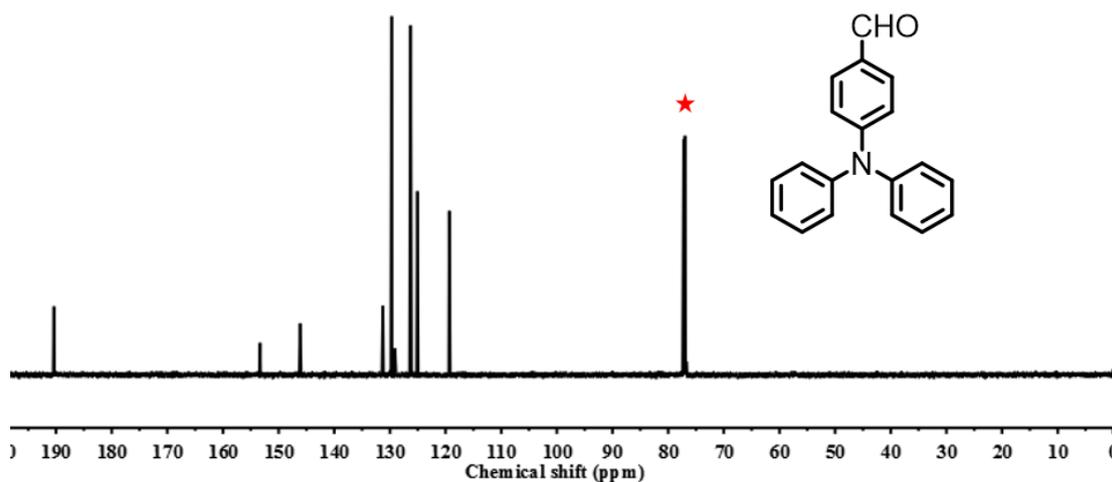


Figure S1-b. The ^{13}C NMR data spectrum of TPA-CHO-1

^1H NMR (600 MHz, CDCl_3) δ 9.88 (s, 2H), 7.77 (d, $J = 8.7$ Hz, 4H), 7.39 (t, $J = 7.9$ Hz, 2H), 7.25 (t, $J = 7.4$ Hz, 1H), 7.17 (dd, $J = 7.9, 5.9$ Hz, 6H).

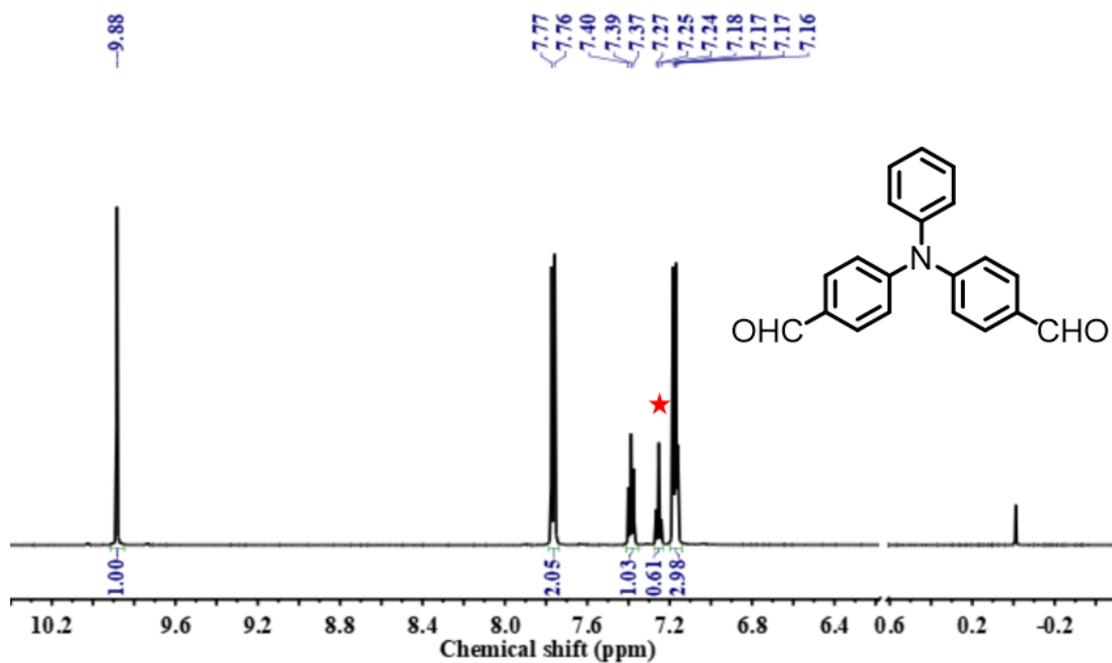


Figure S2-a. The ^1H NMR data spectrum of TPA-CHO-2

^{13}C NMR (150 MHz, CDCl_3) δ 190.46 (s), 151.98 (s), 145.48 (s), 131.27 (s), 130.12 (s), 127.03 (s), 126.23 (s), 122.74 (s).

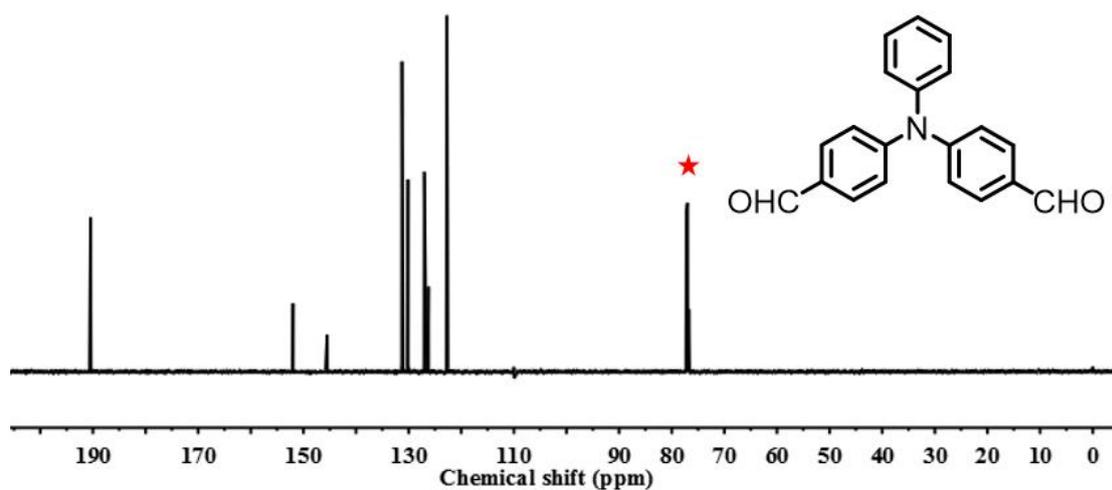


Figure S2-b. The ^{13}C NMR data spectrum of TPA-CHO-2

^1H NMR (600 MHz, CDCl_3) δ 9.94 (s, 3H), 7.83 (d, $J = 8.6$ Hz, 6H), 7.24 (d, $J = 8.5$ Hz, 6H).

