# Rapid and simple purification of elastin-like polypeptides directly from whole cells and cell lysates by organic solvent extraction

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# SUPPLEMENTARY DATA

#### Materials & Methods

#### 1.1 Materials & Methods

All organic solvents (methanol, ethanol, isopropanol, 1-butanol, 2-butanol, 2methyl-2-propanol, acetone, acetonitrile, and ethyl acetate) were HPLC grade (Fisher Scientific) and used without additional purification or drying. Unless otherwise noted, water was ultra-filtered. BL21 and 5-alpha *E. coli* cells as well as BseRI, Acul and DNA T4 ligase were purchased from New England Biolabs. Plasmid miniprep and gel extraction kits were purchased from Zymo Research. Alcohol dehydrogenase, lysozyme and lipopolysaccharide were purchased from Sigma-Aldrich. Blue Stain protein ladder and ampicillin were purchased from Gold Biotechnology. Gel Red was purchased from Biotium. B-PER bacterial lysis reagent was purchased from Thermo Fisher Scientific. All media for bacterial cell culture was autoclaved prior to use and contained an appropriate antibiotic. DNA sequences for gene assembly were purchased from IDT.

#### 1.2 Synthesis of ELP construct genes

Briefly pET-25b(+) plasmid was cut with BseRI or AcuI(NEB) to linearize DNA. A leader DNA sequence (IDT) containing the amino acids GHHHHHHNGW was inserted (DNA T4 ligase, NEB) into V12, V24 and S12. A solubilizing domain containing the amino acid sequence (GKG)<sub>4</sub> was then inserted in a similar fashion to the end of V12, V24 and S12. ELP genes were then assembled using plasmid reconstruction recursive

directional ligation (Pre-RDL) techniques<sup>1,2</sup> to iteratively assemble DNA block sequences with themselves. The unmodified V12, V24 and S12 was appended making the final constructs(ie: S12-K4-S12). Final gene constructs were sequence-verified and transformed into BL21(DE3) E. coli for expression.

# **Protein Sequences**

# GFP (green fluorescent protein, UV variant with an N-terminal polyhistidine tag)

5<u>0</u> SKGEELFTGV VPILVELDGD VNGHKFSVSG EGEGDATYGK LTLKFICTTG KLPVPWPTLV TTFSYGVQCF SRYPDHMKRH DFFKSAMPEG YVQERTISFK DDGNYKTRAE VKFEGDTLVN RIELKGIDFK EDGNILGHKL EYNYNSHNVY ITADKQKNGI KANFKIRHNI EDGSVQLADH YQQNTPIGDG PVLLPDNHYL STQSKLSKDP NEKRDHMVLL EFVTAAGITH GMDELYKHHH

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#### TEL (human tropoelastin, main chain)

GGVPGAVPGG VPGGVFFPGA GLGGLGVGGL GPGVKPAKPG VGGLVGPGLG AEGSALPGAF 80 90 110 120 PGGFFGAGGG AAGAAAAYKA AAKAGAAGLG VGGIGGVGGL GVSTGAVVPQ LGAGVGAGVK PGKVPGVGLP GVYPGGVLPG AGARFPGIGV LPGVPTGAGV KPKAQVGAGA FAGIPGVGPF GGQQPGLPLG YPIKAPKLPA GYGLPYKTGK LPYGFGPGGV AGSAGKAGYP TGTGVGPQAA AAAAKAAAKL GAGGAGVLPG VGVGGPGIPG APGAIPGIGG IAGVGAPDAA AAAAAAAKAA KFGAAGGLPG VGVPGVGVPG VGVPGVGVPG VGVPGVGVPG VGVPGVGVPG VGVPGVGVPG 390 400 410 VGVPGALSPA ATAKAAAKAA KFGARGAVGI GGIPTFGLGP GGFPGIGDAA AAPAAAAAKA AKIGAGGVGA LGGVVPGAPG AIPGLPGVGG VPGVGIPAAA AAKAAAKAAQ FGLGPGVGVA PGVGVVPGVG VVPGVGVAPG IGLGPGGVIG AGVPAAAKSA AKAAAKAQFR AAAGLPAGVP GLGVGAGVPG LGVGAGVPGL GVGAGVPGPG AVPGTLAAAK AAKFGPGGVG ALGGVGDLGG AGIPGGVAGV VPAAAAAAKA AAKAAQFGLG GVGGLGVGGL GAVPGAVGLG GVSPAAAAKA AKFGAAGLGG VLGAGQPFPI GGGAGGLGVG GKPPKPFGGA LGALGFPGGA CLGKSCGRKR

