

Supplementary Information

Application of a novel diol-based porous organic polymers on the determination of trace level tetracyclines in water

Changzheng Cui¹, Zan Cao¹, Shenping Zhang², Yaru Hu¹, Lei Jiang³, Shijie Yao¹, Hui Ye³, Yanbo Zhou¹, Jun Hu^{2*}, Kuangfei Lin¹, Tian-Yang Zhang^{1*}

- 1. State Environmental Protection Key Laboratory of Environmental Risk Assessment and Control on Chemical Process, School of Resources and Environmental Engineering, East China University of Science and Technology, Shanghai, China, 200237.*
- 2. School of Chemistry and Molecular Engineering, East China University of Science and Technology, Shanghai, China, 200237.*
- 3. National Engineering Research Center of Urban Water Resources, Shanghai, China, 200082.*

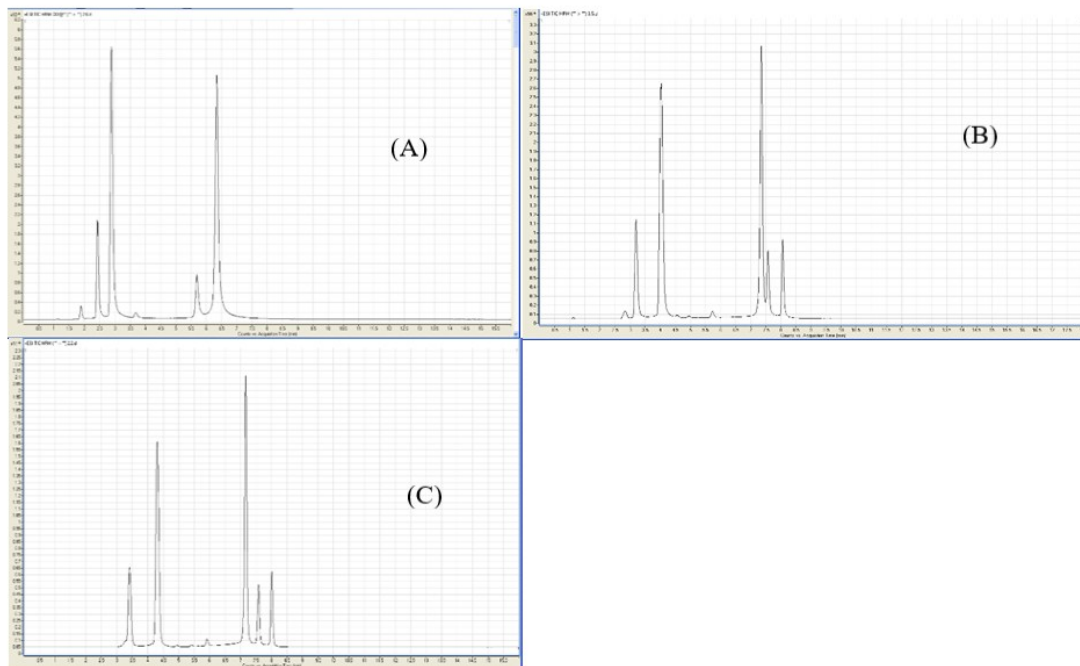


Fig. S1 A total ion chromatogram reflecting the effect of organic mobile phase on chromatographic separation (A) acetonitrile-water; (B) methanol-water; (C) formic acid: acetonitrile (0.1:99.9, v/v)- formic acid: water (0.1:99.9,v/v).Other parameters: flow rate: $0.4 \text{ mL} \cdot \text{min}^{-1}$; injection volume: $10 \mu\text{L}$.

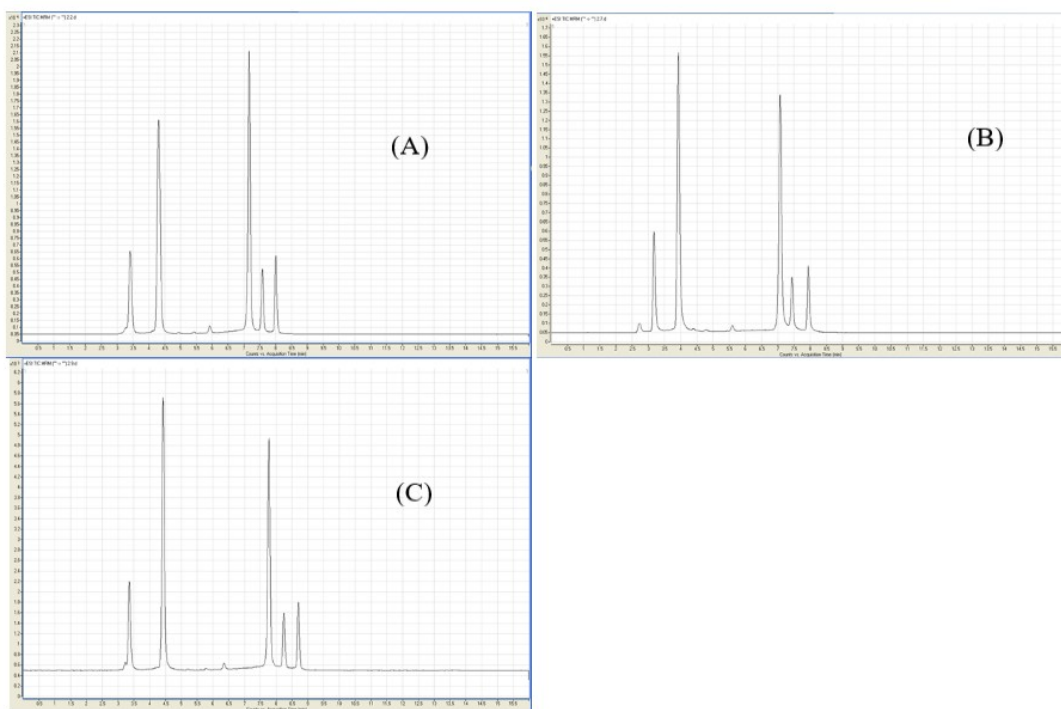


Fig. S2 A total ion chromatogram reflecting the effect of injection volume chromatographic separation (A) 2 μL ; (B) 5 μL ; (C) 10 μL . Other parameters: mobile phase, formic acid: acetonitrile (0.1:99.9, v/v) - formic acid :water (0.1:99.9, v/v); flow rate: 0.2 $\text{mL} \cdot \text{min}^{-1}$

