Electronic Supplementary Information

Mesoporous Pd@Pt core-shell nanoparticles supported on multi-walled carbon nanotubes as a sensing platform: Application to simultaneous electrochemical detection of anticancer drugs doxorubicin and dasatinib

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Fig. S1 Particle size distribution histogram obtained from TEM analysis



Fig. S2The stepwise synthesis of Pd@Pt/MWCNT

1. Formation mechanism of Pd@Pt

The formation mechanism of Pd@Pt is illustrated in Fig. S2. The standard reduction potential of Pt ions $[PtCl6]^{2-}$ /Pt (0.76 V *vs*. SHE) is more positive than Pd ions $[PdCl_4]^{2-}$ (0.59V *vs*. SHE). So with this principle Pt precursor is expected to reduce first than Pd precursor and therefore the formation of Pd@Pt core-shell structure is unexpected. However, the only reason behind formation of Pd@Pt structure can be explained on the basis of difference in reduction rates of two metals (Pd reduction rate > Pt reduction rate). Such unusual behaviour is observed in previous studies for simultaneous reduction of Pd and Pt [1, 2]. Therefore, Pd precursor is reduced first by ascorbic acid forming a Pd seeds and then formation of Pt

nanoparticles on Pd forming a core-shell structure. During this process, F127 molecules are adsorbed on Pd seeds and act as a stabilizing agent. In the present case, the concentration of surfactant is over critical micelle concentration (CMC) forming the spherical micelles. During the growth of Pt on Pd seeds, F127 micelles interact with Pd seeds and acts as structure director forming Pd@Pt core-shell mesoporous structure



Fig. S3 Plot of $E_p vs. \ln v$ obtained from scan rate (V s⁻¹) of doxorubicin



Fig. S4 Effect of accumulation time (t_{acc}) on peak current of (a) 2.4 μ M DOX and (b) 3.0 μ M DAS

2. Optimization of SWV parameters

The effect of accumulation time (t_{acc}) on peak currents of 2.4 µM DOX and 3.0 µM DAS was carefully studied in PBS (0.1M, pH 6.0) through SWV. The optimization was carried out in such a way that one parameter was changed while others were kept constant. Firstly, OCP was considered as E_{acc} at which maximum current response was obtained. Secondly, the effect of t_{acc} on peak currents was investigated in the range 0-150 s at OCP. As can be seen in Fig. S4 that with increase of the t_{acc} the peak currents increases up to 90 s and decreases afterwards may be due to the saturation at the electrode surface. Besides this, other operating parameters such as frequency (f), pulse amplitude (a) and scan increment (ΔE_s) were also explored in order to get maximum analytical signal and better peak separation. Hence, sensitive determination of DOX and DAS was performed by AdSSWV at $E_{acc} = OCP$, $t_{acc} = 90$ s, f = 5 Hz, a = 30 mV, $\Delta E_s = 4$ mV and equilibrium time = 10 s.

Tables

Table S1 Analytical parameters for electrochemical determination of DOX and DAS in PBS (0.1 M, pH 6.0)

No.	Molecule	LWR (nM)	LRE	R ²	LOD (nM)			
(A) Statistical data for individual molecules								
1	DOX	3.8-8410	$I_p(\mu A)=0.021C (10^{-8} M) + 1.451$	0.993	0.73			
2	DAS	37-9720	$I_p(\mu A) = 0.015 C(10^{-8}M) + 1.128$	0.993	5.83			
(B) Statistical data for DOX when the concentration of DAS is kept constant (4.1 μ M)								
3	DOX	4.2-8520	$I_p(\mu A) = 0.02 \text{ C} (10^{-8}\text{M}) + 1.506$	0.991	0.8			
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(C) Statistical data for DAS when the concentration of DOX is kept constant (2.4 μ M)								
4	DAS	37.5-9820	$I_p(\mu A) = 0.014 C(10^{-8}M) + 1.222$	0.992	6.25			
(D) Statistical data for DOX and DAS simultaneously								
5	DOV	1 1 8580	$L(uA) = 0.019 C(10^{-8}M) + 1.501$	0.001	0.86			
5	DOX	4.4-8380	$I_p(\mu A) = 0.019 C(10^{-1}M) + 1.501$	0.991	0.80			
	DAS	38-9880	$I_p(\mu A)=0.013 C(10^{-8}M) + 1.228$	0.991	6.72			

