

Electronic Supplementary Information (ESI)

for

A CoOOH nanoflake-based light scattering probe for the simple and selective detection of uric acid in human serum

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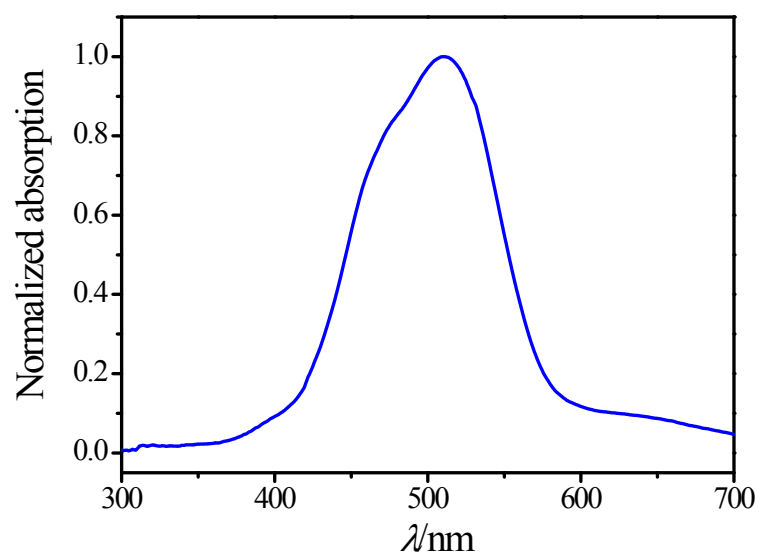


Fig. S1 The normalized absorption spectrum of Co^{2+} .

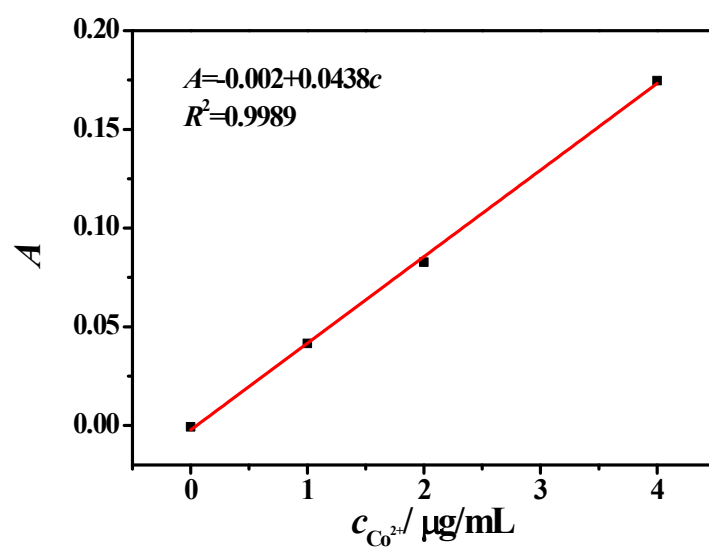


Fig. S2 The absorption standard linear curve of Co^{2+} detected by AA-7000 atomic absorption spectrophotometer. Working conditions: Wavelength: 240.7 nm; slit width: 0.2 nm; time constant: 1.0 s; lamp current: 15 mA; photomultiplier voltage: 719 V; flame Type: air- C_2H_2 ; fuel flow: 2.2 L/min; oxidant: 15.0 L/min at 160 kPa; burner height: 7.5 mm; delay time: 5 s; measurement time: 5 s.

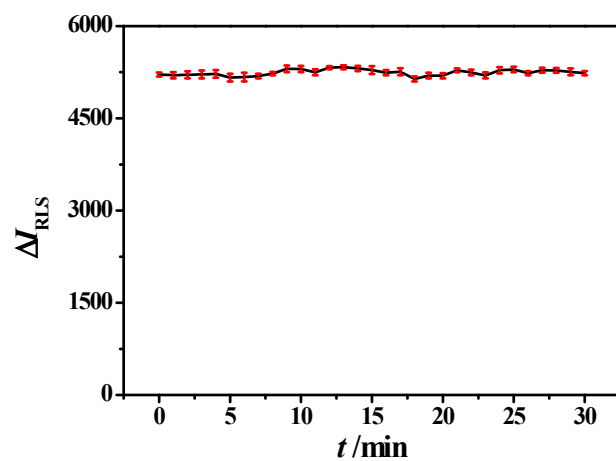


Fig. S3 The light scattering intensity CoOOH nanoflakes at pH 6.64. Conditions: CoOOH nanoflakes, 8 μ M; pH 6.64.