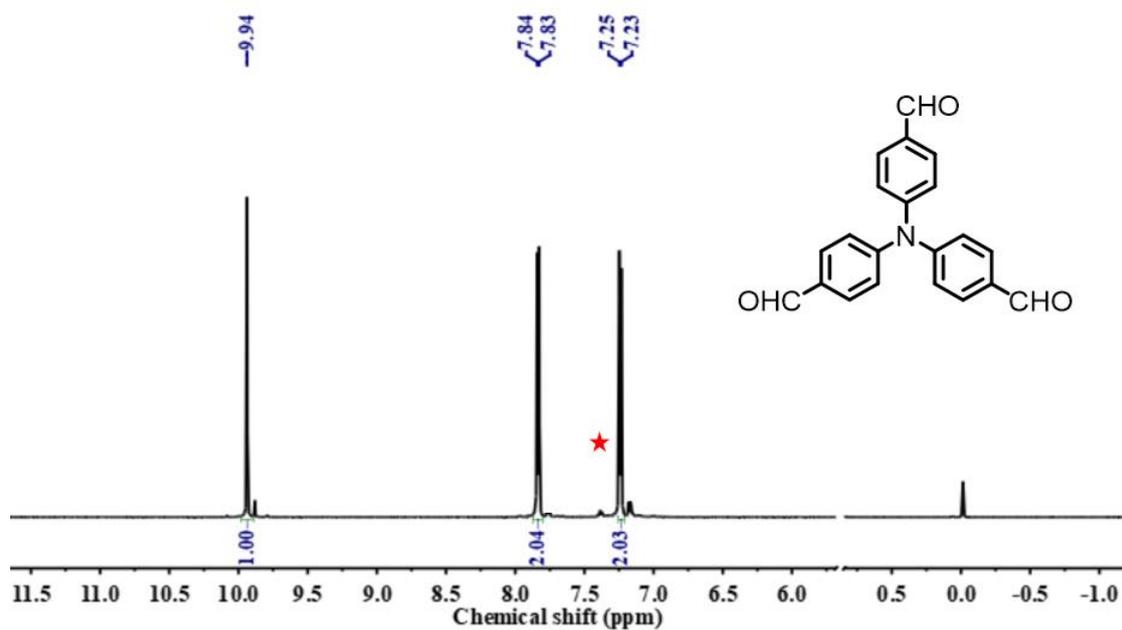


Figure S3-a. The ^1H NMR data spectrum of TPA-CHO-3

^{13}C NMR (150 MHz, CDCl_3), d (TMS, ppm): 190.42, 151.14, 132.58, 131.41, 124.51

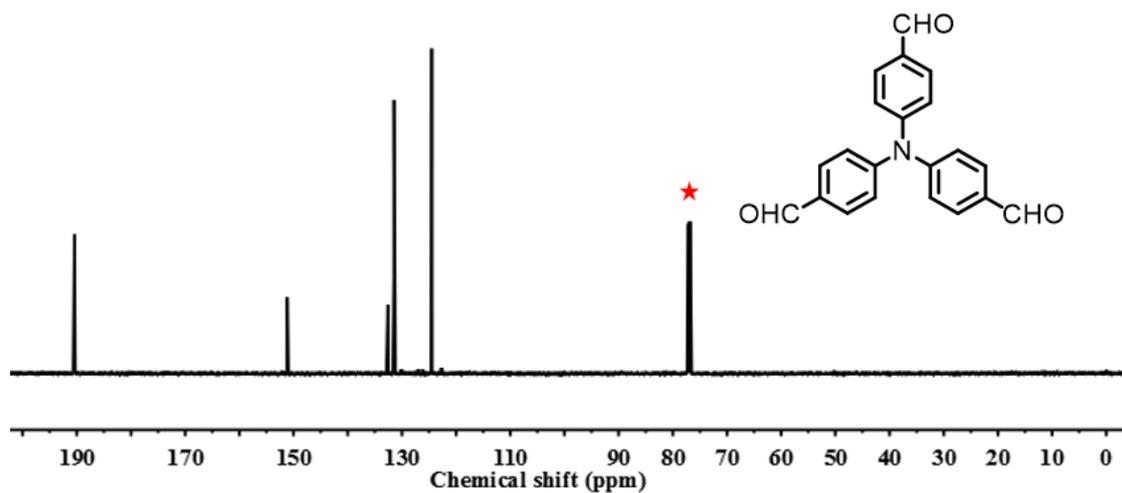


Figure S3-b. The ^{13}C NMR data spectrum of TPA-CHO-3

^1H NMR (600 MHz, dms o) δ 8.72 (s, 2H), 7.16 (t, J = 7.8 Hz, 2H), 6.99 (d, J = 8.4 Hz, 4H), 6.89 (t, J = 7.3 Hz, 1H), 6.81 (d, J = 7.9 Hz, 2H), 6.73 (d, J = 8.4 Hz, 4H), 4.79 (s, 2H), 3.97 (ddd, J = 53.4, 10.8, 7.1 Hz, 8H), 2.22 (s, 12H), 1.08 (t, J = 7.1 Hz, 12H).

