

Electronic Supplementary Information

Several fluorescent probes based on hemicyanine for the detection of SO₂ derivatives

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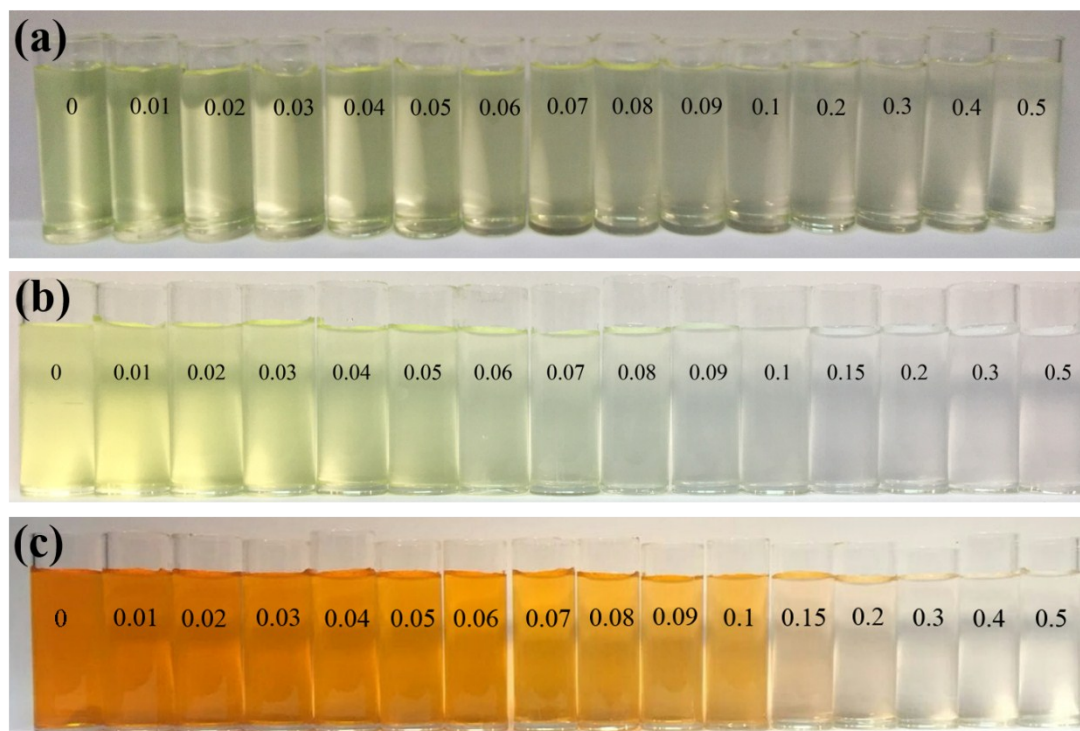


Fig. S1 Visual color changes of (a)probe **Hcy-NO₂**, (b)probe **Hcy-CH₃**, (c)probe **Hcy-NH₂** upon addition of HSO₃⁻ (0-0.5 mM) in DMSO/PBS solution (v/v = 1/9, pH 7.4).

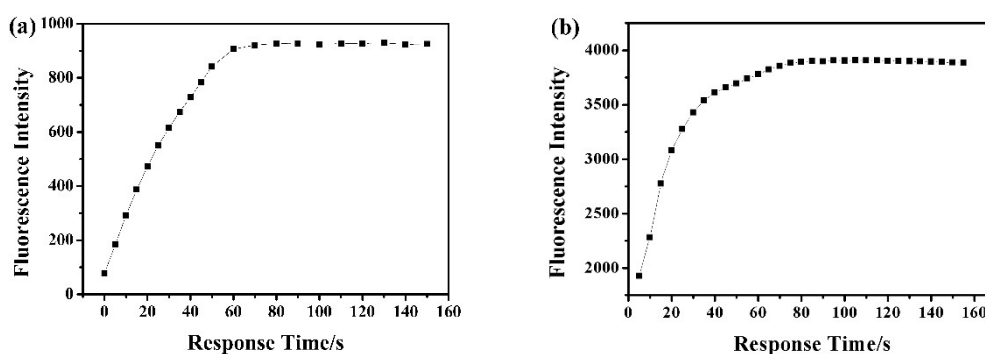


Fig. S2 (a)Time-dependent fluorescence spectra of probe **Hcy-CH₃** (1×10⁻⁴ M) in DMSO/PBS solution (v/v = 1/9, pH 7.4) with the addition of HSO₃⁻ (0.1 mM) λ_{ex} = 290 nm. (b)Time-dependent fluorescence spectra of probe **Hcy-NH₂** (1×10⁻⁴ M) with the addition of HSO₃⁻ (0.2 mM) λ_{ex} = 300 nm.