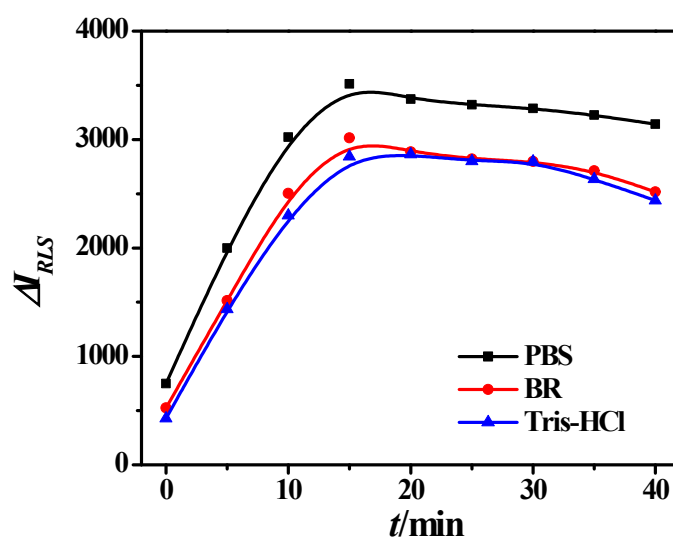


Fig. S4 The light scattering intensity change of CoOOH nanoflakes induced by H₂O₂ in different media (PBS, BR and Tris-HCl) at pH 6.64. Conditions: CoOOH nanoflakes, 8 μM; H₂O₂, 250 μM; pH 6.64; T, 45 °C.

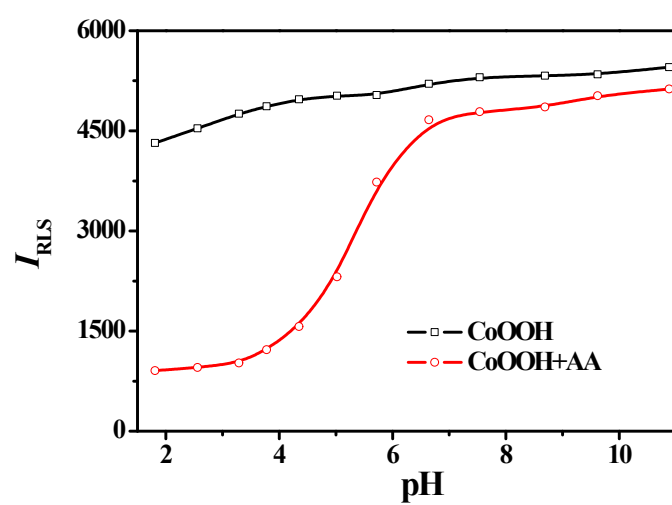


Fig. S5 The light scattering intensity of CoOOH nanoflakes in the absence or presence of AA in PBS buffer at different pH. Conditions: CoOOH nanoflakes, 8 μM ; AA, 750 μM ; pH 6.64; T, 45 $^{\circ}\text{C}$.

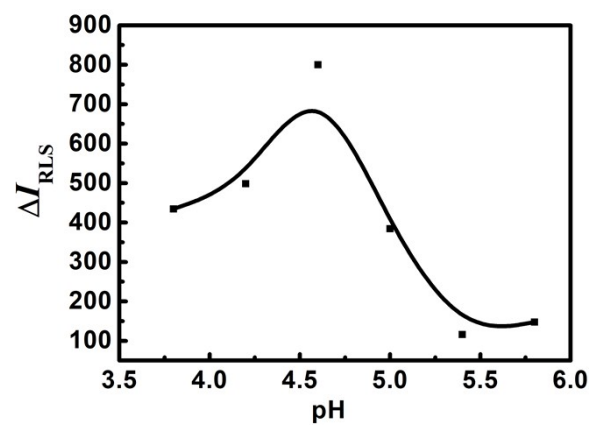


Fig. S6 The light scattering intensity change of CoOOH nanoflakes induced by GSH in PBS buffer at different pH. Conditions: CoOOH nanoflakes, 8 μM ; GSH, 750 μM ; pH 6.64; T, 45 $^{\circ}\text{C}$.