Κ

## V12

Υ

#### V24

Υ

#### S12-K<sub>4</sub>-S12

130 140 VPGSGVPGSG VPGSGVPGSG VPGSGY

### V12-K<sub>4</sub>-V12

130 140 VPGVGVPGVG VPGVGVPGVG VPGVGY

#### V24-K<sub>4</sub>-V24

190 200 210 220 230 240 VPGVGVPGVG VPGVGVPGVG VPGVGVPGVG VPGVGVPGVG VPGVGVPGVG VPGVGVPGVG

250 260 VPGVGVPGVG VPGVGVPGVG VPGVGY

#### CryS96 (αB-crystallin peptide fused with ELP S96)

49<u>0</u> 50<u>0</u> SGVPGSGVPG SGVPGSGVPG SGY

## 1.3 Copper staining SDS-PAGE gels

For samples requiring copper staining for detection, the following procedure was used after electrophoresis. Gels were rinsed in deionized water for 45 sec, stained with 10 % CuCl<sub>2</sub> (w/v) for 5 min with gentle shaking, and minimally rinsed again with water. Finally, gels were imaged using a BioRad Chemidoc Touch imaging system.



Figure S1. SDS-PAGE analysis of S12-K<sub>4</sub>-S12 extraction screen. S = protein MW standards, L = clarified lysate. Extractants used are listed in Table 2 of the paper.



Figure S2. SDS-PAGE analysis of V24-K<sub>4</sub>-V24 extraction screen. S = protein MW standards, L = clarified lysate, "+" = 3X ITC-purified V24-K<sub>4</sub>-V24. Extractants used are listed in Table 2 of the paper.



Figure S3. SDS-PAGE analysis of V24 extraction from lysate, visualized by copper staining. L = clarified lysate; BA = 1:1 nBuOH:IPA.



Figure S4. Photos depicting the steps of the non-optimized extraction from cell lysate (A) and whole cells (B). Coomassie Blue was added after back-extraction to enhance visibility of the aqueous subphase.



Figure S5. SDS-PAGE analysis showing extraction optimization of V24-K<sub>4</sub>-V24 with varied lysate:extractant ratios and salt effects. L = clarified lysate.



Figure S6. SDS-PAGE analysis of lead extractant blend ratio optimization for V24-K<sub>4</sub>-V24. Blend ratios (v/v) are as indicated for each solvent combination.



Figure S7. SDS-PAGE analysis of 15 vs 60 min direct extraction of V24-K<sub>4</sub>-V24 from whole cells. 60-I = IPTG-induced cells.



Figure S8. SDS-PAGE analysis of scaled-up V24-K<sub>4</sub>-V24 extraction from lysate (35 mL,  $\sim$  8.75 g wet pellet weight), visualized by copper staining. L = clarified lysate; BA = 1:2 nBuOH:IPA.



Figure S9. SDS-PAGE analysis of scaled-up (A) S12-K<sub>4</sub>-S12, (B) V12-K<sub>4</sub>-V12, and (C) V24-K<sub>4</sub>-V24 extractions from whole cells with extraction blend ratios (2.5 g of cell pellet).

Method	Benefits	Limitations
ITC	High recovery potential	Limited to ELP with convenient <i>T<sub>t</sub></i> ; requires aqueous solution
Back-extraction	High efficiency; solvent exchange to aqueous solution; target never dried	Care must be taken to avoid target precipitation; process may require optimization
Passive evaporation	High efficiency	Relatively slow process, depending on solvent; may lose structure/function upon drying
Rotary evaporation	Highly scalable for large volumes; high efficiency	May lose structure/function upon drying; may be slow, depending on solvent
Vacuum drying	High throughput; high efficiency	Limited utility with highly volatile solvents; may lose structure/function upon drying
Speed vac	High throughput; high efficiency	May lose structure/function upon drying
Lyophilization	High throughput; high efficiency	Slow; requires transition to aqueous solution; may lose structure/function without the use of cryoprotectants

Table S1. Comparison of ELP Recovery Methods.



Figure S10. Nucleic acid contamination of V24-K<sub>4</sub>-V24 extraction from lysates assessed by agarose gel electrophoresis and stained with GelRed. P = plasmid DNA control; "+" = 3X ITC-purified V24-K<sub>4</sub>-V24. Extractants employed are listed in Table 2 of the paper.



Figure S11. Agarose gel electrophoresis assessment of nucleic acid contamination of optimized extractant blends for V24-K<sub>4</sub>-V24 purification from lysates. Blend ratios (v/v) are as indicated for each solvent combination. P = plasmid DNA; "–" = empty lane.