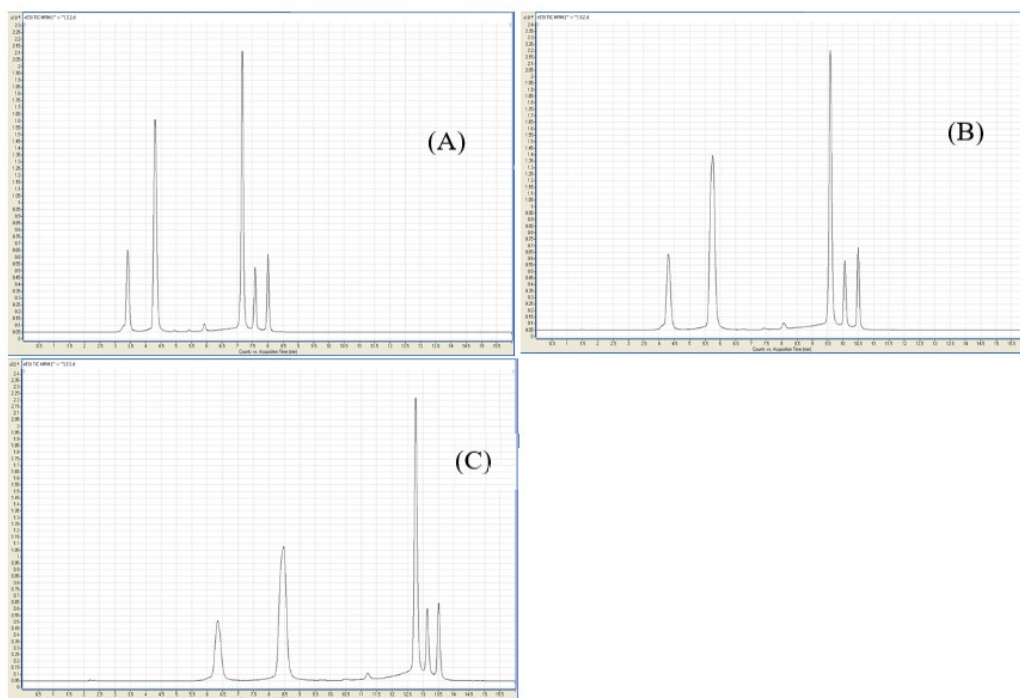
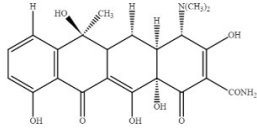
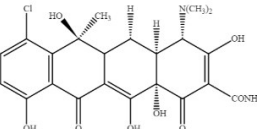
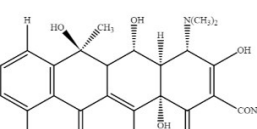
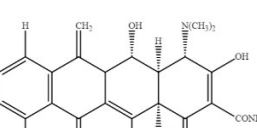


Fig. S3 A total ion chromatogram reflecting the effect of flow rate on chromatographic separation (a) 0.4 mL·min⁻¹; (b) 0.3 mL·min⁻¹; (c) 0.2 mL·min⁻¹. Other parameters: mobile phase, formic acid: acetonitrile (0.1:99.9, v/v)- formic acid: water (0.1:99.9, v/v); injection volume: 10 μL.

Table S1 Development of TCs and their properties

Classification	antibiotics	time to market	Characterizations
First generation	chlortetracycline	1948	Natural broad-spectrum antibiotics containing phenanthranes, widely used in infection caused by gram-positive and negative bacteria, intracellular mycoplasma, chlamydia and rickettsidia
	oxytetracycline	1948	
	tetracycline	1953	
Second generation	demeclocycline	1965	modify the inactivated essential groups in natural tetracyclines to change their antimicrobial activity, chemical stability. These antibiotics are more lipophilic and easier to absorb
	metacycline	1965	
	doxycycline	1967	
Third generation	minocycline	1972	The d-ring of natural tetracyclines was replaced by a variety of groups, such as glycy, dimethylamine and fluorine. This kind of structure is very difficult to construct by the previous semi-synthetic method
	tigecycline	2005	

Table S2 Target TCs and their properties

Compounds	pK _a	Molecular form	MW (g/mol)	CAS number	Structure
TC	3.32/7.78/9.58	C ₂₂ H ₂₄ N ₂ O ₈	480.90	64-75-5	
CTC	3.33/7.55/9.33	C ₂₂ H ₂₃ ClN ₂ O ₈	478.88	57-62-5	
OTC	3.22/7.46/8.94	C ₂₂ H ₂₄ N ₂ O ₉	460.44	2058-46-0	
MTC	2.88/7.44	C ₂₂ H ₂₂ N ₂ O ₈	442.42	914-00-1	
DOC	3.52/7.75/9.57	C ₂₂ H ₂₄ N ₂ O ₈	444.43	564-25-0	