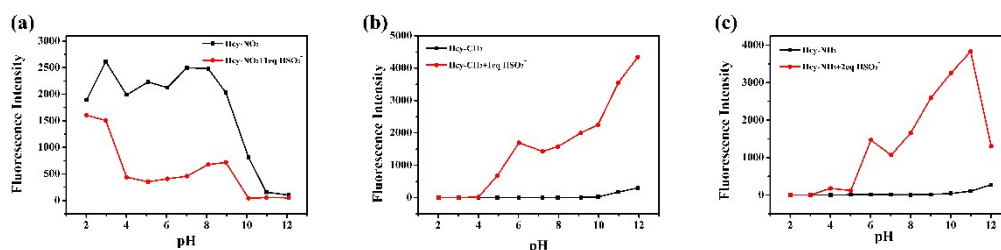


Fig. S3 Effect of pH on the fluorescence intensity of (a)probe **Hcy-NO₂**, (b)probe **Hcy-CH₃**, (c)probe **Hcy-NH₂** in the absence (black line) or presence (red line) of HSO₃⁻.

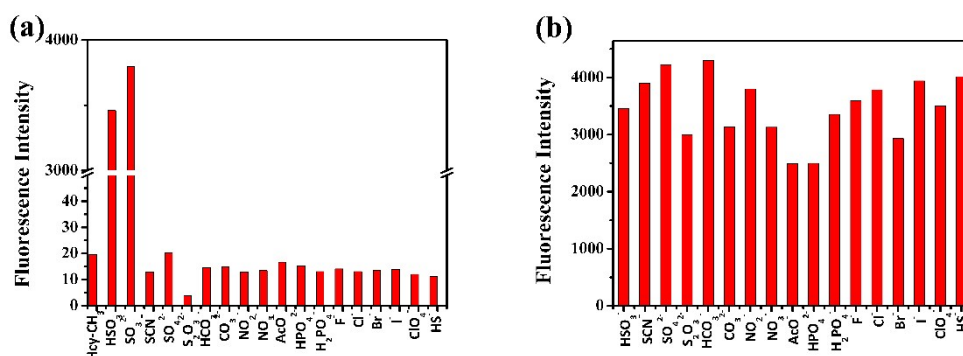


Fig. S4 Fluorescence intensity at 390 nm of probe **Hcy-CH₃** (a)with different anions and (b)with 1 equiv. HSO₃⁻ and 10 equiv. competitive anions in DMSO/PBS solution (v/v = 1/9, pH 7.4).

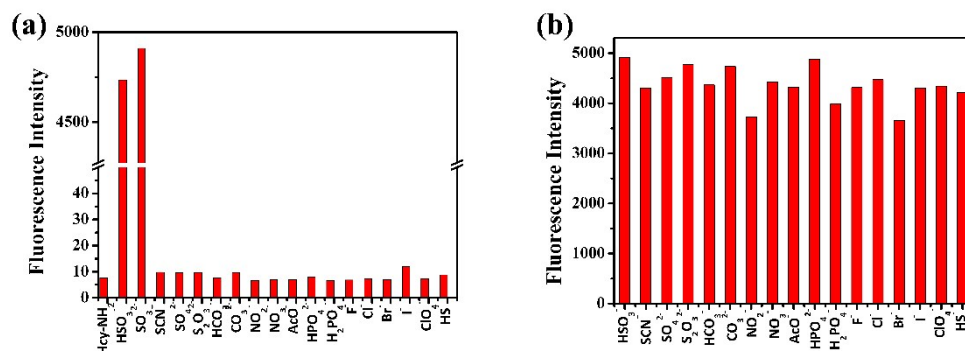


Fig. S5 Fluorescence intensity at 381 nm of probe **Hcy-NH₂** (a)with different anions and (b)with 2 equiv. HSO₃⁻ and 10 equiv. competitive anions in DMSO/PBS solution (v/v = 1/9, pH 7.4).

