

Temperature-induced amperometric glucose biosensor based on a poly (N-vinylcaprolactam)/graphene oxide composited film

Chao Chen,^{ab} Pengcheng Zhao,^{bc} Meijun Ni,^b Chunyan Li,^b Yixi Xie^{bc*} and Junjie Fei^{ac*}

* Corresponding authors

^a Key Laboratory of Environmentally Friendly Chemistry and Applications of Ministry of Education, College of Chemistry, Xiangtan University, Xiangtan 411105, People's Republic of China

E-mail: fei_junjie@xtu.edu.cn

^b Key Laboratory for Green Organic Synthesis and Application of Hunan Province, Xiangtan 411105, People's Republic of China

E-mail: xieyixige@xtu.edu.cn

^c Hunan Institute of Advanced Sensing and Information Technology, Xiangtan University, Xiangtan 411105, People's Republic of China

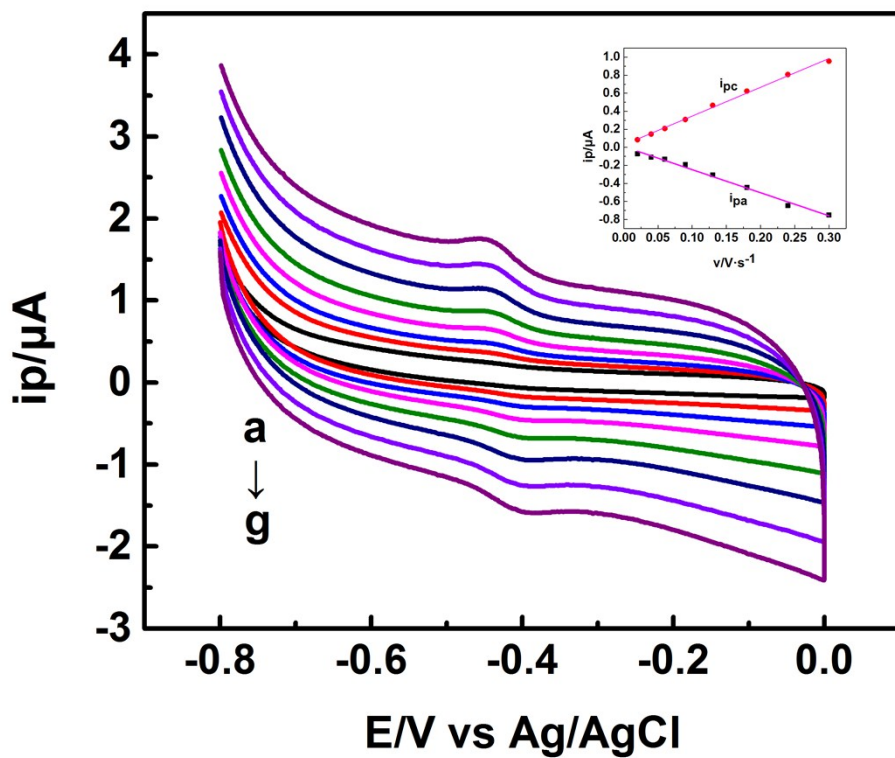


Fig. S1. Cyclic voltammograms of PVCL / GO / GOD / GCE at various scan rates from 20, 40, 60, 90, 130, 180, and 240 to 300 $\text{mV}\cdot\text{s}^{-1}$ respectively (from a to g). Inset: plot of peak current (i_p) vs scan rate. Supporting electrolyte: $0.1 \text{ mol}\cdot\text{L}^{-1}$ phosphate buffer (pH 7.0)