Figure S12. Agarose gel electrophoresis of V24-K<sub>4</sub>-V24 extracted directly from whole cells. (A) Nucleic acids are visualized by GelRed staining and (B) proteins by
Coomassie R-250 staining. P = probe tip sonication lysate, B = B-PER lysate, U = urea lysate, BA = 1:2 nBuOH:IPA, BF = 1:2 nBuOH:Ace, "-" = without salt, "+" = with salts.



Figure S13. SDS-PAGE analysis of LPS content in V24-K<sub>4</sub>-V24 extracted from lysates with 1:2 nBuOH:EtOH. Lanes 1-5 are a serial dilution of commercial LPS from 500 ng-31.25 ng. Lane 7, 9, and 10 contain 5, 30, and 50  $\mu$ g of V12-K4-V12 extract, respectively.

Solvent	Notation	Formula	MM	Density (g/mL) <sup>a</sup>	Solubility in Water (g/100g) <sup>b</sup>	Dielectric R Constant <sup>c</sup> S P	elative olvent olarity <sup>d</sup>	Polarity Parametei (kcal/mol)	Polarity r Index <sup>f</sup>	Log P <sup>g</sup>
Acetone	Ace	(CH <sub>3</sub> ) <sub>2</sub> CO	58.09	0.784	Miscible	21.0 0.	.355	42.2	5.1	-0.24
Acetonitrile	ACN	<b>CH</b> <sup>3</sup> <b>CN</b>	41.06	0.782	Miscible	36.6 0.	.460	45.6	5.8	-0.34
n-Butanol	nBuOH	C <sub>4</sub> H <sub>9</sub> OH	74.14	0.806	6.32	17.8 0.	.586	49.7	3.9	0.88
sec-Butanol	sBuOH	CH <sub>3</sub> CHOHCH <sub>2</sub> C	74.14	0.802	18.1	17.3 0.	.506	47.1	I	0.61
tert-Butanol	tBuOH	(CH <sub>3</sub> ) <sub>3</sub> COH	74.14	0.8	Miscible	12.5 0.	389	43.3	3.9	0.35
Diethyl ether	Et <sub>2</sub> O	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O	74.14	0.708	6	4.27 0.	117	34.5	2.8	0.89
Ethanol	EtOH	C <sub>2</sub> H <sub>5</sub> OH	46.07	0.794	Miscible	25.3 0.	.654	51.9	5.2	-0.31
Ethyl acetate	EtOAc	$C_4H_8O_2$	88.12	0.895	8	6.08 0.	.228	38.1	4.4	0.73
MeOH	MeOH	CH <sub>3</sub> OH	32.05	0.786	Miscible	33.0 0.	.762	55.4	5.1	-0.77
Isopropanol	IPA	CH <sub>3</sub> CHOHCH <sub>3</sub>	60.11	0.786	Miscible	20.2 0.	.546	48.4	3.9	0.05
Water	H <sub>2</sub> O	H <sub>2</sub> O	18.01	~	I	80.1 1.	00	63.1	10.2	-1.38
<sup>a</sup> Density (specif <sup>b</sup> Solvent solubili	fic gravity). <sup>3 3 3 3</sup> . ty in water. <sup>3 3 3 3</sup>	33								

Table S2. Organic Solvents and Properties.

<sup>c</sup> Dielectric constant (relative permittivity,  $\epsilon$ ).<sup>44 44 4</sup> <sup>d</sup> Relative solvent polarity ( $E_T^N$ ) measured at 25 °C, except for tBuOH at 30 °C.<sup>55555</sup>

<sup>e</sup> Polarity parameter ( $E_7$ ).<sup>33333</sup> <sup>f</sup> Polarity Index (P') measure at 25 °C.<sup>66666</sup> <sup>g</sup> Partition coefficient of octanol/water (Log  $P_{ow}$ ).<sup>7777</sup>

1 J. R. McDaniel, J. A. MacKay, F. G. Quiroz and A. Chilkoti, *Biomacromolecules*, 2010, **11**, 944-952.

2 D. E. Meyer and A. Chilkoti, *Biomacromolecules*, 2002, **3**, 357-367.

3 G. Wypych, ed., *Knovel Solvents - A Properties Database*, ChemTec Publishing, 2008; 2012.

4 CRC Handbook of Chemistry and Physics, CRC Press/Taylor & Francis, Boca Raton, FL, 2017.

- 5 C. Reichardt and T. Welton, eds., Solvents and Solvent Effects in Organic Chemistry, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany, 2010.
- 6 L. R. Snyder, *Journal of chromatography*. A, 1974, **92**, 223-230.
- 7 C. L. Yaws, ed., Yaws' Handbook of Thermodynamic and Physical Properties of Chemical Compounds, Knovel, 2003.