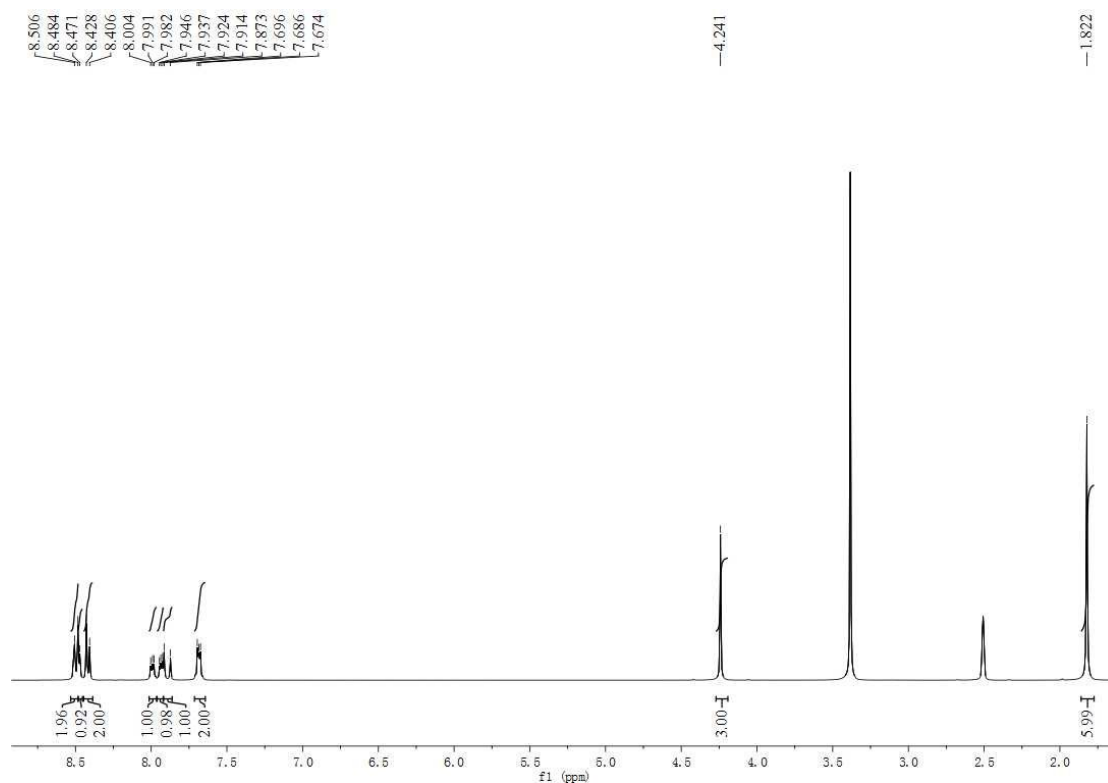


Fig. S6 ¹H NMR spectrum of probe **Hcy-NO₂** in DMSO-*d*₆.