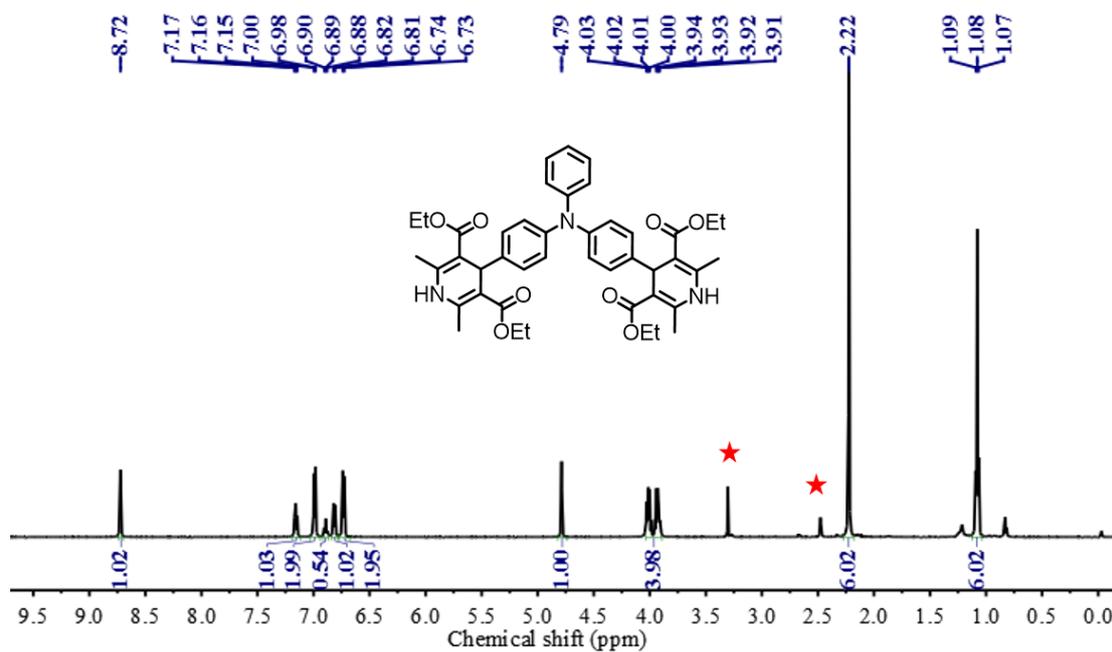


Figure S5-a. The ^1H NMR data spectrum of TPA-DHP-2

^{13}C NMR (150 MHz, dms o) δ 167.40 (s), 147.99 (s), 145.68 (s), 145.42 (s), 143.27 (s), 129.63 (s), 128.81 (s), 123.80 (s), 123.17 (s), 122.39 (s), 102.21 (s), 59.35 (s), 18.71 (s), 14.57 (s).

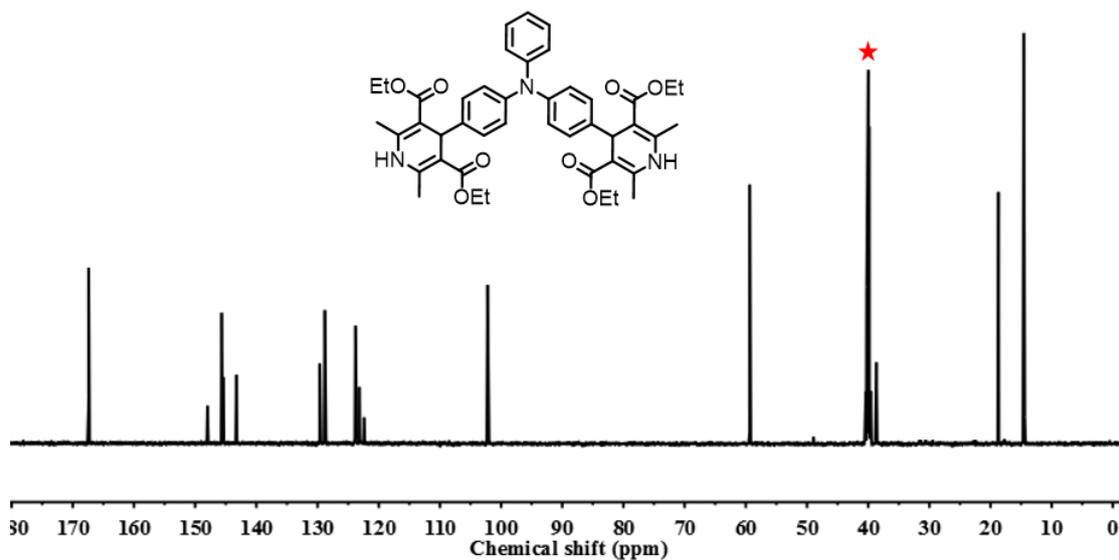


Figure S5-b. The ^{13}C NMR data spectrum of TPA-DHP-2

^1H NMR (600 MHz, d_6 -DMSO) δ 8.78 (s, 3H), 7.61 (d, $J = 8.8$ Hz, 6H), 6.66 (d, $J = 8.7$ Hz, 6H), 4.83 (s, 3H), 3.97 (dd, $J = 46.9, 7.1$ Hz, 12H), 2.23 (s, 18H), 1.07 (t, $J = 7.1$ Hz, 18H).

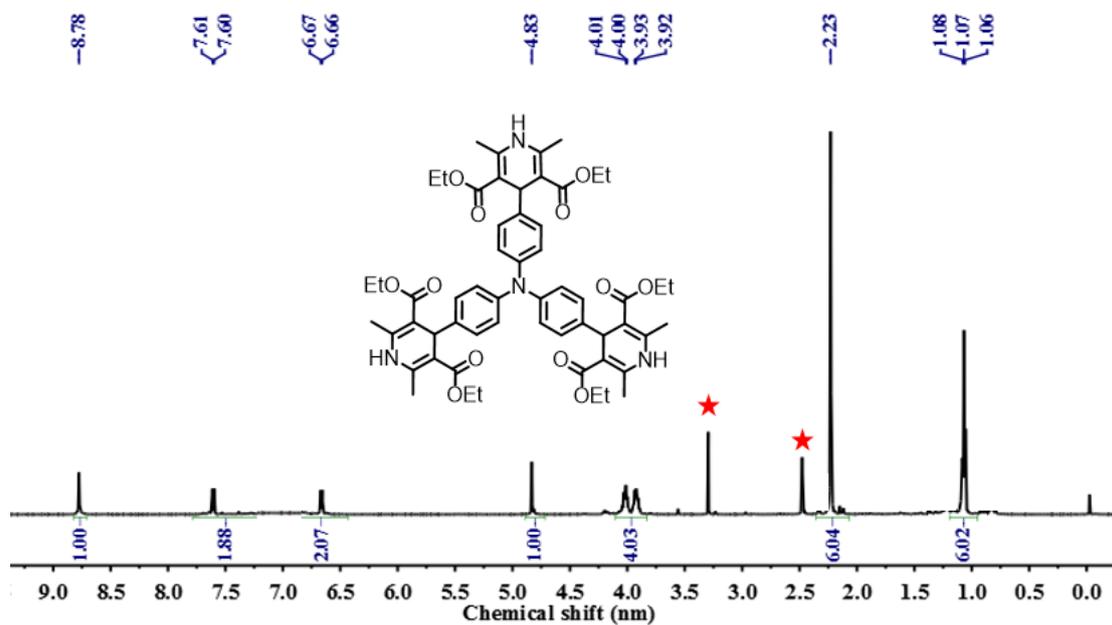


Figure S6-a. The ^1H NMR data spectrum of TPA-DHP-3

^{13}C NMR (150 MHz, d_6 -DMSO) δ 167.24 (s), 147.48 (s), 145.97 (s), 131.62 (s), 129.28 (s), 126.47 (s), 117.37 (s), 102.00 (s), 59.40 (s), 18.68 (s), 14.56 (s).

