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Supplementary Material

Hydrothermal synthesis of carbon quantum dots as fluorescent probe for sensitive and rapid detection of picric acid

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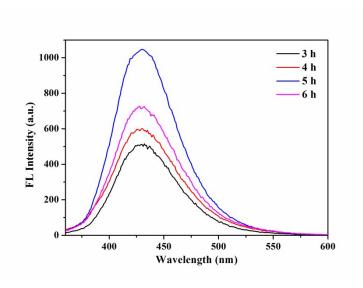


Fig S1. Fluorescence emission spectra of N-CDs with different reaction time.

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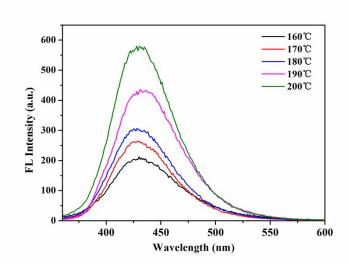


Fig S2. Fluorescence emission spectra of N-CDs with different temperature.

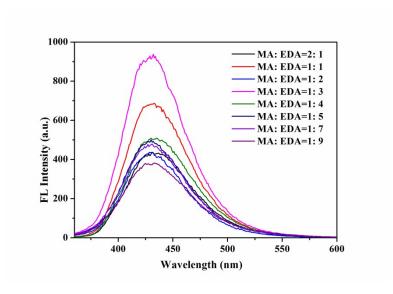


Fig S3. Fluorescence emission spectra of N-CDs with different ratio of mandelic acid with ethylenediamine.

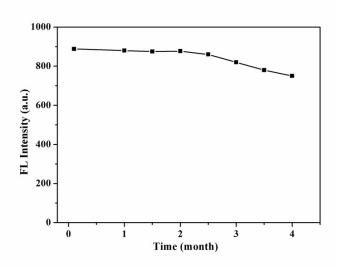


Fig S4. Effects of the storage time on the fluorescence intensity of the N-CDs at 429 nm.

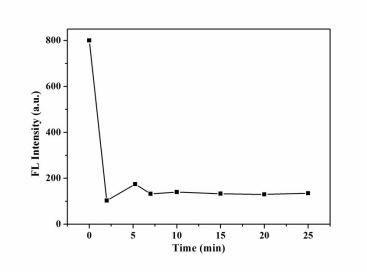


Fig S5. The effect of reaction time.

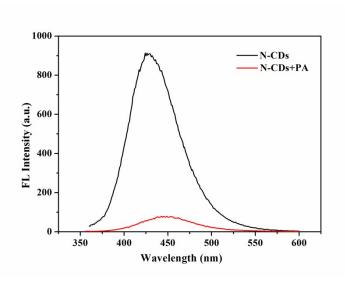


Fig S6. Fluorescence emission spectra of N-CDs in the abscence and present of PA, respectively.

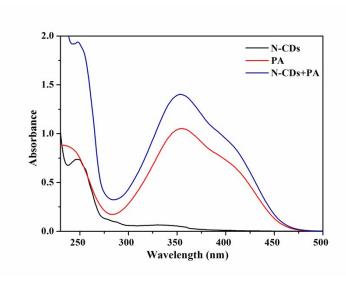


Fig S7. UV-vis absorption spectra of N-CDs (black), PA (red), and N-CDs in the presence of PA (blue).

| System | $\tau_{l}(ns)$ | Percentage (%) | τ_2 (ns) | Percentage (%) | τ_3 (ns) | Percentage (%) | $	au_{avg}$ (ns) |
|--------------|----------------|----------------|---------------|----------------|---------------|----------------|------------------|
| N-CDs | 10.62 | 19.06 | 0.09261 | 38.92 | 1.729 | 42.01 | 2.787 |
| N- CDs+PA | 2.554 | 61.76 | 0.082 | 12.4 | 1.0349 | 25.84 | 1.855 |

Table S1. The fluorescence lifetimes of N-CDs in the absence and presence of PA.