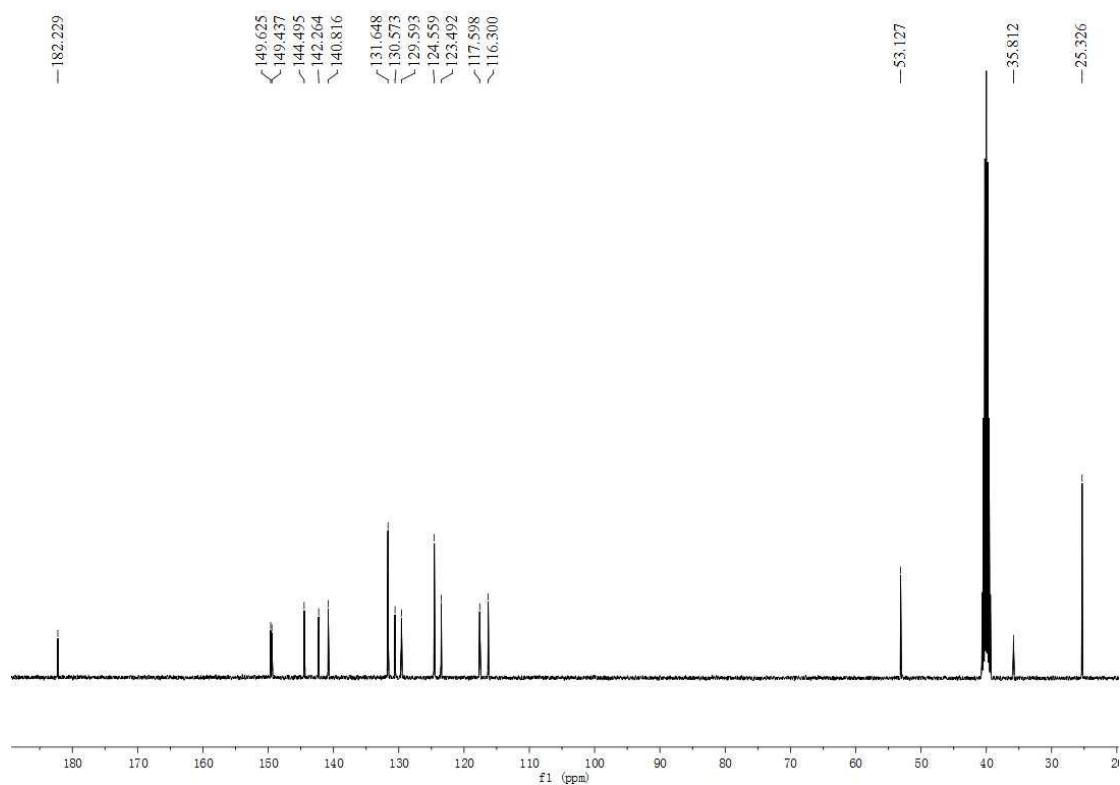


Fig. S7 ^{13}C NMR spectrum of probe **Hcy-NO₂** in $\text{DMSO-}d_6$.

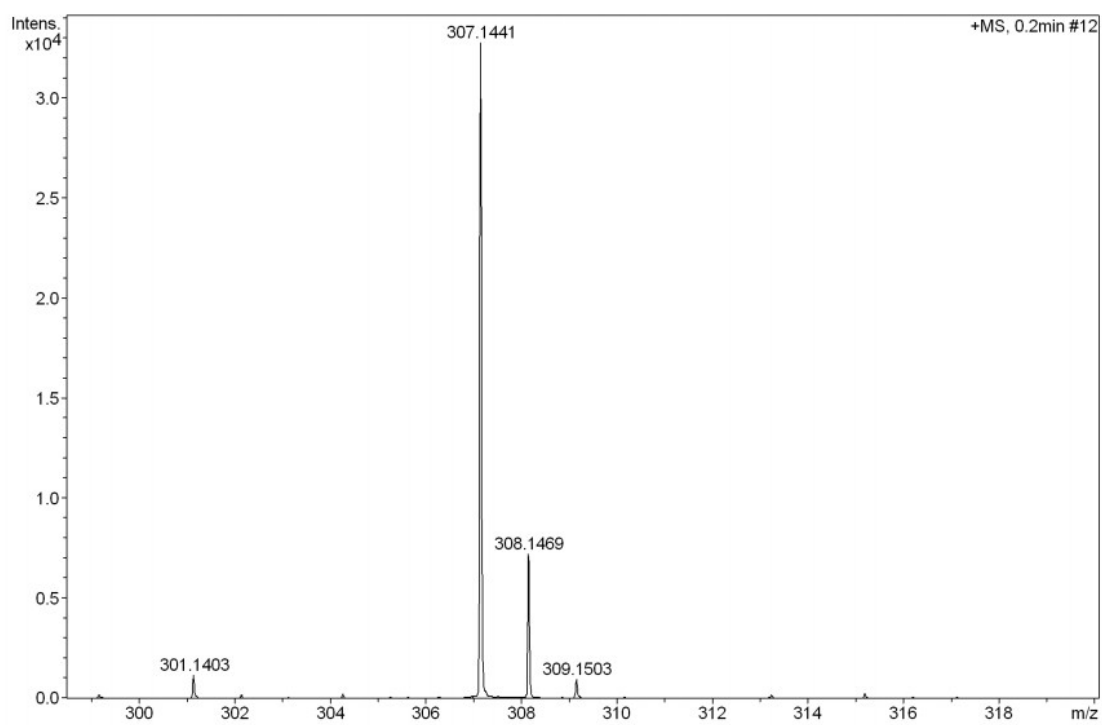


Fig. S8 HRMS spectrum of probe **Hcy-NO₂**.