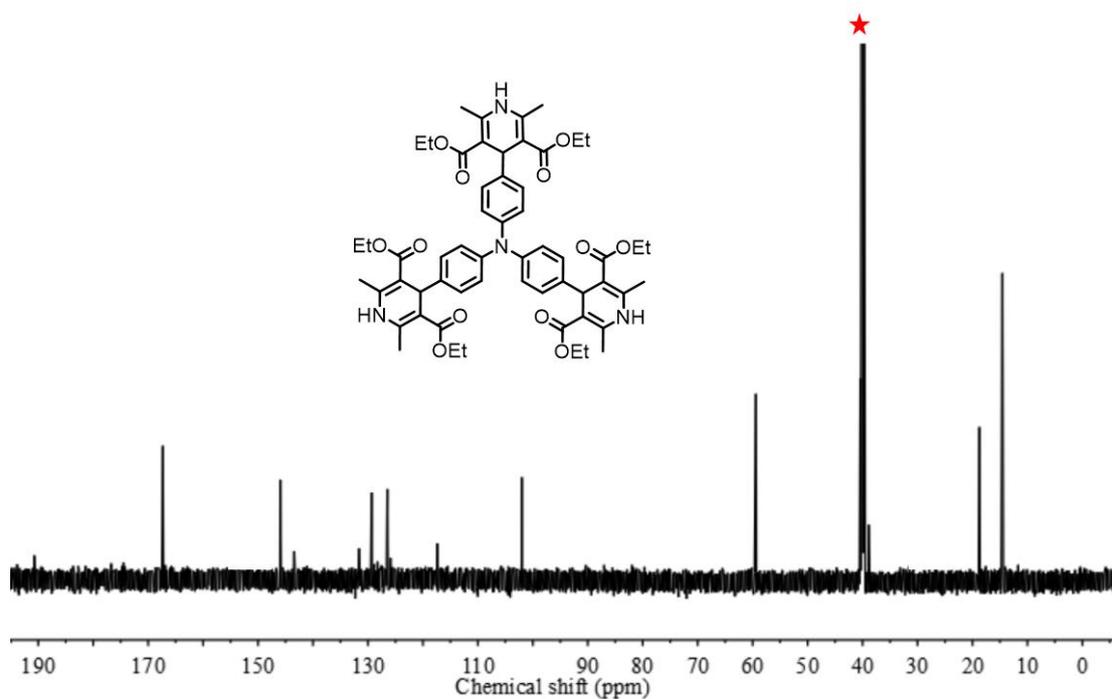


Figure S6-b. The ^{13}C NMR data spectrum of TPA-DHP-3

Photophysical Property

Table 1. Photophysical Properties of TPA-CHO-1~3, TPA-DHP-1~3 and TPA-PPA-1~3.

| | $\lambda_{\text{ex}}/\text{nm}$ | $\lambda_{\text{em}}/\text{nm}$ | $\Phi_{\text{F}}/\%$ | | Identification of $\text{O}_2^{\cdot-}$ |
|-----------|---------------------------------|---------------------------------|----------------------|-----|---|
| TPA-CHO-1 | 317 | 448 | 31.36 | ACQ | - |
| TPA-CHO-2 | 325 | 468 | 2.49 | ACQ | - |
| TPA-CHO-3 | 354 | 500 | 7.20 | ACQ | - |
| TPA-DHP-1 | 313 | 463 | 0.55 | AIE | + |
| TPA-DHP-2 | 346 | 470 | 1.47 | AIE | + |
| TPA-DHP-3 | 357 | 560 | 7.10 | AIE | + |
| TPA-PPA-1 | 432 | 546 | 14.23 | AIE | + |
| TPA-PPA-2 | 437 | 598 | 8.57 | AIE | + |
| TPA-PPA-3 | 451 | 597 | 13.86 | AIE | + |

In this table, λ_{ex} was the maximum absorption wavelength; λ_{em} was the maximum emission wavelength excited by each λ_{ex} ; Φ_{F} was the absolute fluorescence quantum yield.