Table S1 The free cobalt ions in different samples

Samples	The concentration of free cobalt ions ($\mu\text{g/mL}$)
CoOOH nanoflakes in H_2O	0.0048
CoOOH nanoflakes in BR buffer (pH 1.81)	0.0259
CoOOH nanoflakes in BR buffer (pH 6.64)	0.0066
CoOOH nanoflakes reacts with H_2O_2 in BR buffer (pH 6.64)	4.0500

Working conditions: wavelength: 240.7 nm; slit width: 0.2 nm; time constant: 1.0 s; lamp current: 15 mA; photomultiplier voltage: 719 V; flame type: air- C_2H_2 ; fuel flow: 2.2 L/min; oxidant: 15.0 L/min at 160 kPa; burner height: 7.5 mm; delay time: 5 s; measurement time: 5 s.

Table S2 The comparison of different methods for the determination of UA

Methods	Concentration range	LOD	Real samples	Ref.
Fluorometric assay	2 -300 μM	500 nM	Human serum and urine	¹
Fluorometric assay	5 -100 μM	1.7 μM	Diluted human serum samples	²
Fluorometric assay	0.66-3.3 μM	0.044 μM	Human urine	³
Fluorometric assay	0.05-1.0 μM	20 nM	Human urine and serum	⁴
Fluorometric assay	0.7-80 μM	120 nM	Human blood	⁵
Fluorometric assay	10-800 μM	6.6 μM	Human serum	⁶
Fluorometric assay	2-100 μM	0.75 μM	Diluted serum samples	⁷
Fluorometric assay	0.1–100 μM	0.05 μM	Human serum and urine	⁸
Fluorometric assay	1.00×10^{-5} - 5.00×10^{-5} M	5.80×10^{-6} M	Human urine	⁹
Colorimetric detection	4.5-60 μM	1.3 μM	Human urine and serum	¹⁰
Colorimetric detection	10-50 ppm	0.56 ppm	Human blood serum	¹¹
Colorimetric detection	1.0-120 μM	0.25 μM	Human serum	¹²
Colorimetric detection	1.0×10^{-6} - 1.0×10^{-4} M	10×10^{-7} M	Human serum	¹³
Colorimetric detection	1.0-40 μM	0.24 μM	Human serum	¹⁴
Electrochemical detection	1.0×10^{-5} - 1.0×10^{-4} M	1.0×10^{-5} M	Human serum	¹⁵
Electrochemical method	5.0×10^{-7} - 5.0×10^{-5} M	1.0×10^{-7} M	Human serum and urine	¹⁶
Electrochemical method	6.6-112.4 μM	3.1 μM	Human urine	¹⁷
Electrochemical method	5-425 μM	1.0 μM	----	¹⁸
Electrochemical method	0.15- 11.4 μM	0.07 μM	----	¹⁹
Capillary electrophoresis	10-500 μM	3.3 μM	Human urine and plasma	²⁰
Flow injection	9.1×10^{-6} - 9.1×10^{-2} M	0.67 $\mu\text{g/mL}$	Human urine	²¹

capillary electrophoresis with chemiluminescence	6.0×10^{-7} - 3.0×10^{-5} M	3.5×10^{-7} M	Human urine	22
electrochemiluminescence	1.62×10^{-8} - 8.30×10^{-6} M	75 pM	porphyra and kelp	23
High performance liquid chromatography	0-500 μ M	0.04 μ M	Human plasma and urine	24
Reversed-phase high-performance liquid chromatography	0.07-10 mg/L	21 μ g/L	Human saliva	25

‘----’ means none real samples were detected.

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