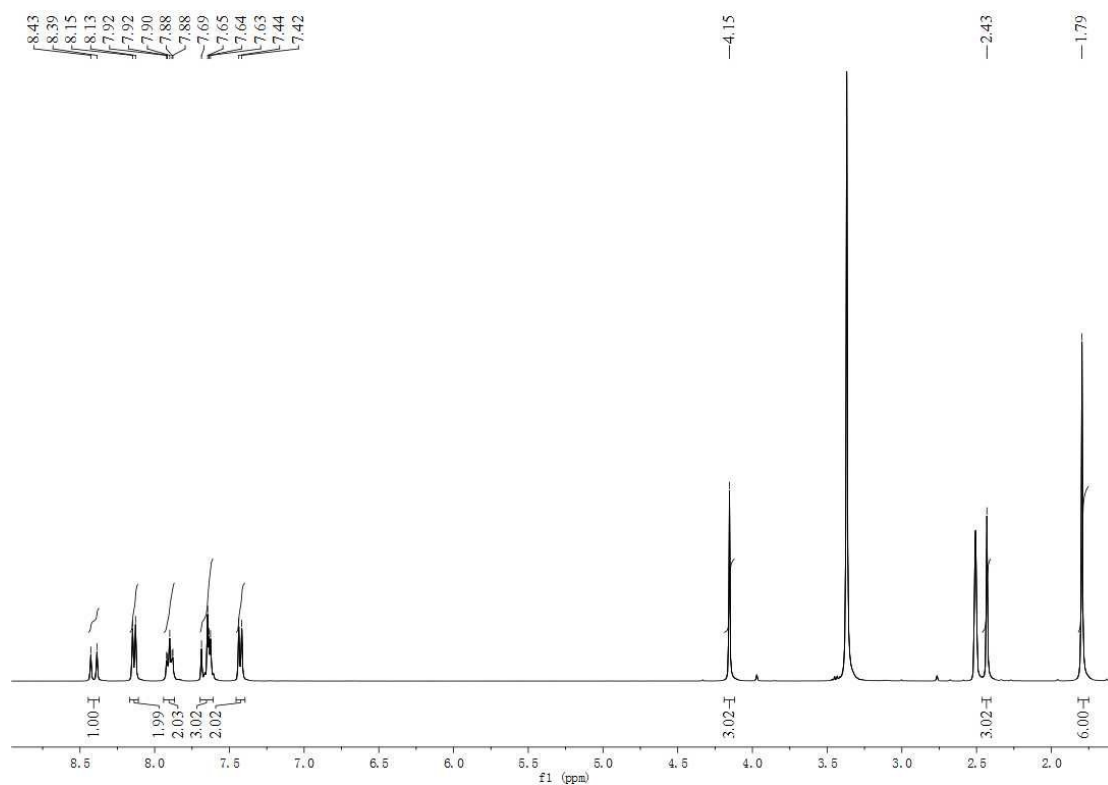


Fig. S9 ^1H NMR spectrum of probe **Hcy-CH₃** in $\text{DMSO-}d_6$.