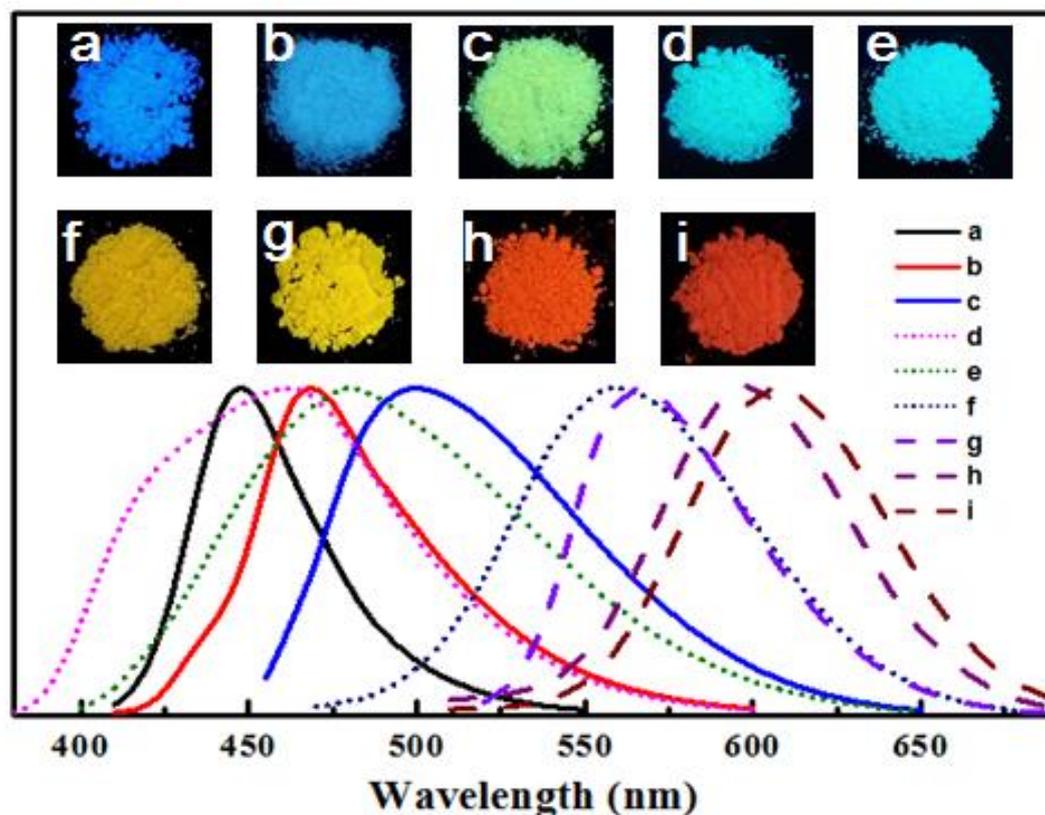


Figure S7 PL spectra of TPA-CHO-1~3, TPA-DHP-1~3 and TPA-PPA-1~3. Inset: The

corresponding photographs of TPA-CHO-1~3(a, b, c), TPA-DHP-1~3(d, e, f) and TPA-PPA-1~3(g, h, i) taken under illumination at 365 nm.

AIE performance

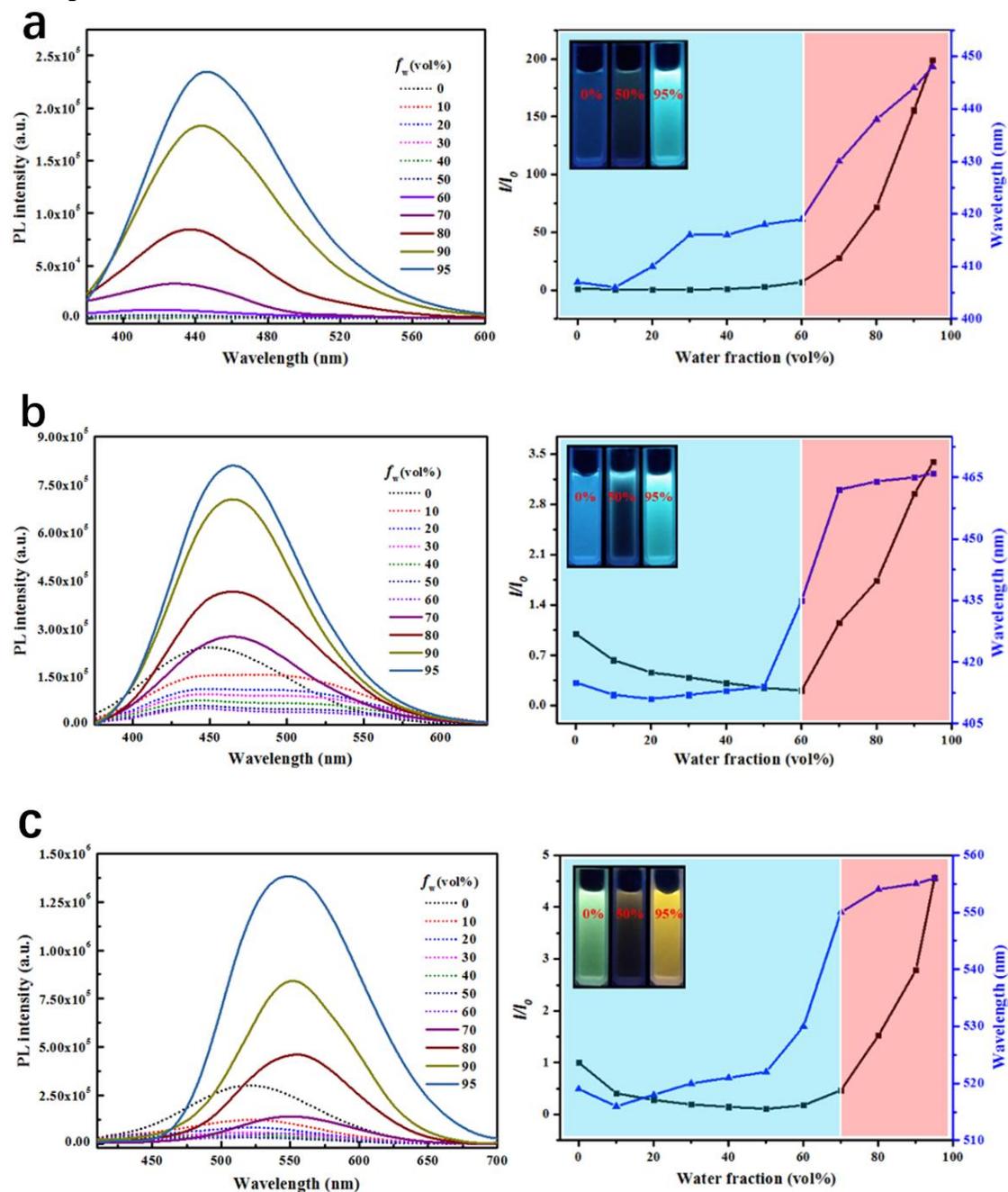


Figure S8 PL spectra of TPA-DHP-1, TPA-DHP-2 and TPA-DHP-3 in Acetone and mixture of Acetone & H₂O, and the Plots of corresponding PL intensity and emission wavelengths versus the H₂O fractions, Inset shown the images in different H₂O fractions (0, 50%, 95%) under UV light (concentration: 20 μM).

