Electronic Supporting Information for

A solid-state Sb/Sb_2O_3 biosensor for the in-situ measurement of extracellular acidification associated with the multidrug resistant phenotype in the breast cancer cells

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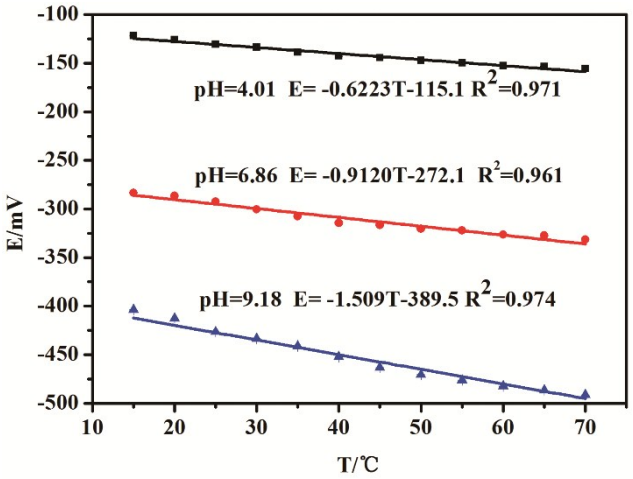

Fig. S1 The relationship between the potential and temperature of the electrode in three standard buffers (pH=4.01, 6.86, 9.18).

Table S1 Results of pH measurements after the interference substance was added with pH=2.00, 4.01 buffers (n=3).

solution	2.00 buffer +KCl	2.00 Buffer +Na_2SO_4	2.00 buffer +KNO_3	2.00 buffer +$Ca(NO_3)_2$	4.01 buffer +KCl	4.01 buffer +Na_2SO_4	4.01 buffer +KNO_3	4.01buffer +$Ca(NO_3)_2$
SSO	2.10±0.02	2.22±0.02	2.06±0.01	*	4.05±0.02	3.89±0.01	4.03±0.01	3.69±0.02
glass electrode	2.05±0.01	2.14±0.01	2.03±0.01	1.87±0.01	4.00±0.01	3.88±0.01	3.98±0.01	3.67±0.01
ΔpH	0.05±0.03	0.08±0.03	0.03±0.02	*	0.05±0.03	0.01±0.02	0.05±0.02	0.02±0.03

* represents that the pH value of the solution was not measured by the fabricated SSO electrode due to the linear relationship between the potential and pH in the range of 2-12.