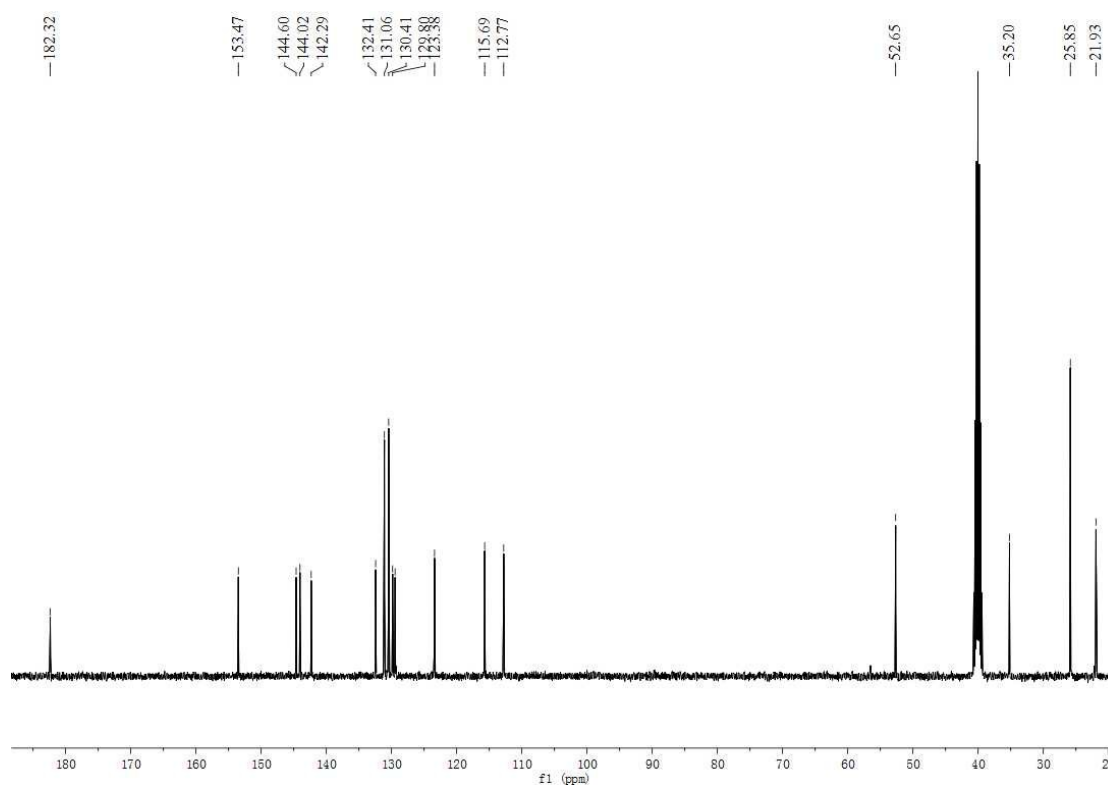


Fig. S10 ^{13}C NMR spectrum of probe **Hcy-CH₃** in $\text{DMSO-}d_6$.

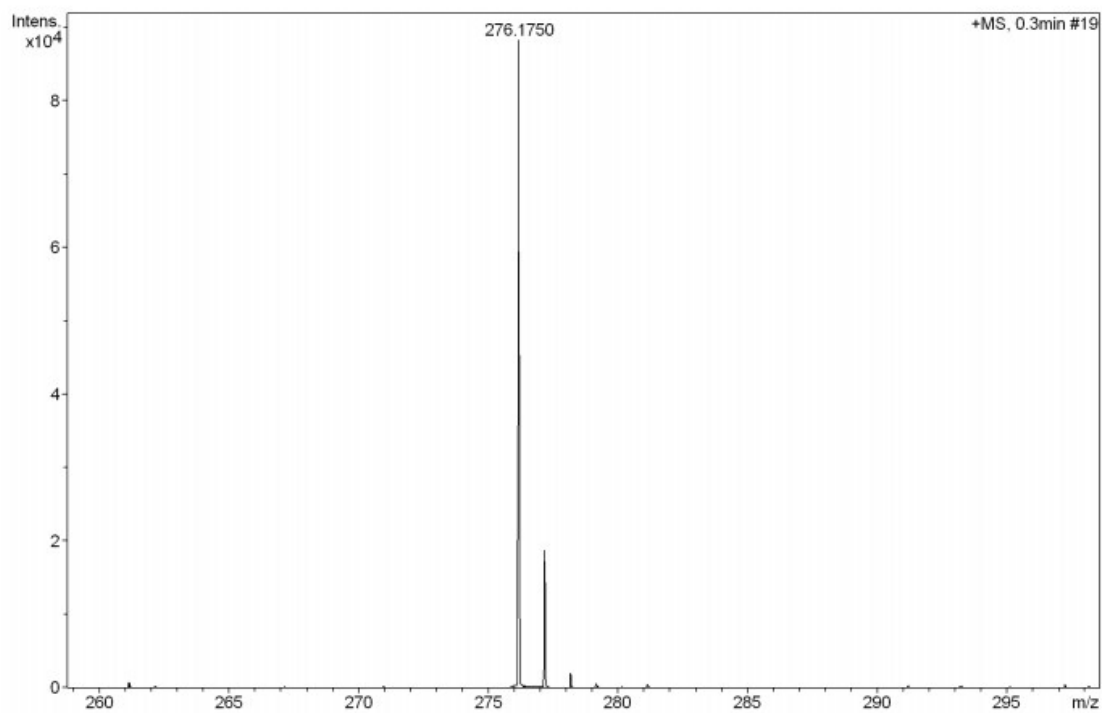


Fig. S11 HRMS spectrum of probe **Hcy-CH₃**.