Solvent-induced effect

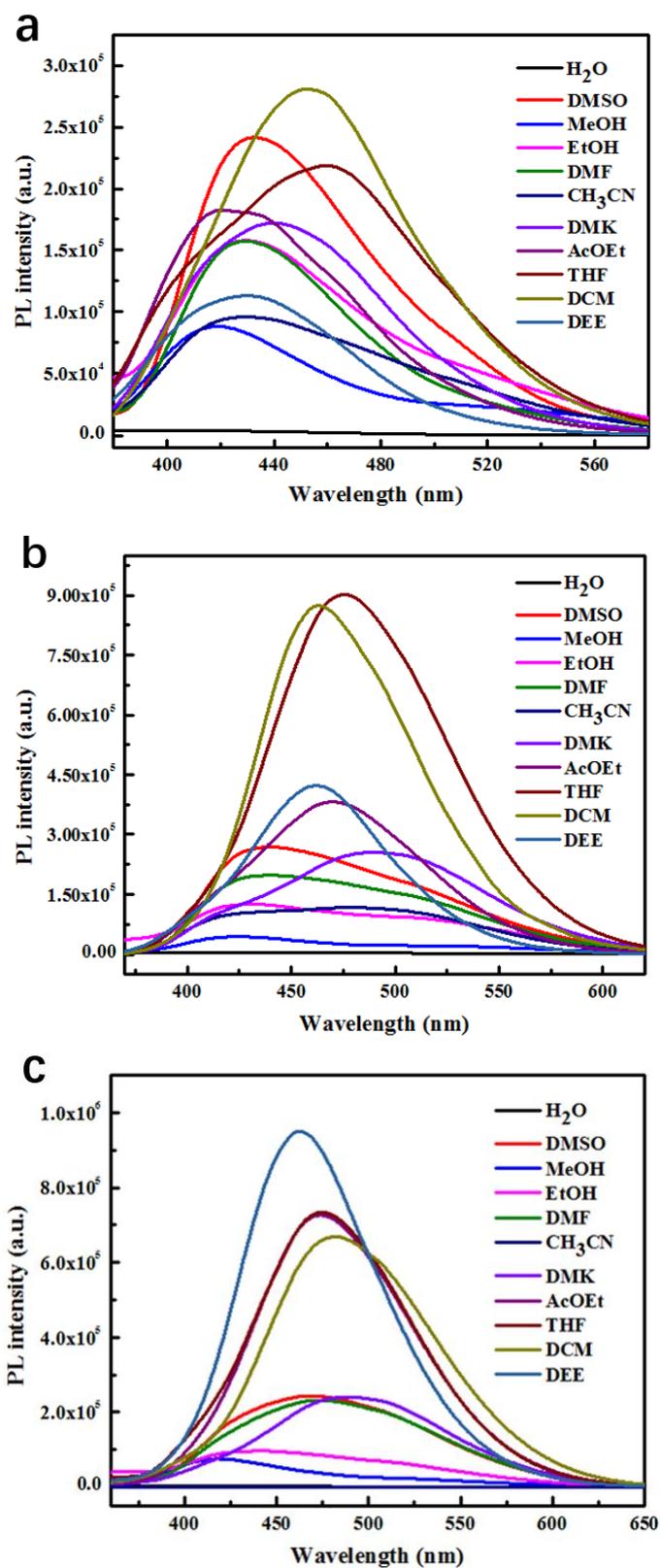


Figure S9 Fluorescence spectra of TPA-DHP-1 (a), TPA-DHP-2 (b) and TPA-DHP-3 (c) at 20 μM in different solvent.

Theoretical calculations

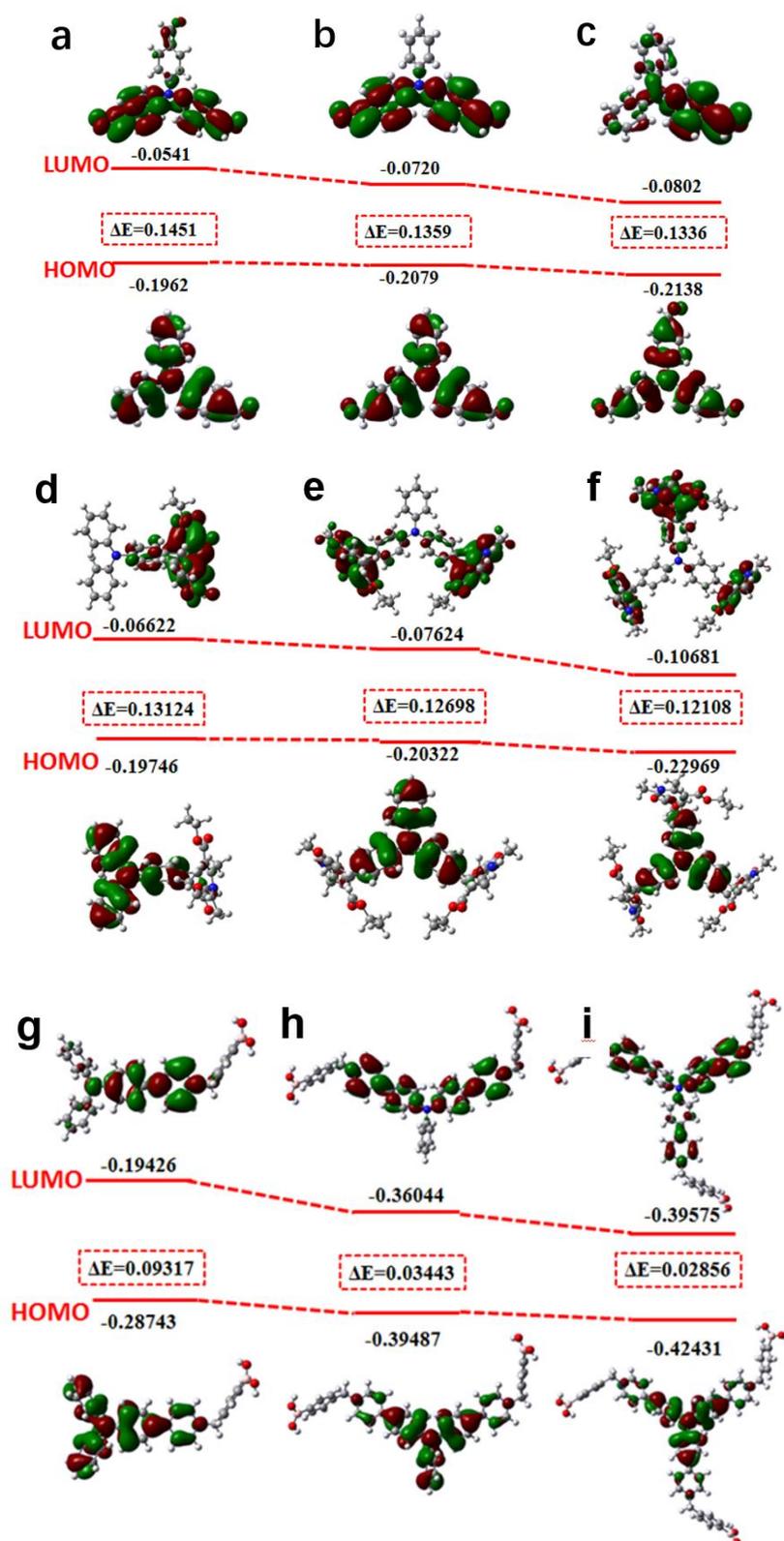


Figure S10 HOMO-LUMO energy levels of TPA-CHO-1~3 (a, b, c), TPA-DHP-1~3(d, e, f) and TPA-PPA-1~3(g, h, i), as estimated in Gaussian 09 using the B3LYP modification with the 6-31G* basis set.