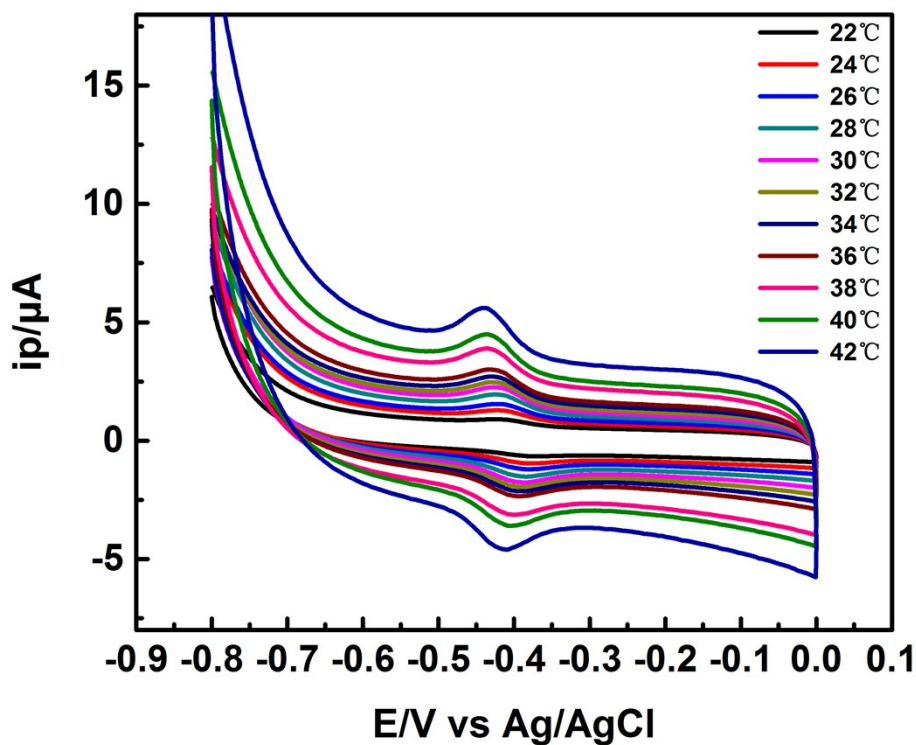


Fig. S2. CV plots of PVCL / GO / GOD / GCE at various temperatures from 22 to 42 $^{\circ}\text{C}$; Scan

rate: $50 \text{ mV}\cdot\text{s}^{-1}$; Supporting electrolyte: $0.1 \text{ mol}\cdot\text{L}^{-1}$ phosphate buffer (pH 7.0).