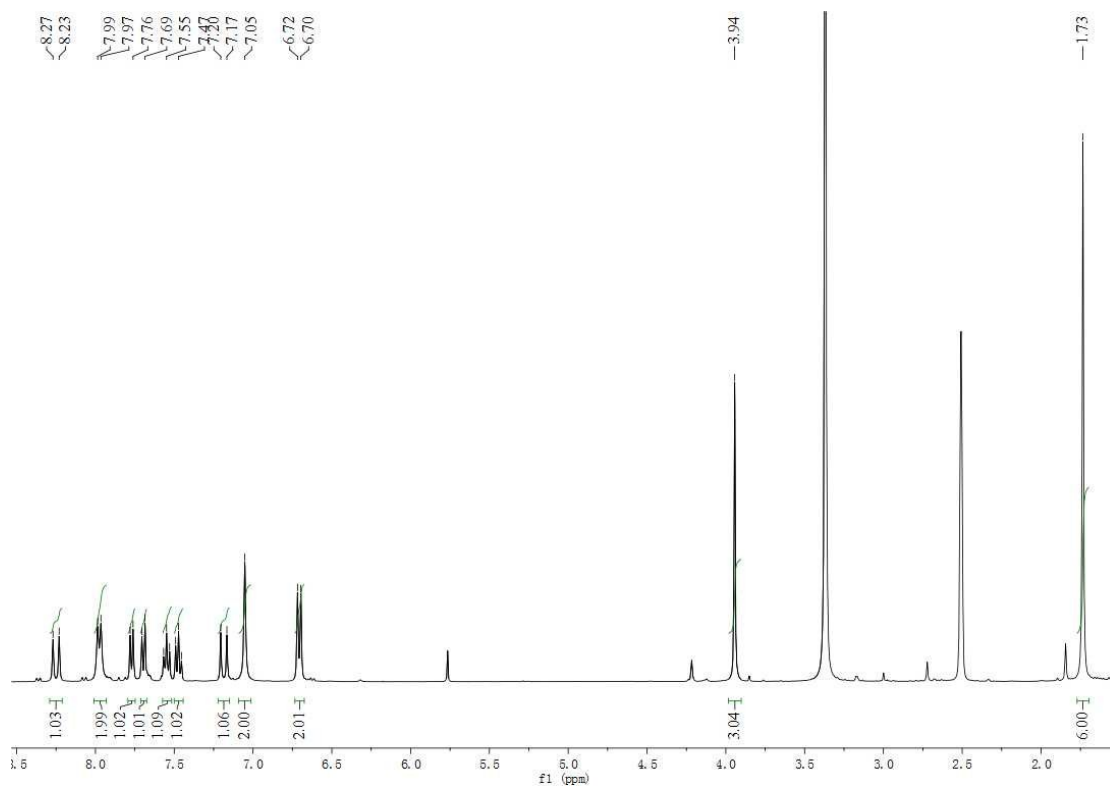


Fig. S12 ¹H NMR spectrum of probe **Hcy-NH₂** in DMSO-*d*₆.