Reaction mechanism

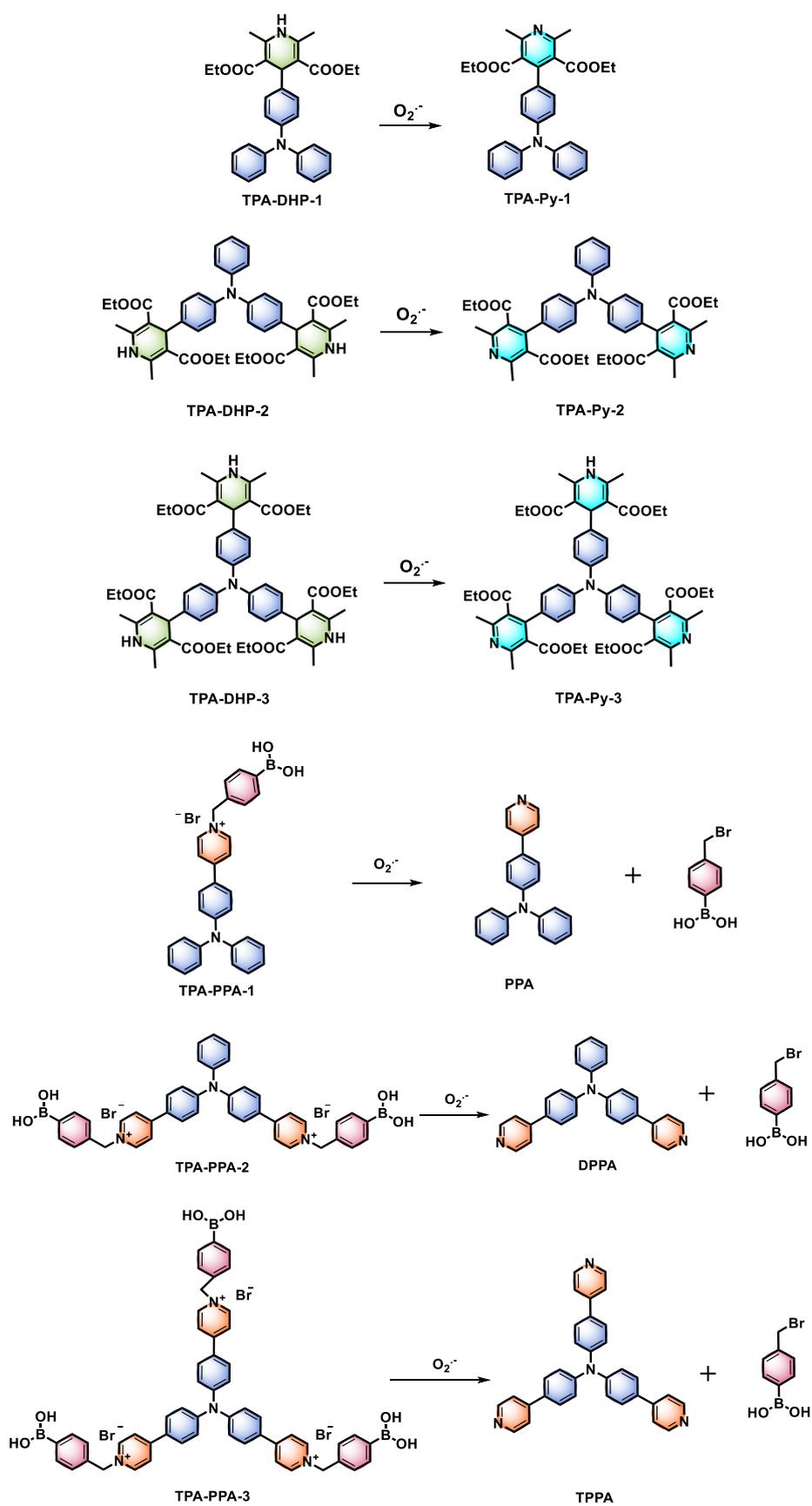


Fig. S11 molecular structures and proposed reaction mechanism for $O_2^{\cdot -}$ detection.