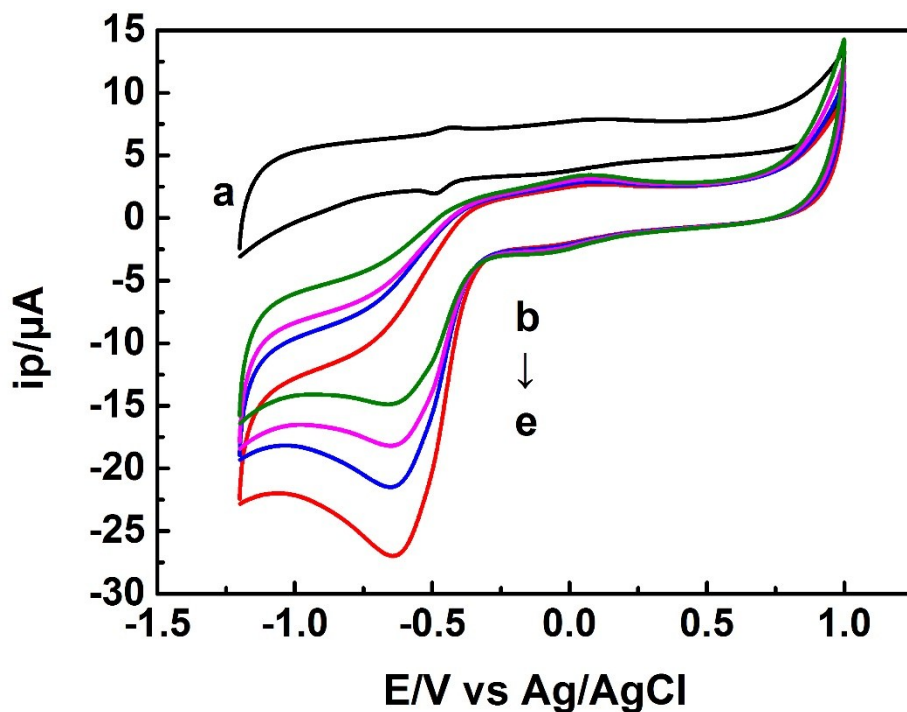


Fig. S3 CV plots of PVCL / GO / GOD / GCE at (a) N_2 -saturated and at (b)(c)(d)(e) O_2 -saturated with 0, 0.3, 0.5, 0.6 mM glucose Scan rate: $50 \text{ mV}\cdot\text{s}^{-1}$; Supporting electrolyte: $0.1 \text{ mol}\cdot\text{L}^{-1}$ phosphate buffer (pH 7.0)