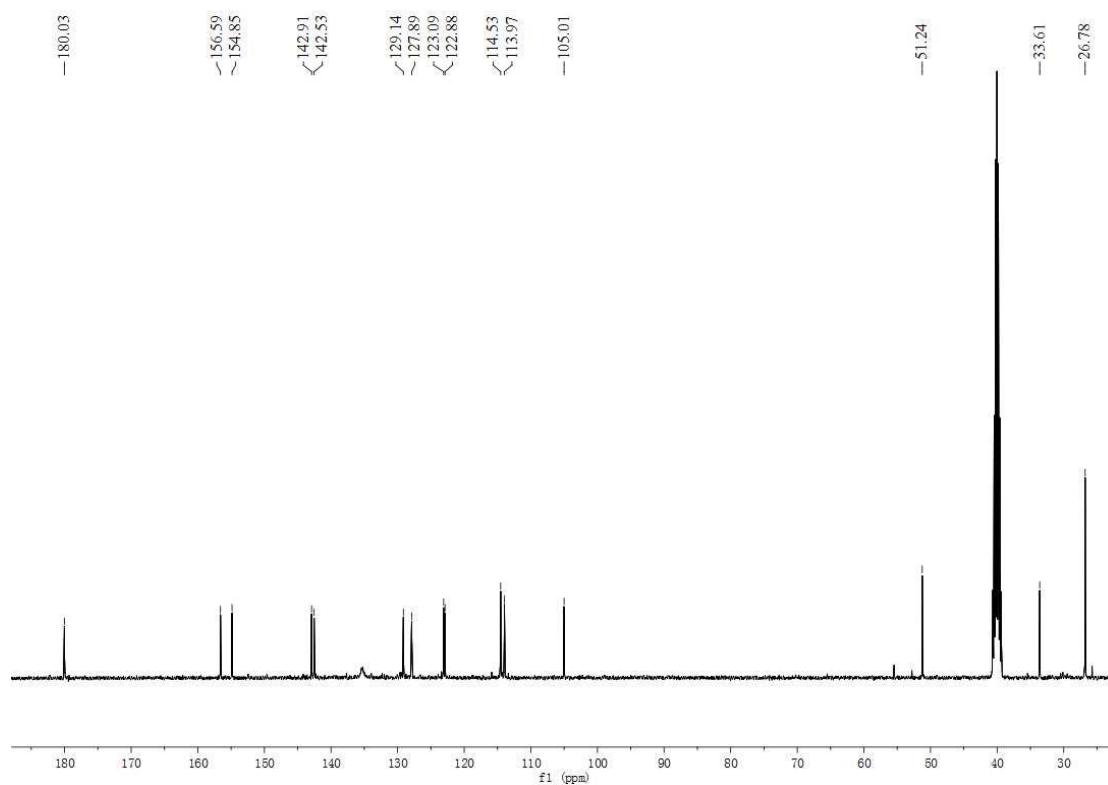


Fig. S13 ^{13}C NMR spectrum of probe **Hcy-NH₂** in $\text{DMSO-}d_6$.

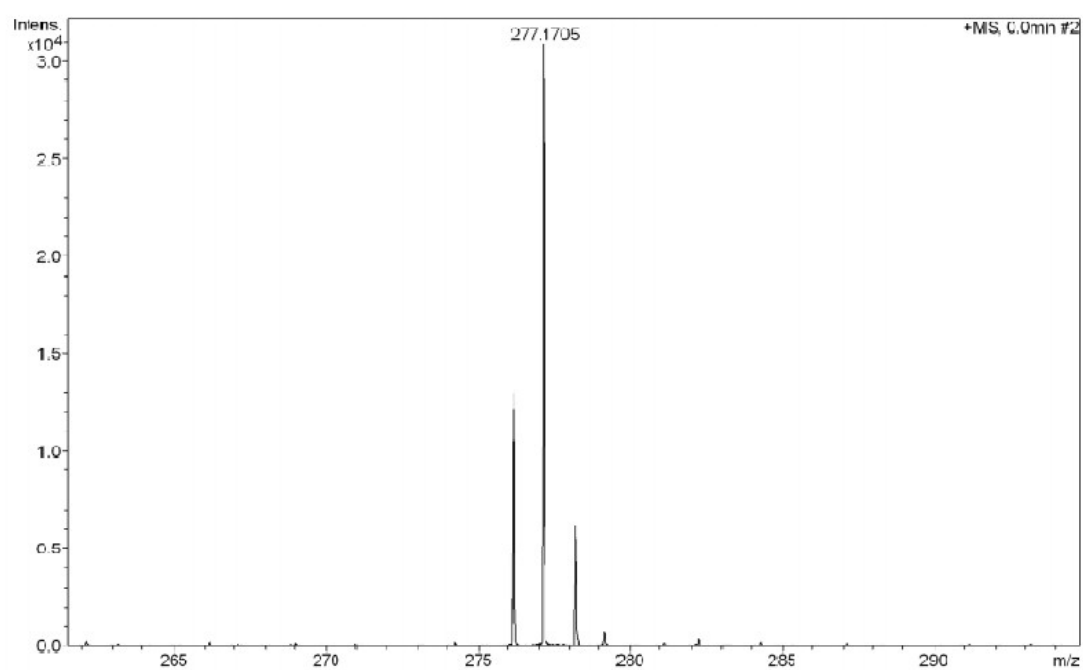


Fig. S14 HRMS spectrum of probe **Hcy-NH₂**.