	2		1.05		D 4
Molecule	Sensor	Technique	LOD	Analytical Range	References
	GQD	CV	0.016 µM	0.018~3.60 μM	[3]
DOX	OMWCNT	CV	$9.4\times 10^{-3}\mu M$	0.04~90 µM	[4]
	MWCNT/CoFe ₂ O ₄ /CPE	CV	10 pM	0.05~1150 nM	[5]
	Mg ₂ Al-Cl-LDHs	CV	0.02 nM	10.0~2110.0 nM	[6]
	Hg(Ag)FE	SWV	9.89 ng mL^{-1}	4.99~59.64 ng mL ⁻¹	[7]
	MAb/GNP/TB Sol-Gel	CV	$0.09 \ pg \ mL^{-1}$	2.5~50.0 pg mL ⁻¹	[8]
	MWNT-COOH	CV	$10 \text{ ngm} L^{-1}$	$0.02{\sim}1.28 \text{ mu gmL}^{-1}$	[9]
	GNP/RGD-MAP-C	CV	$0.01 \mu g/mL$	$0.5 \sim 1.0 \ \mu g/mL$	[10]
	GNPs/HDT	CV	$1.74 \times 10^{-11} M^{-1}$	$1.0 \sim 160.0 \text{ pg mL}^{-1}$	[11]
	ds–DNA/Au	CV	1 pM	0.08~0.5 M	[12]
	Ferrofluid	CV	—	$0.05 \sim 0.5 \text{ mgm}l^{-1}$	[13]
	AgNPs-CDs-rGO/GCE	DPV	2 nM	0.01~2.5 µM	[14]
	ZnO/BMTFB/CPE	SWV	9 nM	0.07–500 μΜ	[15]
	Pd@Pt/MWCNT	AdSSWV	0.86 nM	4.4-8580 nM	This work#
	Pd@Pt/MWCNT	AdSSWV	0.73 nM	3.8-8410 nM	This work*
DAS	GCE	DPV	0.13 μΜ	0.20 - 2.00 μM	[16]
	Pt/MWCNTs- BMIHFP	SWV	1.0 µM	5.0-500 μM	[17]
	ZnO/BMTFB/CPE	SWV	0.5 μΜ	1.0–1200 µM	[15]
	Pd@Pt/MWCNT	AdSSWV	6.72 nM	38-9880 nM	This work#
	Pd@Pt/MWCNT	AdSSWV	5.83 nM	37-9720 nM	This work*

 Table S2 Comparison of the proposed sensor with literature for the electrochemical determination of DOX and DAS

#: Simultaneous determination; *: Individual determination

Abbreviations: graphene quantum dots (GQD); oxidized multiwalled carbon nanotube (OMWCNT); magnetic graphene oxide grafted with chlorosulfonic acid (Fe_3O_4 -GO-SO_3H); spinel-structured cobalt ferrite (CoFe₂O₄); carbon paste electrode (CPE); layered double hydroxides (LDHs); silver amalgam film electrode (Hg(Ag)FE); gold nanoparticles (GNPs); thiol base sol-gel (TBSol-Gel); carboxylated multiwalled carbon nanotubes (MWNT-COOH); glycine-aspartic acid (RGD); 1,6-hexanedithiol (HDT); double-stranded DNA (ds-DNA); magnetic nanoparticle colloidal suspension (Ferrofluid); 1-butyl-3-methylimidazolium hexafluoro phosphate (BMIHFP); carbon dots (CDs); 1-butyl-3-methylimidazolium tetrafluoroborate (BMTFB)

Molecule	Concentration (taken) (10 ⁻⁷ M)	Concentration (found) (10 ⁻⁷ M)	Recovery (%) (n = 5)	Bias (%)	Precision % RSD (n = 5)
DOX	Intra - day 1.19	1.175	98.7	1.3	1.98
	Inter – day 2.38	2.34	98.3	1.7	2.24
DAS	Intra – day 1.33	1.31	98.5	1.5	1.96
	Inter – day 2.26	2.22	98.2	1.8	2.16

Table S3 Precision and Bias of assay for DOX and DAS by the proposed voltammetric procedure (n = 5)

Table S4 Recovery studies for individual determination of DOX and DAS in urine and blood serum samples (n=5)

Sample	Drug	Spiked (10 ⁻⁷ M)	Detected (10-7 M)	Recovery (%) ±% RSD
Urine	DOX	Not spiked Not found		
Sample 1		0.42	0.414	98.57 ± 1.89
		0.76	0.75	98.68 ± 1.96
		1.28	1.272	99.37 ± 1.98
Urine	DAS	Not spiked	Not found	
Sample 2		0.47	0.47	100.0 ± 1.93
		0.72	0.714	99.17 ± 1.95
		1.18	1.17	99.15 ± 1.98
Serum	DOX	Not spiked	Not found	
Sample 1		0.35	0.345	98.57 ± 1.90
		0.77	0.765	99.35 ± 2.06
		1.32	1.32	100.0 ± 1.98
Serum	DAS	Not spiked	Not found	
Sample 2		0.38	0.376	98.94 ± 2.04
		0.65	0.651	100.15 ± 2.11
		1.12	1.12	100.00 ± 2.08

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