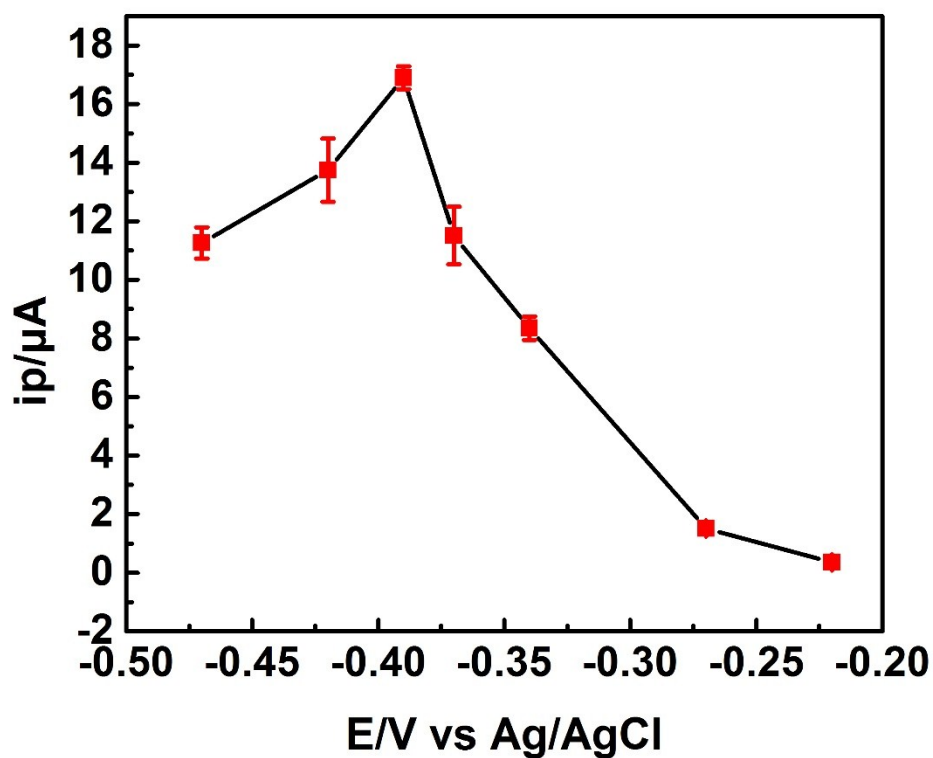


Fig.S4 under different potential ampere of glucose catalysis

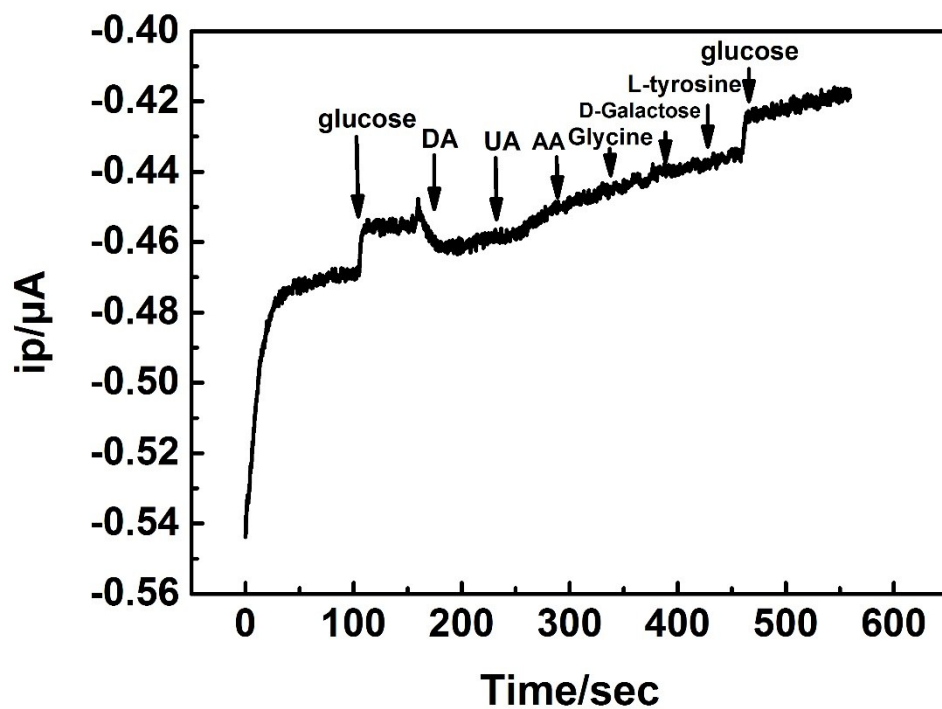


Fig. S5 Amperometric response of PVCL / GO / GOD / GCE at different interference

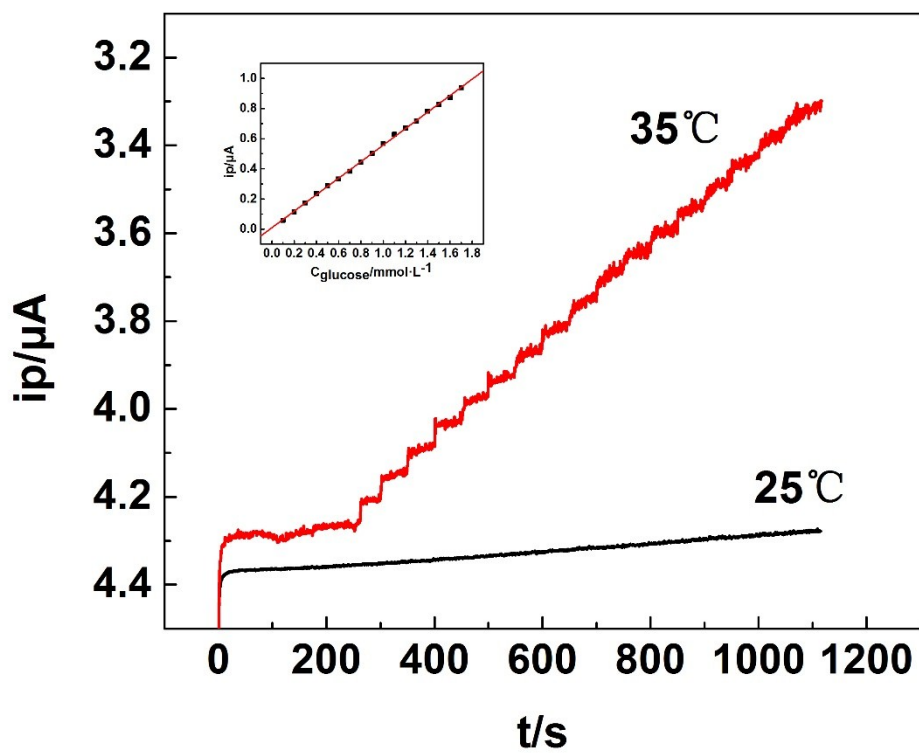


Fig.S6 Amperometric response of PVCL / GO/ GOD / GCE to glucose compare 25°C to 35°C.the

inset: the current was proportional to the concentration of glucose at 35°C.