Theoretical calculations

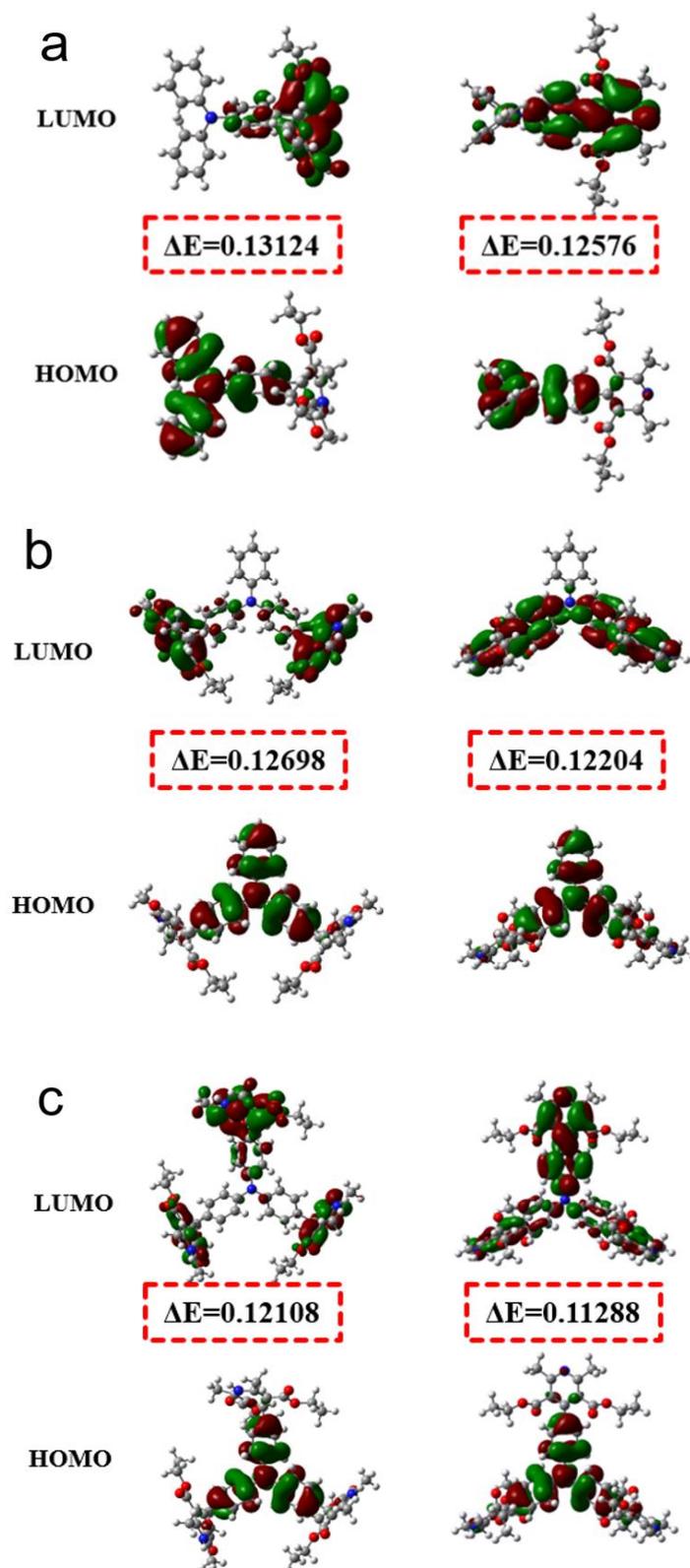


Figure S12 HOMO-LUMO energy levels of TPA-DHP-1(a), TPA-DHP-2(b), TPA-DHP-3(c) (left) and corresponding aromatization products (right) as estimated in Gaussian 09 using the B3LYP modification with the 6-31G* basis set.

Cyclic voltammograms

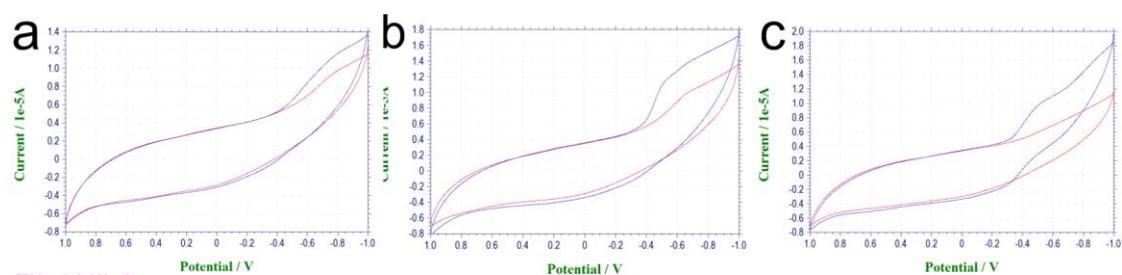


Figure S13 Cyclic voltammograms of TPA-DHP-1(a), TPA-DHP-2(b) and TPA-DHP-3(c) in N_2 -saturated 0.2 M PBS (pH 7.0), at 50 mV/s .

Structural characterization

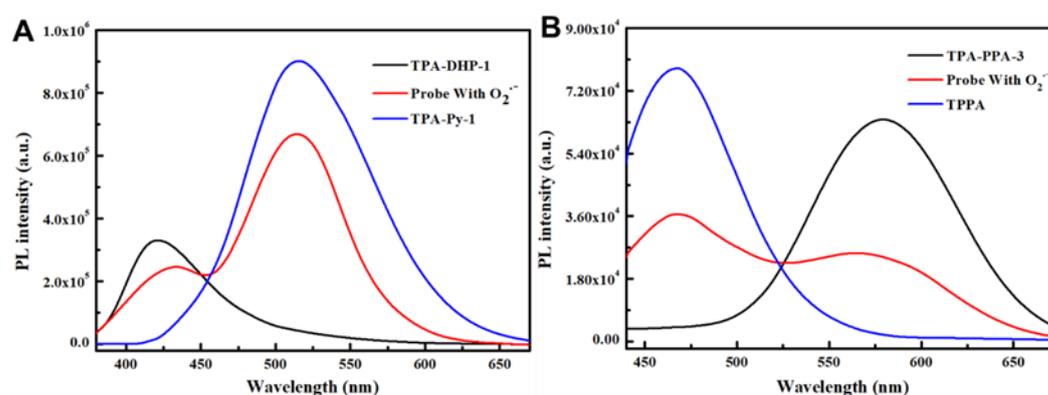


Fig. S14 (A) PL spectra of TPA-DHP-1 before the oxidation, with $O_2^{\cdot-}$ and the oxidation product TPA-Py-1; (B) PL spectra of TPA-PPA-3 before the oxidation, with $O_2^{\cdot-}$ and the oxidation product TPPA.

Cell viability

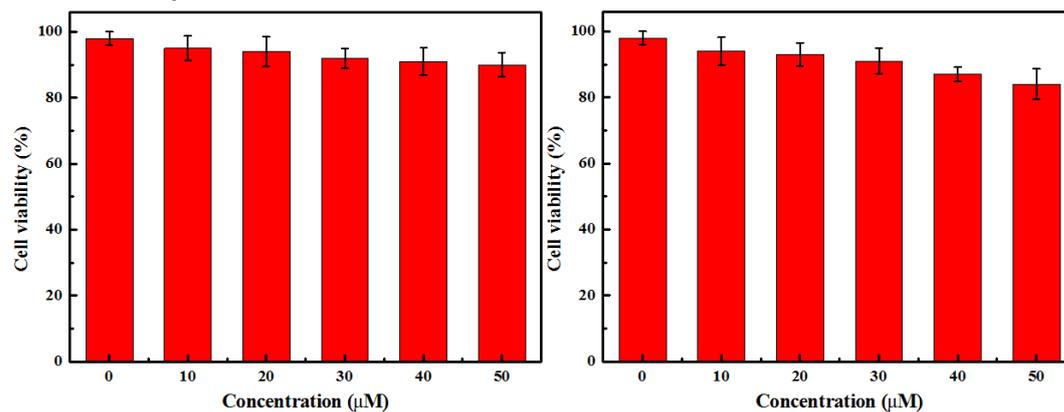


Figure S15 (c) Cell viabilities of HeLa cells treated with different concentrations of TPA-DHP-1 (left) and TPA-PPA-3 (right) for 96 h by MTT assay

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